

Colorimetric and fluorescent sensing of pyrophosphate in 100% aqueous solution using
rhodamine B derivative and Al³⁺ complex

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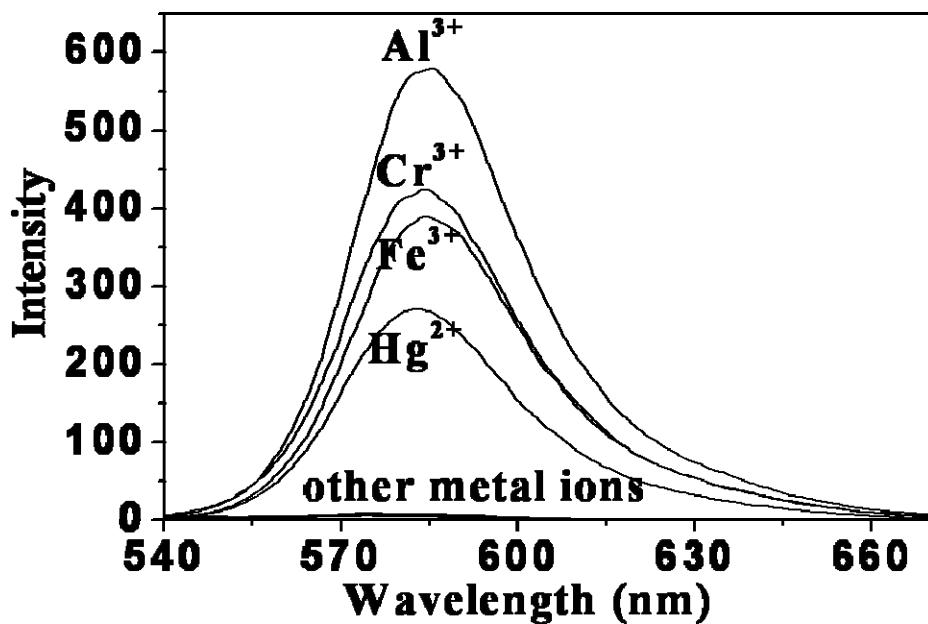


Figure S1. Fluorescent emission change of **1** (40 μM) upon addition of various metal ions (10 equiv) at pH 7.4 (1 mM HEPES) ($\lambda_{\text{ex}} = 510 \text{ nm}$, Slit: 10 nm/5 nm)

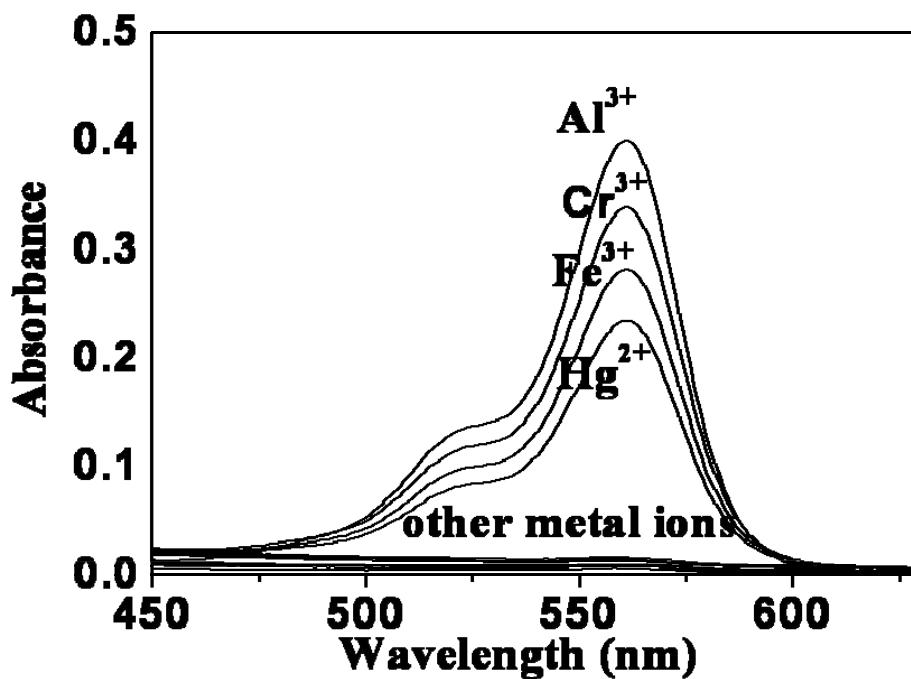


Figure S2. Colorimetric change of **1** (40 μM) upon addition of various metal ions (10 equiv) at pH 7.4 (1 mM HEPES) ($\lambda_{\text{ex}} = 510 \text{ nm}$, Slit: 10 nm/5 nm)

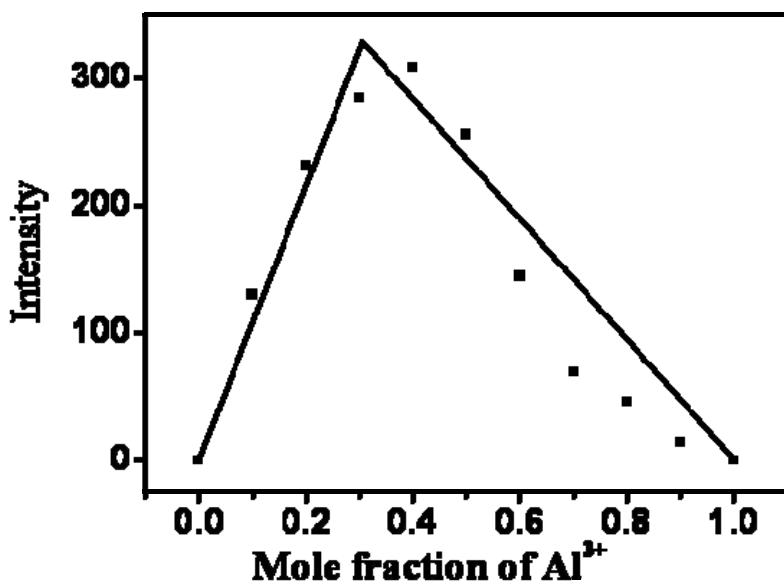


Figure S3. Job plot for **1**- Al^{3+} complexes in 1 mM HEPES buffer at pH 7.0.
Total concentration $[\mathbf{1}] + [\text{Al}^{3+}] = 100 \mu\text{M}$.

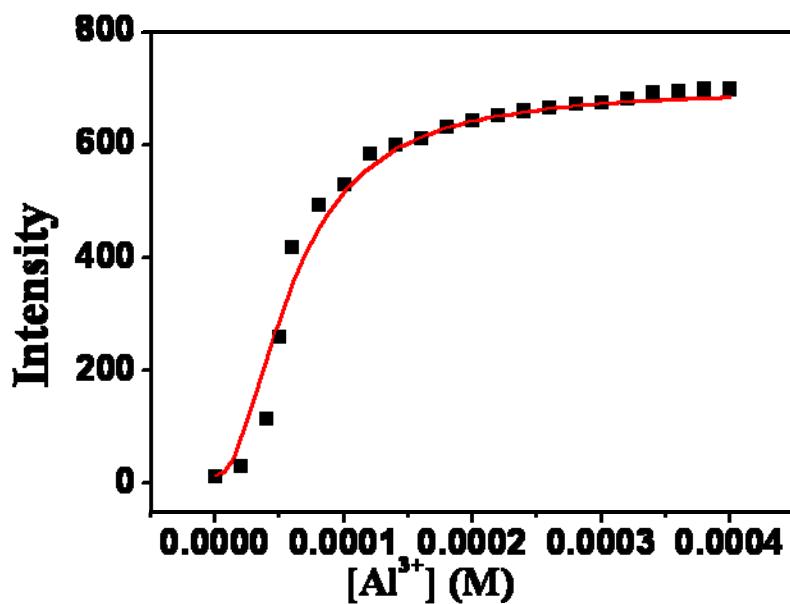
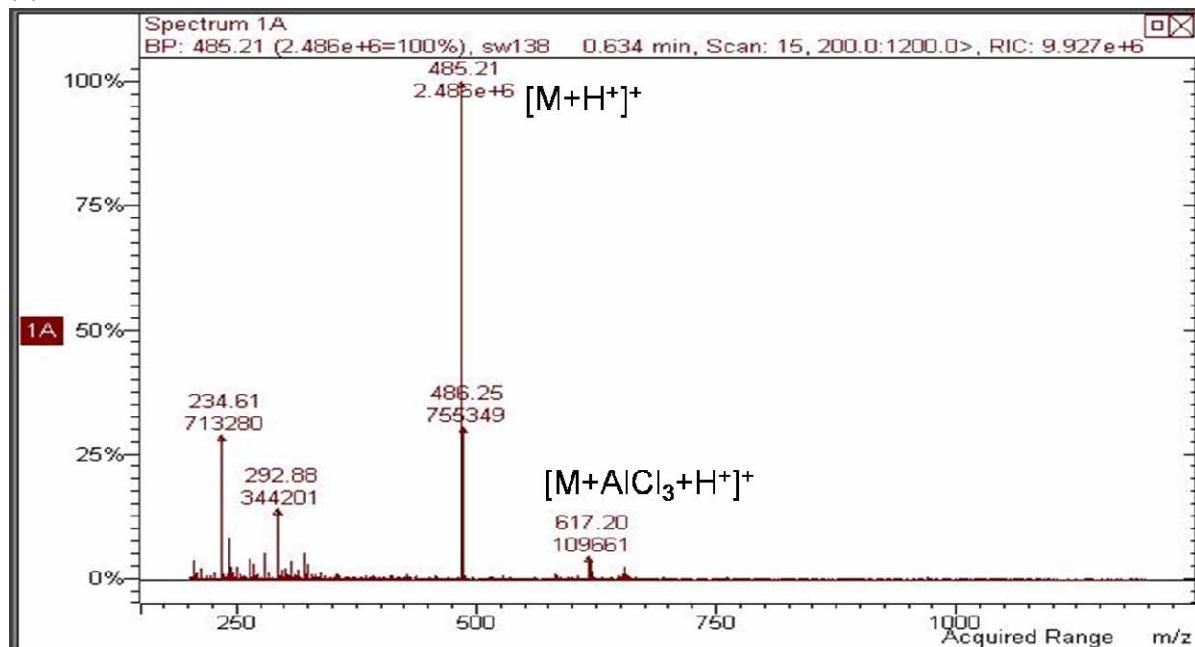


Figure S4. Theoretical fit to the experimental data (■) obtained from the maximum intensity as a function of the addition of Al³⁺ to **1** (40 μM) in 1 mM HEPES buffer at pH 7.0.

(a)



(b)

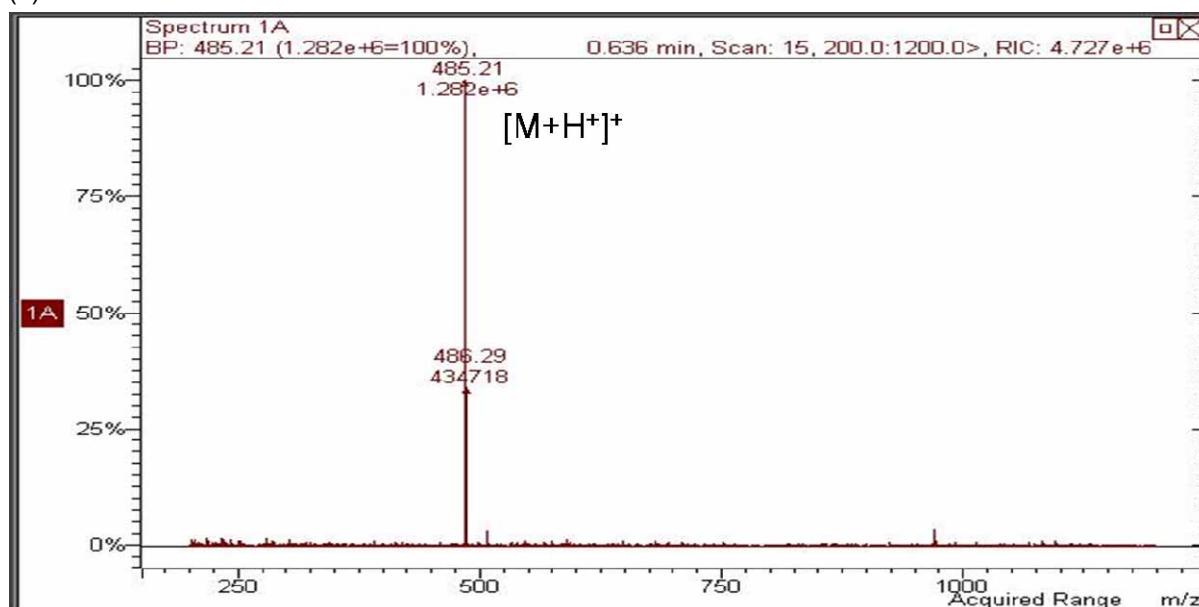


Figure S5. ESI mass spectra of **1** (40 μ M) with Al^{3+} (10 equiv) in the absence (a) and presence (b) of PPi (10 equiv) in MeOH.