Electronic Supplementary Information for:

Long-lived Emissive Intra-ligand Triplet Excited States (³IL): Next Generation Luminescent Oxygen Sensing Scheme and a Case Study With Red Phosphorescent Diimine Pt(II) Bis(acetylide) Complexes Containing Ethynylated Naphthalimide or Pyrene Subunits

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General inf	ormation	S2
Figure S1.	¹ H NMR of Pt-1	S3
Figure S2.	¹³ C NMR of Pt-1	S3
Figure S3.	TOF ESI MS of Pt-1	S4
Figure S4.	¹ H NMR of Pt-2	S4
Figure S5.	TOF ESI MS of Pt-2	S5
Figure S6.	¹ H NMR of Pt-3	S5
Figure S7.	¹³ C NMR of Pt-3	S6
Figure S8.	Emission intensity changes of the complexes at different concentrations in acetonitrile	S7
Figure S9.	Oxygen sensing properties of complexes Pt-1	S8
Figure S10.	Oxygen sensing properties of complexes PtOEP	S8
Figure S11.	Two-sites model plots for sensing films of Pt-1 and PtOEP in polymer IMPEK-C	S9
Figure S12.	Emission spectra of complexes Pt-1, Pt-2 and Pt-3 in IMPEK-C films under N_2 and O_2	S10
Z-matrix of	Pt-1	S11
Z-matrix of	[•] Pt-2	S21
Z-matrix of	Pt-3	S29

Experimental section General information

NMR spectra were taken on a 400 MHz Varian Unity Inova spectrophotometer. Mass spectra were recorded with a Q-TOF Micro MS spectrometer. UV-Vis spectra were taken on a HP8453 UV-visible spectrophotometer. Fluorescence spectra were recorded on a JASCO FP-6500 or a Sanco 970 CRT spectrofluorometer. Luminescence quantum yields were measured with Ru(bpy)₂(Phen) as the reference (ϕ = 6.0 % in acetonitrile). Emission decay measurements at RT was performed with a pulsed diode laser (PicoQuant GmbH Model LDH-P-C-375, λ ex= 372 nm, pulse width 100 ps), attached to the fluorescence spectrophotometer (Jobin Yvon Fluorolog 3).

The structures of the complexes were optimized using density functional theory (DFT) with B3LYP functional and 6-31G(d)/LanL2DZ basis set. The excited state related calculations were carried out with the time dependent DFT (TD-DFT) with the ground state geometry. The 6-31G(d) basis set was employed for C, H, N, O, S and the LanL2DZ basis set was used for Ru(II). There are no imaginary frequencies for all optimized structures. All these calculations were performed with Gaussian 09 (M. J. Frisch, Trucks, et al, Gaussian 09, Revision A. I., Gaussian, Inc., Wallingford CT, 2009.).

Synthesis of Complexes Pt-1, Pt-2 and Pt-3.

All the compounds were synthesized according to reported method (Pomestchenko, I. E.; Luman, C. R.; Hissler, M.; Ziessel, R.; Castellano, F. N. *Inorg. Chem.*, 2003, 42(5), 1394-1396). For the detail synthesis of Pt-1, please refer to our recent paper (H. Guo, M. L. Muro-Small, S. Ji, J. Zhao and F.N. Castellano. *Inorg. Chem.*, **2010**, DOI:10.1021/ ic101107b.).

For **Pt-1**, ¹H NMR (400 MHz, CDCl₃): δ 9.68 (d, *J* = 6.0 Hz, 2H), 9.14 (d, *J* = 8.0 Hz, 2H), 8.51 (t, *J* = 7.6 Hz, 4H), 8.05 (s, 2H), 7.88 (d, *J* = 7.6 Hz, 2H), 7.68 (d, *J* = 5.2 Hz, 2H), 7.48 (t, *J* = 8.0 Hz, 2H), 4.16 (t, *J* = 7.6 Hz, 4H), 1.71 (m, 4H), 1.48 (m, 24H), 0.96 (t, *J* = 7.6 Hz, 4H). ¹³C NMR (100 MHz, CDCl₃): δ 164.63, 164.43, 156.40, 151.45, 134.26, 33.26, 133.15, 131.24, 130.99, 130.10, 128.52, 126.64, 125.19, 122.90, 119.60, 119.26, 101.55, 100.12, 40.34, 36.16, 30.44, 20.61, 14.03. HR-MALDI-MS: C₅₄H₅₆NaN₄Pt, Calculated *m/z* =1038.3534, found, *m/z* =1038.3608.

For **Pt-2**, ¹H NMR (400 MHz, d_6 -acetone): δ 9.64 (d, J = 5.6 Hz, 2H), 8.60 (s, 2H), 7.85 (d, J = 6.0 Hz, 2H), 7.35 (d, J = 6.8 Hz, 4H), 7.19 (t, J = 8.0 Hz, 4H), 7.07 (t, J = 7.6 Hz, 2H), 1.44 (s, 18H). HR-MALDI-MS: C₃₄H₃₄N₂Pt, Calculated, m/z = 665.2370, found m/z = 665.2318.

For **Pt-3**, ¹H NMR (400 MHz, CDCl₃): δ 9.95 (d, *J* = 6.0 Hz, 2H), 9.19 (d, *J* = 9.2 Hz, 2H), 8.29 (d, *J* = 8.0 Hz, 2H), 8.09 (d, *J* = 7.6 Hz, 4H), 8.01 (m, 10H), 7.91(t, *J* = 7.6 Hz, 2H), 7.62 (d, *J* = 6.0 Hz, 2H), 1.45 (s, 18H). ¹³C NMR (100 MHz, CDCl₃): δ 163.73, 156.53, 151.65, 132.53, 131.77, 131.73, 130.40, 129.43, 128.48, 128.03, 127.82, 127.64, 127.29, 126.66, 125.82, 124.97, 124.75, 124.54, 124.50, 124.11, 118.99, 101.31, 93.74, 30.48, 29.90.



Figure S1. ¹H NMR of **Pt-1** (CDCl₃, 400 MHz).



ppm (1)0 190 180 170 160 150 140 130 120 110 100 90 80 70 60 50 40 30 20 10 0 -10 -20

Figure S2. ¹³C NMR of **Pt-1** (CDCl₃, 100 MHz).



Figure S3. HR-MALDI-MS of Pt-1.





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Figure S5. FR-MALDI-MS of Pt-2.



Figure S6. ¹HNMR of Pt-3 (CDCl₃, 400 MHz).

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Figure S7. ¹³CNMR of Pt-3 (CDCl₃, 100 MHz).



Figure S8. Emission intensity changes of Pt-1, Pt-2 and Pt-3 with different concentrations in acetonitrile. 25 °C.





Figure S9. Phosphorescent intensity response of sensing films of the **Pt-1** in IMPEK-C to (a) O_2/N_2 saturation cycles and (b) dynamic response to variation of the oxygen partial pressures. $\lambda ex = 458$ nm, $\lambda em = 630$ nm. The numbers indicate the O_2 concentration in mixed O_2/N_2 gas (v/v). 25 °C.



Figure S10. Phosphorescent intensity response of sensing films of the **PtOEP** in IMPEK-C to (a) O_2/N_2 saturation cycles and (b) dynamic response to variation of the oxygen partial pressures. λ ex= 549 nm, λ em = 648 nm. The numbers indicate the O_2 concentration in mixed O_2/N_2 gas (v/v). 25 °C.



PtOEP



Figure S11. Two-sites model plots for sensing films of **Pt-1** and **PtOEP** in polymer IMPEK-C. The solid lines are the fitting results of the data with the two-site model.



Figure S12. Emission spectra of complexes Pt-1, Pt-2 and Pt-3 in IMPEK-C films under N₂ and O₂. For Pt-1, $\lambda ex = 458$ nm; Pt-2, $\lambda ex = 430$ nm; Pt-3, $\lambda ex = 468$ nm.

С

25

B41

2

A40

1

D39

Pt-1							
							N O
Symbolic Z-i	matrix:						
Charge = 0	Multiplicity = 1					Pt	
01							
С							
С	1	B1					0
С	2	B2	1	A1			
С	3	B3	2	A2	1	D1	
С	1	B4	2	A3	3	D2	
С	4	B5	3	A4	2	D3	
С	6	B6	4	A5	3	D4	
С	7	B7	6	A6	4	D5	
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Ν	5	B15	1	A14	2	D13	
Pt	15	B16	10	A15	9	D14	
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С	20	B20	17	A19	15	D18	
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н	38	B40	25	A39	2	D38	

					-	
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Н	42	B43	25	A42	2	D41
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Н	46	B47	25	A46	2	D45
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С	75	B88	70	A87	67	D86

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31 32 33

S19

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83
84
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86
87
88
89 92 2.0 93 1.0
90 91 2.0 93 1.0
91
92
93 94 1.0
94 95 1.0 96 1.0 97 1.0
95
96
97

Pt-2

01 C

Symbolic Z-matrix: Charge = 0 Multiplicity = 1 0 1

Pt

С	1	B1				
С	2	B2	1	A1		
С	3	B3	2	A2	1	D1
С	1	B4	2	A3	3	D2
С	4	B5	3	A4	2	D3
С	6	B6	4	A5	3	D4
С	7	B7	6	A6	4	D5
С	8	B8	7	A7	6	D6
С	9	B9	8	A8	7	D7
Н	1	B10	5	A9	4	D8
Н	5	B11	1	A10	2	D9
Н	9	B12	8	A11	7	D10
Н	10	B13	9	A12	8	D11
Ν	10	B14	9	A13	8	D12
Ν	5	B15	1	A14	2	D13

Pt	16	B16	5	A15	1	D14
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С	17	B19	16	A18	5	D17
С	20	B20	17	A19	16	D18
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С	22	B23	19	A22	18	D21
С	23	B24	22	A23	19	D22
н	23	B25	22	A24	19	D23
С	24	B26	22	A25	19	D24
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С	27	B28	24	A27	22	D26
Н	25	B29	23	A28	22	D27
Н	27	B30	24	A29	22	D28
Н	29	B31	27	A30	24	D29
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С	33	B34	21	A33	20	D32
С	34	B35	33	A34	21	D33
Н	34	B36	33	A35	21	D34
С	35	B37	33	A36	21	D35
Н	35	B38	33	A37	21	D36
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Н	36	B40	34	A39	33	D38
Н	38	B41	35	A40	33	D39
Н	40	B42	36	A41	34	D40
Н	7	B43	6	A42	4	D41
Н	3	B44	2	A43	1	D42
С	8	B45	7	A44	6	D43
С	2	B46	1	A45	5	D44
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С	46	B55	8	A54	7	D53
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66 67 68 69 1.0 70 1.0 71 1.0 69 70 71

Pt-3

Symbolic Z-matrix:

Charge = 0 Multiplicity = 1

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С	2	B2	1	A1		
С	3	В3	2	A2	1	D1
С	1	B4	2	A3	3	D2
С	4	B5	3	A4	2	D3
С	6	B6	4	A5	3	D4
С	7	B7	6	A6	4	D5
С	8	B8	7	A7	6	D6
С	9	B9	8	A8	7	D7
н	1	B10	5	A9	4	D8
Н	5	B11	1	A10	2	D9
н	9	B12	8	A11	7	D10
н	10	B13	9	A12	8	D11
Ν	10	B14	9	A13	8	D12
Ν	5	B15	1	A14	2	D13
Pt	16	B16	5	A15	1	D14
С	17	B17	16	A16	5	D15
С	18	B18	17	A17	16	D16
С	17	B19	16	A18	5	D17
С	20	B20	17	A19	16	D18
н	7	B21	6	A20	4	D19
н	3	B22	2	A21	1	D20
С	8	B23	7	A22	6	D21
С	2	B24	1	A23	5	D22
С	24	B25	8	A24	7	D23



н	26	B26	24	A25	8	D24
н	26	B27	24	A26	8	D25
н	26	B28	24	A27	8	D26
С	24	B29	8	A28	7	D27
н	30	B30	24	A29	8	D28
н	30	B31	24	A30	8	D29
н	30	B32	24	A31	8	D30
С	24	B33	8	A32	7	D31
н	34	B34	24	A33	8	D32
н	34	B35	24	A34	8	D33
н	34	B36	24	A35	8	D34
С	25	B37	2	A36	1	D35
Н	38	B38	25	A37	2	D36
н	38	B39	25	A38	2	D37
н	38	B40	25	A39	2	D38
С	25	B41	2	A40	1	D39
н	42	B42	25	A41	2	D40
н	42	B43	25	A42	2	D41
н	42	B44	25	A43	2	D42
С	25	B45	2	A44	1	D43
н	46	B46	25	A45	2	D44
н	46	B47	25	A46	2	D45
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С	19	B49	18	A48	17	D47
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н	51	B53	50	A52	19	D51
С	52	B54	50	A53	19	D52
С	52	B55	50	A54	19	D53
С	53	B56	51	A55	50	D54
н	53	B57	51	A56	50	D55
С	55	B58	52	A57	50	D56
С	56	B59	52	A58	50	D57
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С	59	B62	55	A61	52	D60
С	59	B63	55	A62	52	D61
н	60	B64	56	A63	52	D62
С	62	B65	57	A64	53	D63
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Н	66	B69	62	A68	57	D67
С	69	B70	64	A69	59	D68
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н	69	B72	64	A71	59	D70

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С	76	B77	75	A76	21	D75
н	76	B78	75	A77	21	D76
С	77	B79	75	A78	21	D77
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58