

Supporting Information

Electrochemical studies on the permeable characteristics of thiol-modified double-stranded DNA self-assembled monolayers on gold

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Content

1. Molecular structure of thiol-modified double-stranded DNA (HS-(CH₂)₆-5'-AGT ACA GTC ATC GCG-3' and complementary chain)
2. Confirmation of our experimental results in Table 1 for ds-DNA-SAMs assembled on gold under different NaCl concentrations (0.025 ~ 2 M)
3. The hexagonal packing of ds-DNA-SAMs on Au(111) with different interaxial distance L_a (n S-S, n = 4 ~ 12) (Ignoring the tilted angle φ)
4. The hexagonal packing of lying ds-DNA-SAMs on Au(111) with different site distance L_a (n S-S, n = 8 ~ 12) (Tilted angle $\varphi = 90^\circ$)
5. The change of channel diameter from erecting to tilted ds-DNA-SAMs on Au
6. Confirmation of the data from the model by *CC* experiment and literature reports
7. Electrochemical response of redox probes (negative, positive and neutral) on ds-DNA-SAMs on gold from literature
8. List of redox probes to be investigated in our future work
9. Derivations of the equations in Table 2 in this article
10. References

1. Molecular structure of thiol-modified double-stranded DNA (HS-(CH₂)₆-5'-AGT ACA GTC ATC GCG-3' and complementary chain)

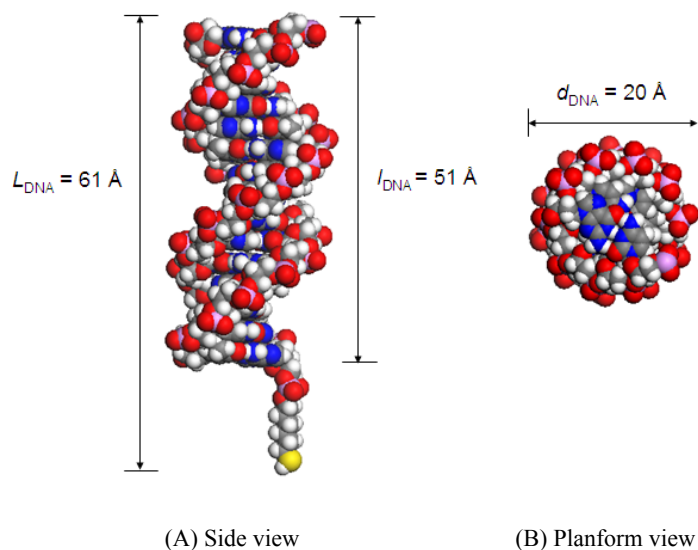


Fig. S1 Molecular structure (A, side view; B, planform view) of thiol-modified double-stranded DNA (HS-(CH₂)₆-5'-AGT ACA GTC ATC GCG-3' and complementary chain) (drawn by Materials Studio software 3.2) ^[1]. L_{DNA} (the geometrical length of thiol-modified ds-DNA molecule, 61 Å in our study), l_{DNA} (ds-DNA geometrical length, 51 Å in our study) and d_{DNA} (ds-DNA geometrical diameter, 20 Å).

2. Confirmation of our experimental results in Table 1 for ds-DNA-SAMs assembled on gold under different NaCl concentrations (0.025 ~ 2 M)

(1) Experimentally investigating the relative standard deviation (*RSD*) of experimental data

We investigated the *RSD* of experimental data by choosing two representative assembly NaCl concentrations (lower C_{NaCl} , 0.12 M and higher C_{NaCl} , 2 M) (Table S1) with $n = 3$. The relative standard deviation (*RSD*) of Γ_m is smaller than 5%, consistent with < 10% from literature^[2-6], the *RSD* of C obtained at 0.1 V s⁻¹, $\Phi_{1\text{ Hz}}$ and R_{it}^* is about 20%, consistent with the *RSD* values (< 30%) from literature^[7,8], the *RSD* of Δj is < 10% and k_{et} is much bigger and even reaches 46%, which is due to its sensitivity for the defects in SAMs^[9-11].

Table S1. Parameters of ds-DNA-SAMs on gold assembled under C_{NaCl} (0.12 and 2 M) ($n = 3$)

C_{NaCl} (M)	Ions penetration (5 mM phosphate-50 mM NaCl, pH 7.0)						Charge transfer (2 mM $\text{Fe}(\text{CN})_6^{3-/4-}$)		Surface coverage		
	CV						EIS		CC		
	C ($\mu\text{F cm}^{-2}$)						$\Phi_{1\text{ Hz}}$	R_{it}^*	Δj	k_{et}	Γ_m
	Assembly concentration	0.1 V s ⁻¹	0.5 V s ⁻¹	1 V s ⁻¹	5 V s ⁻¹	20 V s ⁻¹	50 V s ⁻¹	(^o)	($\Omega\text{ cm}^2$)	($\mu\text{A cm}^{-2}$)	(cm s^{-1})
0.12	18.2 ± 2.9	12.6 ± 2.6	10.8 ± 2.4	7.9 ± 1.9	6.2 ± 1.4	5.5 ± 1.2	73 ± 4	$(1.6 \pm 0.3) \times 10^4$	355 ± 34	$(1.3 \pm 0.6) \times 10^{-4}$	0.67 ± 0.02
2	35.2 ± 8.7	24.5 ± 5.7	21.0 ± 4.7	15.7 ± 2.7	12.5 ± 1.6	10.9 ± 1.0	75 ± 2	$(8.6 \pm 2.3) \times 10^3$	25 ± 1	$(1.9 \pm 0.3) \times 10^{-6}$	1.95 ± 0.02

(2) Experimentally cross-checked the experimental data (C values) by CV and EIS

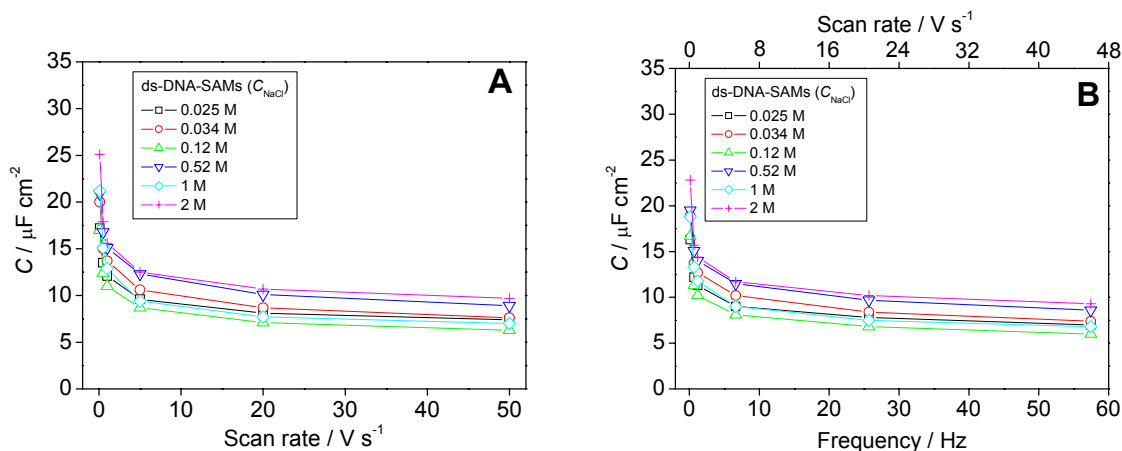


Fig. S2 Relationships of C of ds-DNA-SAMs on gold with C_{NaCl} (0.025 ~ 4 M). (A) C obtained by CV with scan rate v ; (B) C obtained by EIS based on the equation $C = \frac{1}{2\pi f_m A Z''}$ with frequency f_m , whereas

Z'' was the imaginary parts of impedance at the frequency f_m and A was the real area of electrode.^[12] In order to compare the C obtained from EIS with CV , f_m was designated close to $f_m = \frac{v}{2\Delta E_m} = \frac{v}{2(E_{\text{max}} - E_{\text{min}})}$ based on the potential zone ($2\Delta E_m = 2(E_{\text{max}} - E_{\text{min}}) = 0.8 \text{ V}$) of a CV

scan period from -0.2 V (E_{min}) to $+0.2 \text{ V}$ (E_{max}) and scan rate v (0.1, 0.5, 1, 5, 20, 50 V s^{-1}). f_m was calculated to be 0.12, 0.66, 1.18, 6.64, 25.7, 57.4 Hz. *Surface coverage Γ_m of ds-DNA-SAMs on gold assembled under different C_{NaCl} (0.025 ~ 2.0 M) was 5×10^{-12} , 6×10^{-12} , 7×10^{-12} , 1.3×10^{-11} , 1.8×10^{-11} and $1.9 \times 10^{-11} \text{ mol cm}^{-2}$ from CC measurements.

(3) Experimental investigation of ds-DNA-SAMs assembled on gold under different NaCl concentrations (0.025 ~ 2 M) with 5 mM $\text{NaH}_2\text{PO}_4\text{-Na}_2\text{HPO}_4$ (pH 7.0)

In order to further confirm our experimental phenomena (Table 1) for ds-DNA-SAMs formed under different C_{NaCl} (0.025 ~ 2 M) solution, we also assembled ds-DNA on gold under different C_{NaCl} (0.025 ~ 2 M) with 5 mM $\text{NaH}_2\text{PO}_4\text{-Na}_2\text{HPO}_4$ (pH 7.0) and investigated the relationships of parameters (C , $\Phi_{1 \text{ Hz}}$, R_{it}^* , Δj and k_{et}) with C_{NaCl} (Fig. S3 and Table S2). The experimental results proved the experimental phenomena as Table 1 indicated.

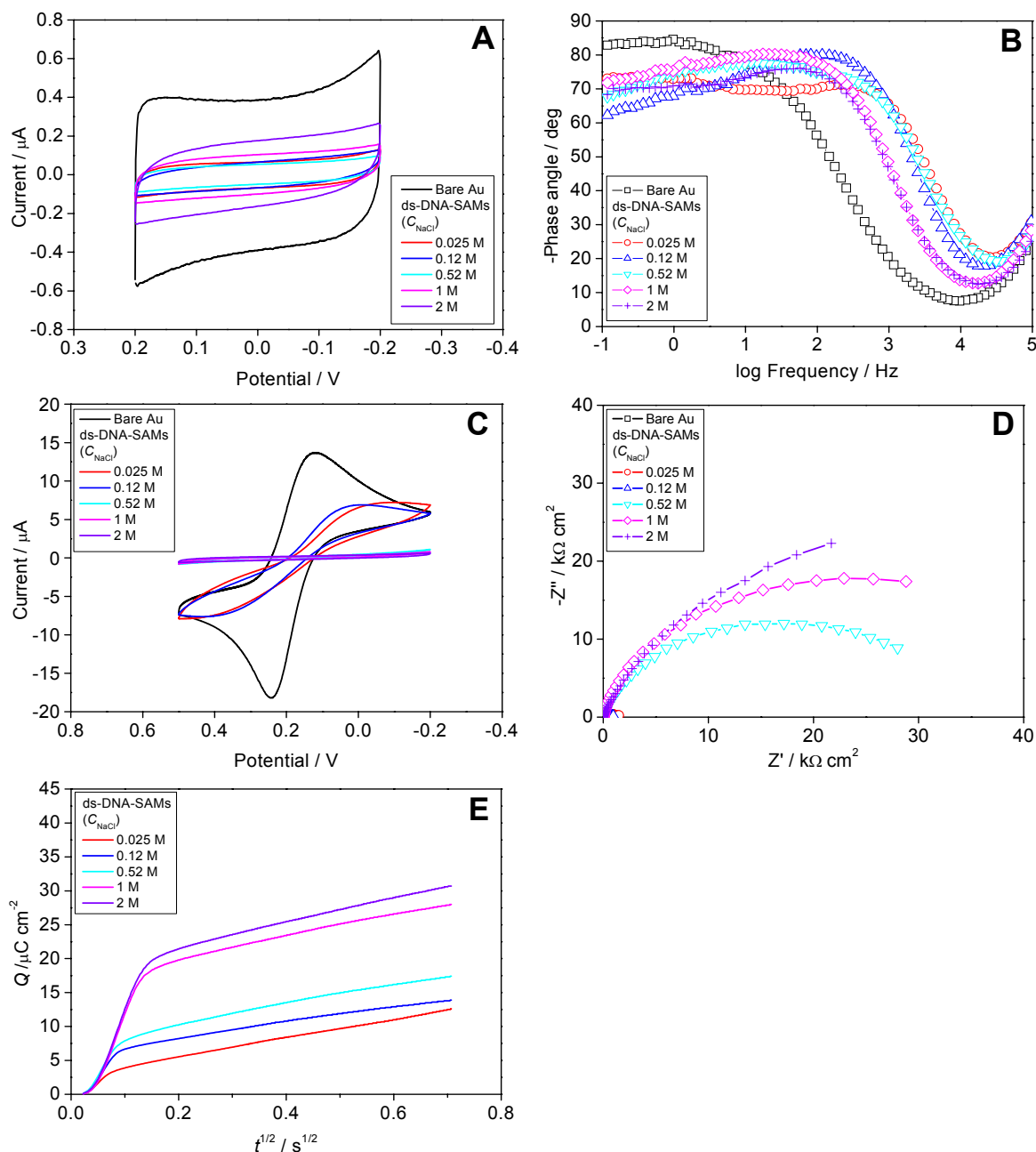


Fig. S3 Electrochemical characteristics of bare Au and ds-DNA-SAMs assembled under different C_{NaCl} (0.025 ~ 2 M) with 5 mM $\text{NaH}_2\text{PO}_4\text{-Na}_2\text{HPO}_4$ (pH 7.0). (A) CV plots at 0.1 V s^{-1} and (B) the plots of phase angle with log frequency in blank PBS (5 mM phosphate-50 mM NaCl, pH 7.0) solution; (C) CV at 0.1 V s^{-1} and (D) EIS plots in 2 mM $\text{Fe}(\text{CN})_6^{3-/4-}$ PBS (5 mM phosphate-50 mM NaCl, pH 7.0) solution; (E) CC plots (background subtraction) in 50 μM $\text{Ru}(\text{NH}_3)_6^{3+}$ TBS (10 mM tris-HCl, pH 7.4) solution. The colors of the curves in the figures were: bare Au, black; Assembly C_{NaCl} : 0.025 M, red; 0.12 M, blue; 0.52 M, cyan; 1 M, magenta; 2 M, violet. *Surface coverage Γ_m of ds-DNA-SAMs on gold assembled under different C_{NaCl} (0.025 ~ 2.0 M) with 5 mM $\text{NaH}_2\text{PO}_4\text{-Na}_2\text{HPO}_4$ (pH 7.0) was 3×10^{-12} , 7×10^{-12} , 9×10^{-12} , 1.8×10^{-11} and $1.9 \times 10^{-11} \text{ mol cm}^{-2}$ from CC measurements.

Table S2. Parameters for characterizing the permeability of ds-DNA-SAMs on gold assembled under different C_{NaCl} (0.025 ~ 2 M) with 5 mM NaH_2PO_4 - Na_2HPO_4 (pH 7.0)

C_{NaCl} (M)	Ions penetration (5 mM phosphate-50 mM NaCl, pH 7.0)								Charge transfer (2 mM $\text{Fe}(\text{CN})_6^{3-/4-}$)		Surface coverage
	<i>CV</i>						<i>EIS</i>		<i>CC</i>	<i>EIS</i>	<i>CC</i>
	<i>C</i> ($\mu\text{F cm}^{-2}$)						$\Phi_{1\text{ Hz}}$	R_{it}^*	Δj	k_{et}	Γ_{m}
	0.1 V s ⁻¹	0.5 V s ⁻¹	1 V s ⁻¹	5 V s ⁻¹	20 V s ⁻¹	50 V s ⁻¹	($^\circ$)	($\Omega \text{ cm}^2$)	($\mu\text{A cm}^{-2}$)	(cm s^{-1})	($10^{-11} \text{ mol cm}^{-2}$)
0.025	16.3	12.5	10.8	7.2	4.9	3.8	73	1.9×10^4	372	1.2×10^{-4}	0.3
0.12	16.3	10.2	8.4	5.8	4.6	4.4	69	2.0×10^4	357	1.6×10^{-4}	0.7
0.52	12.4	8.7	7.6	5.7	4.7	4.0	75	2.5×10^4	46	4.1×10^{-6}	0.9
1.0	24.8	17.9	15.6	12.1	10.4	9.6	76	1.1×10^4	38	3.3×10^{-6}	1.8
2.0	41.5	28.3	22.7	15.5	11.8	10.4	71	8.2×10^3	30	2.6×10^{-6}	1.9

3. The hexagonal packing of ds-DNA-SAMs on Au(111) with different interaxial distance L_a (n S-S, n = 4 ~ 12) (Ignoring the tilted angle φ)

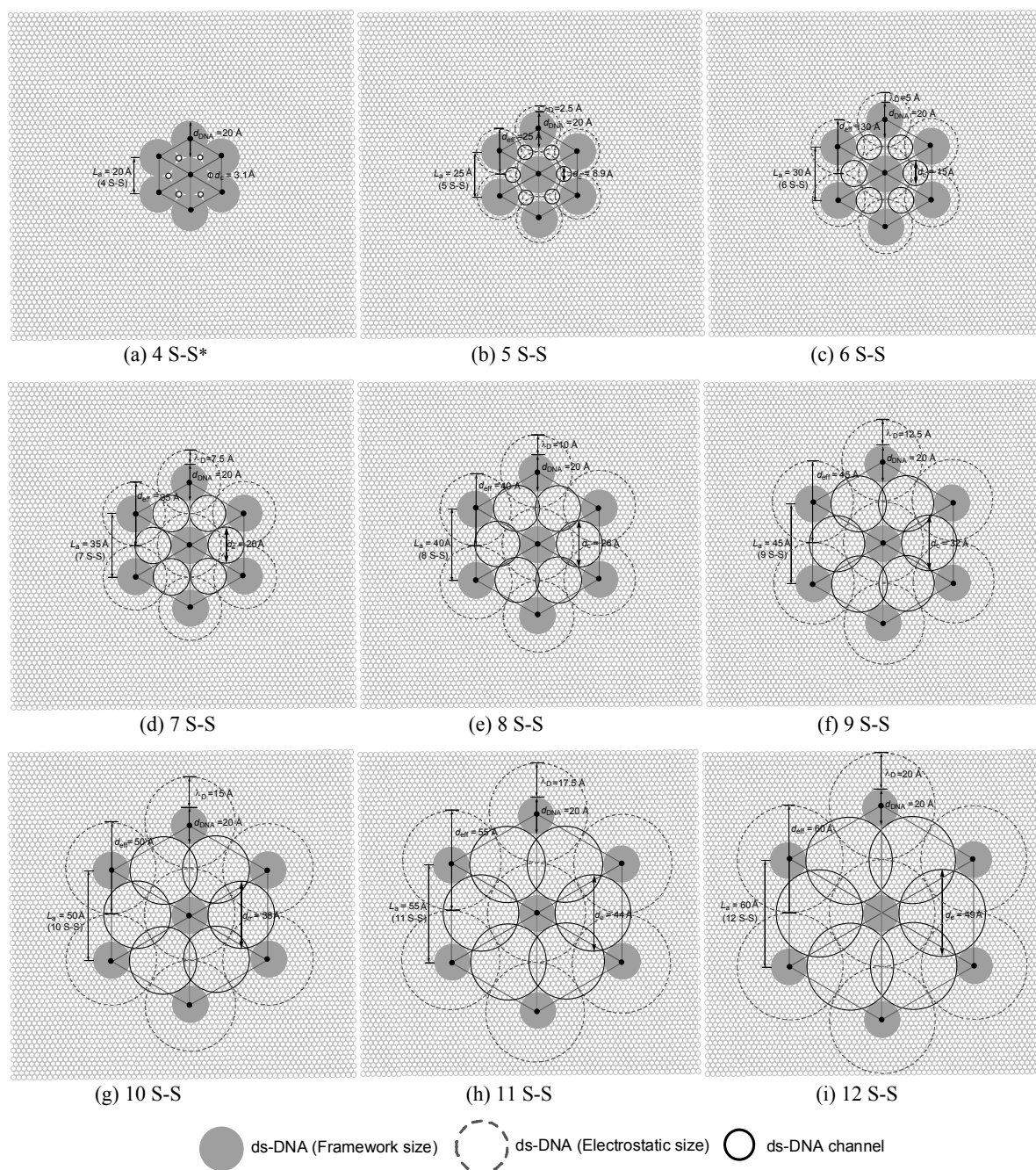


Fig. S4 The hexagonal packing of ds-DNA-SAMs on Au(111) with different interaxial distance L_a for different C_{NaCl} (n S-S, n = 4 ~ 12). *ds-DNA-SAMs with $L_a = 20 \text{ \AA}$ (4 S-S) were impossible due to the existence of repulsive force between adjacent ds-DNA molecules. Three-aggregate channels were the solid circles of adjacent three ds-DNA molecules on Au(111).

4. The hexagonal packing of lying ds-DNA-SAMs on Au(111) with different site distance L_a (n S-S, n = 8 ~ 12) (Tilted angle $\varphi = 90^\circ$)

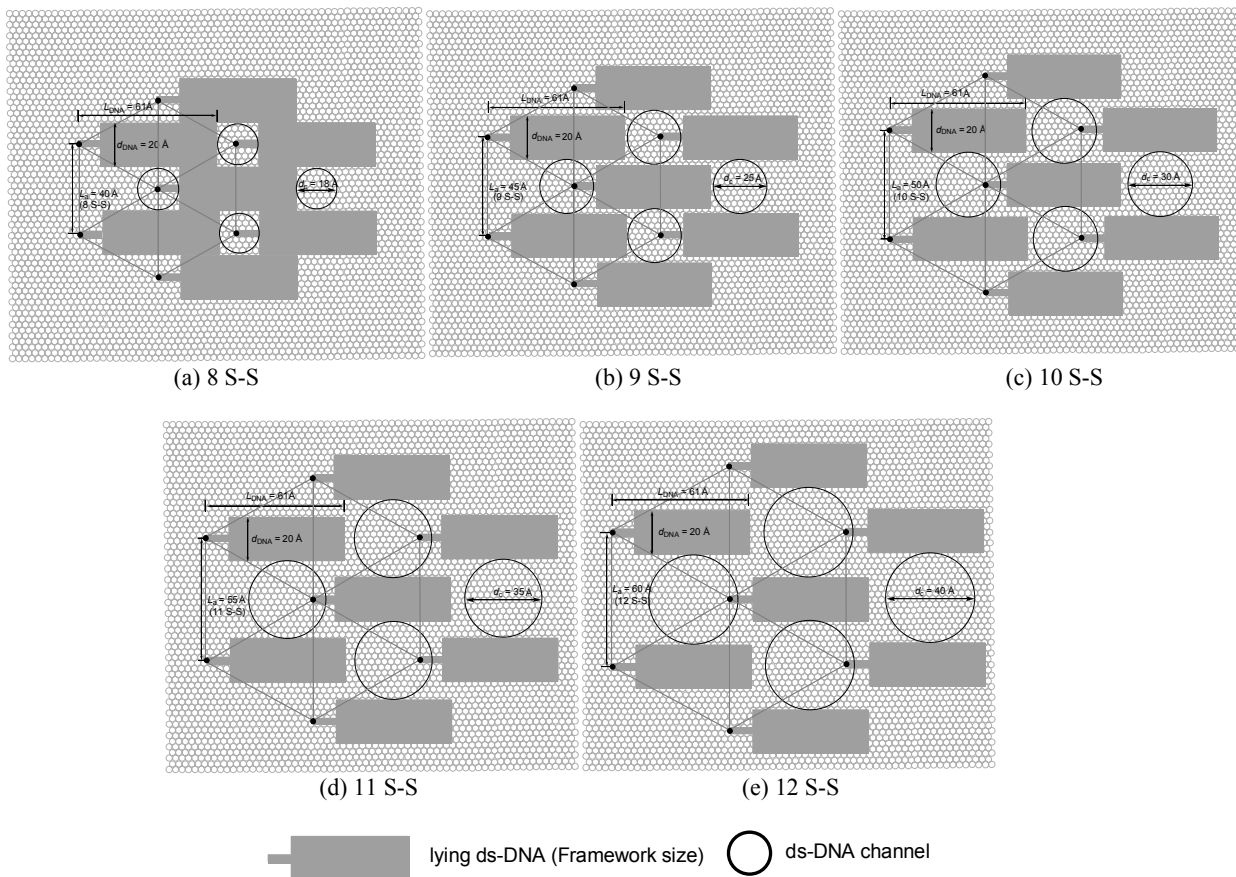


Fig. S5 Simple model for lying ds-DNA-SAMs on Au(111) with different site distance L_a (n S-S, n = 8, 9, 10, 11, 12). Four-aggregate channels were the solid circles of adjacent four ds-DNA molecules on Au(111). The area of joint section (HS-(CH₂)₆-) (length 10 Å × width 2.3 Å = 23 Å²) was much smaller than that of ds-DNA (length 51 Å × width 20 Å = 1020 Å²), thus, we ignored the joint section (HS-(CH₂)₆-) and obtained $S_{\text{DNA}} = l_{\text{DNA}} d_{\text{DNA}} = 51 \times 20 = 1020 \text{ Å}^2$ when calculating the θ_{DNA}^* and d_c^* in Table 3 based on $\theta_{\text{DNA}}^* = S_{\text{DNA}} N_m = \frac{l_{\text{DNA}} d_{\text{DNA}}}{0.866 L_a^2} = \frac{115.5 l_{\text{DNA}} d_{\text{DNA}}}{L_a^2} \%$ ($\varphi = 90^\circ$) and $d_c^* = L_a - d_{\text{DNA}}$ ($\varphi = 90^\circ$).

5. The change of channel diameter from erecting to tilted ds-DNA-SAMs on Au

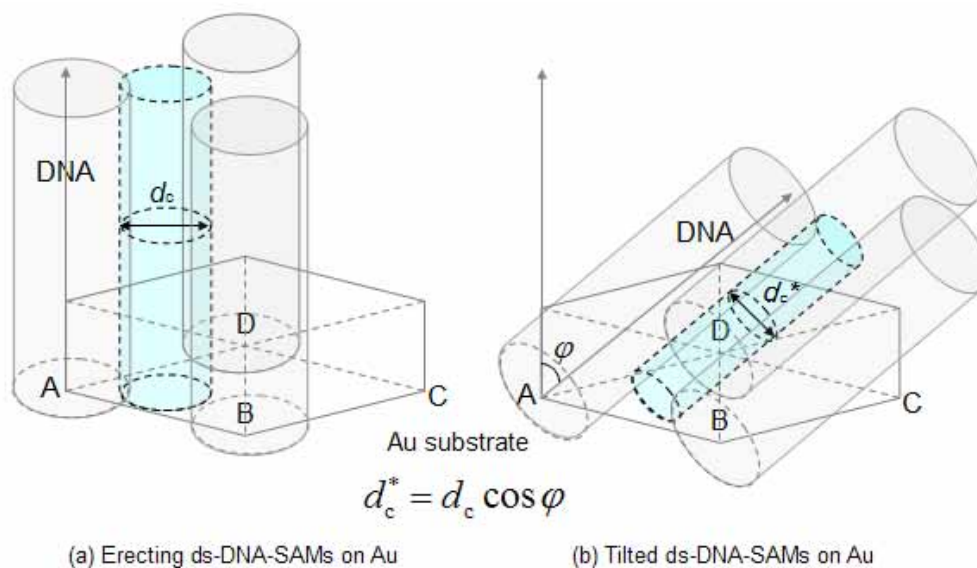


Fig. S6 The change of channel diameter from erecting to tilted ds-DNA-SAMs on Au. The φ is the tilted angle of ds-DNA-SAMs on Au. The channel diameters of erecting and tilted ds-DNA-SAMs are d_c and d_c^* respectively with $d_c^* = d_c \cos \varphi$. When the perpendicular interaxial distance a of adjacent tilted ds-DNA molecules (as Fig. S6b showed) was equal to 20 Å, the maximal tilted angle $\varphi_{\max}$ was obtained with $d_c^* = d_c \cos \varphi_{\max}$.

6. Confirmation of the data from the model by CC experiment and literature reports

Table S3. Interfacial parameters for ds-DNA-SAMs on gold calculated by our proposed model, literature reports and our experimental results^a

Interfacial parameters for ds-DNA-SAMs on gold calculated by model, literature and our experiment						
Proposed model ^b						
C_{NaCl} (M)	2	1	0.52	0.12	0.034	0.025
d_{eff} (Å)	24	26	28	38	53	58
L_a (Å) ^a	30 (6 S-S)	30 (6 S-S)	30 (6 S-S)	40 (8 S-S)	55 (11 S-S)	60 (12 S-S)
Γ_m (mol cm ⁻²)	2.1×10^{-11}	2.1×10^{-11}	2.1×10^{-11}	1.2×10^{-11}	6.3×10^{-12}	5.3×10^{-12}
ϕ_{max} (°)	59	59	59	90	90	90
Literature reports ^c						
C_M^+ (M)		1 ^[13,14]	0.69 ^[15]		0.055 ^[18]	0.01 ^[2]
			0.43 ^[17]			
Γ_m (mol cm ⁻²)		1.8×10^{-11} ^[13] (MD)	1.7×10^{-11} ^[15] (³² P-radiolabeling)		1.3×10^{-12} ^[18] (³² P-radiolabeling)	7.6×10^{-12} ^[2] (CV)
ϕ (°)		45 ^[14] (NEXAFS)	51 ^[15] (Ellipsometry) 51~58 ^[17] (Ellipsometry) 65~72 ^[17] (AFM)			
Our experiment ^d						
C_{NaCl} (M) ^e	2	1	0.52	0.12	0.034	0.025
Γ_m (mol cm ⁻²)	1.9×10^{-11}	1.8×10^{-11}	1.3×10^{-11}	7.0×10^{-12}	6.0×10^{-12}	5.0×10^{-12}

* (a) The data difference from the model, literature and our experiment was also perhaps due to the simplification of the model, the difference of experimental conditions or not considering the roughness factor R_f of gold electrodes when calculating the parameters.

(b) Proposed model: C_{NaCl} was the ionic strength for ds-DNA assembly ($C_{\text{NaCl}} = 0.025 \sim 2$ M). d_{eff} was equal to $d_{\text{DNA}} + 2\lambda_D$, where λ_D was calculated by Equation 4 in Table 2. L_a was n S-S (1 S-S = 5 Å) and assumed to be the nearest integer of d_{eff} (e.g. for $d_{\text{eff}} = 28$ Å, $L_a = 30$ Å, 6 S-S). Γ_m and ϕ were calculated based on Equation 7 and Equation 12 in Table 2. If considering the hydration force in the model, it was assumed that hydration force dominated the packing of ds-DNA molecules on gold with $L_m = 10$ Å for $C_{\text{NaCl}} \geq 0.37$ M due to $2\lambda_D \leq 10$ Å [19–22]. Thus, the smallest ds-DNA interaxial distance L_a should be 6 S-S ($d_{\text{DNA}} + L_m = 20 + 10 = 30$ Å) with $\Gamma_m 2.1 \times 10^{-11}$ mol cm⁻² for assembly C_{NaCl} (0.52 M, 1 M and 2 M). In this Table, we considered the hydration force when calculating the Γ_m for assembly C_{NaCl} (0.52 M, 1 M and 2 M).

(c) Literature reports: Because cations for ds-DNA self-assembly on gold differed from literatures (e.g. Na⁺, K⁺), we used C_M^+ to represent the total concentration of all monovalent cations in assembly solution and did not consider the cations' types. The abbreviation of methods for calculating the Γ_m and ϕ were in the parentheses: Molecular dynamics simulation (MD), Near-edge X-ray absorption fine structure (NEXAFS), Atomic force microscopy (AFM) and Cyclic voltammetry (CV).

Ref. [13]: ds-DNA sequences, 5'-HS-(CH₂)₆-(T)₂₅-3' with its complementary chain (25 base pairs); Experimental conditions, Molecular dynamics simulation (MD) in 1 M NaCl solution; Gold type, Au(111).

Ref. [14]: ds-DNA sequences, 5'-HS-(CH₂)₆-TGC AGT TCC GGT GGC TGA TC-SH-3' with its complementary chain (20 base pairs); Assembly conditions: 1 μM ds-DNA in TE buffer (10 mM tris-HCl, 1 mM EDTA, and 1 M NaCl, pH 7.0) for long (>24 h) and short (1 h); Experimental conditions: the ϕ was from NEXAFS; Au type: gold-coated Si (111) samples (A' and R_f not reported); Pretreatment of gold electrodes: Atomically flat terraces (60 ± 10 nm) separated by facets of 13–15 bunched steps were obtained by a special heating sequence.

Ref. [15]: ds-DNA sequences, 3'-HS-ds-DNA-5' (15 base pairs); Assembly conditions, 50 μM ds-DNA in 0.4 M phosphate buffer (pH 7.2) for 24 h; Experimental conditions, Γ_{DNA} was determined by ³²P radioactive labeling and ϕ was obtained based on the thickness of ds-DNA-SAMs by ellipsometry and $L_{\text{DNA}} = 57$ Å; Gold type, Polycrystalline gold (A' and R_f not reported); Pretreatment of gold electrodes: Gold substrates were placed for 20 min in an ultraviolet ozone cleaner, and then immersed for 20 min in absolute ethanol. Finally, they were rinsed with ethanol [16].

Ref. [17]: ds-DNA sequences, 5'-CTA AGA TTT TCT GCA TAG CAT TAA-TG-3'-(CH₂)₃-SH (26 base pairs); Assembly conditions, 15 μM ds-DNA in 25 mM Tris, 0.4 M NaCl for 2–12 h; Experimental conditions, ϕ was obtained based on the thickness of ds-DNA-SAMs from ellipsometry or AFM and $L_{\text{DNA}} = 94$ Å; Gold type, not mentioned; Pretreatment of gold electrodes: not mentioned.

Ref. [18]: ds-DNA sequences, 5'-TTT TTT TTT TTT TTT-(CH₂)₆-SH-3' (15 base pairs); Assembly conditions, 0.1 mM ds-DNA in 5 mM phosphate-50 mM NaCl (pH 7.4) for 18 ~ 48 h; Experimental conditions, Γ_{DNA} was determined by ³²P-radiolabeling; Gold type, Au(111) (A' and R_f not reported); Pretreatment of gold electrodes: not mentioned.

Ref. [2]: ds-DNA sequences, HS-(CH₂)₆-5'-TCGATCTGACGTCAGCTAAA-3' with its complementary chain (20 base pairs); Assembly conditions, 10 μM ds-DNA in 10 mM tris buffer (pH 7.4) for 18 ~ 48 h; Experimental conditions, Γ_{DNA} was determined by CV in 3.5 ~ 12.0 μM Ru(NH₃)₆³⁺/10 mM Tris buffer (pH 7.4); Gold type, evaporated gold films ($A' = 0.69$ cm², R_f not reported); Pretreatment of gold electrodes: Gold substrates were cleaned by immersion in a 3:1 mixture of concentrated H₂SO₄ and 30% H₂O₂ for 5 min at ~ 90 °C and then rinsed with copious amounts of water.

(d) Our experiment: The Γ_m in our experiment was calculated by CC experiment in TBS (10 mM tris-HCl, pH 7.4) solution.

7. Electrochemical response of redox probes (negative, positive and neutral) on ds-DNA-SAMs on gold from literature

Table S4. Electrochemical response of redox probes ($\text{Fe}(\text{CN})_6^{3-/4-}$, $\text{Ru}(\text{NH}_3)_6^{3+/2+}$ and Ferrocenemethanol (FcMeOH)) on ds-DNA-SAMs on gold from literature

Amount of base pairs n_{DNA}	Assembly electrolytes	d_c (Å)	d_c^* (Å)	ϕ_{max} (°)	Responses of probes ^(a)	Ref.
<i>(I) Negatively charged redox probes ($\text{Fe}(\text{CN})_6^{3-/4-}$, $E^o = 0.12 \text{ V}$)</i>						
<i>Hydration diameter d_{Hyd} of $\text{Fe}(\text{CN})_6^{3-/4-}$ ($6.0 \text{ Å}^{[29]}$, $9.5 \text{ Å}^{[30]}$, $\sim 9 \text{ Å}^{[31]}$)</i>						
30	0.1 M phosphate buffer, 1 M NaCl	15 ~ 20	6.5 ~ 7.7	59 ~ 71	×	[23]
20	20 mM tris- ClO_4 (pH 8.6), 0.4 M NaClO_4	~26	~4.5	~80	×	[24]
15	5 mM phosphate-50 mM NaCl (pH 7.4)	120	100	~90	×	[18]
20	50 mM tris- ClO_4 (pH 8.7)	~44	~35	~90	×	[25]
20	20 mM tris- ClO_4 (pH 8.7), 20 mM NaClO_4	~44	~28	~50	×	[26]
25	20 mM tris- ClO_4 (pH 8.6)	~49	~40	~90	×	[27]
20	0.1 M tris- ClO_4 , 0.1 M NaClO_4	~38	~30	~90	×	[28]
<i>(II) Positively charged redox probes ($\text{Ru}(\text{NH}_3)_6^{3+/2+}$, $E^o = -0.19 \text{ V}$)</i>						
<i>Hydration diameter d_{Hyd} of $\text{Ru}(\text{NH}_3)_6^{3+/2+}$ ($6.4 \text{ Å}^{[29]}$, $7.8 \text{ Å}^{[32]}$)</i>						
30	0.1 M phosphate buffer, 1 M NaCl	15 ~ 20	6.5 ~ 7.7	59 ~ 71	✓	[23]
20	20 mM tris- ClO_4 (pH 8.6), 0.4 M NaClO_4	~26	~4.5	~80	✓	[24]
20	0.1 M tris- ClO_4 , 0.1 M NaClO_4	~38	~30	~90	✓	[28]
<i>(III) Neutral redox probe (FcMeOH, $E^o = 0.16 \text{ V}$)</i>						
<i>Hydration diameter d_{Hyd} of FcMeOH ($6.8 \text{ Å}^{[33]}$)</i>						
20	20 mM tris- ClO_4 (pH 8.6), 0.4 M NaClO_4	~26	~4.5	~80	✓	[24]

(a) (×) The redox peaks of CV and the Warburg line of electrochemical impedance spectroscopy (EIS) disappeared indicating that the negatively charged redox probes could not arrive at gold surface and react due to the electrostatic repulsion; (✓) Obvious redox peaks of cyclic voltammetry (CV) appeared for diffusion-controlled reaction indicating that the positively or neutral redox probes could diffuse to gold surface and react.

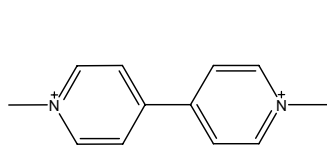
(b) The d_c was calculated based on Equation 11 and the d_c^* was calculated based on Equation 16 and Equation 17 in Table 2. The ϕ was reported as 50° for Ref. [26] and the other ϕ_{max} was calculated based on Equation 12 for $L_a < L_{\text{DNA}}/\sqrt{3}$ or Equation 13 for $L_a \geq L_{\text{DNA}}/\sqrt{3}$ in Table 2 and the Γ_m values: Ref. [18] was $1.3 \times 10^{-12} \text{ mol cm}^{-2}$ from ^{32}P -radiolabeling and the other literature have not reported the Γ_m values, which are roughly estimated by our experimental data of Γ_m in Table 1.

8. List of redox probes to be investigated in our future work

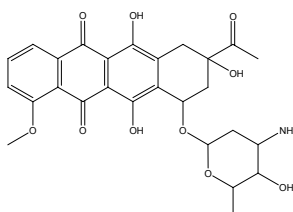
Table S5. List of redox probes

No.	Redox probes	E^0 (V vs.SCE)	Size (Å)	
			Framework Size	Van der Waals Size
(a)	Methyl vologen	-0.69	11.3 × 3.4 × 4.9	13.1 × 5.2 × 6.7
(b)	Daunorubicin	-0.60	14.9 × 4.9 × 11.1	16.7 × 6.7 × 12.9
(c)	AQS	-0.46	11.8 × 3.4 × 7.0	13.6 × 5.2 × 8.9
(d)	Methylene blue	-0.23	13.7 × 4.7 × 5.6	15.5 × 6.5 × 7.4
(e)	$\text{Ru}(\text{NH}_3)_6^{3+/2+}$	-0.19	3.3 × 3.3 × 4.4	5.1 × 5.1 × 6.2
(f)	TMPD	0.03	4.9 × 4.7 × 8.4	6.7 × 6.5 × 10.2
(g)	1, 4-Benzoquinone	0.04	4.9 × 1.5 × 6.8	6.7 × 3.4 × 8.7
(h)	$\text{Fe}(\text{CN})_6^{3-/4-}$	0.12	7.4 × 7.4 × 7.4	9.4 × 9.4 × 9.4
(i)	FcMeOH	0.16	6.7 × 4.7 × 5.9	8.5 × 6.5 × 7.8
(j)	Dopamine	0.33	5.9 × 3.6 × 8.6	7.7 × 5.4 × 10.4
(k)	FcDA	0.40	7.3 × 4.7 × 8.2	9.1 × 6.5 × 10.1

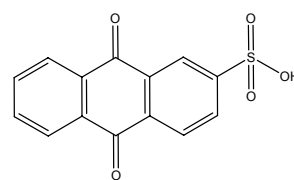
*The redox probes with E^0 (-0.46 ~ 0.4 V) would be chose to study electron transfer through ds-DNA-SAMs on gold because ds-DNA would be desorbed at potential smaller than -0.60 V in 0.1 M phosphate buffer solution (pH 6.9) by literature reports^[34,35] Anthraquinone-2-sulfonic acid (AQS), N,N,N',N'-Tetramethyl-p-phenylenediamine (TMPD), Ferrocenmethanol (FcMeOH), 1,1'-Ferrocenedicarboxylic acid (FcDA). Framework size from ball-and-stick model of Gaussview software 3.0; Van der Waals size from CPK model of Materials Studio software 3.2.^[11] The standard potential E^0 : Daunorubicin from literature^[36], Dopamine from literature^[37], FcMeOH from literature^[38]; other redox probes from literature^[39].



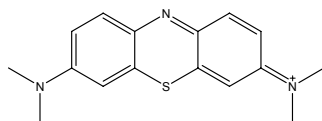
(a) Methyl vologen



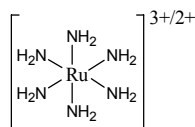
(b) Daunorubicin



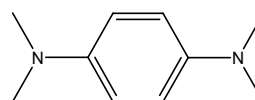
(c) AQS



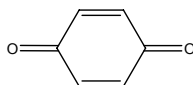
(d) Methylene blue



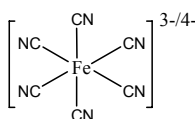
(e) $\text{Ru}(\text{NH}_3)_6^{3+/2+}$



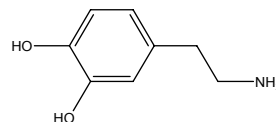
(f) TMPD



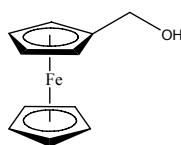
(g) 1, 4-Benzoquinone



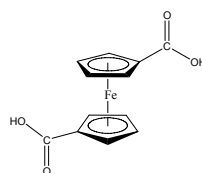
(h) $\text{Fe}(\text{CN})_6^{3-/4-}$



(i) Dopamine



(j) FcMeOH



(k) FcDA

9. Derivations of the equations in Table 2 in this article

Table S6. Derivations of the equations in Table 2 in this article

Equations	Derivation of the equations
(1) $\lambda_D = \left(\frac{\epsilon_0 \epsilon_r RT}{2 F^2 C_{NaCl}} \right)^{1/2}$	The equation was from the literature [40].
(2) $d_{eff} = d_{DNA} + 2 \lambda_D$	As Fig. 2A shown that d_{DNA} was the ds-DNA geometrical diameter (20 Å), λ_D was the Debye length (Å), d_{eff} was the ds-DNA effective diameter (Å). For example, for ds-DNA under 1 M NaCl solution, λ_D was 3 Å, then, d_{eff} was calculated to be 26 Å.
(3) $L_a = n \text{ S-S } (1 \text{ S-S} = 5 \text{ Å}, n = 5, 6, 7, 8, \dots)$	As Fig. S4 shown
(4) $L_m = L_a - d_{DNA}$	As Fig. 2A shown
(5) $\Gamma_m = \frac{1}{A_{average} N_A} = \frac{10^{16}}{0.866 N_A L_a^2}$	From the Fig. 2A, the average possessive area ($A_{average}$) of each ds-DNA molecule (different from the section area of ds-DNA molecule calculated based on the geometrical diameter $d_{DNA} = 20 \text{ Å}$) on Au(111) was equal to the area of parallelogram ABCD. According to the side length of parallelogram ABCD (L_a) and $\angle ABC = 120^\circ$, we obtained that $A_{average} = 1/2 (AC \times BD) \times 10^{-16}$ where $BD = L_a$ and $AC = \sqrt{3} L_a$. So $A = 1/2 \times \sqrt{3} L_a^2 = 0.866 \times 10^{-16} L_a^2 (\text{cm}^2)$. Then $\Gamma_m (\text{mol cm}^{-2}) = 10^{16}/0.866 N_A L_a^2$.
(6) $N_m = \Gamma_m N_A = \frac{10^{16}}{0.866 L_a^2}$	$N_m (\text{molecules cm}^{-2}) = \Gamma_m N_A = 10^{16}/0.866 L_a^2$
(7) $\theta_{DNA} = A_{DNA} N_m = \frac{90.7 d_{DNA}^2}{L_a^2} \%$	The section area of ds-DNA molecule (A_{DNA}) was $A_{DNA} = \pi (d_{DNA}/2)^2 \times 10^{-16}$. According to the amount of ds-DNA molecules each square centimeter (Equation 6, $N_m = \frac{10^{16}}{0.866 L_a^2}$), we obtained $\theta_{DNA} = A_{DNA} N_m \times 100\% = \frac{\pi d_{DNA}^2}{4 \times 0.866 L_a^2} \times 100\% = \frac{90.7 d_{DNA}^2}{L_a^2} \%$.
(8) $N_c = 2 N_m = \frac{10^{16}}{0.433 L_a^2}$	Based on Fig. 2A (two channels in parallelogram ABCD), we obtained that the amount of ds-DNA channels N_c was two times as big as that of ds-DNA molecules N_m , so $N_c = 2 N_m = 2 \times \frac{10^{16}}{0.866 L_a^2} = \frac{10^{16}}{0.433 L_a^2}$
(9) $d_c = \frac{2}{\sqrt{3}} L_a - d_{DNA} = 1.155 L_a - d_{DNA}$	Based on Fig. S7(a), for $\triangle AED$, $AE = DE = (d_{DNA} + d_c)/2$, $AD = L_a$, $\angle DAE = 30^\circ$, we obtained: $DE^2 = AD^2 + AE^2 - 2 AD \times AE \cos \angle DAE$, $AD = 2 AE \cos 30^\circ$. Then $L_a = 0.866 (d_{DNA} + d_c)$, we obtained $d_c = 1.155 L_a - d_{DNA}$
(10) $\varphi_{max} = \arcsin \left(1.155 \sqrt{1 - 0.866 \times 10^{-16} N_A a^2 \Gamma_m} \right)$ $(L_a = n \text{ S-S} < \frac{L_{DNA}}{\sqrt{3}})$	As shown in Fig. S7(b): (AA' , BB' and CC' were the middle axial of tilted ds-DNA molecules; $BG \square AA'$, $FE \square AB$, $A'F \square ABCD$; Tilted angle of ds-DNA-SAMs was φ ; BG was the perpendicular interaxial distance of adjacent tilted ds-DNA molecules assigned as a , AB and BC were the interaxial distance of adjacent ds-DNA molecules, L_a) Based on $A'F \square ABCD$, we obtained $A'F \square AB$, $FE \square AB$, So $AB \square \triangle A'EF$, then $AB \square A'E$. we obtained: $\cos \varphi = \cos \angle AA'F = A'F/AA'$ and $\sin \angle A'AE = A'E/AA'$. Combined the two equation, so $\sin \angle A'AE / \cos \varphi = A'E/A'F$. Because $A'E^2 = EF^2 + A'F^2$, $EF = AF \sin 30^\circ = 1/2 AF$ and $\tan \varphi = AF/A'F$, we obtained: $A'E^2 = (1 + 1/4 \tan^2 \varphi) A'F^2$ Then $\sin \angle A'AE / \cos \varphi = \sqrt{1 + \frac{\tan^2 \varphi}{4}}$ Because $\sin \angle A'AE = \sin \angle GAB = BG/AB = a/L_a$, we obtained: $\frac{a}{L_a \cos \varphi} = \sqrt{1 + \frac{\tan^2 \varphi}{4}}$ Then $\sin \varphi = \sqrt{\frac{4}{3} (1 - \frac{a^2}{L_a^2})}$ Based on Equation 5 $\Gamma_m = \frac{10^{16}}{0.866 N_A L_a^2}$, we obtained $L_a^2 = \frac{10^{16}}{0.866 N_A \Gamma_m}$. Then $\sin \varphi_{max} = \sqrt{\frac{4}{3} (1 - 0.866 \times 10^{-16} N_A a^2 \Gamma_m)} = 1.155 \sqrt{1 - 0.866 \times 10^{-16} N_A a^2 \Gamma_m}$ $\varphi_{max} = \arcsin \left(1.155 \sqrt{1 - 0.866 \times 10^{-16} N_A a^2 \Gamma_m} \right) \quad (a = 20 \text{ Å})$

Table S6. Derivations of the equations in Table 2 in this article (continued)

Equations	Derivation of the equations
(11) $\varphi = 90^\circ$ ($L_a = nS - S \geq \frac{L_{DNA}}{\sqrt{3}}$)	As Fig. 2B and Fig. S5 shown, ds-DNA could lie flat on gold for $L_{AC} \geq L_{DNA}$. Based on $L_{AC} = \sqrt{3}L_a$, we obtained $L_a = nS - S \geq \frac{L_{DNA}}{\sqrt{3}}$.
(12) $\theta_{DNA}^* = S_{DNA} N_m = \frac{l_{DNA} d_{DNA}}{0.866 L_a^2} = \frac{115.5 l_{DNA} d_{DNA}}{L_a^2} \%$ ($\varphi = 90^\circ$)	As Fig. S5 shown, the area of ds-DNA molecule lying on gold was $S_{DNA} = l_{DNA} d_{DNA} \times 10^{-16} \text{ cm}^2$ and the ds-DNA surface coverage was $N_m = 10^{16}/0.866 L_a^2$ molecules cm^{-2} . Then $\theta_{DNA}^* = S_{DNA} N_m = \frac{l_{DNA} d_{DNA}}{0.866 L_a^2} = \frac{115.5 l_{DNA} d_{DNA}}{L_a^2} \%$ ($\varphi = 90^\circ$)
(13) $N_c^* = \Gamma_m = \frac{10^{16}}{0.866 L_a^2}$ ($\varphi = 90^\circ$)	As Fig. S5 shown
(14) $d_c^* = (1.155 L_a - d_{DNA}) \cos \varphi$ ($0^\circ < \varphi < 90^\circ$)	When ds-DNA tilted with the angle ($0^\circ < \varphi < 90^\circ$), based on Fig. 2B, $d_c = (1.155 L_a - d_{DNA}) \cos \varphi$.
(15) $d_c^* = L_a - d_{DNA}$ ($\varphi = 90^\circ$)	When ds-DNA lied flat ($\varphi = 90^\circ$), based on Fig. S5, $d_c^* = L_a - d_{DNA}$.

Meanings of parameters in the equations and units: λ_D (debye length, Å), d_{eff} (ds-DNA effective diameter, Å), L_a (ds-DNA interaxial distance, Å), L_m (ds-DNA marginal distance, Å), Γ_m (ds-DNA surface coverage, mol cm^{-2}), N_m (ds-DNA surface coverage, molecules cm^{-2}), θ_{DNA} (area ratio of ds-DNA cross section with gold surface), d_c (channel diameter, Å), N_c (amount of ds-DNA channels, channels cm^{-2}), φ (tilted angle of ds-DNA-SAMs, $^\circ$), θ_{DNA}^ (ds-DNA area percentage on gold surface, $\varphi = 90^\circ$), d_c^* (channel diameter, Å, $\varphi = 90^\circ$), N_c^* (channel amount, channels cm^{-2} , $\varphi = 90^\circ$), R (gas constant, $8.314 \text{ J mol}^{-1} \text{ K}^{-1}$), T (temperature, K), F (Faraday constant, 96485 C mol^{-1}), C_{NaCl} (NaCl concentration, M), ϵ_0 (the permittivity of free space, $8.85 \times 10^{-14} \text{ C}^2 \text{ J}^{-1} \text{ cm}^{-1}$), ϵ_r (the dielectric constant of water, 78), L_{DNA} (the geometrical length of thiol-modified ds-DNA molecule, 61 Å in this study), l_{DNA} (ds-DNA geometrical length, 51 Å in this study), d_{DNA} (ds-DNA geometrical diameter, 20 Å), $A_{average}$ (the average possessive area of each ds-DNA molecule on Au(111) which was equal to the area of parallelogram ABCD (Fig. 2A), $A_{average} = \sqrt{3}/2 \times 10^{-16} L_a^2 \text{ cm}^2$), A_{DNA} (area of ds-DNA section, $A_{DNA} = \pi (d_{DNA}/2)^2 \times 10^{-16} \text{ cm}^2$), S_{DNA} (area of lying ds-DNA on Au(111), $S_{DNA} = l_{DNA} d_{DNA} \times 10^{-16} \text{ cm}^2$), N_A (Avogadro constant, $6.02 \times 10^{23} \text{ mol}^{-1}$), a (the perpendicular interaxial distance of adjacent tilted ds-DNA molecules, $a \geq 20 \text{ Å}$), 1 S-S was the distance (5 Å) of adjacent hollow sites (f_{cc}) on Au(111).

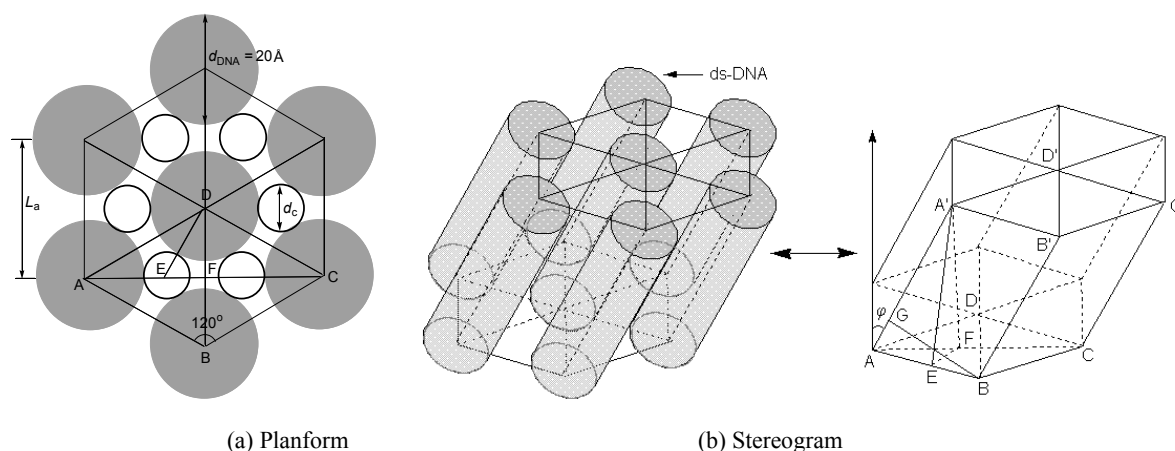


Fig. S7 Planform and stereogram of ds-DNA-SAMs on Au(111).

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