

Supplementary Information

Rapid and highly sensitive protein extraction via cobalt oxide nanoparticle based liquid-liquid microextraction coupled with MALDI mass spectrometry

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Calculation of concentration of Co₃O₄/CTA⁺ NP

Number of surface atoms of Co₃O₄/CTA⁺ NP and concentration were calculated according to reported methods [1–3]. The average size of the Co₃O₄/CTA⁺ NP was about 50 nm measured by SEM. We can assume NP as a large spherical metal clusters with the following conditions (R. L. Johnston, “Atomic and Molecular Clusters” Taylor & Francis, 2002):

$$V_{\text{cluster}} = NV_{\text{atom}} \quad \dots \quad (1)$$

$$\frac{4}{3}\pi (R_{\text{cluster}})^3 = N \frac{4}{3} \pi (R_{\text{atom}})^3 \quad \dots \quad (2)$$

Where V_{cluster} = cluster or atomic volume; R_{atom} = cluster or atomic radius; N = total number of atoms within the cluster.

Further,

$$R_{\text{cluster}} = N^{1/3} R_{\text{atom}} \quad \dots \quad (3)$$

Then, we can calculate the cluster radius by calculating

Calculation of Concentration of Co₃O₄/CTA⁺ NP

From SEM result, we know that R_{cluster} = 50 nm and R_{Co₃O₄}=141 pm. The number of Co₃O₄ atoms per nanoparticle is estimated by Eq. 3.

$$N = (R_{\text{cluster}} / R_{\text{Co}_3\text{O}_4/\text{CTA}^{++}})$$

$$= (50 \times 10^{-9} \text{ m} / 141 \times 10^{-12} \text{ m})^3$$

44591588 Co and O atoms per Co₃O₄ /CTA⁺ NP

1 mg of $\text{Co}_3\text{O}_4/\text{CTA}^+$ NP prepared was dissolved in 100 mL of toluene, so the molar (M) concentration was 0.041M and the number of the Co_3O_4 particles in the solutions:

$$N_{\text{atom}} = 100 \times 0.041 \times N_A \quad (N_A \text{ is the Avagadro's number})$$

$$= 100 \times 0.041 \times 6.022 \times 10^{23}$$

$$= 2.47 \times 10^{24}$$

$$N_{\text{NP}} = N_{\text{atom}} / N$$

$$N_{\text{NP}} = (2.47 \times 10^{24}) / (44591588)$$

$$N_{\text{NP}} = 5.5 \times 10^{16}$$

Six different dilute concentrations of $\text{Co}_3\text{O}_4/\text{CTA}^+$ particles were prepared in 100 μL toluene for the separation of biomolecules from the sample solution and in 1 μL (used for MALD-MS analysis) the particle/ μL would be:

1). $0.7 \times 10^{12}/ \mu\text{L}$

2). $1.4 \times 10^{12}/ \mu\text{L}$

3). $4.0 \times 10^{12}/ \mu\text{L}$

4). $6.7 \times 10^{12}/ \mu\text{L}$

5) $8.0 \times 10^{12}/ \mu\text{L}$

5). $9.4 \times 10^{12}/ \mu\text{L}$

This values of concentration of $\text{Co}_3\text{O}_4/\text{CTA}^+$ particles/ μL used in the experiments throughout the manuscript were calculated based on the following references:

1. J. D. Lewis, M. T. Day, V. J. MacPherson, Z. Pikeramenou, *Chem. Comm.* 2006, **13**, 1433 - 1435.
2. A. Murugadoss, A. Chattopadhyay, 2008, **9**, Nanotechnology (9 pages).
3. I.Yu, T. Isobe, M. Senna, *J. Phys. Solids*, 1996, **57**, 373 – 379.