

# Chiral recognition of 2-alkyl alcohols with porphyrin

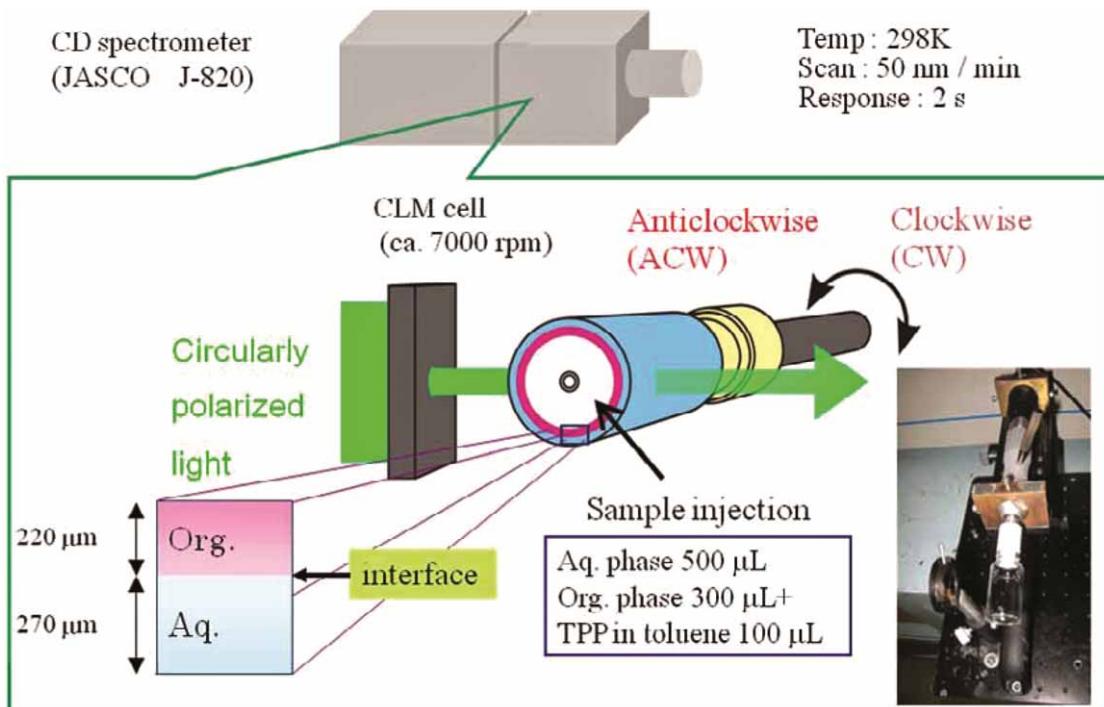
## J-nanoaggregates at liquid-liquid interface

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**Figure S1** Schematic illustration of Centrifugal Liquid Membrane-Circular Dichroism (CLM-CD) method.

Table S1. Correlations of CD intensity of  $\text{H}_4\text{TPP}^{2+}$  aggregate and 2-nonaol concentration. These spectra were measured 1 min after introducing the  $\text{H}_2\text{TPP}$ -dodecane solution.  $[\text{TPP}]_{\text{org}} = 2.8 \times 10^{-5} \text{ M}$ ,  $[2\text{-nonenol}]_{\text{org}} = 0 - 1.0 \times 10^{-3} \text{ M}$ ,  $[\text{H}_2\text{SO}_4]_{\text{aq}} = 4 \text{ M}$ .  $\Delta\text{CD}$  is defined as the sum of the absolute values of the 1st and 2nd Cotton intensities.

Concentration / M	S-(+)-2-nonanol			R-(-)-2-nonanol		
	1st Cotton / mdeg	2nd Cotton / mdeg	$\Delta\text{CD}$ / mdeg	1st Cotton / mdeg	2nd Cotton / mdeg	$\Delta\text{CD}$ / mdeg
0	3.6	N/A	3.6	3.6	N/A	3.6
$1.0 \times 10^{-8}$	6.2	N/A	6.2	8.7	N/A	8.7
$1.0 \times 10^{-7}$	12.3	N/A	12.3	12.4	N/A	12.4
$1.0 \times 10^{-6}$	11.7	-8.3	20.0	-3.2	10.0	13.2
$1.0 \times 10^{-5}$	7.7	-2.5	10.2	-4.4	11.5	15.9
$1.0 \times 10^{-4}$	15.9	-11.6	27.5	-18.3	17.2	35.5
$1.0 \times 10^{-3}$	16.8	-3.4	20.2	-5.3	5.0	10.3
$1.0 \times 10^{-2}$	10.5	-8.5	19.0	-8.4	11.6	20.0

Table S2. Correlations of CD intensity of H<sub>4</sub>TPP<sup>2+</sup> aggregate and carbon chain length of chiral alcohol. These data were measured 1 min after introducing the H<sub>2</sub>TPP-dodecane solution. [TPP]<sub>org</sub> = 2.8 × 10<sup>-5</sup> M, [2-alkyl alcohol]<sub>org</sub> = 1.0 × 10<sup>-2</sup> M, [H<sub>2</sub>SO<sub>4</sub>]<sub>aq</sub> = 4 M.

Chiral alcohol	Number of carbon atoms	S-(+)-2-alkyll alcohol		
		1st Cotton / mdeg	2nd Cotton / mdeg	ΔCD / mdeg
2-butanol	4	5.8	N/A	5.8
2-pentanol	5	3.8	-2.2	6.0
2-hexanol	6	4.6	-2.0	6.6
2-heptanol	7	3.4	-4.7	8.1
2-octanol	8	4.7	-3.7	8.4
2-nonanol	9	10.5	-8.5	19.0

Table S3. Dependence of  $\Delta CD$  decreasing rate constant of  $H_4TPP^{2+}$  aggregates on 2-alkyl alcohol concentration.  $[TPP]_{org} = 1.5 \times 10^{-5} M$ ,  $[H_2SO_4]_{aq} = 4 M$ .

Chiral Alcohol	Decreasing rate constant / min <sup>-1</sup>					
	2-butanol	2-pentanol	2-hexanol	2-heptanol	2-octanol	2-nonanol
Number of carbon atoms	4	5	6	7	8	9
$1.0 \times 10^{-6} M$	N/A	N/A	N/A	N/A	$8.7 \times 10^{-2}$	$8.1 \times 10^{-2}$
$1.0 \times 10^{-5} M$	N/A	N/A	N/A	$9.0 \times 10^{-2}$	$8.1 \times 10^{-2}$	$7.1 \times 10^{-2}$
$1.0 \times 10^{-4} M$	N/A	N/A	$9.3 \times 10^{-2}$	$8.2 \times 10^{-2}$	$7.4 \times 10^{-2}$	$4.8 \times 10^{-2}$
$1.0 \times 10^{-3} M$	N/A	$5.0 \times 10^{-2}$	$8.3 \times 10^{-2}$	$7.5 \times 10^{-2}$	$6.9 \times 10^{-2}$	$2.4 \times 10^{-2}$

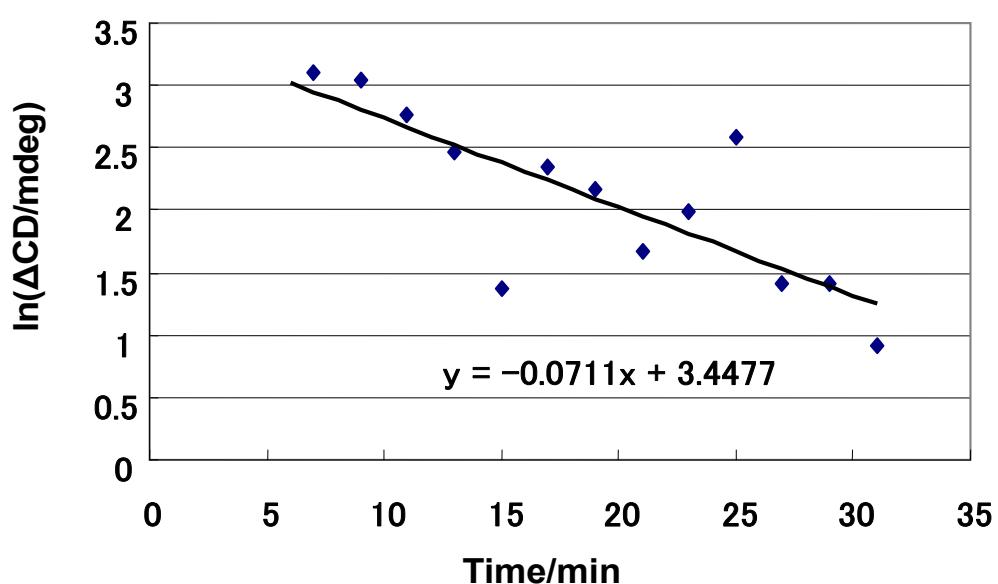


Figure S2. An example of the time dependence of  $\Delta CD$  intensity of the aggregate. The t = 0 min was set to the time of addition of the TPP-dodecane solution.  $[TPP]_{org} = 1.5 \times 10^{-5} M$ ,  $[(S)-(+)-2\text{-nonanol}]_{org} = 1.0 \times 10^{-5} M$ ,  $[H_2SO_4]_{aq} = 4 M$ .