

Electronic Supplementary Information

BSA-templated MnO₂ nanoparticles as both peroxidase and oxidase mimics

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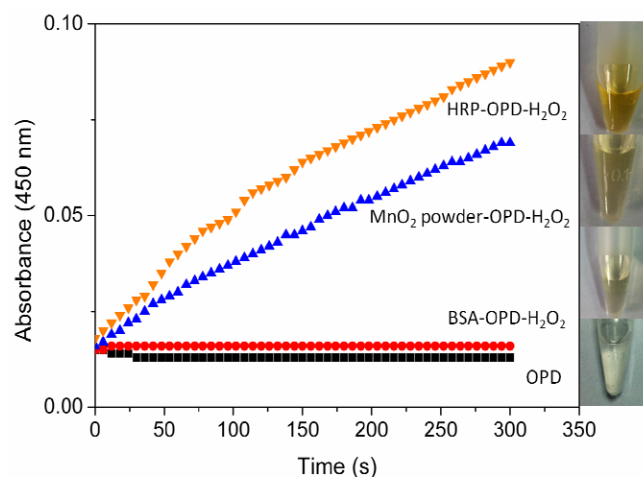


Fig. S1. The time-dependent absorbance changes at 450 nm of OPD in the presence of H_2O_2 with $3.3 \mu\text{g mL}^{-1}$ HRP or $3.7 \mu\text{g mL}^{-1}$ micrometer sized MnO_2 powder or $3.7 \mu\text{g mL}^{-1}$ BSA.

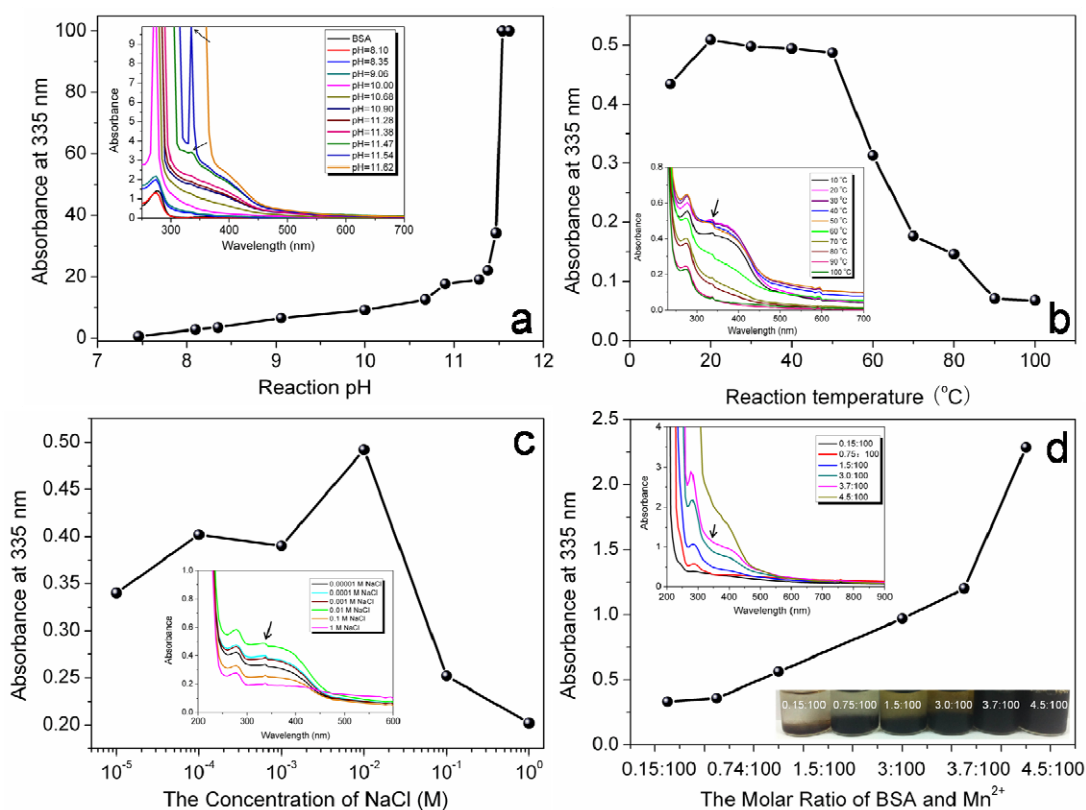


Fig. S2. Optimization of a) reaction pH from 8.10 to 11.62; b) the reaction temperature of BSA- MnO_2 NPs; c) the ionic strength of BSA- MnO_2 NPs; and d) the molar ratio of BSA and Mn^{2+} precursor. Inset: UV-vis spectra at 335 nm for MnO_2

NPs in different conditions and the styleonme of different molar ratio of BSA and Mn^{2+} precursor.

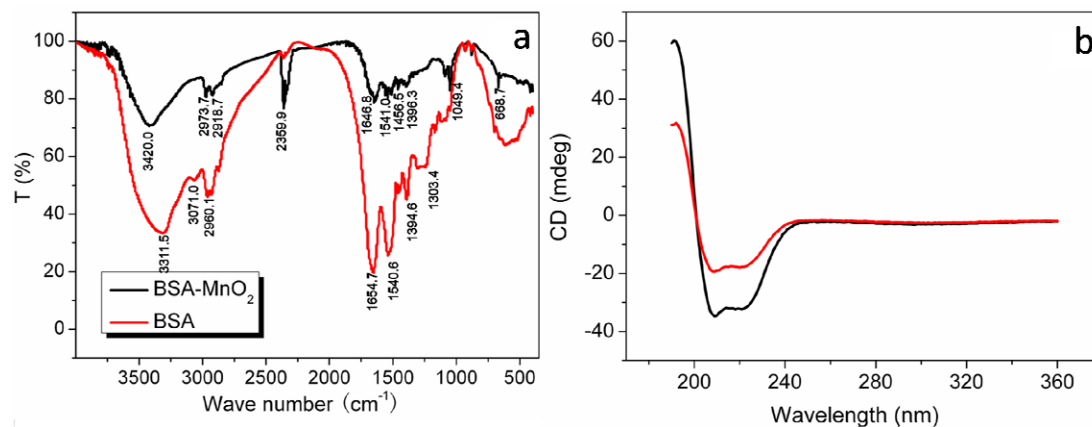


Fig. S3. a) FTIR spectra of BSA-MnO₂ NPs (black line) and pure BSA (red line); and b) CD spectra of BSA-MnO₂ NPs (red line) and pure BSA (black line).

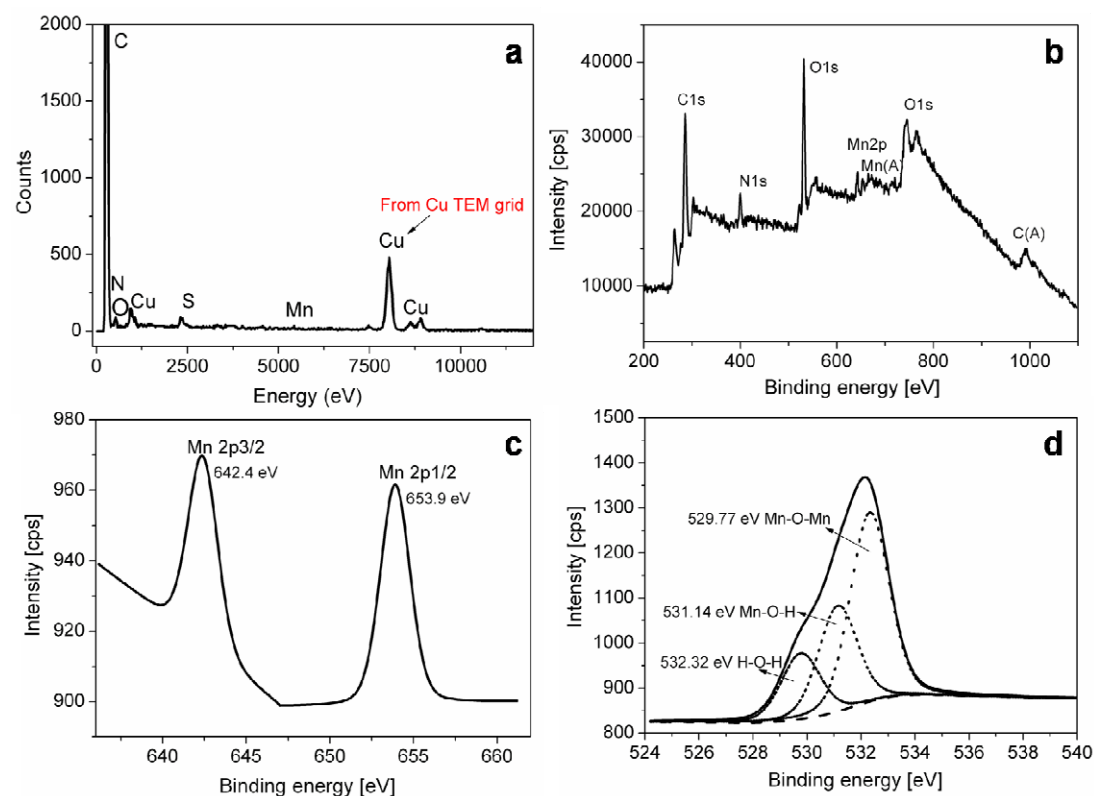


Fig. S4. a) EDX spectra of BSA-MnO₂ NPs; XPS spectra of b) BSA-MnO₂ NPs; c) Mn 2p; and d) O 1s.

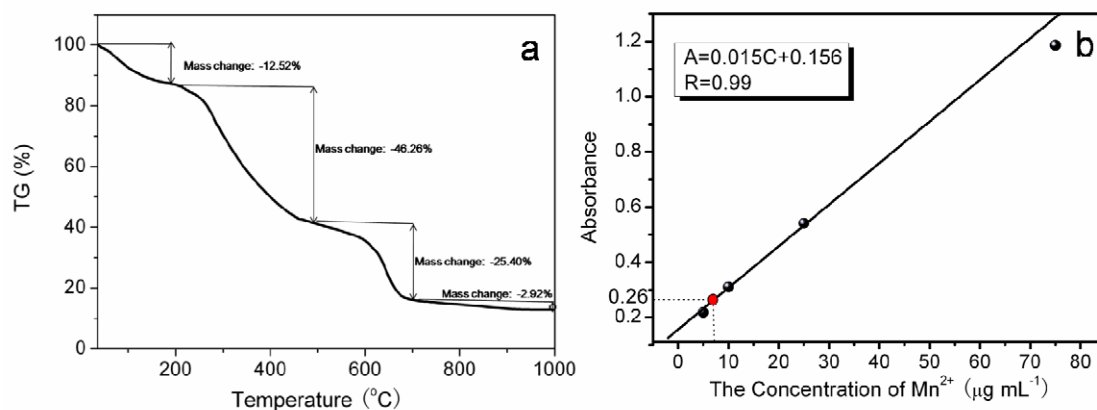


Fig. S5. a) TG of BSA-MnO₂ NPs; and b) the concentration of Mn element in BSA-MnO₂ NPs diluted 10 times by FAAS detection.

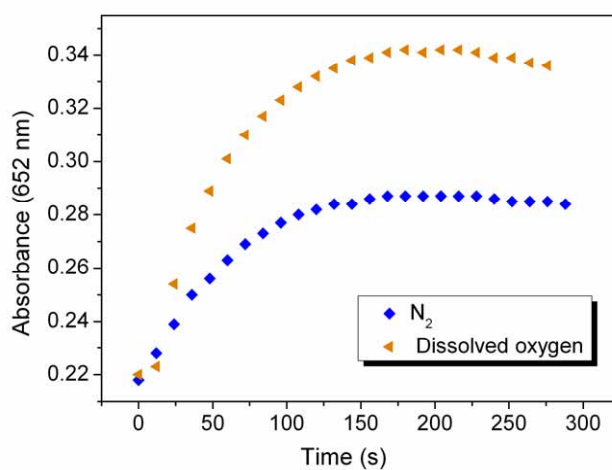


Fig. S6. Effect of dissolved oxygen (yellow line) on TMB oxidation at 35 °C and pH 4.0. The reaction rate after bubbling with high purity nitrogen for 30 minutes is greatly reduced (blue line).

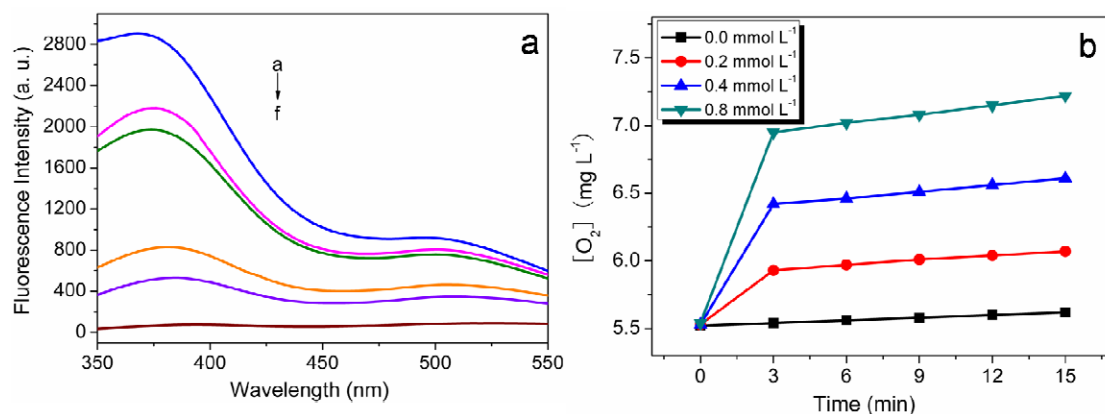


Fig. S7. a) The effect of the BSA-MnO₂ NPs on the formation of hydroxyl radical with terephthalic acid as a fluorescence probe. a-f: 7.3, 11.0, 14.0, 22.0, 27.5, 55.0 $\mu\text{g mL}^{-1}$. 10 mM H₂O₂, 0.5 mM terephthalic acid and different concentrations of the BSA-MnO₂ NPs were first incubated in 100 mM NaAc buffer (pH 5.0) exposed to UV light at 365 nm for 20 min; and b) the effect of the BSA-MnO₂ NPs concentration on the generation of O₂ by decomposition of H₂O₂. Reaction conditions: 50 mM H₂O₂ and different concentrations of BSA-MnO₂ NPs in 100 mM NaAc buffer (pH 5.0).

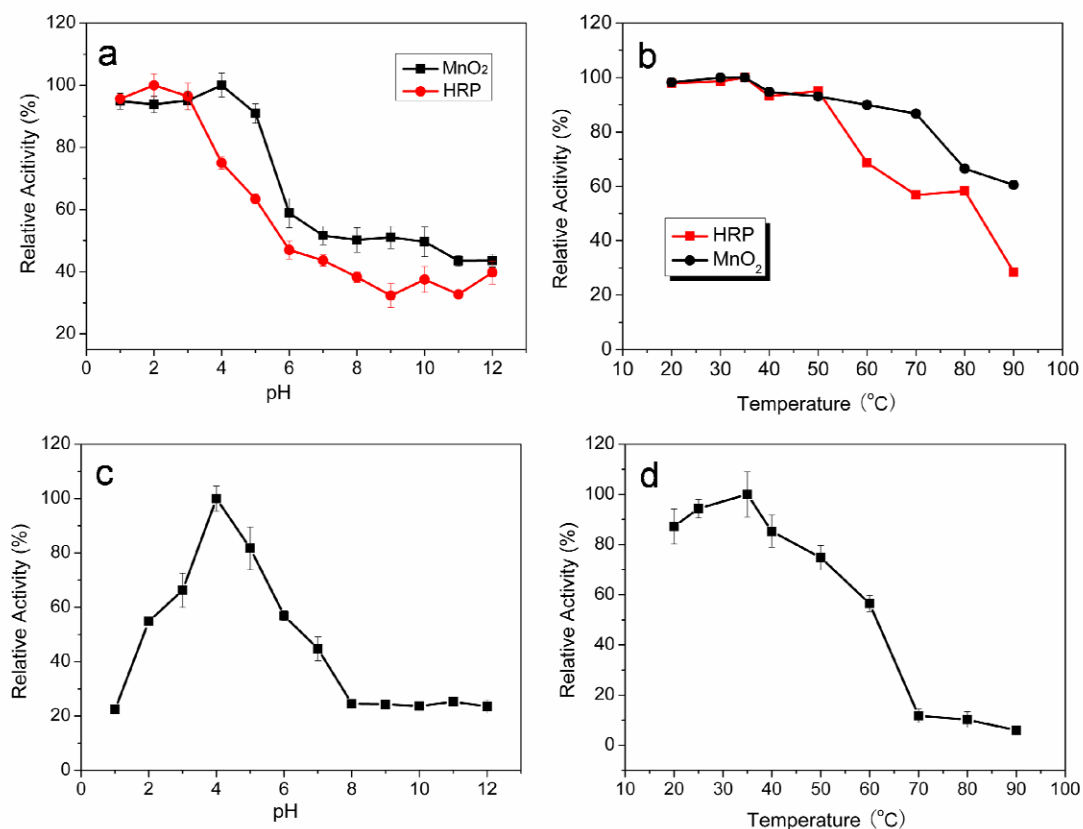


Fig. S8. Dependency of the peroxidase-like activity of BSA-MnO₂ NPs and HRP on pH a) of OPD oxidation and temperature b) of OPD oxidation with H₂O₂; and oxidase-like activity of BSA-MnO₂ NPs on pH c) of TMB oxidation and temperature d) of TMB oxidation. Experiments were carried out using 3.7 $\mu\text{g mL}^{-1}$ BSA-MnO₂ NPs or 3.3 $\mu\text{g mL}^{-1}$ HRP in 300 μL NaAc buffer contains 3.3 mM OPD substrate, and 3.7 $\mu\text{g mL}^{-1}$ BSA-MnO₂ NPs in 300 μL citrate buffer contains 1.7 mM TMB substrate. The H₂O₂ concentration was 0.3 mM at pH 4.0 and temperature 35 °C unless otherwise stated. The maximum point in each curve was set as 100%.

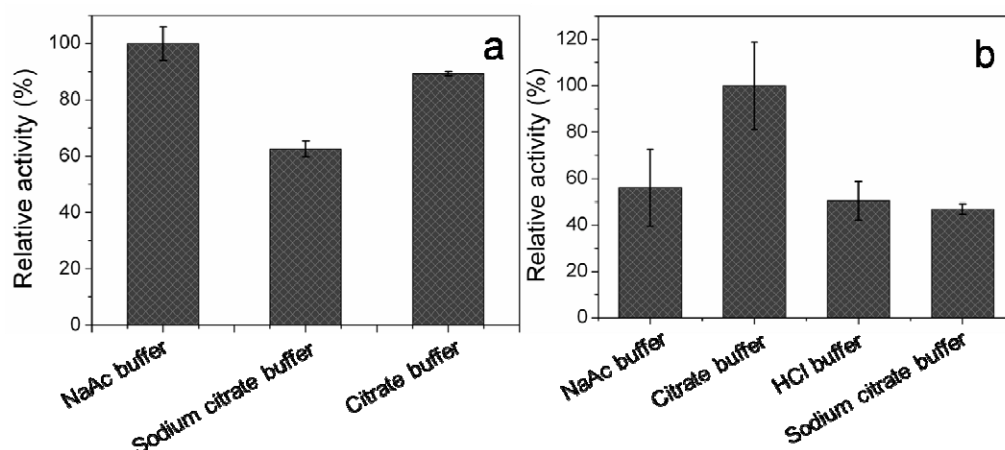


Fig. S9. Optimization conditions of all kinds of buffers a) for OPD and b) for TMB under standard conditions. 0.2 mol L⁻¹ NaAc buffer (8.204 g NaAc in 500 mL deionized water using HAc adjusting pH 4.0); 0.2 mol L⁻¹ citrate buffer (7.300 g Na₂HPO₄·2H₂O and 4.665 g citrate acid in 500 mL deionized water); 0.2 mol L⁻¹ sodium citrate buffer (6.885 g sodium citrate in 500 mL deionized water using citrate acid adjusting pH 4.0); and HCl buffer (7.300 g Na₂HPO₄·2H₂O in 500 mL deionized water using HCl adjusting pH 4.0).

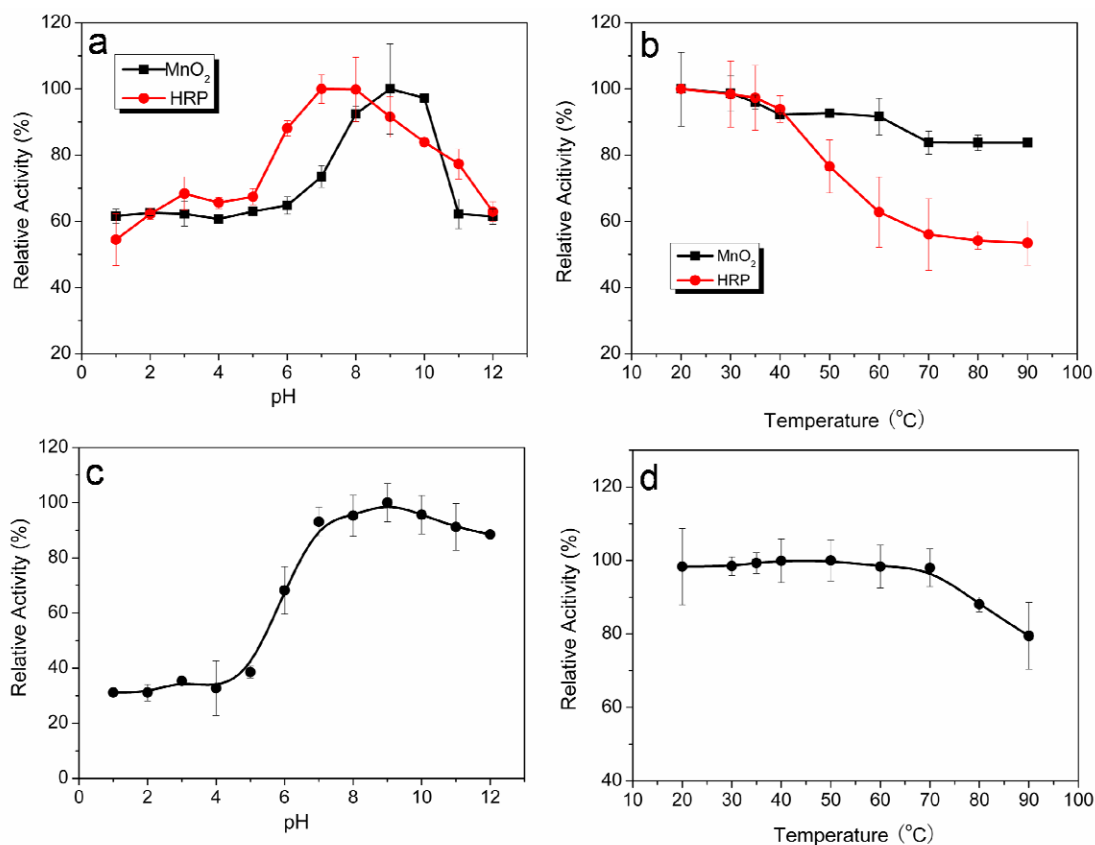


Fig. S10. a, b) The stability of BSA-MnO₂ NPs and HRP with OPD oxidation; and c, d) the stability of BSA-MnO₂ NPs with TMB oxidation. a, c) BSA-MnO₂ NPs and HRP were first incubated at pH from 1.0 to 12.0 for 2 h and then their peroxidase- and oxidase-like activities were measured; b, d) BSA-MnO₂ NPs and HRP were first incubated at temperature from 20 to 90 °C for 2 h and then the peroxidase- and oxidase-like activities were measured under standard conditions.

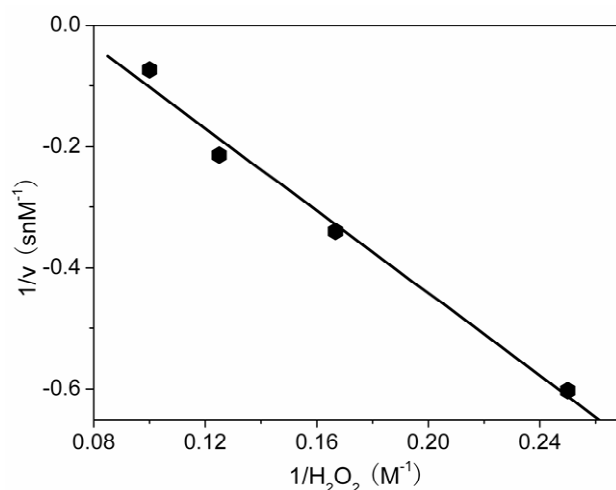


Fig. S11. Steady state kinetic assays with $3.7 \mu\text{g mL}^{-1}$ BSA-MnO₂ NPs. Double reciprocal plots of catalase activity of BSA-MnO₂ NPs and H₂O₂.

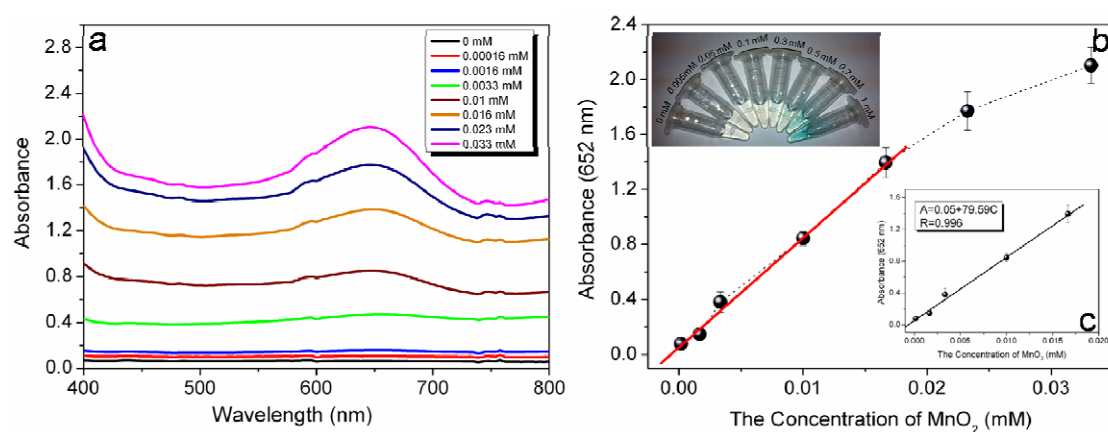


Fig. S12. a) UV-vis spectra for a solution of TMB-MnO₂ NPs in citrate buffer. The BSA-MnO₂ NPs concentrations were 0.00016, 0.0016, 0.0033, 0.010, 0.016, 0.023, 0.033 mM, respectively under standard conditions; and b) a dose-response curve for BSA-MnO₂ NPs detection. Inset: c) a linear calibration plot for detection of BSA-MnO₂ NPs.

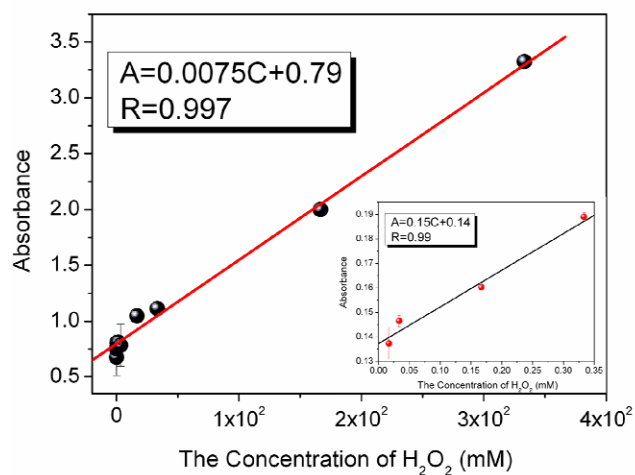


Fig. S13. A linear calibration plot for H₂O₂ detection using BSA-MnO₂ NPs as an artificial enzyme under standard conditions. Inset: a linear calibration plot for H₂O₂ using HRP as a comparison.