

Electronic Supplementary Information

Bio-inspired sensor based on surfactant film and Pd nanoparticles

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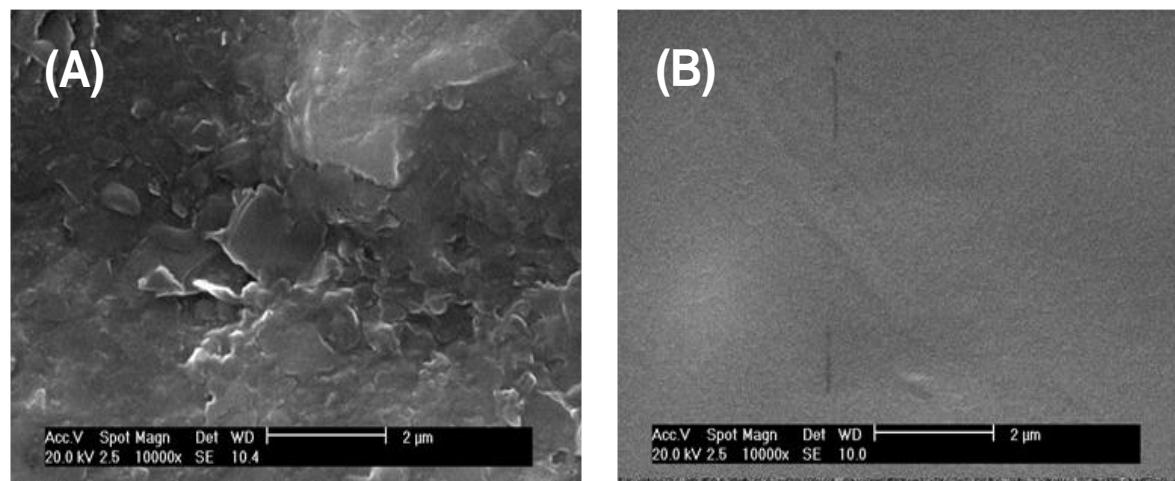


Fig. S1. SEM micrograph of (A) bare GCE (B) and GCE coated with Pd/ImS3-14 observed at 10 kV with magnification of 10K.

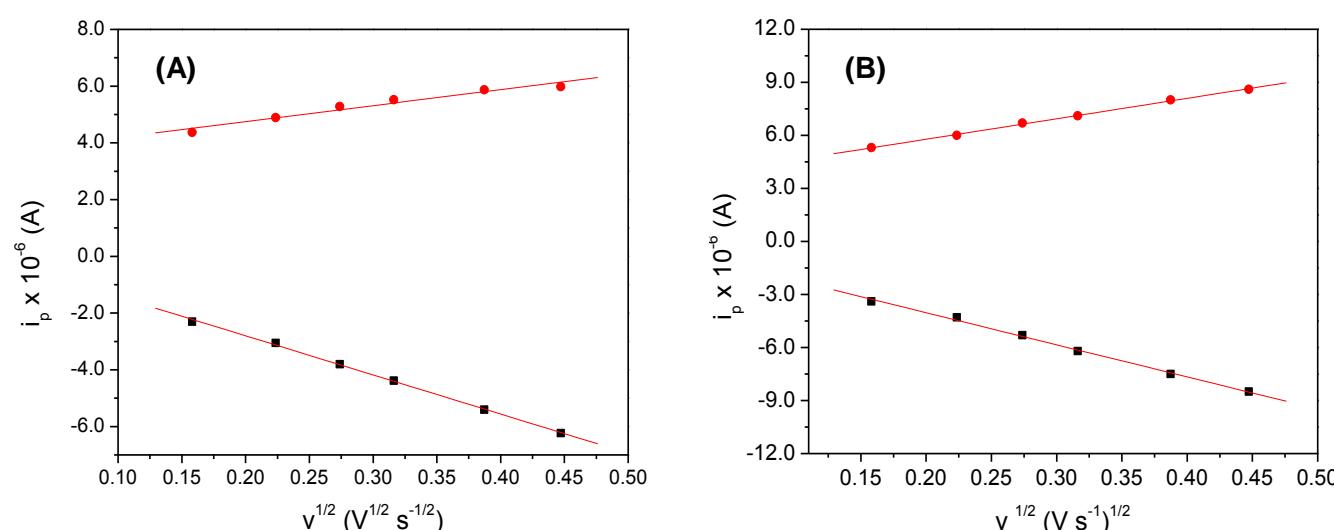


Fig. S2. Randles-Sevcik plot of (A) HQ and (B) CC, for Pd/ImS3-14/(Fe^{III}Cu^{II})/GCE with a scan rate in range of 25–200 mV s⁻¹.

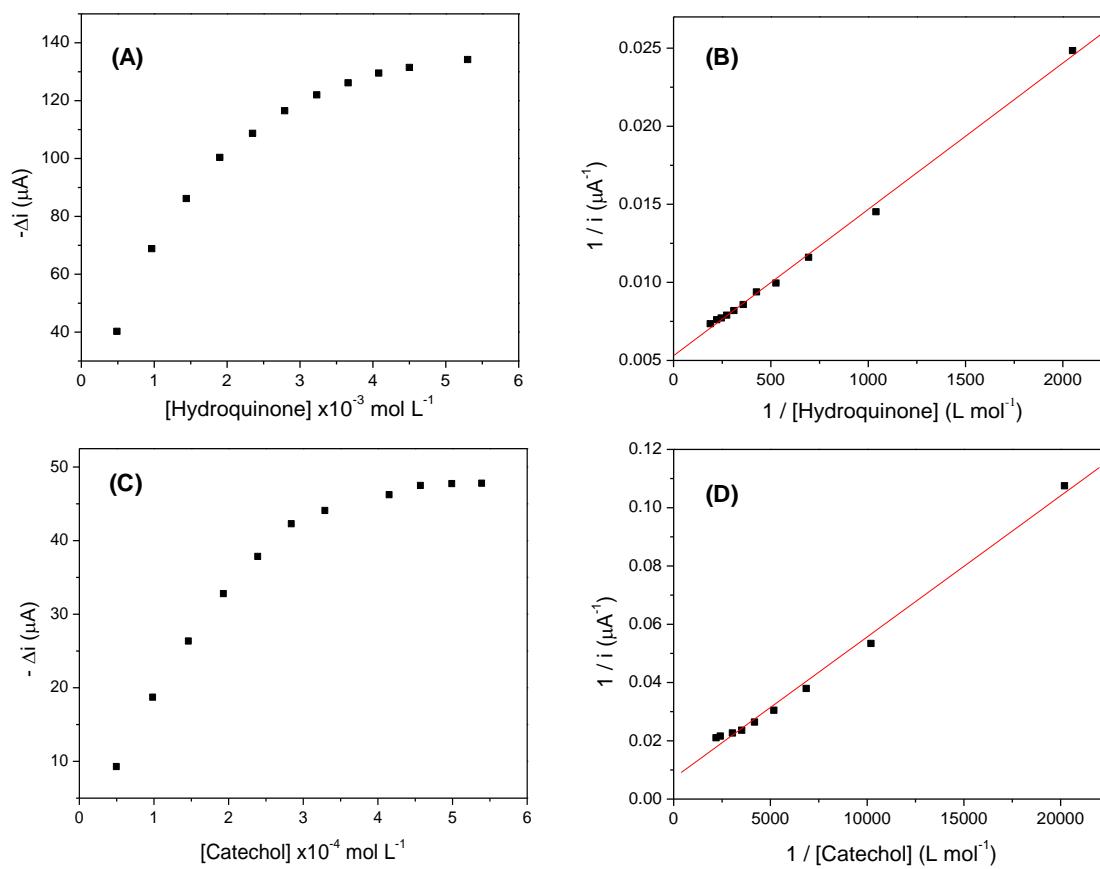


Fig. S3. Kinetic activity of complex as catecholoxidase and Lineweaver-Burk plot using solutions of HQ (A and B) and CC (C and D) in phosphate buffer (0.1 mol L⁻¹, pH 7.0).

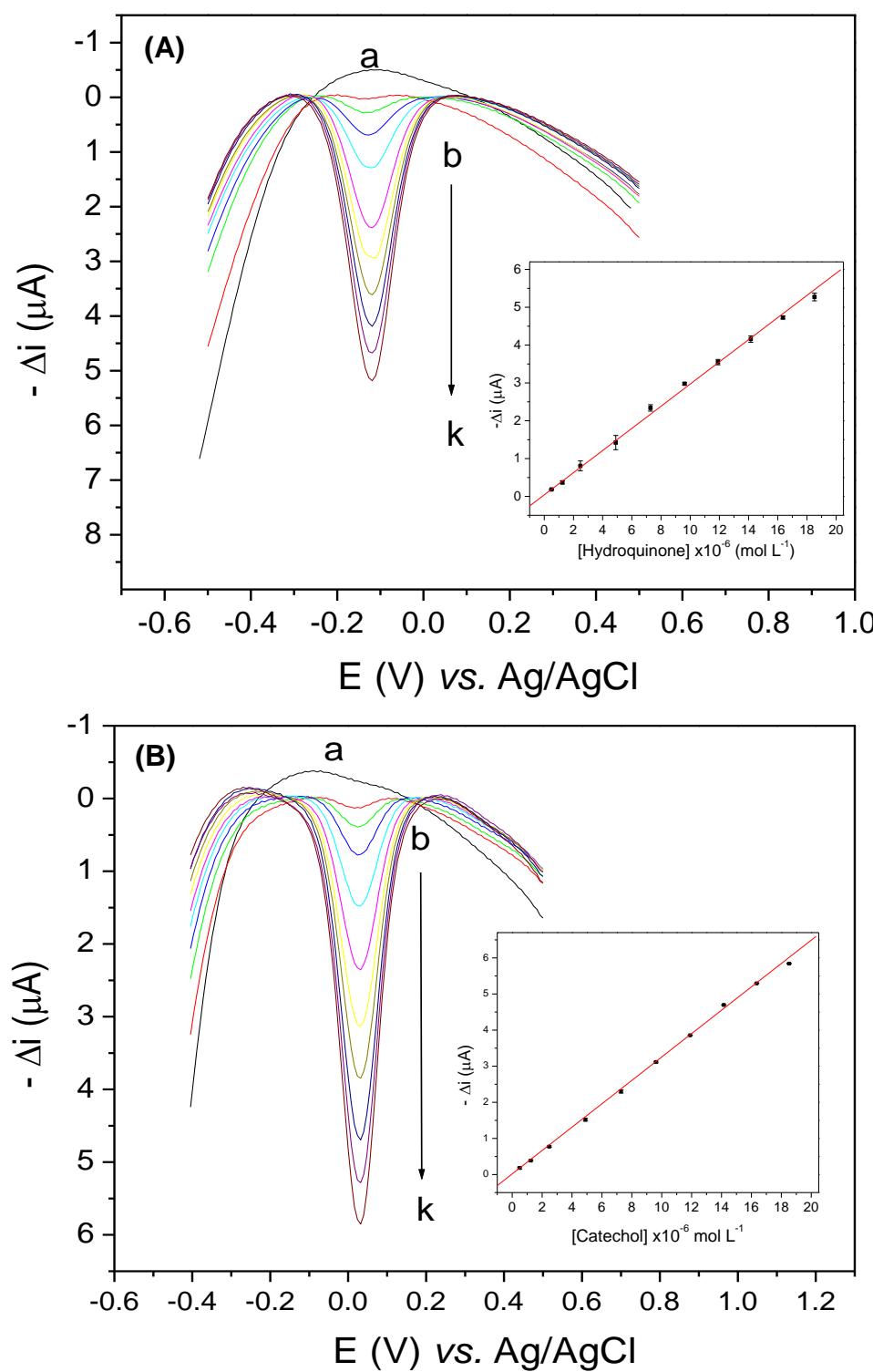


Fig. S4. Square-wave voltammograms obtained using the Pd/ImS3-14/(Fe^{III}/Cu^{II})/GCE sensor in (a) phosphate buffer solution only (0.1 mol L^{-1} , pH 7.0) and in solutions containing diphenol in the range of 4.99×10^{-7} to $1.85 \times 10^{-5} \text{ mol L}^{-1}$ (curves b-k). The insert shows the calibration curve. (A) HQ; (B) CC.

Table S1. CC an HQ recovery in cigarette filter extracts using the proposed sensor

Sample	CC ($\times 10^{-6}$ mol L $^{-1}$)			HQ ($\times 10^{-6}$ mol L $^{-1}$)		
	Added	Found ^a	Recovery (%) ^b	Added	Found ^a	Recovery (%) ^b
A	1.87	1.78 \pm 0.17	98.20 \pm 3.74	1.87	1.72 \pm 0.10	91.67 \pm 5.40
	3.70	3.92 \pm 0.13	105.90 \pm 3.53	3.70	4.09 \pm 0.15	99.42 \pm 4.09
	5.50	5.41 \pm 0.03	98.40 \pm 0.51	5.50	5.80 \pm 0.04	105.55 \pm 3.17
B	1.87	1.97 \pm 0.04	105.30 \pm 1.91	1.87	1.79 \pm 0.04	95.70 \pm 2.02
	3.70	3.88 \pm 0.12	104.80 \pm 3.34	3.70	3.60 \pm 0.05	97.30 \pm 1.29
	5.50	5.48 \pm 0.05	99.60 \pm 0.84	5.50	5.49 \pm 0.05	99.80 \pm 0.96
C	1.87	1.91 \pm 0.01	102.10 \pm 1.91	1.87	1.93 \pm 0.07	103.20 \pm 3.50
	3.70	3.83 \pm 0.06	103.50 \pm 3.83	3.70	3.73 \pm 0.21	100.80 \pm 0.96
	5.50	5.31 \pm 0.11	96.60 \pm 5.31	5.50	5.56 \pm 0.11	101.10 \pm 2.08

^a Mean \pm standard deviation; n=3.

^b Recovery = (mean found value / added value) \times 100%.