

3, 5-Diformyl-Borondipyrromethene for Selective Detection of Cyanide Anion

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S1

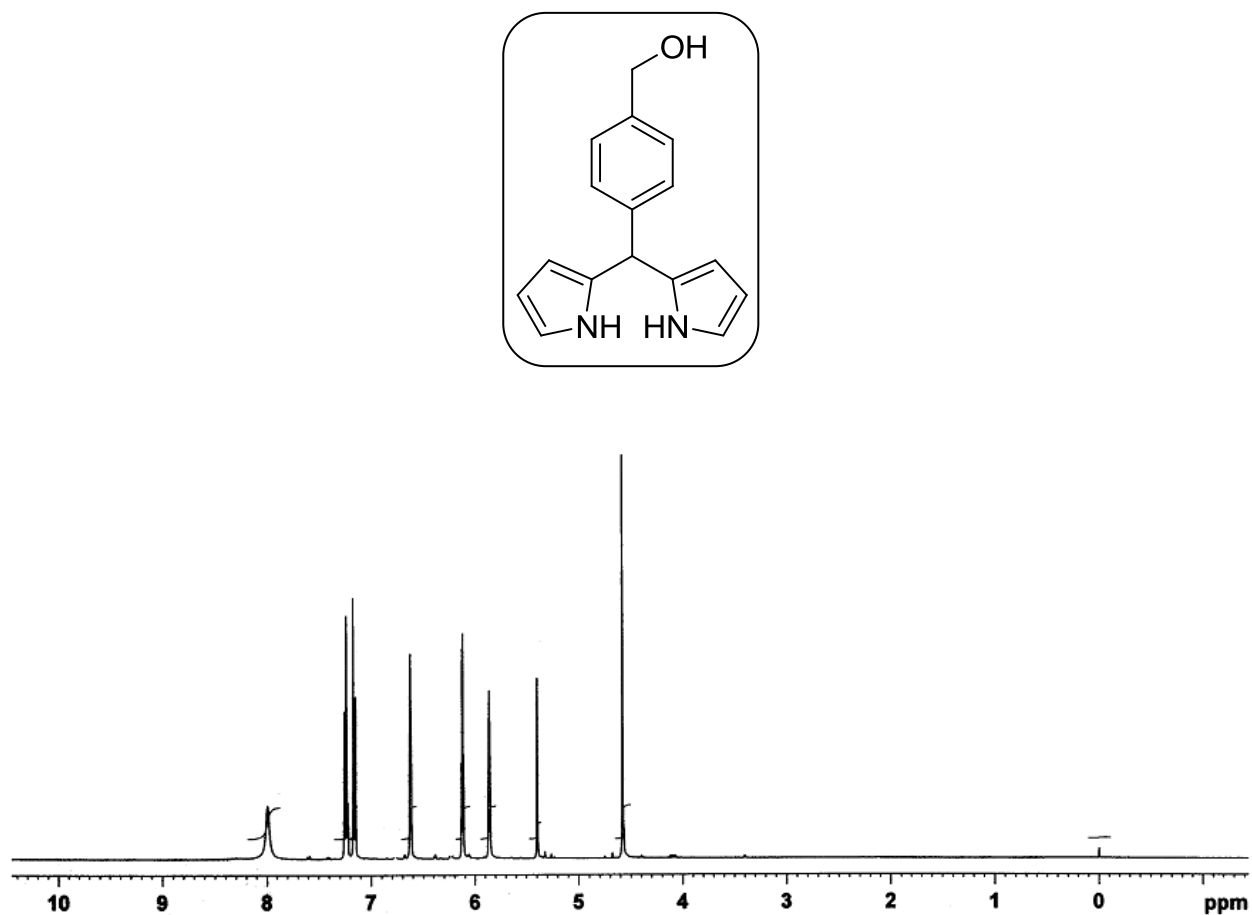


Figure S1: ^1H NMR spectrum of compound **4** recorded in CDCl_3

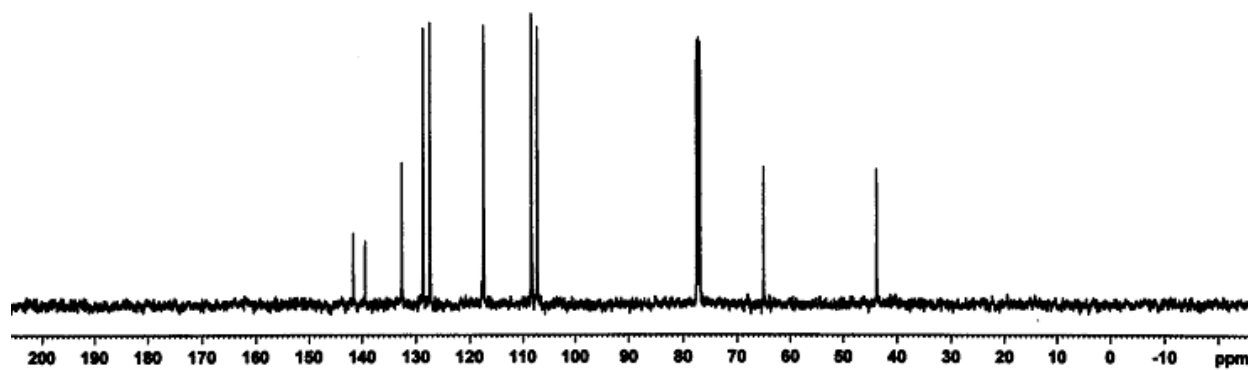
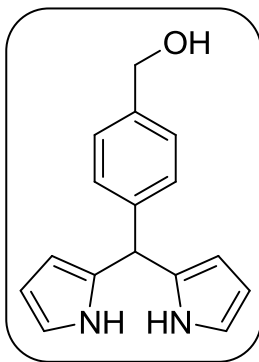


Figure S2: ^{13}C NMR spectrum of compound **4** recorded in CDCl_3

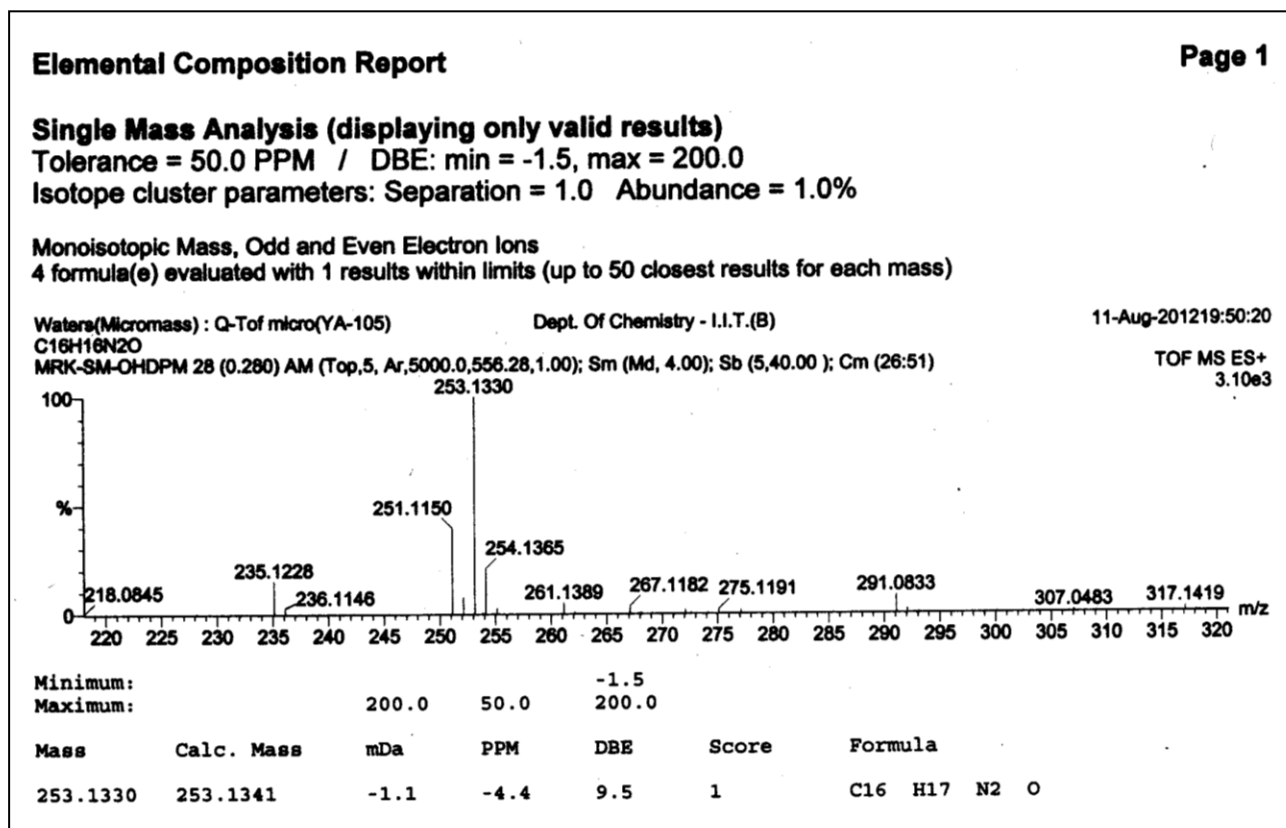
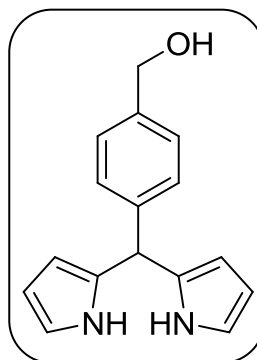


Figure S3: HRMS spectrum of compound 4

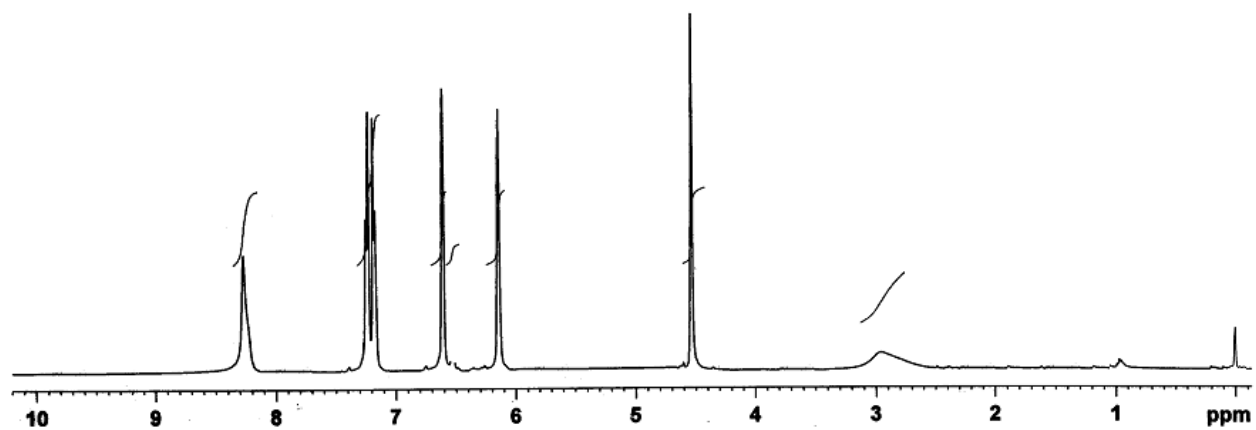
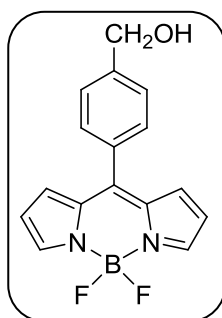


Figure S4: ¹H NMR spectrum of compound **5** recorded in CDCl₃

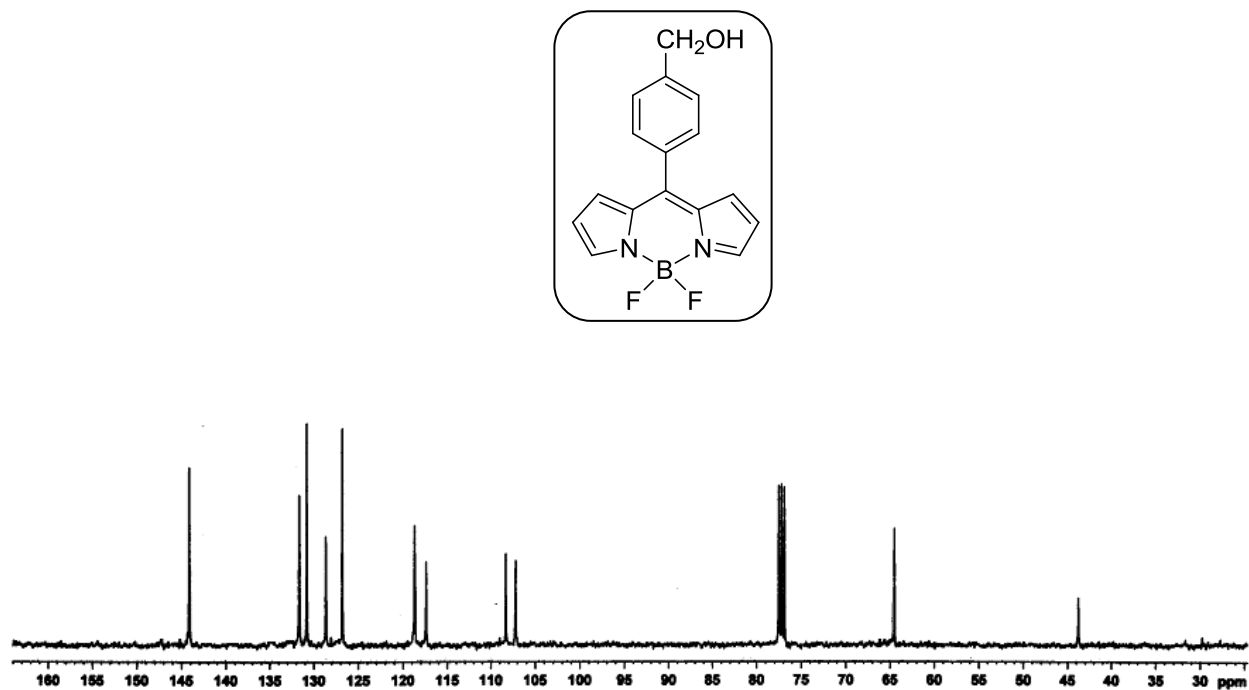


Figure S5: ^{13}C NMR spectrum of compound **5** recorded in CDCl_3

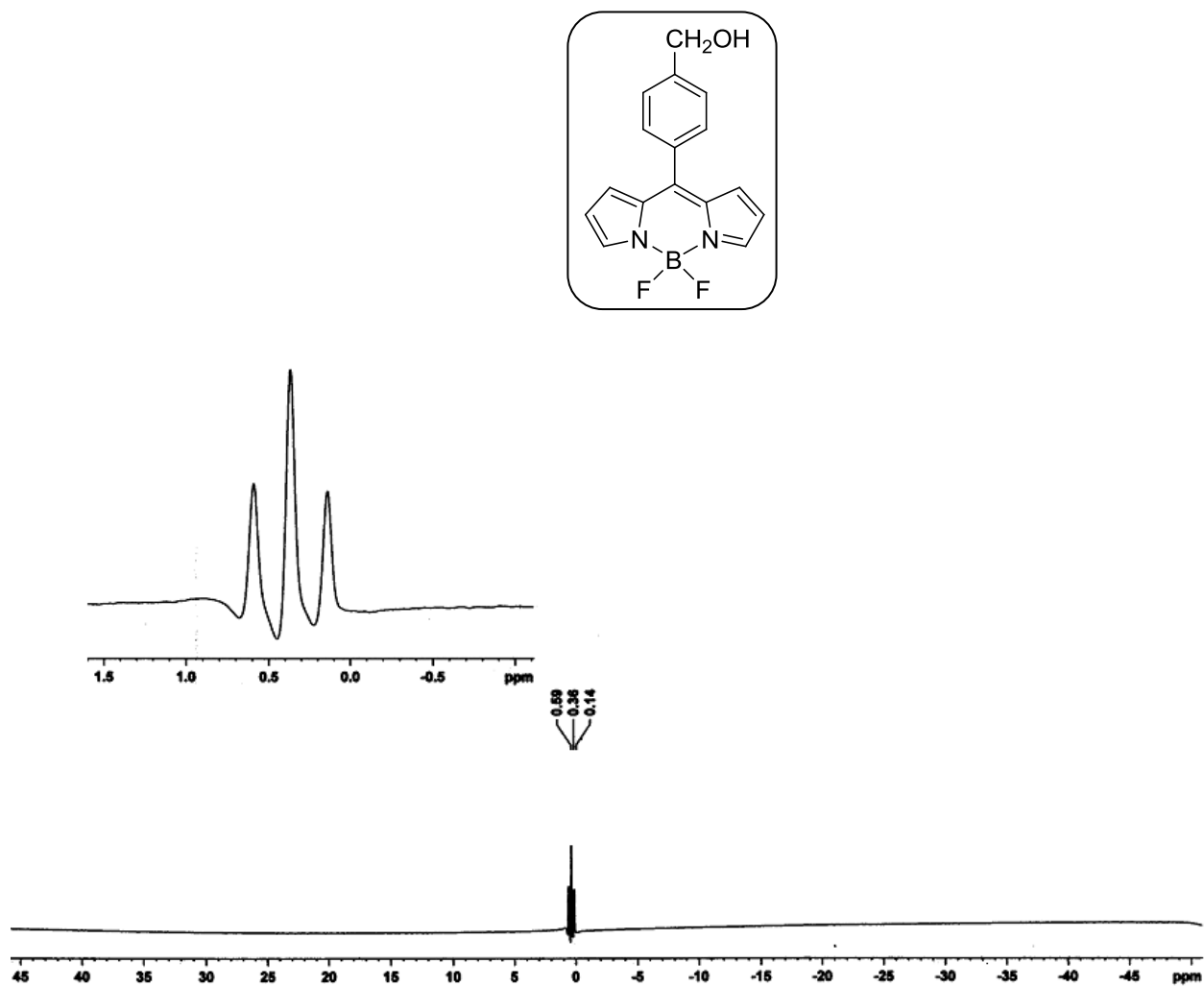


Figure S6: ^{11}B NMR spectrum of compound **5** recorded in CDCl_3 . Inset shows the expansion

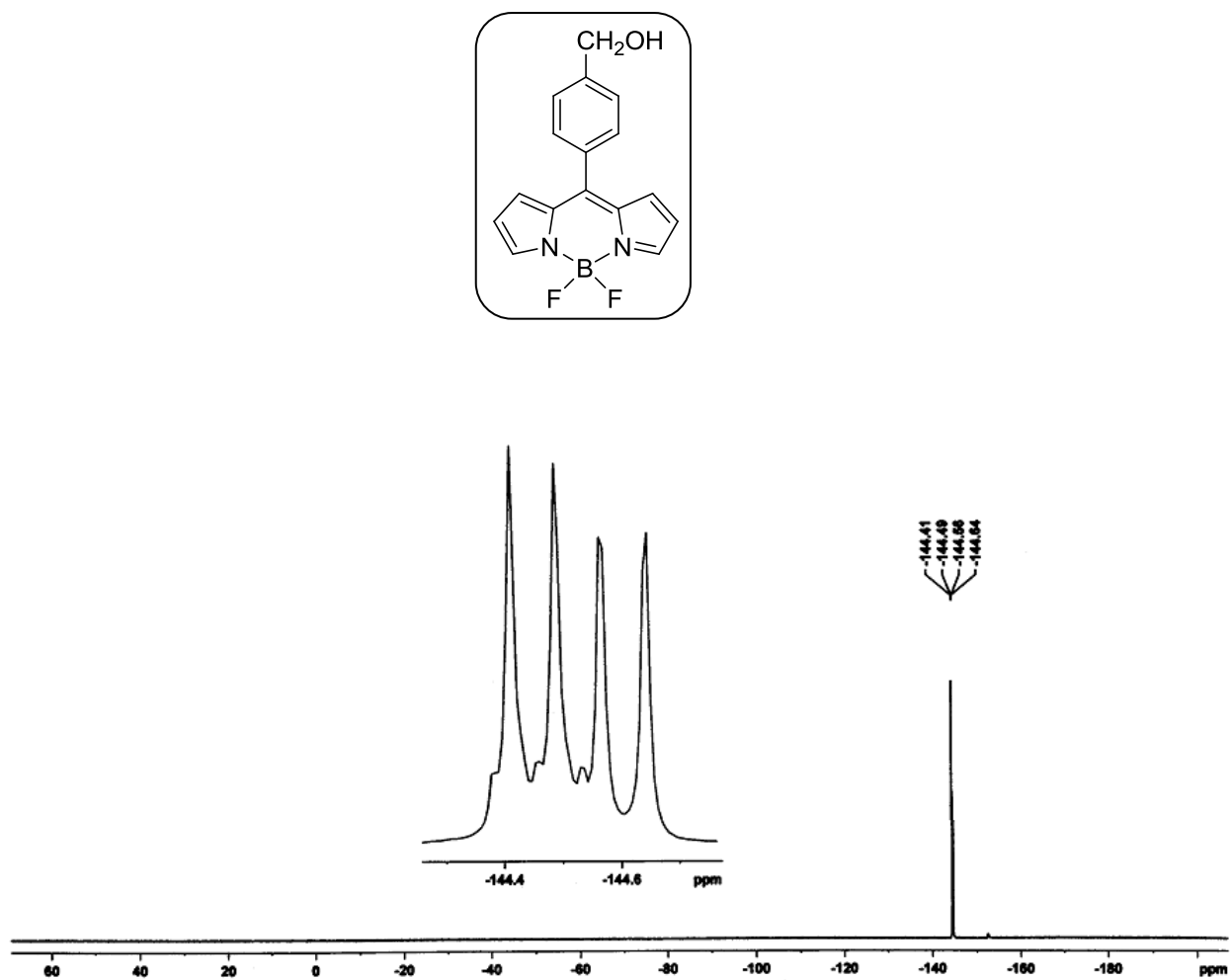


Figure S7: ^{19}F NMR spectrum of compound **5** recorded in CDCl_3 . Inset shows the expansion

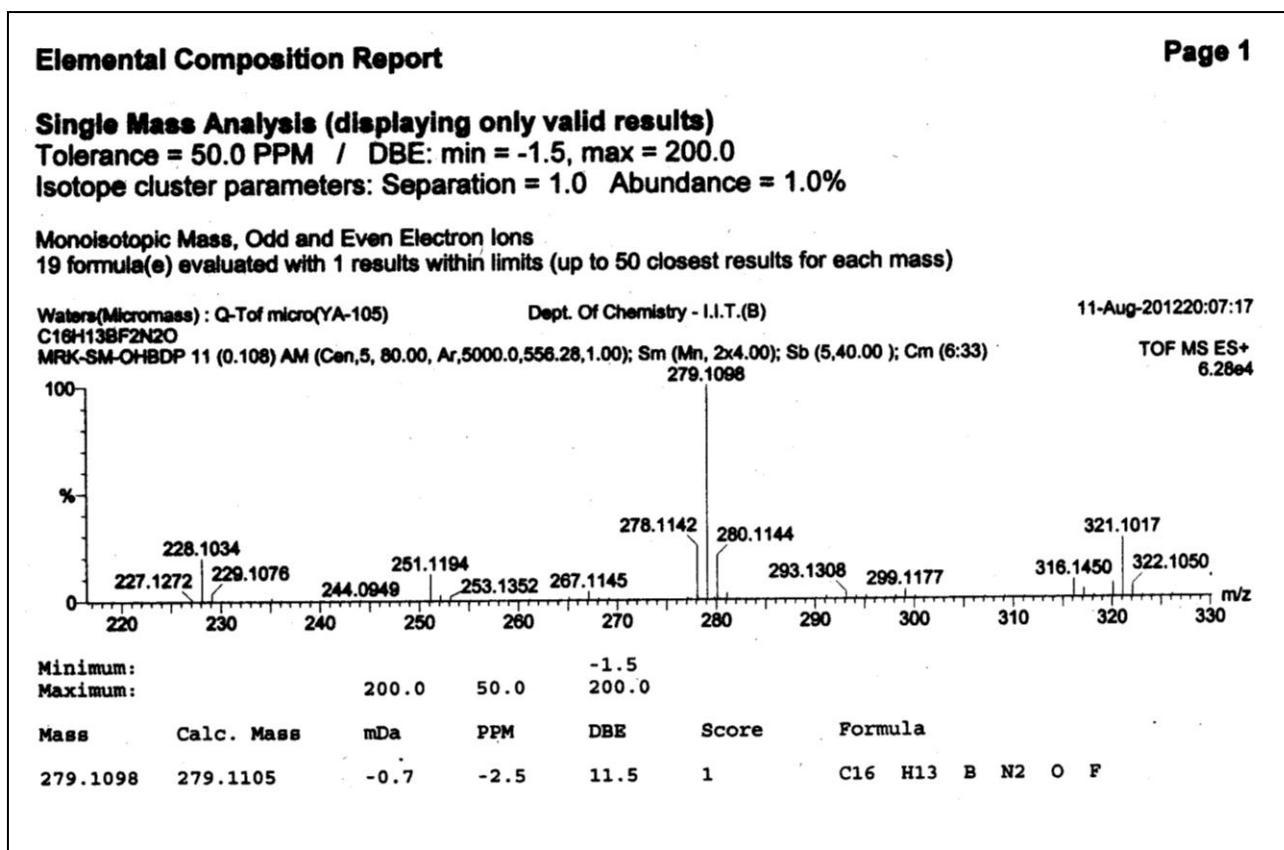
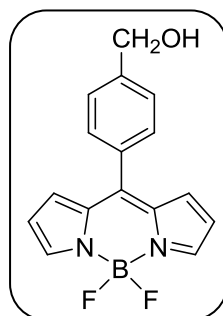


Figure S8: HRMS spectrum of compound 5

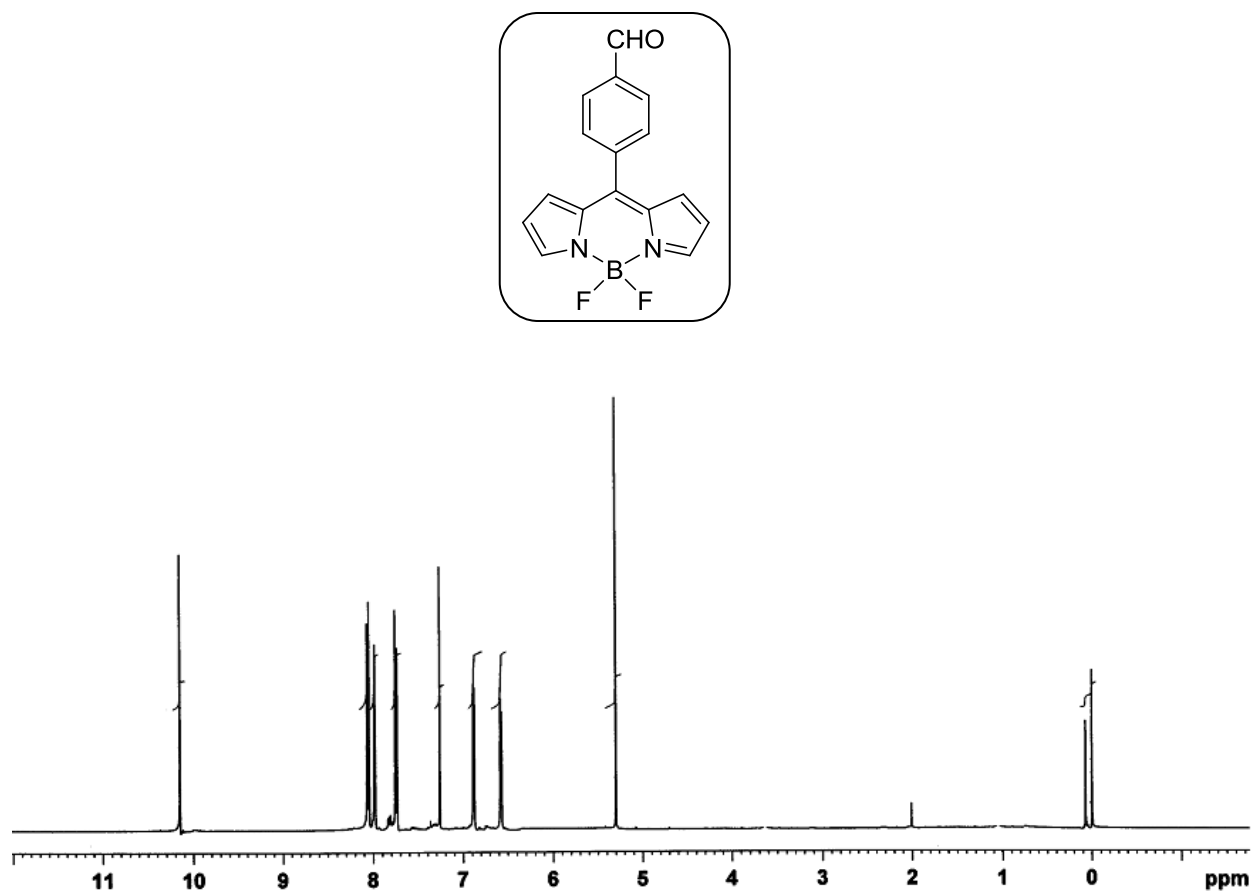


Figure S9: ^1H NMR spectrum of compound **3** recorded in CDCl_3

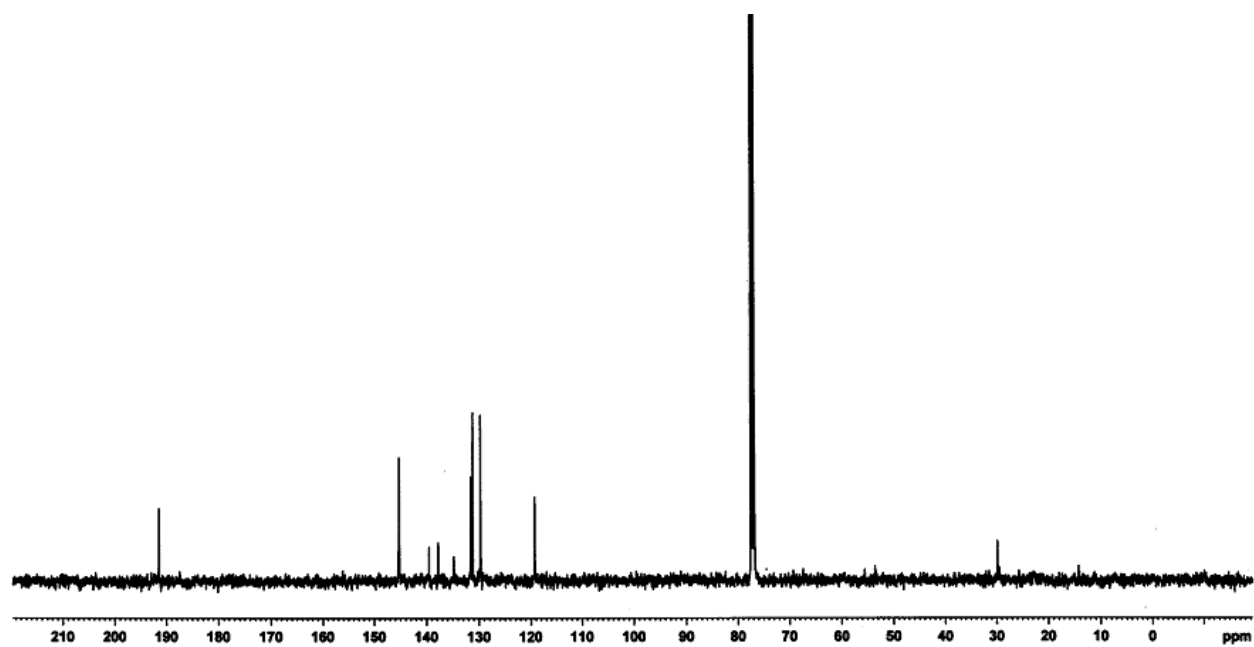
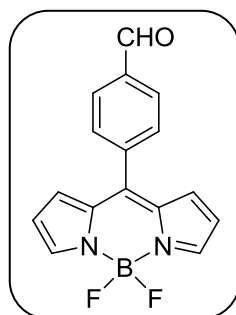


Figure S10: ^{13}C NMR spectrum of compound **3** recorded in CDCl_3

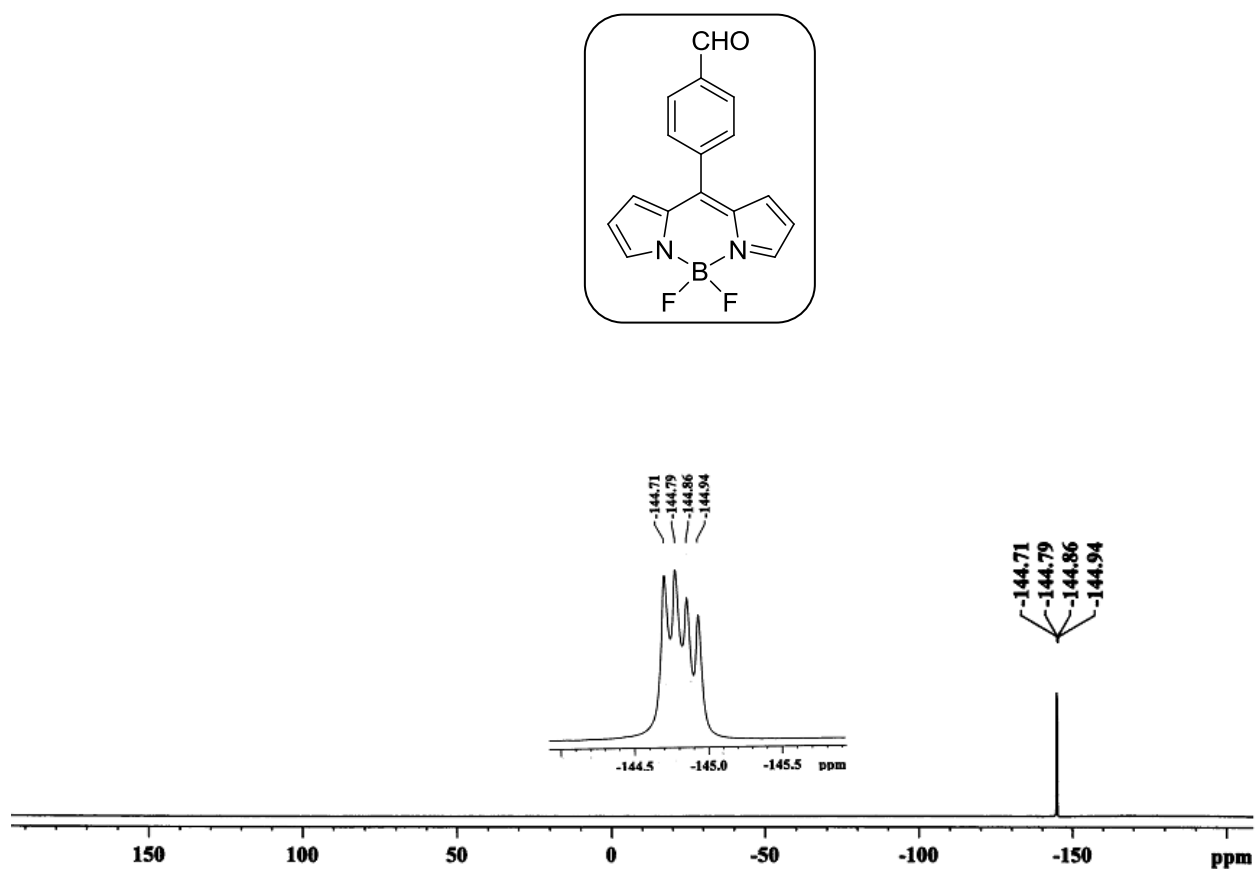


Figure S11: ^{19}F NMR spectrum of compound **3** recorded in CDCl_3 . Inset shows the expansion

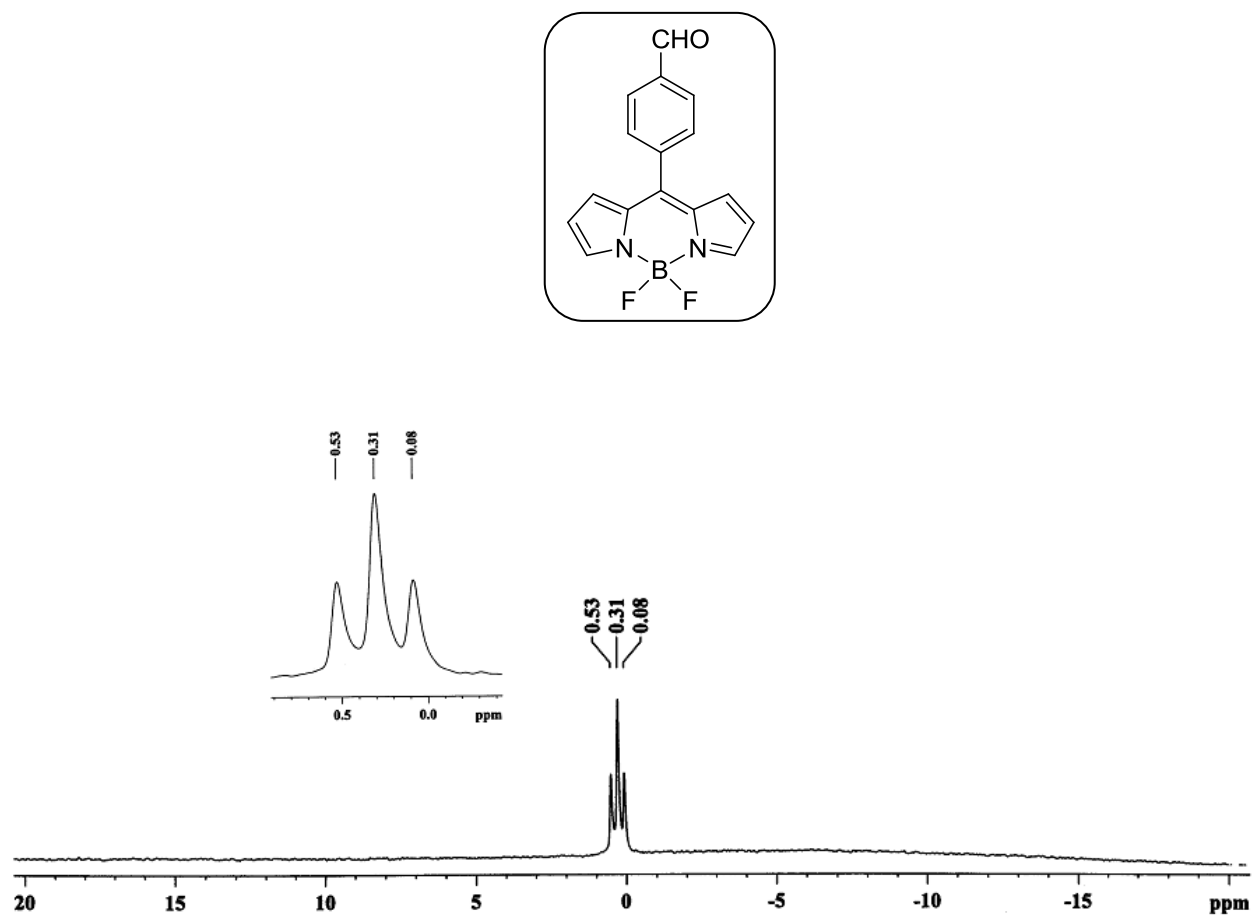
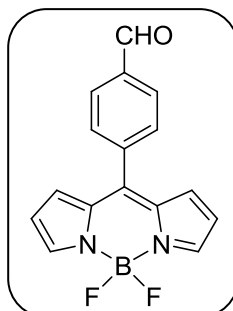


Figure S12: ^{11}B NMR spectrum of compound 3 recorded in CDCl_3 . Inset shows the expansion



Elemental Composition Report

Page 1

Single Mass Analysis (displaying only valid results)

Tolerance = 50.0 PPM / DBE: min = -1.5, max = 200.0

Isotope cluster parameters: Separation = 1.0 Abundance = 1.0%

Monoisotopic Mass, Odd and Even Electron Ions

18 formula(e) evaluated with 1 results within limits (up to 50 closest results for each mass)

Waters(Micromass) : Q-ToF micro(YA-105)

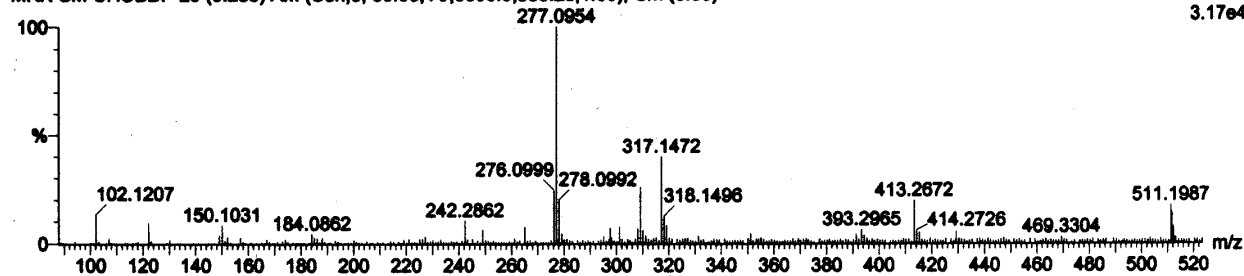
Dept. Of Chemistry - I.I.T.(B)

11-Aug-201220:49:00

C16H11BF2N2O

MRK-SM-CHOBDP 29 (0.288) AM (Cen,5, 80.00, Ar,5000.0,556.28,1.00); Cm (6:36)

TOF MS ES+
3.17e4



Minimum:

Maximum: 200.0 50.0 -1.5 200.0

Mass	Calc. Mass	mDa	PPM	DBE	Score	Formula
277.0954	277.0948	0.5	1.9	12.5	1	C16 H11 B N2 O F

Figure S13: HRMS spectrum of compound 3

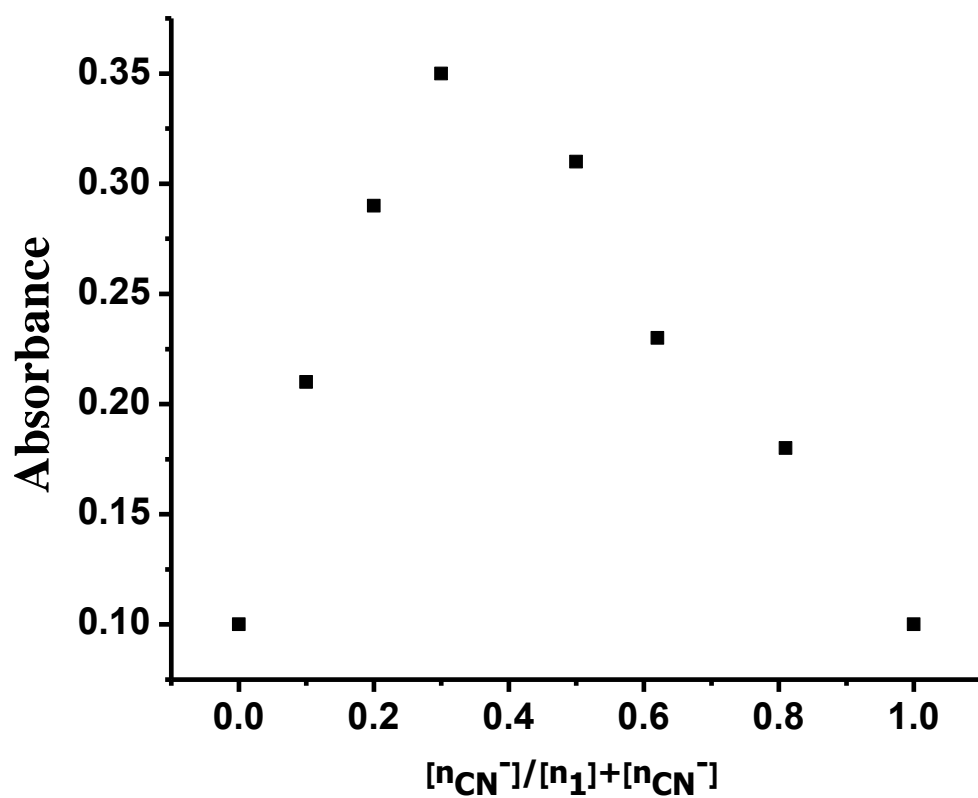


Figure S14: Job's plot for the **1** with CN^- . Where n_{CN^-} , n_1 are mole fractions of CN^- and **1** respectively which forms 1:2 complex.

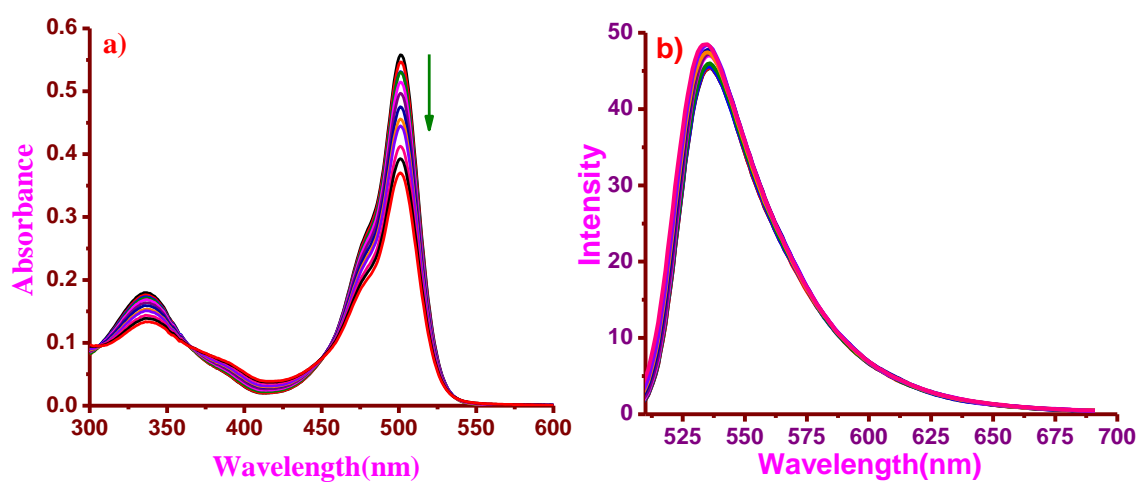


Figure S15: a) Absorption b) Emission spectra of compound **3** (5×10^{-6} M) upon titration with different conc. of CN^- (TBACN) solution (0-50 equiv.) in CH_3CN .

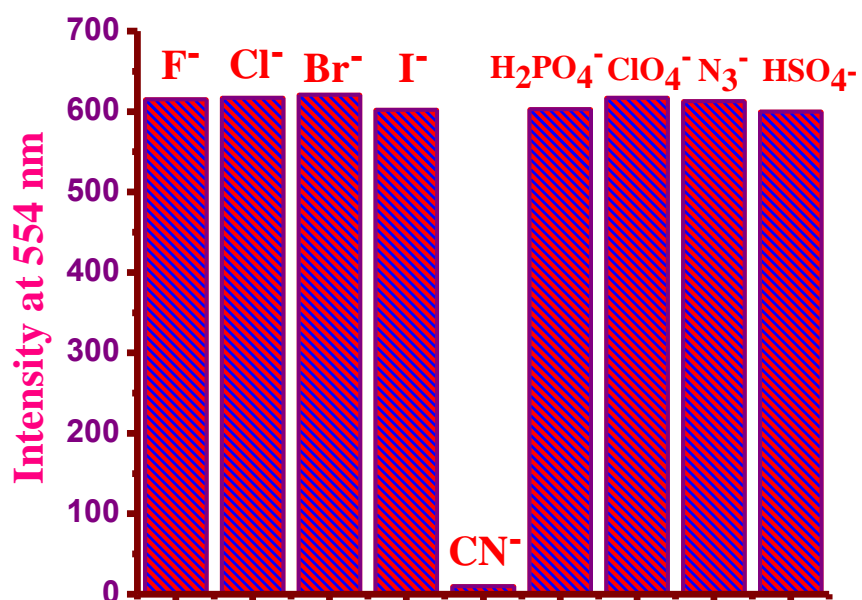


Figure S16: Histogram showing the changes in fluorescence intensity of compound **1** upon addition of various anions (excess of equivalents).

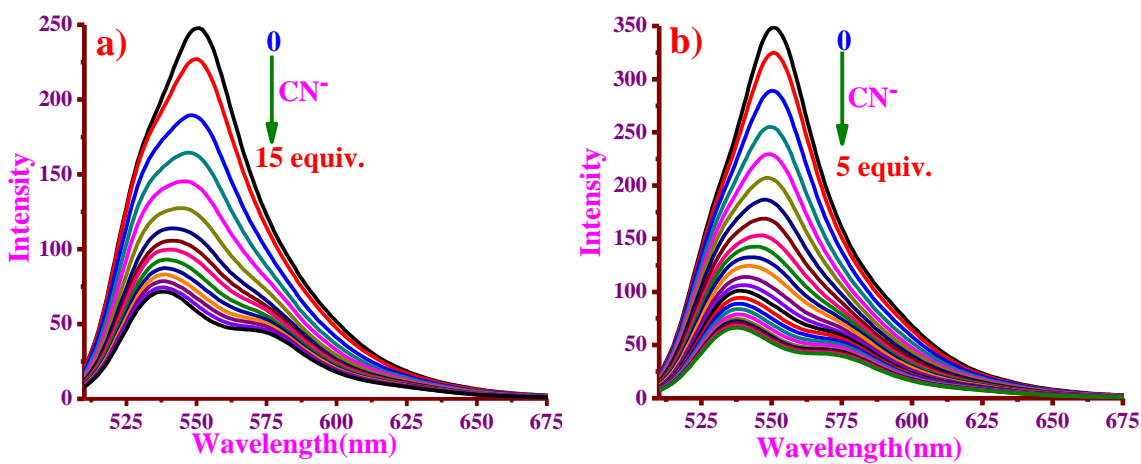


Figure S17: Emission spectra of compound **1** (5x10⁻⁶ M), a) upon titration with different conc. of CN⁻ (TBACN) solution (0-15 equiv.) in (CH₃CN:H₂O; 97:3 v/v), b) upon titration with different conc. of CN⁻ (TBACN) solution (0-5 equiv.) in (CH₃CN:H₂O; 99:1 v/v).