

Supporting Information

Radical Cascades in Electron Transfer Dissociation (ETD) – Implications for Characterizing Peptide Disulfide Regio-Isomers

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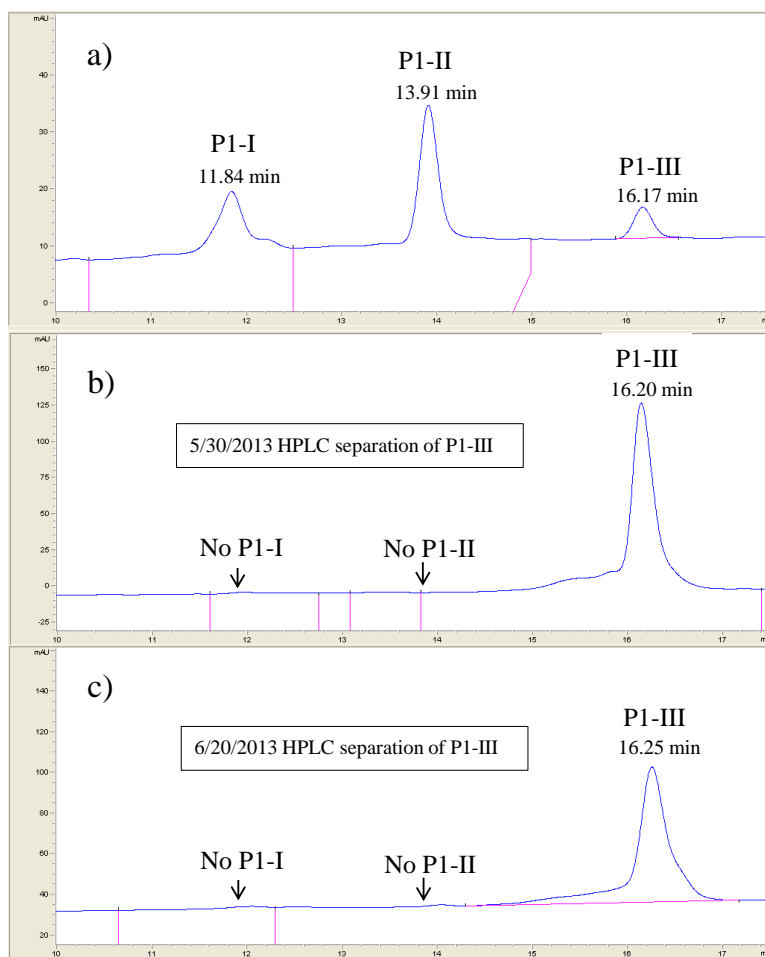
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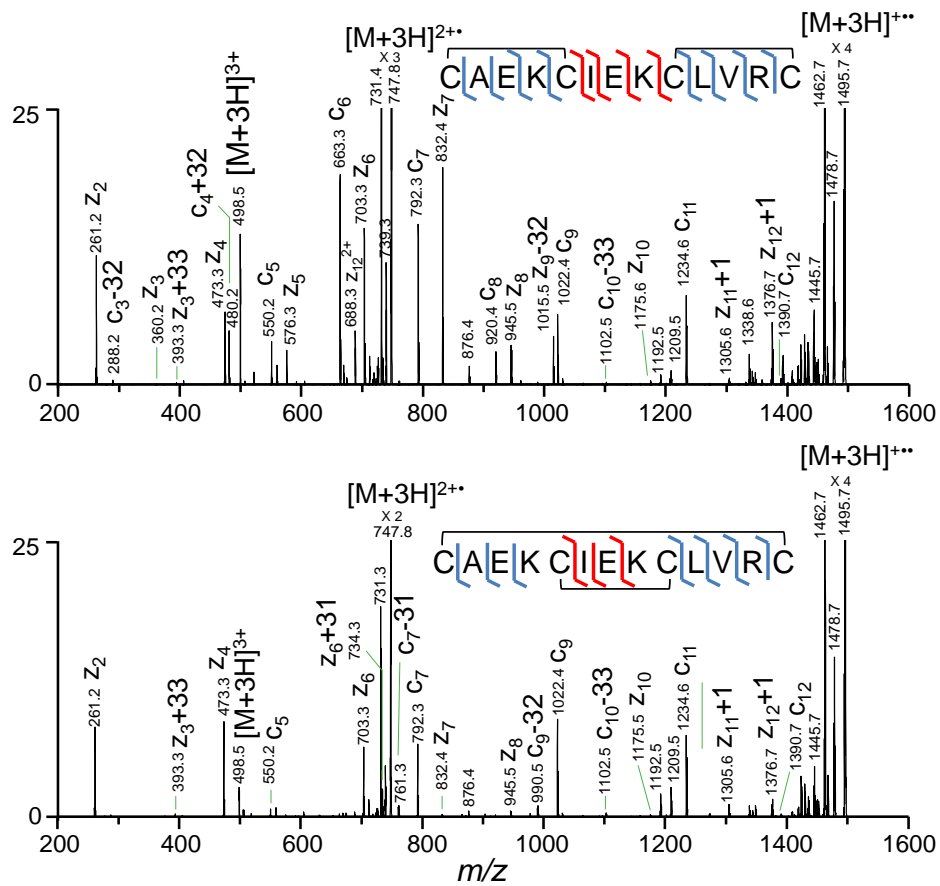
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Stability of the peptide regio-isomers

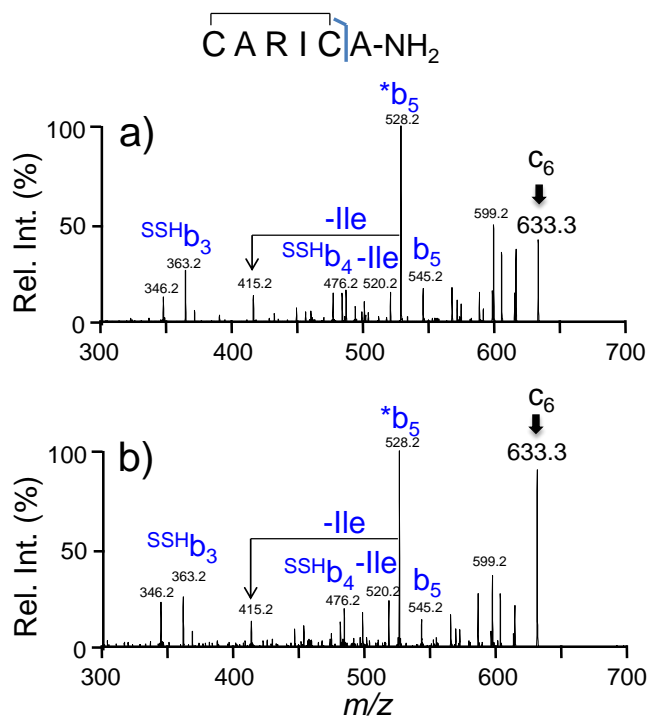
No disulfide bond scrambling was observed for P1 peptides once the two disulfide bonds were generated. The stability of individual regio-isomers was tested by repeated HPLC separations of the purified sample after various periods of storage. Consistent retention time was observed without the detection of other isomers. The data of P1-III were used as an example to illustrate the stability test process. P1-III was first collected after HPLC separation of the oxidation mixture (SI Figure 1a) and vacuum dried. A portion of purified P1-III was redissolved in aqueous solution and immediately subjected to an HPLC run (SI Figure 1b). Note that P1-III was the one with the longest retention time among three regio-isomers. No peak was observed before P1-III, indicating no scrambling to other isomers occurred after vacuum drying. This sample was stored in freezer (-20 °C) for 20 days and redissolved in aqueous solution for another HPLC run (SI Figure 1c) and no scrambling was observed.



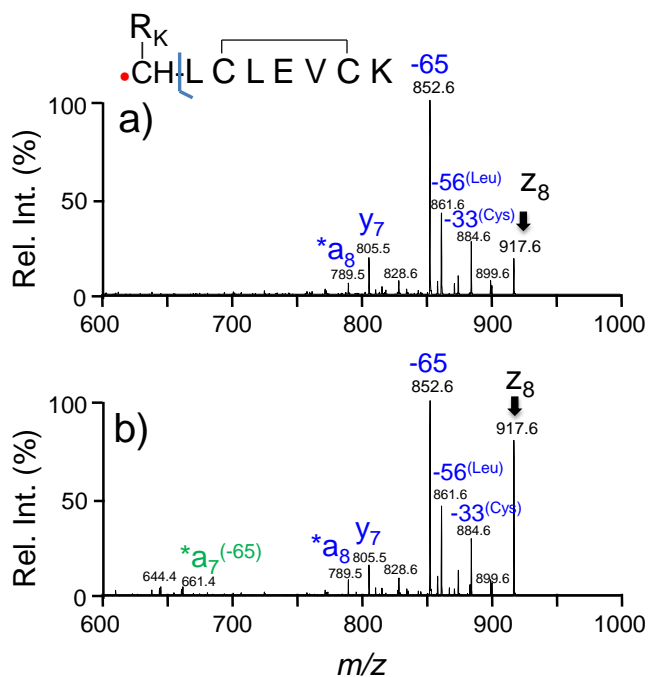
SI Figure 1. HPLC separation chromatograph of: a) mixture of three regio-isomers of P1 after the oxidation reactions, b) P1-III redissolved in 20 μL of water right after being vacuum dried; c) P1-III redissolved in 20 μL of water after storage at $-20\text{ }^{\circ}\text{C}$ freezer for 20 days. All separations were carried out under the same condition: B: 20-35% over 30 min at 0.65 mL/min flow rate (solvent A: 0.1% TFA in water, B: 0.085% TFA in acetonitrile).



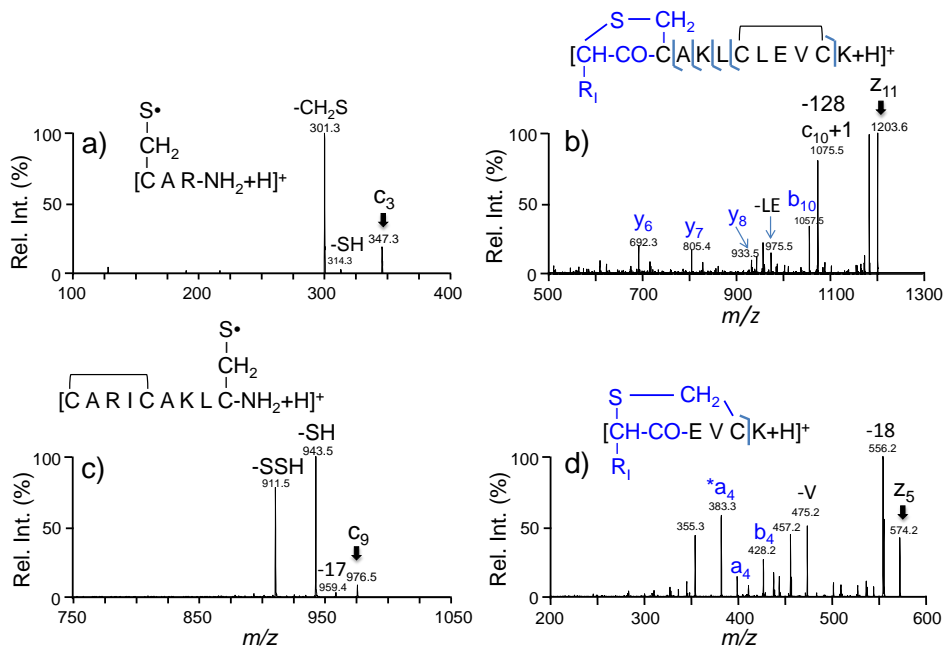
SI Figure 2. ETD of $[M+3H]^{3+}$ of a) P2-I; b) P2-II. ETD reaction time was 100 ms, without supplementary activation.



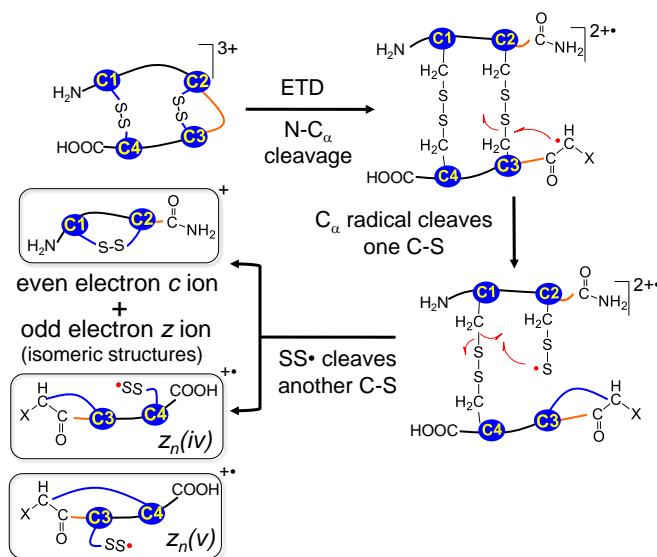
SI Figure 3. MS³ CID of *c*₆ from ETD of [M+3H]³⁺ from a) P1-I; b) P1-III.



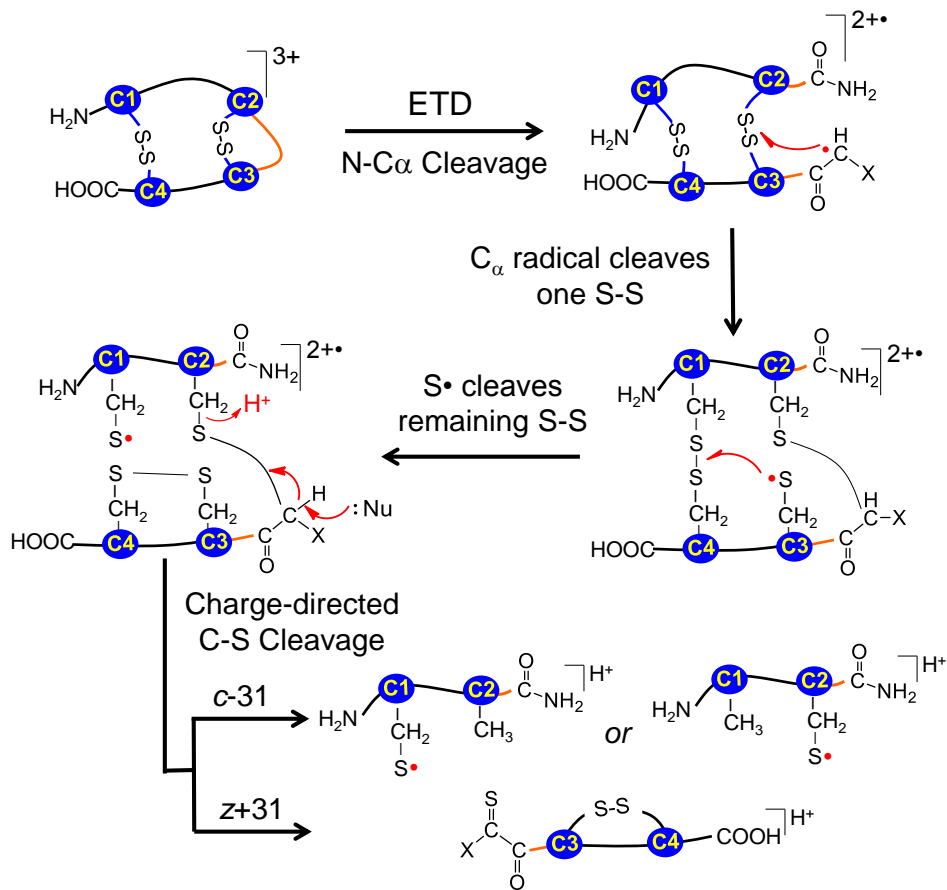
SI Figure 4. MS³ CID of *z*₈ from ETD of [M+3H]³⁺ from a) P1-I; b) P1-III.



SI Figure 5. MS³ CID of a) *c*₃, b) *z*₁₁, c) *c*₉ and d) *z*₅ from ETD of [M+3H]³⁺ from P1-II.



SI Scheme 1. Alternative radical cascades pathway for the formation of cyclic *c* and *z*[•] ion between C2 and C3 in ETD of P1-II.



SI Scheme 2. Proposed pathway for the formation of *c*-31/*z*+31 ion between C2 and C3 in ETD of P1-II.

SI Table 1. Theoretical prediction of the backbone fragments (*c* and *z* ions, 1+) resulting from ETD of fully oxidized P1 peptides. The prediction is based on homolytic cleavage of disulfide bond and carried out with ProSight PTM developed by the Kelleher Group at Northwestern University.

Ion	Theoretical	Ion	Theoretical
n	<i>m/z</i>		<i>m/z</i>
c ₁	120.0	z ₁	131.1
c ₂	191.1	z ₂	233.1
c ₃	347.2	z ₃	332.2
c ₄	460.3	z ₄	461.2
c ₅	562.3	z ₅	574.3
c ₆	633.3	z ₆	676.3
c ₇	761.4	z ₇	789.4
c ₈	874.5	z ₈	917.5
c ₉	976.5	z ₉	988.5
c ₁₀	1089.6	z ₁₀	1090.5
c ₁₁	1218.6	z ₁₁	1203.6
c ₁₂	1317.7	z ₁₂	1359.7
c ₁₃	1419.7	z ₁₃	1430.8
c ₁₄	1547.8	z ₁₄	1532.8

SI Table 2. Assigned peak list for ETD of $[M+3H]^{3+}$ (*m/z* 517.1) of P1-I. Relative ion intensities (%) are normalized to the most abundant product ion peak within each spectrum.

ExpPeak	Charge	Relative	Assignment
<i>m/z</i>	State	Intensity	
331.3	1+	0.3	z ₃ -1
347.3	1+	0.6	c ₃
461.3	1+	0.2	z ₄
494.3	1+	1	z ₄ +33
517.1	3+	82	$[M+3H]^{3+}$
562.3	1+	2.2	c ₅
574.4	1+	2.2	z ₅
633.4	1+	9	c ₆

676.5	1+	0.6	z ₆
697.0	2+	6.5	
710.5	2+	4	c ₁₃
732.5	2+	2	
759.1	2+	100	[M+3H] ²⁺ -SH
761.5	1+	9.5	c ₇
767.0	2+	9.5	[M+3H] ²⁺ -NH ₃
775.5	2+	64	[M+3H] ²⁺
789.5	1+	8	z ₇
805.5	1+	0.7	y ₇
867.6	1+	0.8	
874.6	1+	0.9	c ₈
884.6	1+	1.2	
917.6	1+	8.2	z ₈
933.5	1+	0.4	y ₈
976.6	1+	2.2	c ₉
988.6	1+	2.8	z ₉
1056.7	1+	0.9	c ₁₀₋₃₃
1089.6	1+	0.2	c ₁₀
1090.6	1+	0.3	z ₁₀
1203.6	1+	0.7	z ₁₁
1219.6	1+	0.4	c ₁₁₊₁
1359.7	1+	0.2	z ₁₂
1393.7	1+	1.2	
1420.7	1+	0.4	c ₁₃₊₁
1431.7	1+	0.3	z ₁₃₊₁
1500.9	1+	1.2	
1517.8	1+	12	[M+3H] ⁺ -SH
1533.8	1+	3.2	[M+3H] ⁺ -NH ₃
1550.8	1+	5.5	[M+3H] ⁺

SI Table 3. Assigned peak list for ETD of [M+3H]³⁺ (*m/z* 517.1) of P1-II. Relative ion intensities (%) are normalized to the most abundant product ion peak within each spectrum.

ExpPeak <i>m/z</i>	Charge State	Relative Intensity	Assignment
331.3	1+	2.5	z ₃₋₁
347.3	1+	0.5	c ₃

461.3	1+	0.3	Z ₄
494.3	1+	2.2	Z ₄ +33
517.1	3+	50	[M+3H] ³⁺
556.4	1+	1.5	
562.3	1+	0.8	C ₅
574.4	1+	1.6	Z ₅
602.4	1+	4.5	C ₆ -31
633.4	1+	7	C ₆
676.5	1+	0.1	Z ₆
697.0	2+	4.5	
710.5	2+	2.2	C ₁₃
759.1	2+	100	[M+3H] ²⁺ -SH
761.5	1+	7.5	C ₇
767.0	2+	13	[M+3H] ²⁺ -NH ₃
775.5	2+	70	[M+3H] ²⁺
789.5	1+	4.1	Z ₇
805.5	1+	0.4	Y ₇
821.4	1+	0.7	
867.6	1+	1.8	
874.6	1+	0.1	C ₈
899.5	1+	1.9	
917.6	1+	6.5	Z ₈
948.5	1+	5.8	Z ₈ +31
976.6	1+	1.8	C ₉
988.6	1+	1	Z ₉
1056.7	1+	2.8	C ₁₀ -33
1089.6	1+	0.3	C ₁₀
1090.6	1+	0.4	Z ₁₀
1203.6	1+	1	Z ₁₁
1219.6	1+	5	C ₁₁ +1
1349.7	1+	2.4	
1359.7	1+	0.4	Z ₁₂
1393.7	1+	2.5	
1419.7	1+	0.8	C ₁₃ +1
1431.7	1+	0.2	Z ₁₃ +1
1490.8	1+	2	
1500.9	1+	5	
1517.8	1+	36	[M+3H] ⁺ -SH
1533.8	1+	11	[M+3H] ⁺ -NH ₃
1550.8	1+	21	[M+3H] ⁺

SI Table 4. Assigned peak list for ETD of $[M+3H]^{3+}$ (m/z 517.1) of P1-III. Relative ion intensities (%) are normalized to the most abundant product ion peak within each spectrum.

ExpPeak m/z	Charge State	Relative Intensity	Assignment
301.3	1+	0.7	
314.4	1+	0.7	c ₃ -33
331.3	1+	2.3	z ₃ -1
347.3	1+	2.4	c ₃
461.3	1+	0.7	z ₄
447.3	1+	1.2	
494.3	1+	0.6	z ₄ +33
517.1	3+	53	$[M+3H]^{3+}$
562.3	1+	0.7	c ₅
574.4	1+	1.2	z ₅
602.4	1+	3.5	c ₆ -31
633.4	1+	5	c ₆
644.4	1+	0.5	
664.5	2+	1	
676.5	1+	0.2	z ₆
697.0	2+	1.4	
710.5	2+	2	c ₁₃
759.1	2+	100	$[M+3H]^{2+}$ -SH
761.5	1+	6	c ₇
767.0	2+	26	$[M+3H]^{2+}$ -NH ₃
775.5	2+	50	$[M+3H]^{2+}$
789.5	1+	5.5	z ₇
821.4	1+	1	
867.6	1+	1	
874.6	1+	0.3	c ₈
917.6	1+	5	z ₈
948.5	1+	3.4	z ₈ +31
976.6	1+	1.4	c ₉
988.6	1+	0.7	z ₉
1045.6	1+	1.6	
1056.7	1+	0.7	c ₁₀ -33
1089.6	1+	0.5	c ₁₀
1090.6	1+	0.5	z ₁₀
1203.6	1+	4.5	z ₁₁

1219.6	1+	3.5	c ₁₁ +1
1236.6	1+	0.5	z ₁₁ +33
1349.7	1+	1.1	
1359.7	1+	0.3	z ₁₂
1419.7	1+	0.4	c ₁₃ +1
1431.7	1+	0.2	z ₁₃ +1
1500.9	1+	5.5	
1517.8	1+	34	[M+3H] ⁺⁺⁺ -SH
1533.8	1+	12	[M+3H] ⁺⁺⁺ -NH ₃
1550.8	1+	10	[M+3H] ⁺⁺⁺
