

Online Supporting Information

Determination of free acidic and alkaline residues of protein via moving reaction boundary titration in microdevice electrophoresis

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Table S1 Parameters of seven model proteins.

Protein	MW(KDa)	Isoelectric point	N_{ACR} ^{a)}	N_{ALR} ^{a)}
Mb	17.5	7.07	23	31
Hb	67	7.23	64	84
BSA	66	7.8	149	124
Cyt C	13.8	4.47	17	23
Trypsin	23.3	10.5	30	26
Pepsin	35	8.1	67	27
CT	11.7	4.9	17	16

^{a)}The value of N_{ACR} and N_{ALR} are calculated through the corresponding amino acid sequences of seven proteins according to the protein database of NCBI, the detailed calculation process is shown in the section of Data Analysis.

Chemicals

Two standard reagents of potassium hydrogen phthalate and sodium tetraborate were obtained from Shanghai Chemical reagent Co. (Shanghai, China). Acrylamide and Bis-acrylamide were bought from Fluka (Switzerland). N, N, N', N'-tetramethyl- ethylenediamine (TEMED) was from Sigma (USA). The other chemicals used herein were ammonium persulfate (analytical reagent grade, AR), sodium hydroxide (AR), hydrochloric acid (AR) and potassium chloride (AR) and phenolphthalein (chemical reagent grade, CR) as well as bromophenol blue (CR) were purchased from Shanghai Chemical reagent Co. (Shanghai, China).

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Data Analysis

Different concentrations (0, 1, 2, 3 and 4 mg/mL) of model proteins were chosen for experiments of MRB titration. A blank control (no any protein) was used for obtaining the values of v_{OH^-} and v_{H^+} in PAG, correcting the values of v_{OH^-} and v_{H^+} used in Eq. (5)-(6) and Eq. (7)-(8) when detecting free acidic and alkaline groups of immobilized protein, respectively. A same experiment of MRB titration was repeated three times, and the distance of the boundary motion was referred with a ruler.

After a base titration (Scheme 1A), the boundary velocity could be computed with Eq. (1) and the photographs, and the acidic groups and residues of protein could be calculated with Eq. (6) and Eq. (7), respectively. Similarly, the boundary velocity in the acid titration (Scheme 1B) could be computed with Eq. (1) and the relevant photographs, and the alkaline groups and residues of protein could be calculated with Eq. (9) and Eq. (10), respectively.

The titrating sites of acidic residues in protein concerned Asp (e), Glu (d), Tyr (y) and Cys (c). And the titrating sites of alkaline residues included His (h), Lys (k), 2×Arg (r), Trp (w). The actual values of N_{ACR} and N_{ALR} were calculated from the National Center of Biotechnology Information (NCBI) protein database which was publicly available for community use.

1. The amino acids sequence of Mb from equine skeletal muscle is listed as follow:

**mglsdgewqq vlnvwgkvea diaghgqevl irlftghpet lekfdkfkhl kteaemkase
dlkxhgtvvl talggilkkk ghheaekpl aqshatkhi pikylefisd aiihvlhskh
pgdfgadaqq amtkalelfr ndiaakykel gfqg**

The amino acids sequence can be referred to the URL of NCBI:

(http://www.ncbi.nlm.nih.gov/protein/NP_001157488.1)

The N_{ACR} of Mb is 23 ($8d + 13e + 2y + 1c$).

The N_{ALR} of Mb is 31 ($11h + 16k + 2r + 2w$).

2. The amino acids sequence of one chain of Hb from human is listed as follow:

**mvhltpeeks avtalwgkvn vdevggealg rllvypwtq rffesfgdls tpdavmgnpk
vkahgkklvg afsdglahld nlkgfatls elhcdklhvd penfr**

The amino acids sequence can be referred to the URL of NCBI

(<http://www.ncbi.nlm.nih.gov/protein/ABG47031.1>)

The N_{ACR} of Hb (4Chains) is 64: $4(7d + 7e + 1y + 1c)$.

The N_{ALR} of Hb (4Chains) is 84: $4(5h + 8k + 2 \times 3r + 2w)$.

3. The amino acids sequence of BSA from bovine is listed as follow:

**mkwvtfisll llfsaysrg vfrdthkse iahrfdlge ehfkgvlvia fsqylqqcpl
dehkvlnel tefaktcvad eshagceksl htlfgdelck vaslretygd madccekqep
erneclshk ddspdlpklk pdpntlcdef kadekkwgk ylyeiarrhp yfyapellyy
ankyngvfqe ccqaedkgac lpkietmre kvltssarqr lrcasiqkfg eralkawsva
rlsqkfpkae fvevtklvtd ltkvhkech gdllcaddr adlakyicdn qdtissklke
cdkplleks hciaevekda ipenlplta dfaedkdvck nyqakdafl gsflyeysrr
hpeyavsvll rlakeyeatl eccakddph acystvfdkl khlvdeppnl ikqncdqfek
lgeygfqnal ivrytrkvpq vstptlvevs rslgkvgtre ctkpesermp ctedylslil
nrlevlhektpvsekvtkcc teslvnrrpc fsaltpdety vpkafdeklf tfhadictlp**

dteqikkqt alvellkhkp kateeqktv menfvafvdk ccaaddkeac favegpklvv stqtala

The amino acids sequence can be referred to the URL of NCBI

(<http://www.ncbi.nlm.nih.gov/protein/CAA76847.1>)

The N_{ACR} of BSA is 149 (39d + 58e + 19y + 33c).

The N_{ALR} of BSA is 124 (15h + 57k + 2 × 25r + 2w).

4. The amino acids sequence of Cyt C from bovine is listed as follow:

**mgdvekgkki fvqkcaqcht vekggkhktg pnhlgfgrk tgqapgsyt danknkgitw
geetlmeyle npkkyipgtk mifagikkkk eredliaylk katne**

The amino acids sequence can be referred to the URL of NCBI

(http://www.ncbi.nlm.nih.gov/protein/NP_001039526.1)

The N_{ACR} of Cyt C is 17 (3d + 9e + 3y + 2c).

The N_{ALR} of Cyt C is 23(2h + 18k + 2r + w).

5. The amino acids sequence of trypsin from porcine is listed as follow:

**ivggytcaan sipyqvslns gshfcggsli nsqwvvsaaah cyksriqvrl gehnidvleg
neqfinaaki ithpnfngnt ldndimlikl sspatlxsrv atvslprsc aagteclisg
wgnstkssgss ypsllqlcka pvlsscks sypgqitgnm icvgfleggk dscqgdsggp
vvengqlqgi vswgygcaqk nkpgvytkvc nyvnwiqti aan**

The amino acids sequence can be referred to the URL of NCBI

(http://www.ncbi.nlm.nih.gov/protein/1C9P_A)

The N_{ACR} of trypsin is 30 (6d + 5e + 8y + 11c).

The N_{ALR} of trypsin is 26 (4h + 10k + 2 × 4r + 4w).

6. The amino acids sequence of pepsin from porcine gastric mucosa is listed as follow:

**lvkvplvrkk slrqnlidg klkdfkthk hnpaskyfe aaligdepl enyldteyfg
tigitpaqd ftvifdtgss nlwvpsvycs slacsdhnqf npddsstfea tsqelsityg
tgsmtgilgy dtvqvggisd tnqifglset epgsflyap fdgilglayp sisasgatpv
fdnlwdqglv sqdlfsvyls snddsgsvvl lggidssyyt gslnwvpsv egywqitlds
itmdgetiac sggcqavldt gtslltgpts aianiqsdig asensdgemv iscssidslp
divftidgvq yplspayil qdddsetsgf egmdvptssg elwilgdvfi ryytvfdra
nnkvglapva**

The amino acids sequence can be referred to the URL of NCBI

(http://www.ncbi.nlm.nih.gov/protein/3PSG_A)

The N_{ACR} of pepsin is 67 (32d + 13e + 16y + 6c).

The N_{ALR} of pepsin is 27 (2h + 13k + 2 × 4r + 4w).

7. The amino acids sequence of from CT from bovine is listed as follow:

**ivngeeavpg swpwqvsldg ktghfcggs linenwvta ahcgvttsv vagefdqgs
sseqiklki akvfknskyn sltinnditl lklaasfs qtsavclps asddfaagtt cvttgwglttr y**

The amino acids sequence can be referred to the URL of NCBI

(http://www.ncbi.nlm.nih.gov/protein/1GHB_F)

The N_{ACR} of CT is 17 (6d + 5e + 2y + 4c).

The N_{ALR} of CT is 16 (2h + 8k + 2 × r + 4w).

Optimization of Conditions

To achieve reproducible experiments with the developed method, we at first optimized the electric field strength (E), NaOH and HCl concentrations, and ionic strength in the whole system. Fig. S1A showed the influence of electric field strength on the boundary velocity in the MRB titration system. As for applied voltage, at a relative low field strength (250 V/m), the boundary moved slowly. When high field strength (2500 V/m) was used, the boundary migrated very quickly, but a poor linearity was present between the boundary displacement distance (D) and running time (t). In addition, at high field strength, the capillary was felt very hot. Considering fair boundary displacement but not much Joule heating, 1250 V/m was finally chosen as the optimized field strength (Fig. S1A).

It was further observed in Fig. S1B that the boundary indicated by the acid-base indicator (phenolphthalein, or bromophenol blue) moved slowly and was not clear if a low concentration NaOH or HCl (less than 15 mM) was used; while the boundary migrated fast and became clear if high concentration base or acid (20 mM) was applied in the given system. Whereas, there was a poor linearity between D and t , besides much Joule heating if higher concentration NaOH or HCl was used. Finally, 20 mM NaOH and HCl were chosen for base and acid titration the experiments, respectively (Fig. S1B).

Fig. S1C exhibited the impact of background electrolyte of potassium chloride on the movement of MRB. It was demonstrated in Panel C that high concentration background electrolyte (KCl) had no obvious effect on the boundary velocity, but could result in much more Joule heating. The ionic strength of the anodic solution, cathodic solution and PAG was selected at 100 mM to keep similar conductance through the whole titration system.

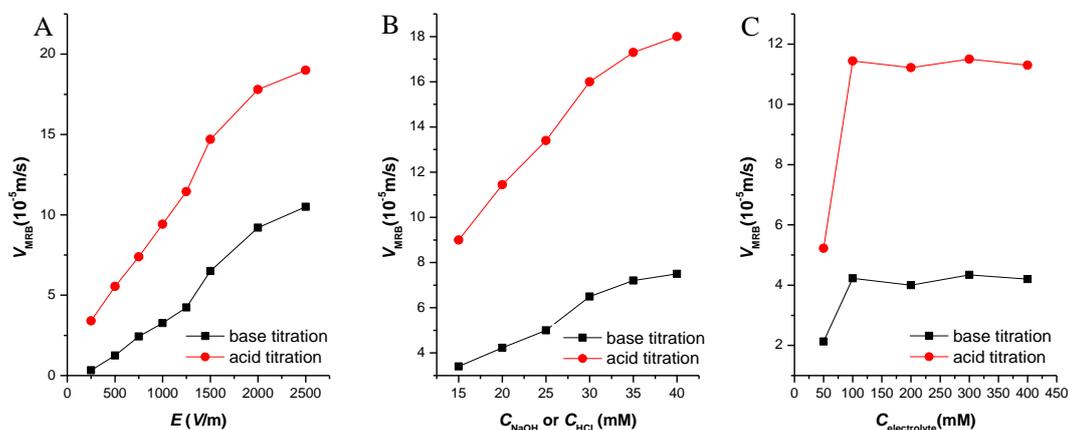


Fig. S1 Influence of electric field strength (E) (A), NaOH and HCl concentration (B) and ionic strength of background electrolyte (C) on boundary velocity (V_{MRB}). Conditions in Panel A: 20 mM NaOH or 20 mM HCl, 2.0 mg/mL BSA, 0.02 μ L 1.0% (w/v) phenolphthalein or 0.1% (w/v) bromophenol blue, 500-2500 V/m, PAG (15% T and 4.0% C), 100 mM background electrolyte, 500 μ m i.d. and 4 cm length capillary, and air-conditioned room (22-25 $^{\circ}$ C). Conditions in Panel B: 15-40 mM NaOH or HCl. the other conditions are the same as those in Panel A. Conditions in Panel C: 50-400 mM background electrolyte. The other conditions are the same as those in Panel A.

Table S2 Linearity curves between the boundary movement (D) and the running time (t) in the MRBT system

Protein	MRB base titration		MRB acid titration	
	Linearity equation ^a	R ^b	Linearity equation ^c	R
Mb	$D=0.0048t-0.0603$	0.9989	$D=0.0103t+0.08$	0.9926
Hb	$D=0.0039t-0.0103$	0.9985	$D=0.0101t+0.17$	0.996
BSA	$D=0.0044t+0.0414$	0.9922	$D=0.0107t+0.095$	0.994
Cyt C	$D=0.0033t+0.1$	0.9999	$D=0.0104t+0.07$	0.9955
Trypsin	$D=0.0044t-0.0029$	0.9901	$D=0.0113t+0.05$	0.9948
Pepsin	$D=0.0036t+0.0633$	0.9947	$D=0.0113t+0.09$	0.9987
CT	$D=0.0034t-0.02$	0.9965	$D=0.0115t+0.05$	0.9921

^a) The linearity equations correspond with those in Fig. 1A.

^b) R is the correlation coefficient.

^c) The linearity equations correspond with those in Fig. 2A.