

Supporting information for

**A turn-on fluorescent probe for hypochlorous acid based on the oxidation of diphenyl  
telluride**

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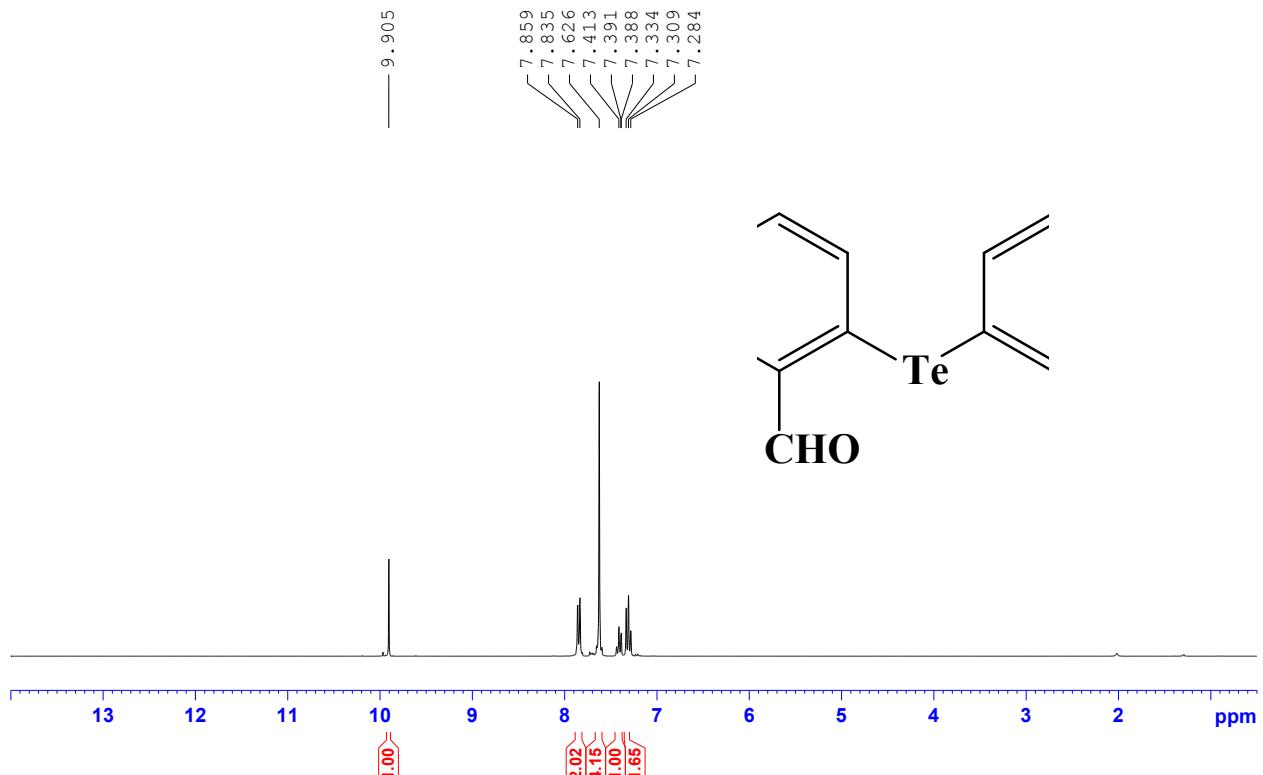
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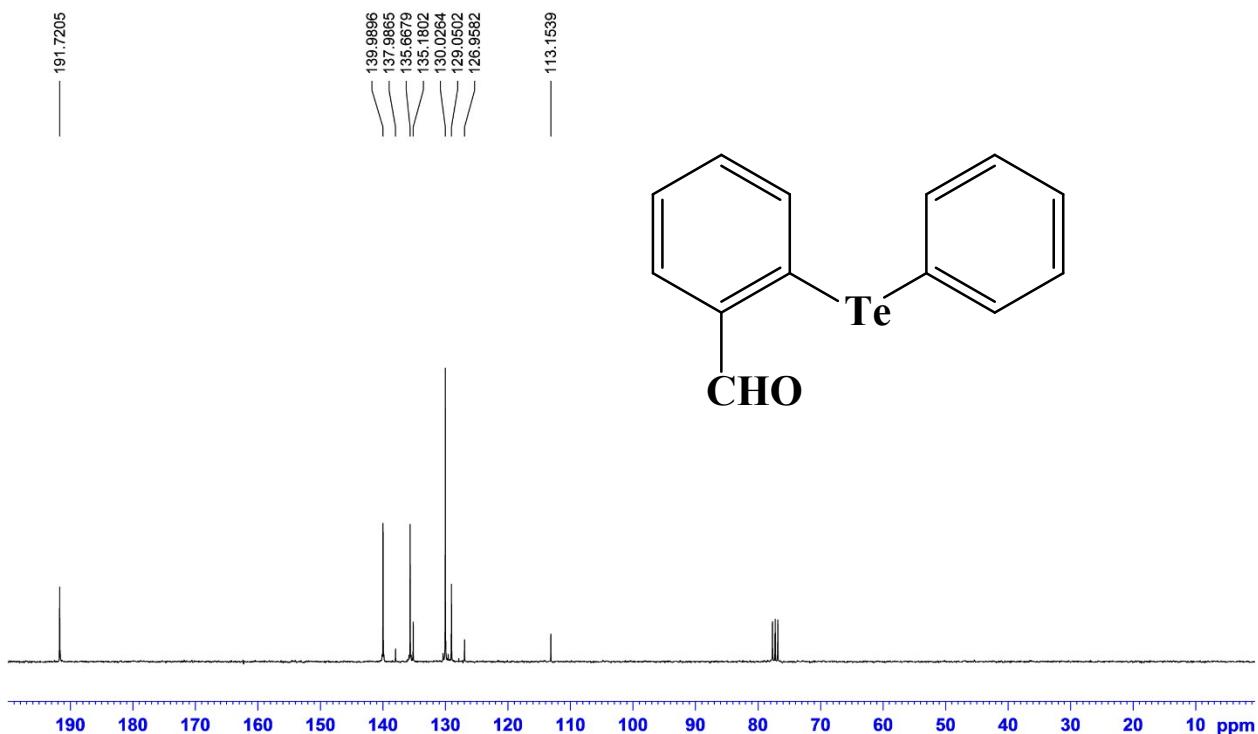
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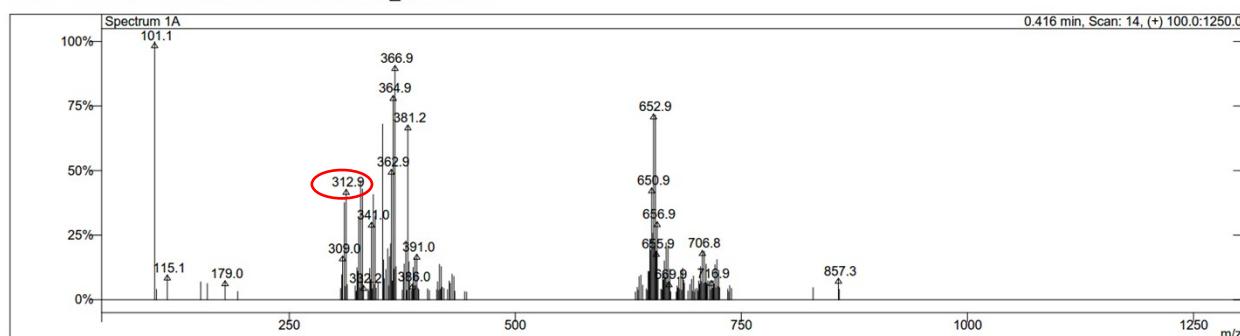


**Figure S1.**  $^1\text{H}$  NMR (300 MHz,  $\text{CDCl}_3$ ) spectrum of 2-(phenyltellanyl) benzaldehyde

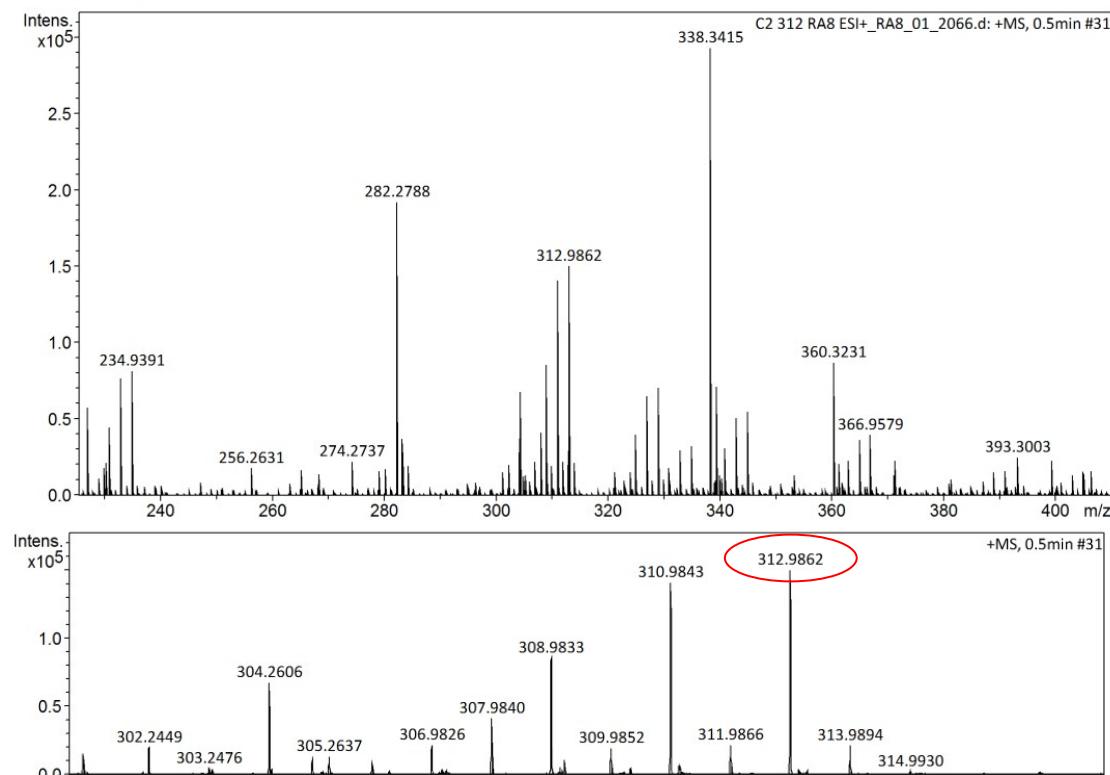


**Figure S2.**  $^{13}\text{C}$  NMR (300 MHz,  $\text{CDCl}_3$ ) spectrum of 2-(phenyltellanyl) benzaldehyde

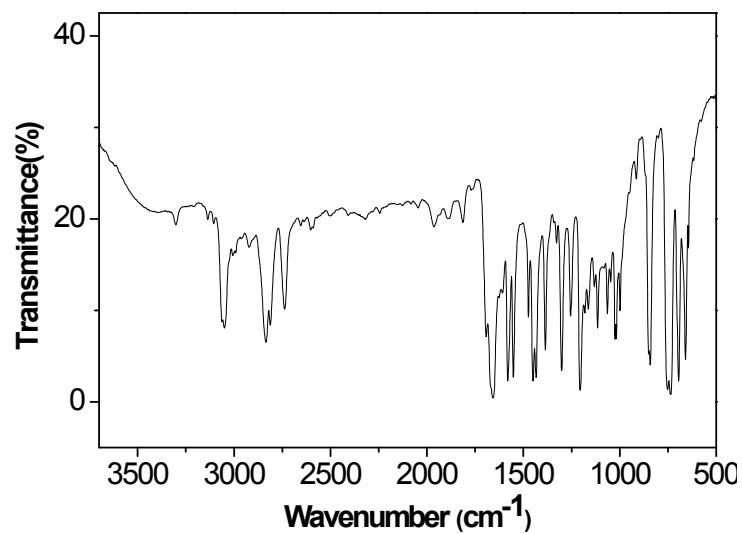
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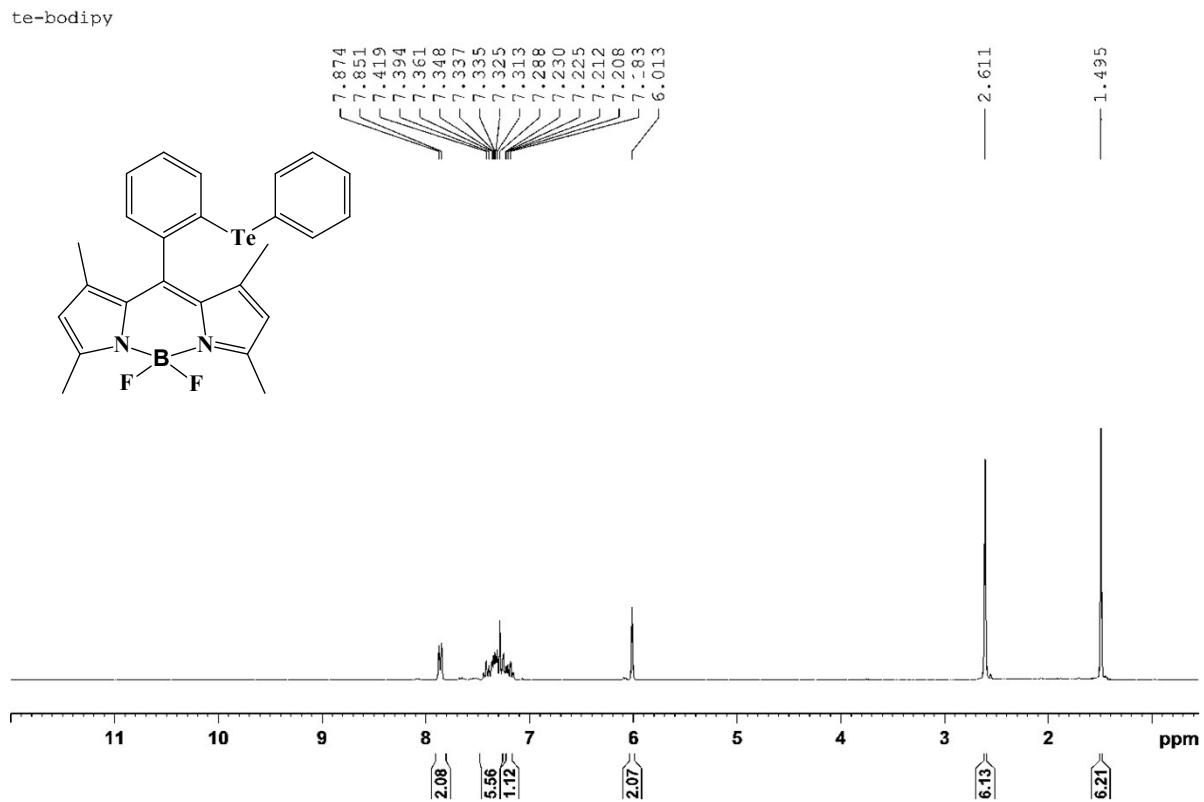
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CAS No: None Acquired Range: 100.0 - 1250.0 m/z



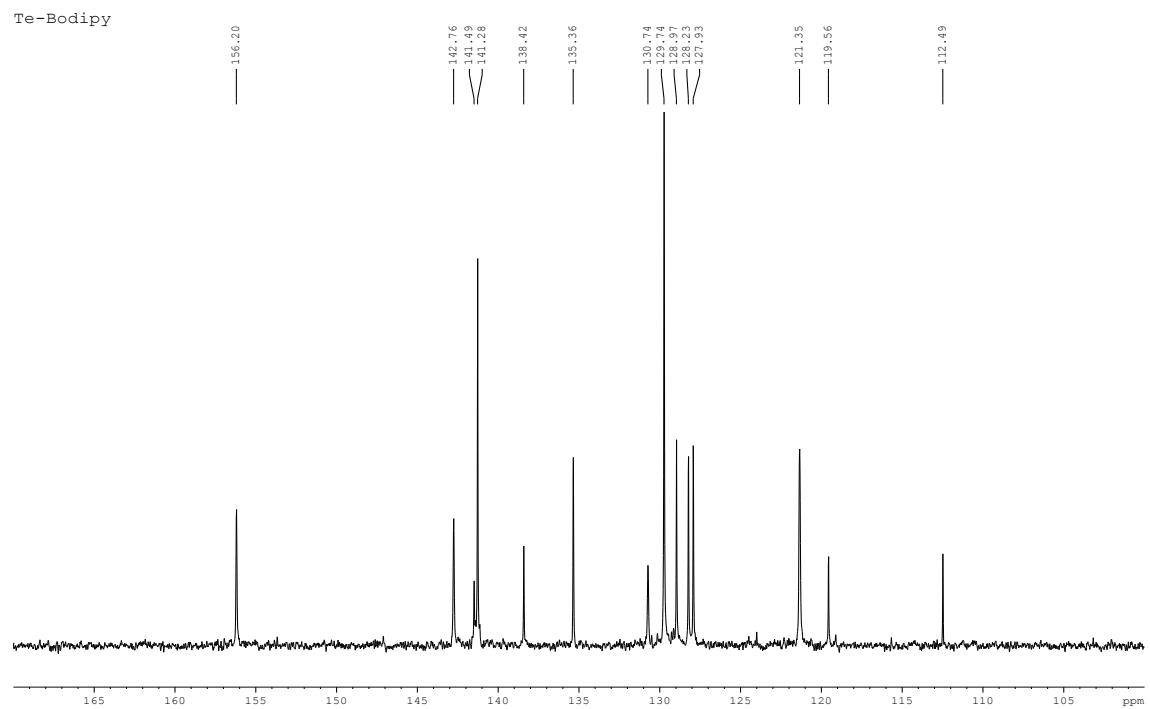
**Figure S3.** Mass spectrum of 2-(phenyltellanyl)benzaldehyde



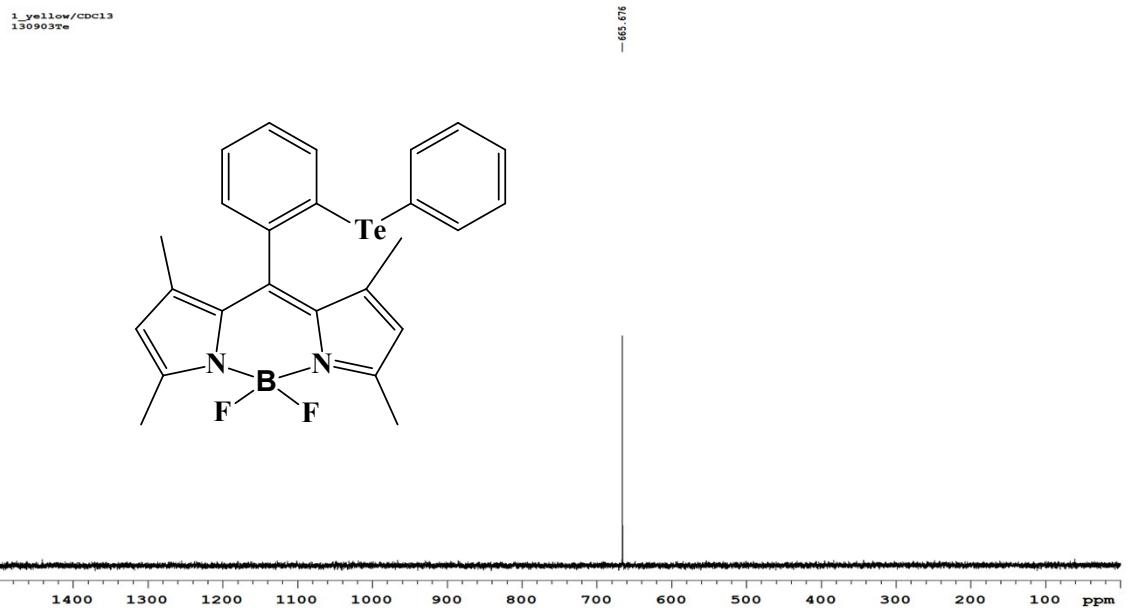
**Figure S4.** IR spectrum of 2-(phenyltellanyl) benzaldehyde



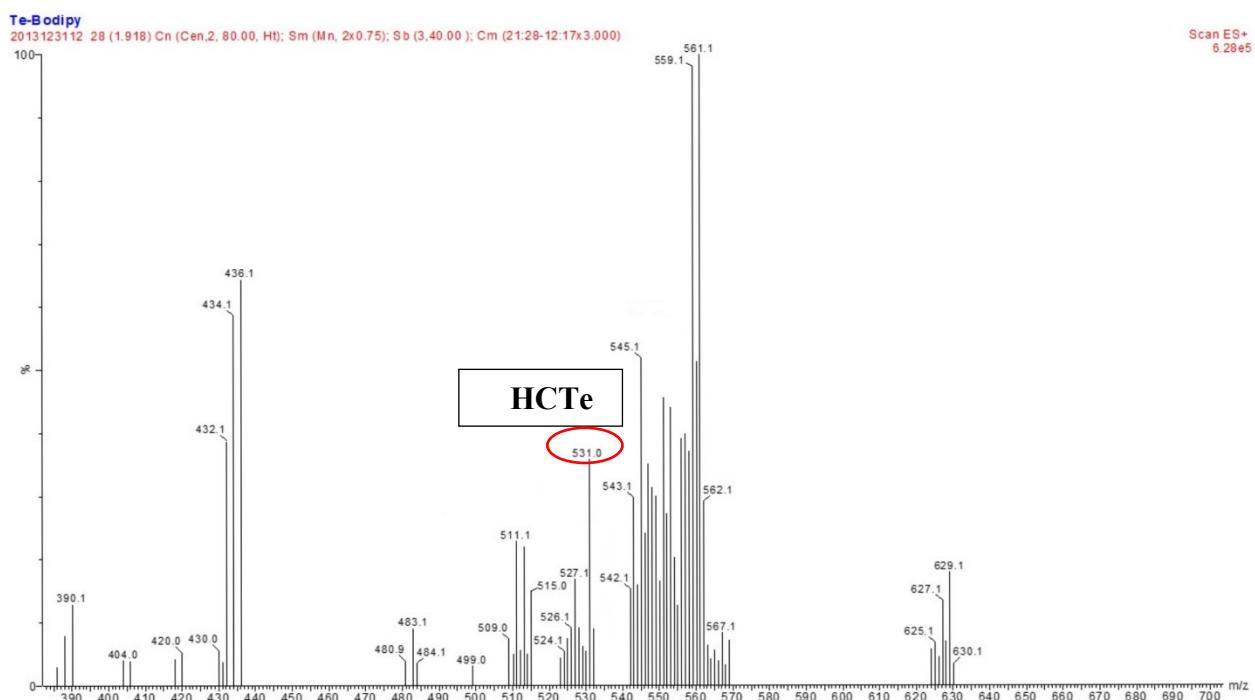
**Figure S5.** <sup>1</sup>H NMR (300 MHz, CDCl<sub>3</sub>) spectrum of HCTe



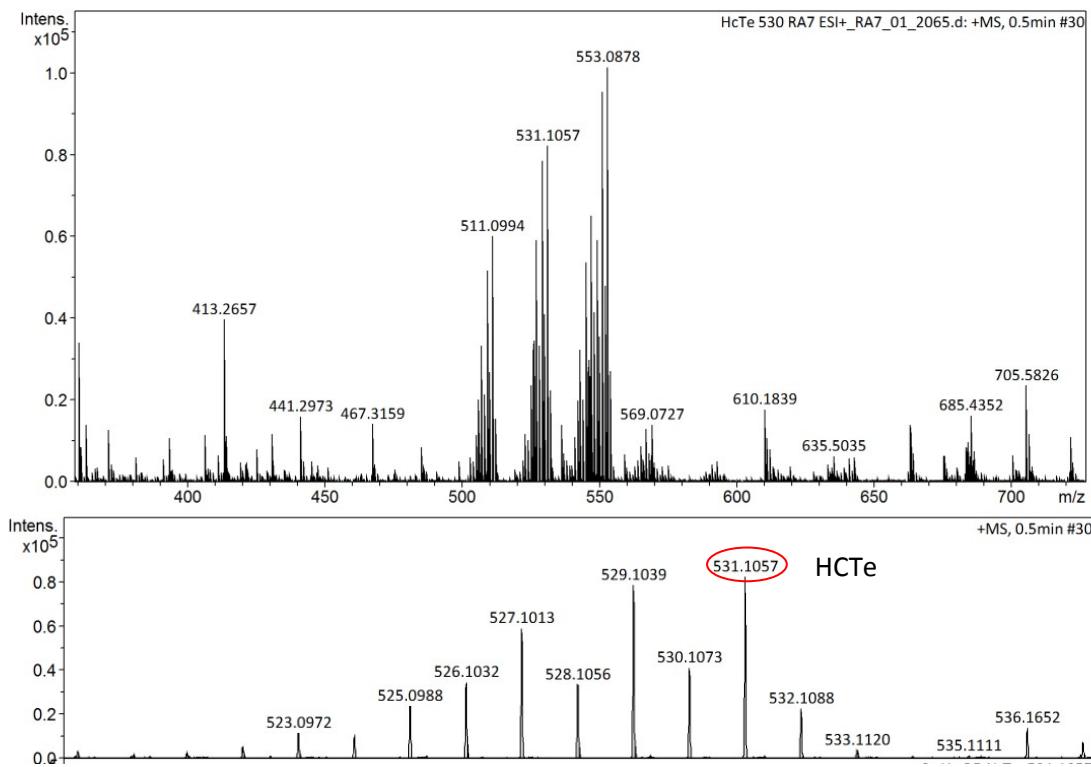
**Figure S6.** <sup>13</sup>C NMR (300 MHz, CDCl<sub>3</sub>) spectrum of HCTe



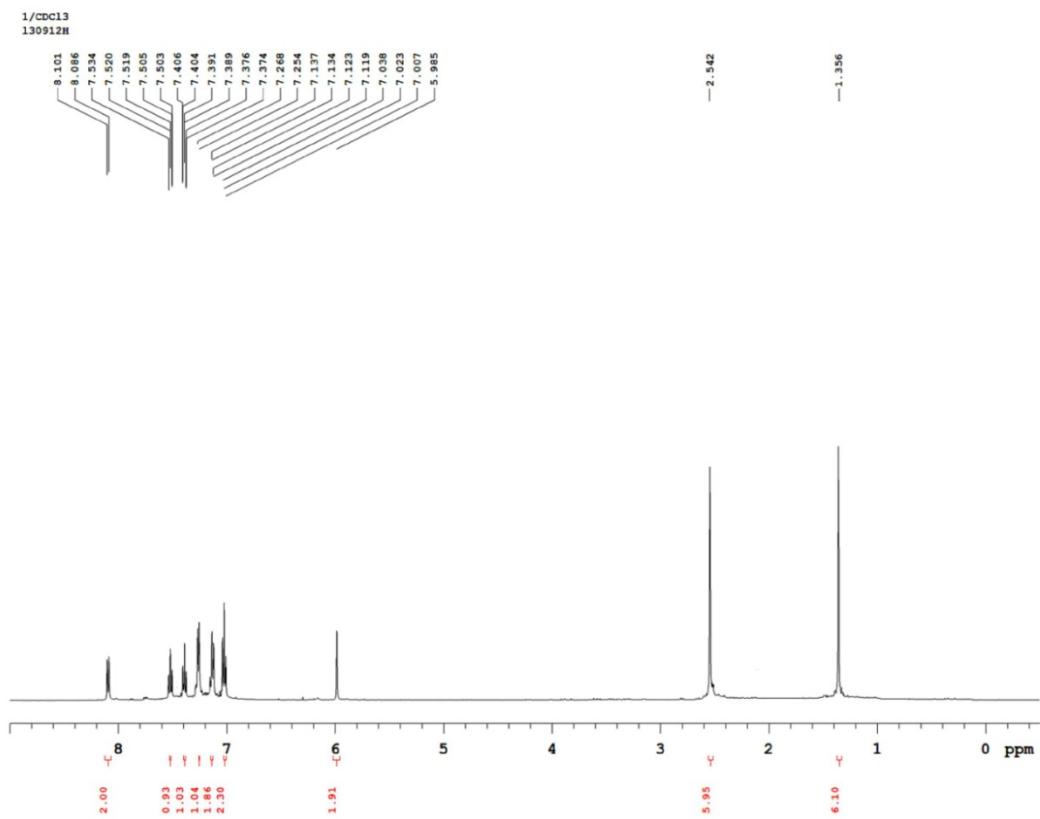
**Figure S7.** <sup>125</sup>Te NMR (500 MHz, CDCl<sub>3</sub>) spectrum of HCTe



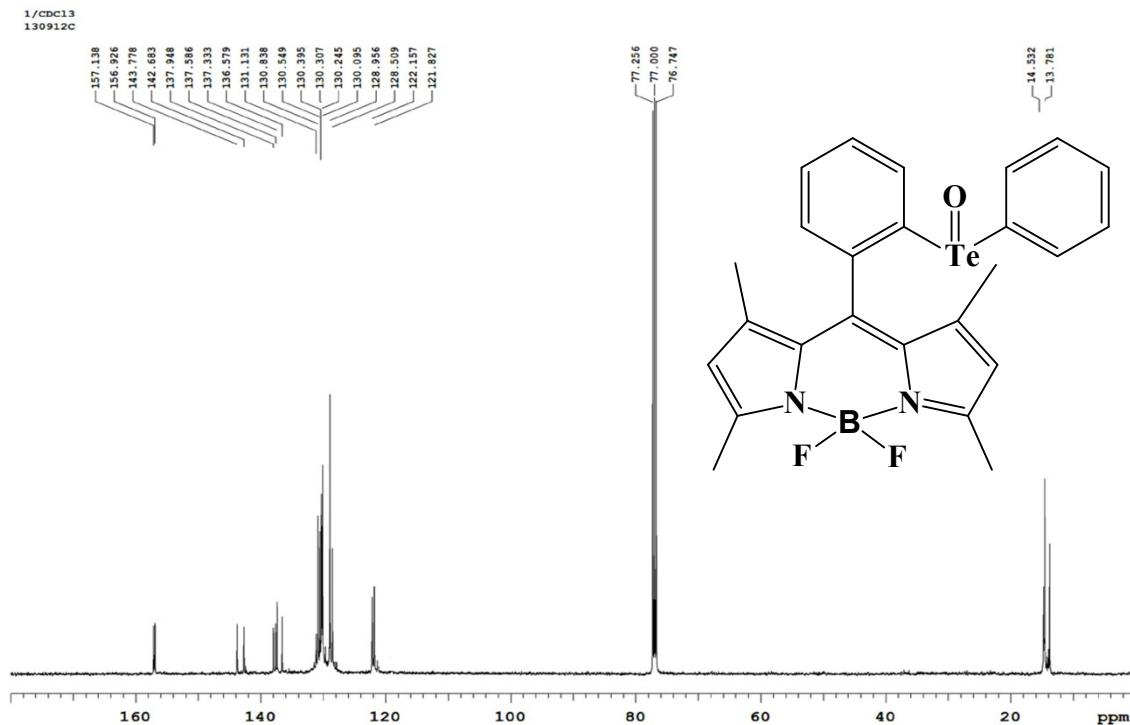
**Figure S8.** Mass spectrum (ESI+) of HCTe



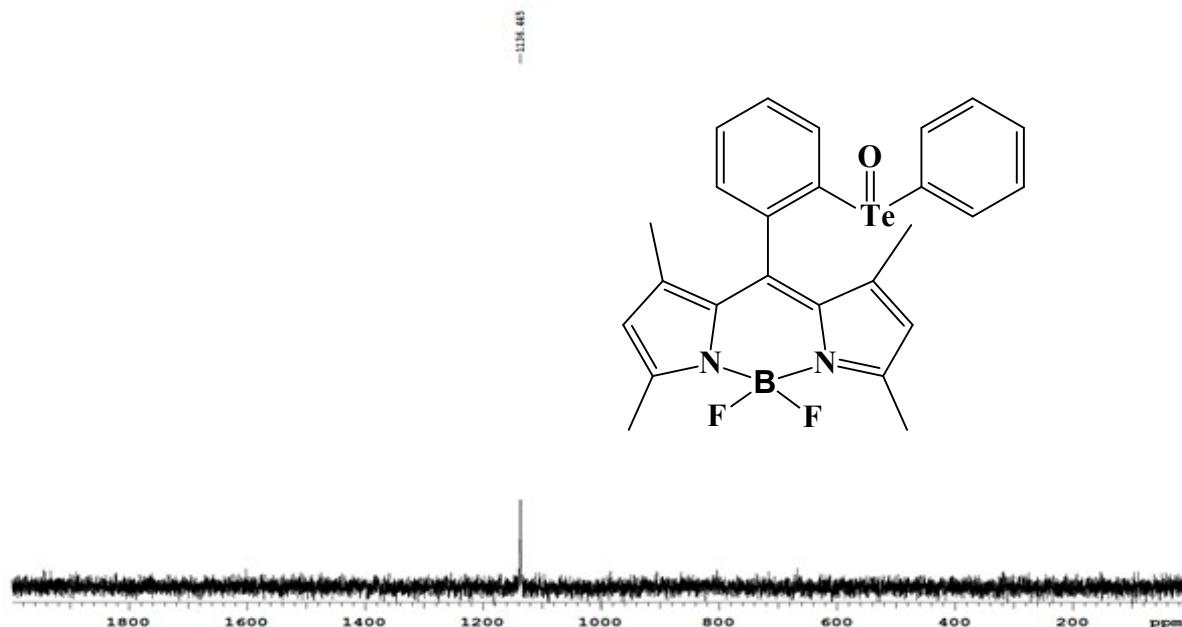
**Figure S9.** High resolution mass spectrum of HCTe



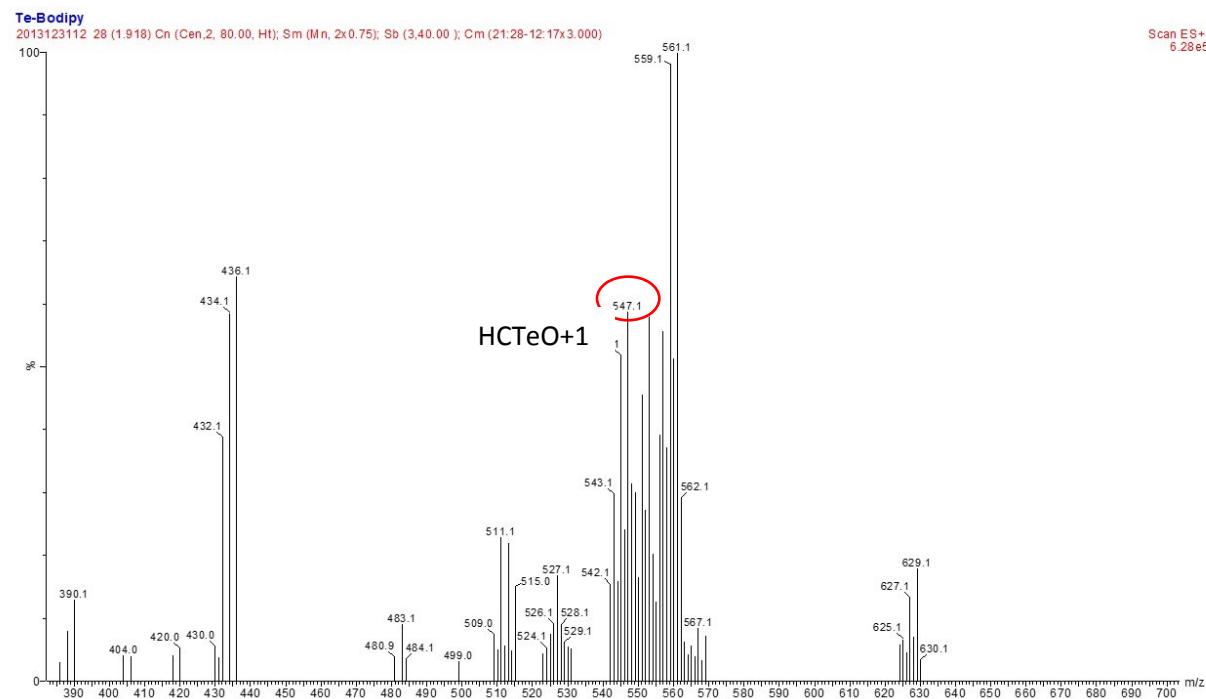
**Figure S10.**  $^1\text{H}$  NMR (500 MHz,  $\text{CDCl}_3$ ) spectrum of HCTeO



**Figure S11.**  $^{13}\text{C}$  NMR (500 MHz,  $\text{CDCl}_3$ ) spectrum of HCTeO



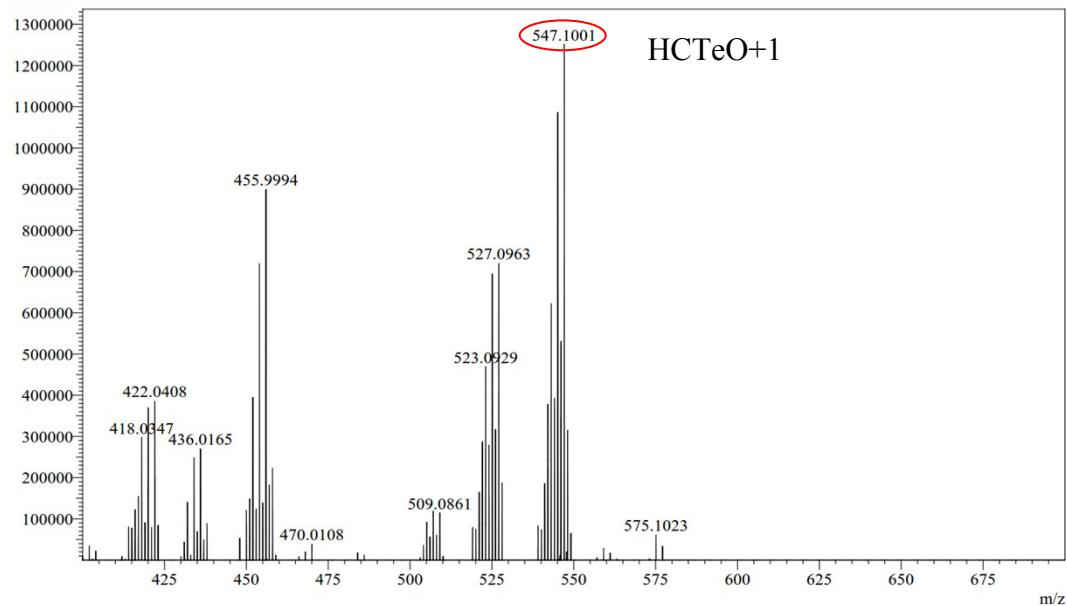
**Figure S12.**  $^{125}\text{Te}$  NMR (500 MHz,  $\text{CDCl}_3$ ) spectrum of HCTeO



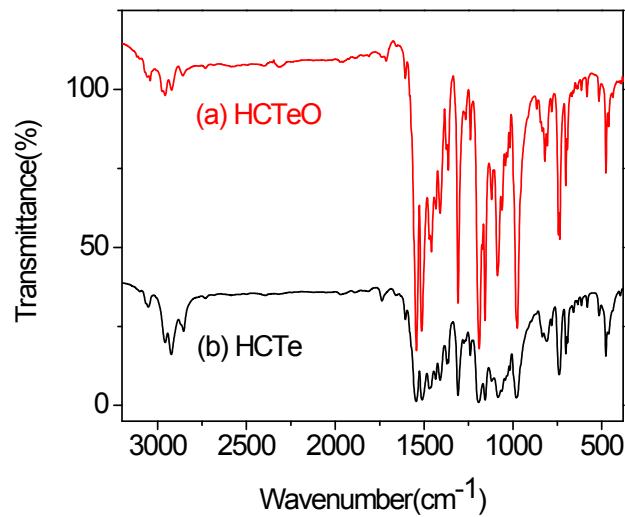
**Figure S13.** Mass spectrum (ESI+) of HCTeO

<Spectrum>Te=oBodipy-02.lcd

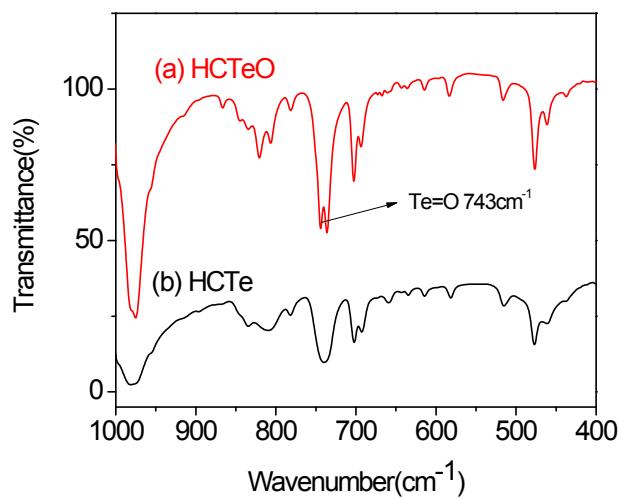
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Intensity



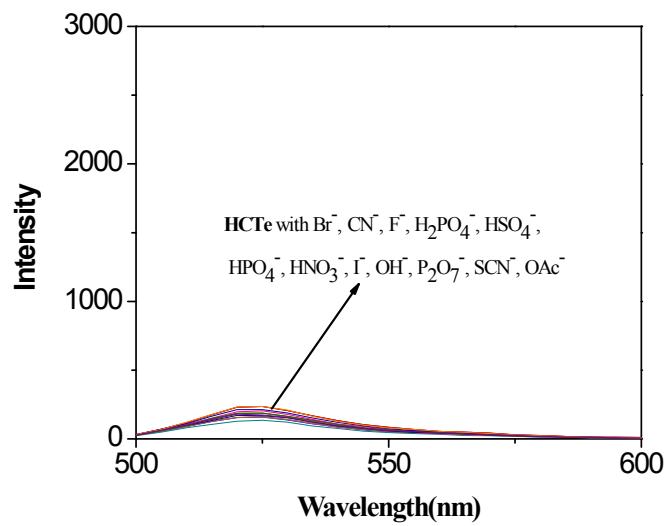
**Figure S14.** Mass spectrum of HR (ESI+) of HCTeO



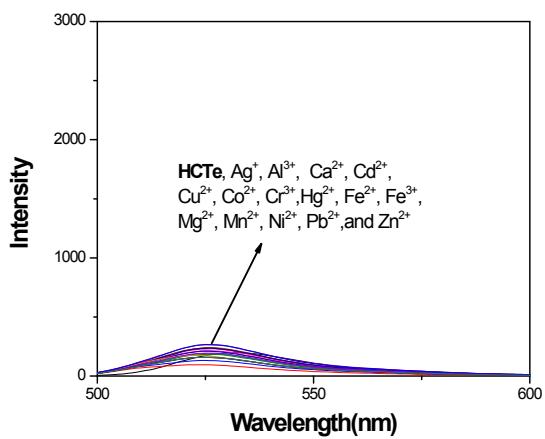
**Figure S15.** IR spectrum of HCTe and HCTeO.



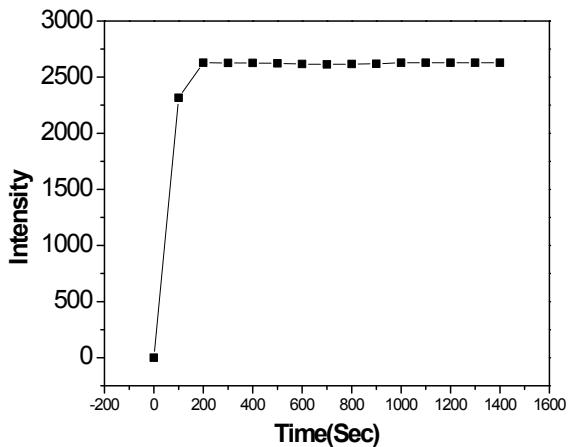
**Figure S16.** Expanded IR spectrum of HCTe and HCTeO.



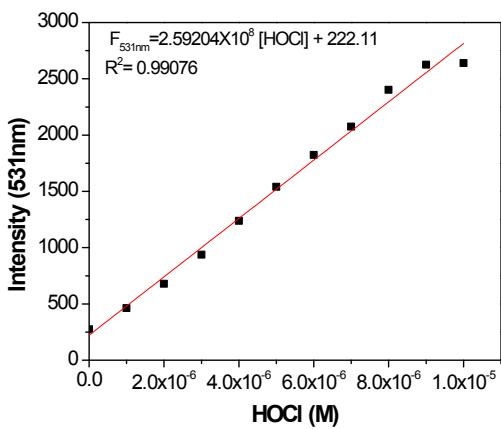
**Figure S17.** Fluorescence changes in HCTe (10  $\mu\text{M}$ ) in response to treatment with various anions (100  $\mu\text{M}$ ) in a water– $\text{CH}_3\text{OH}$  ( $\text{v/v} = 99/1$ , 0.1 M PBS, pH 7.4) solution. The excitation wavelength was 480 nm.



**Figure S18.** Fluorescence changes in HCTe (10  $\mu$ M) in response to treatment with various metal ions (100  $\mu$ M) in a water–CH<sub>3</sub>OH (v/v = 99/1, 0.1 M PBS, pH 7.4) solution. The excitation wavelength was 480 nm.

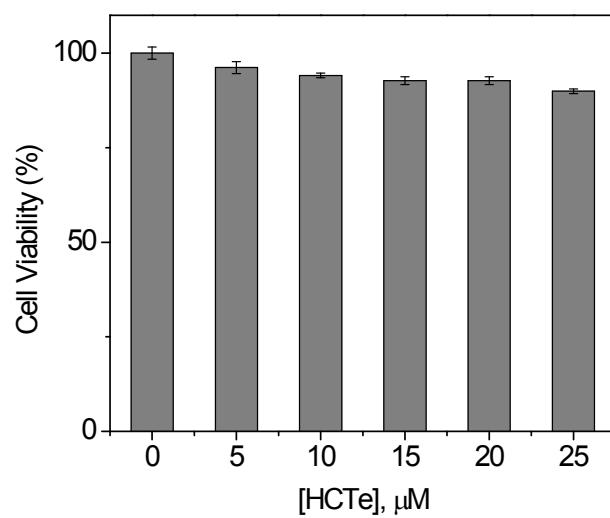


**Figure S19.** Time courses of the response of **HCTe** to HOCl. The excitation wavelength was 480 nm.



**Figure S20.** Calibration curve of **HCTe**–NaOCl in a water-CH<sub>3</sub>OH(v/v = 99/1, 0.1 M PBS, pH 7.4) solution. The detection limit (DL) of **HCTe** with HOCl was determined from the following equation: DL = K \* S<sub>b</sub> / S, where K = 3; S<sub>b</sub> is the standard deviation of the blank solution; S is the slope of the calibration curve.

$$DL = 3 * 3.5678 / (2.59 \times 10^8) = 4.13 \times 10^{-8} \text{ M (41.3 nM)}$$



**Figure S21.** Cell viability values (%) estimated by an MTT assay versus incubation concentrations of HCTe. RAW264.7 cells were cultured in the presence of **HCTe** (0–25 $\mu\text{M}$ ) at 37°C for 24 h.