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12	60mM.
13	Fig. S2 Effect of injection time on the organotin compounds peak intensity. The data was obtained
14	by determining 1 mg L ⁻¹ mixed solution of TMT, TBT, DBT and MBT with CE-ICP-MS under
15	optimized condition except sample injection time.
16	Fig. S3 Effect of HNO_3 concentrations in the make up solution on peak intensity. The data was
17	obtained by determining 1 mg L ⁻¹ mixed solution of TMT, TBT, DBT and MBT with CE-ICP-MS
18	under optimized condition except HNO ₃ concentrations in the make up solution.
19	Fig. S4 Electropherograms of tap water samples. Electropherograms: (A) tap water samples
20	spiking with 0.5 mg L ⁻¹ each individual organotin compound; (B) blank tap water sample. The
21	data was obtained under the optimized condition by CE-ICP-MS.

Table S1 ICP-MS parameters

Parameters	Settings
Plasma RF power /W	1500
Plasma gas flow /(L min ⁻¹)	15.0
Carrier gas flow /(L min ⁻¹)	0.7
Makeup gas flow /(L min ⁻¹)	0.4
Peak pattern	Full-Quant
Integration time per mass / s	0.3
Nebulizer	micromist
Isotope monitored	¹¹⁷ Sn, ¹¹⁸ Sn, ¹¹⁹ Sn, ¹²⁰ Sn

- **Table S2** Method detection limits for organotin Species using different CE-hyphenated techniques

Technique	Technique Analyts		Reference and Note	
CE LIV	TMT, TET, TBT, and	2.20 umol I^{-1}	(13)	
CE-UV	TPhT	2–20 µmoi L		
CE fluorogaonaa	TMT, TET, TPT,	9 19 um al I ⁻¹	(14)	
CE-muorescence	TBT, and TPhT	δ-18 μποι L		
CE LIV	TBT, TPhT, TPrT,	$0.4.14$ um cl I^{-1}	(15)	
CE-UV	DPhT	0.4–14µmoi L		
	TMT, MBT, DBT, and	and	(24)	
CE-HU-AFS	TBT	1–10 μ moi L	(24)	
CE-ICP-MS	TMT,TBT,DBT,MBT	0.31–0.943 μmol L ⁻¹	This work	

- 32 Table S3 Spiked recoveries of organotin compounds in groundwater samples by CE-ICP-MS
- 33 under optimized condition.

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	Recovery (%)					
Compound	Tap water		River water 1		River water 2	
	0.1 mg L^{-1}	0.5 mg L^{-1}	0.1 mg L ⁻¹	0.5 mg L^{-1}	0.1 mg L^{-1}	0.5 mg L^{-1}
TMT	87.8	104.6	73.0	105.7	73.4	104.5
TBT	90.3	99.5	76.3	104.8	86.9	99.7
DBT	96.7	97.6	77.9	103.5	84.6	105.2
MBT	90.2	99.8	87.8	99.7	90.8	93.2

36 Table S4 Recoveries and concentrations of organotin compounds in wine and Mya arenaria

37 *Linnaeus* samples by CE-ICP-MS.

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compound	Concentrations of organotin compounds in wine sample (blank) (ug L ⁻¹)	Recovery (%) of wine sample spiked with 0.5 mg L ⁻¹ (as Sn)	Concentrations (µg g ⁻¹) of organotin compounds in <i>Mya arenaria</i> <i>Linnaeus</i> samples	Recovery (arenaria Linn 0.1 mg L ⁻¹ (as Sn)	(%) of <i>Mya</i> naeus samples 0.5 mg L ⁻¹ (as Sn)
			(blank)		
TMT	nd	98.7	nd	81.7	80.0
TBT	nd	96.8	1.02	77.5	68.6
DBT	nd	62.3	nd	78.0	70.3
MBT	325	31.8	nd	61.7	73.7

- 40 Fig. S1 Electropherograms of the four organotin compounds (TMT, TBT, DBT, MBT), each as
- 41 $1\mu g mL^{-1}$ as Sn, under different separation electrolyte buffers. (1) tartaric acid 20mM and
- 42 methanol 20% (v/v); (2) Na₂HPO₄ 20mM and boric acid 20mM; (3) boric acid 60mM and Tris
- 43 60mM.



- 46 Fig. S2 Effect of injection time on the organotin compounds peak iintensity. The data was
- 47 obtained by determining 1 mg L⁻¹ mixed solution of TMT, TBT, DBT and MBT with CE-ICP-MS
- 48 under optimized condition except the injection time.
- 49



- 52 Fig. S3 Effect of HNO₃ concentrations in the make-up solution on peak intensity. The data was
- 53 obtained by determining 1 mg L^{-1} mixed solution of TMT, TBT, DBT and MBT with CE-ICP-MS
- 54 under optimized condition except HNO₃ concentrations in the make-up solution.
- 55



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- 58 Fig. S4 Electropherograms of tap water samples. Electropherograms: (A) tap water samples
- 59 spiking with 0.5 mg L^{-1} each individual organotin compound; (B) blank tap water sample. The
- 60 data was obtained under the optimized condition by CE-ICP-MS.
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