

Supplementary Information

Novel pH sensor based on anthraquinone–ferrocene modified free standing gold nanowire array electrode

Louise Devlin, Mamun Jamal* and Kafil M. Razeeb

Received (in XXX, XXX) Xth XXXXXXXXXX 20XX, Accepted Xth XXXXXXXXXX 20XX

DOI: 10.1039/b000000x

Tyndall National Institute, University College Cork, Lee Maltings, Dyke Parade, Cork, Ireland

*Email: mamun.jamal@tyndall.ie

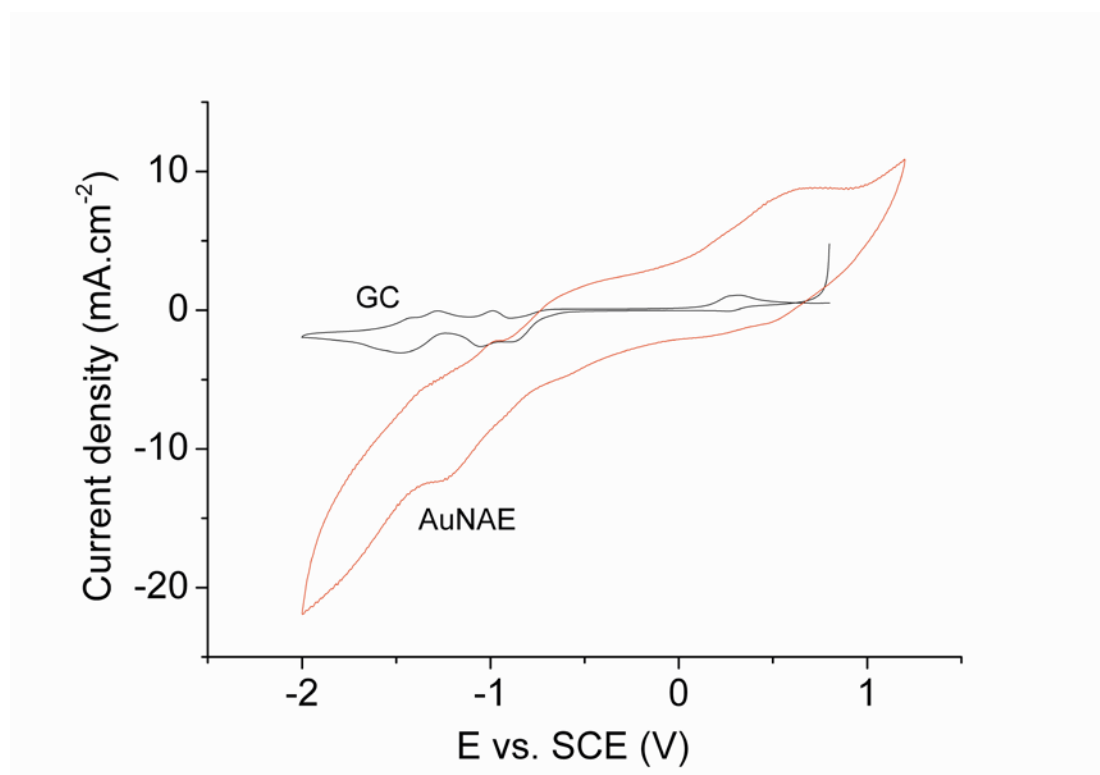


Fig. S1 Cyclic voltammograms of GC and AuNAE in AQ-Fc dissolved in acetonitrile (electrolyte: 0.1 M tetrabutyl ammonium hexafluorophosphate).

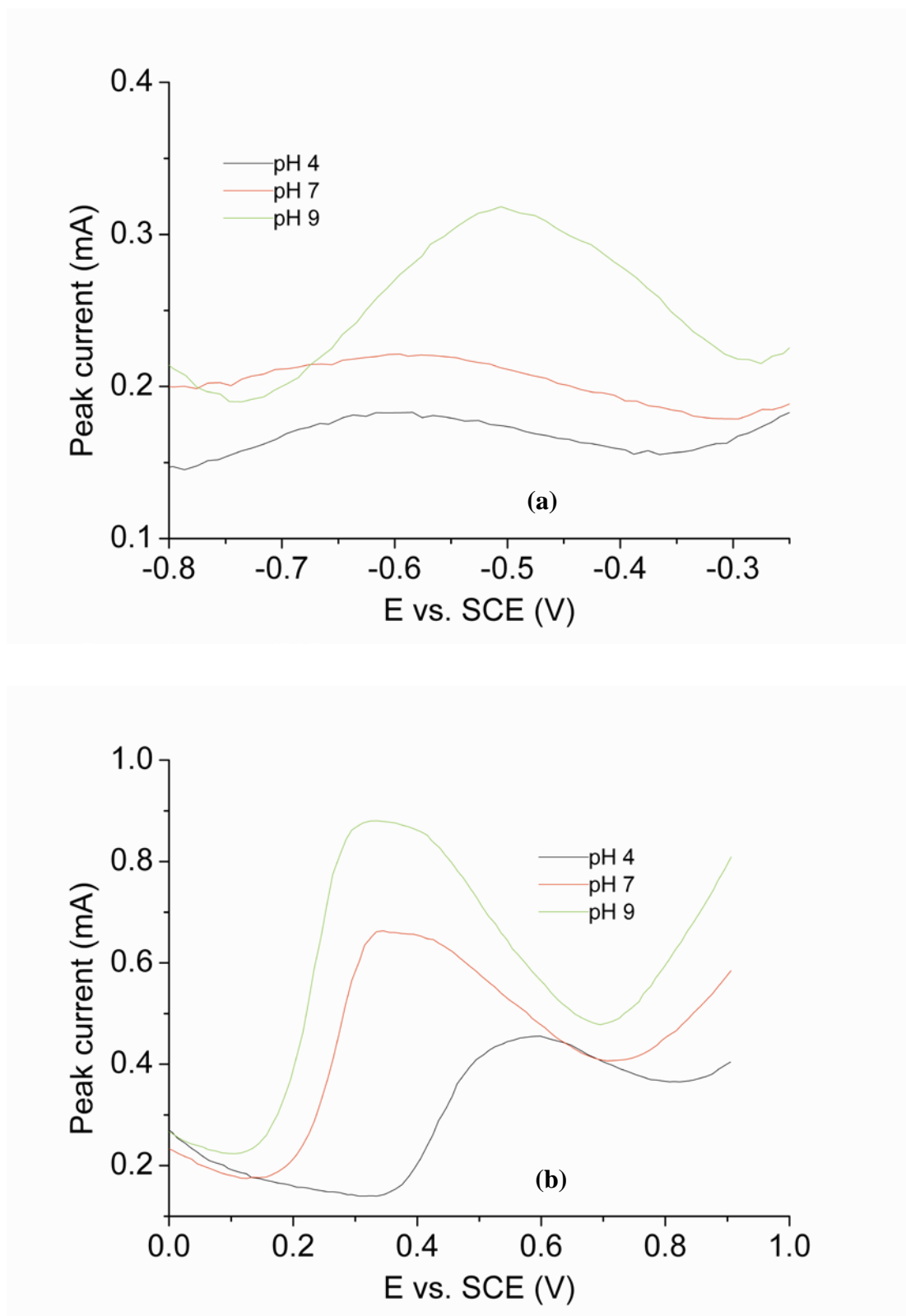


Fig. S2 SWV of AQ-Fc modified AuNAE in various pH solutions (a) potential range from -0.8 to -0.35 V; (b) potential range from 0 to 0.9 V.

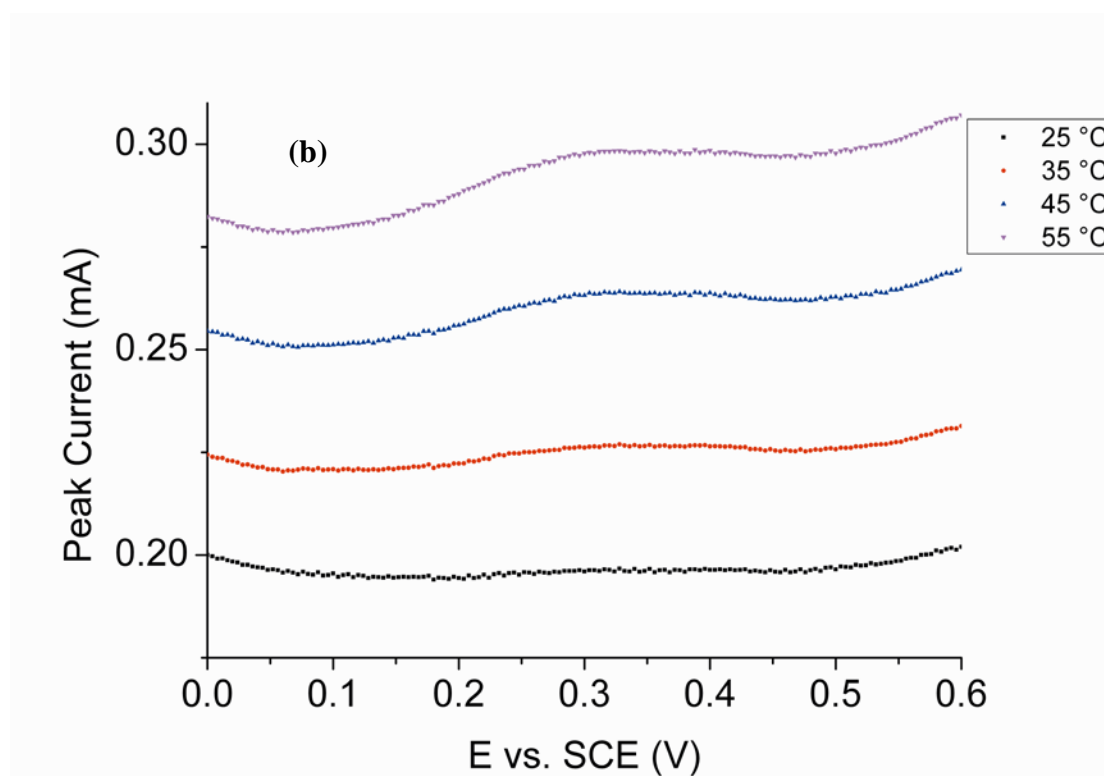
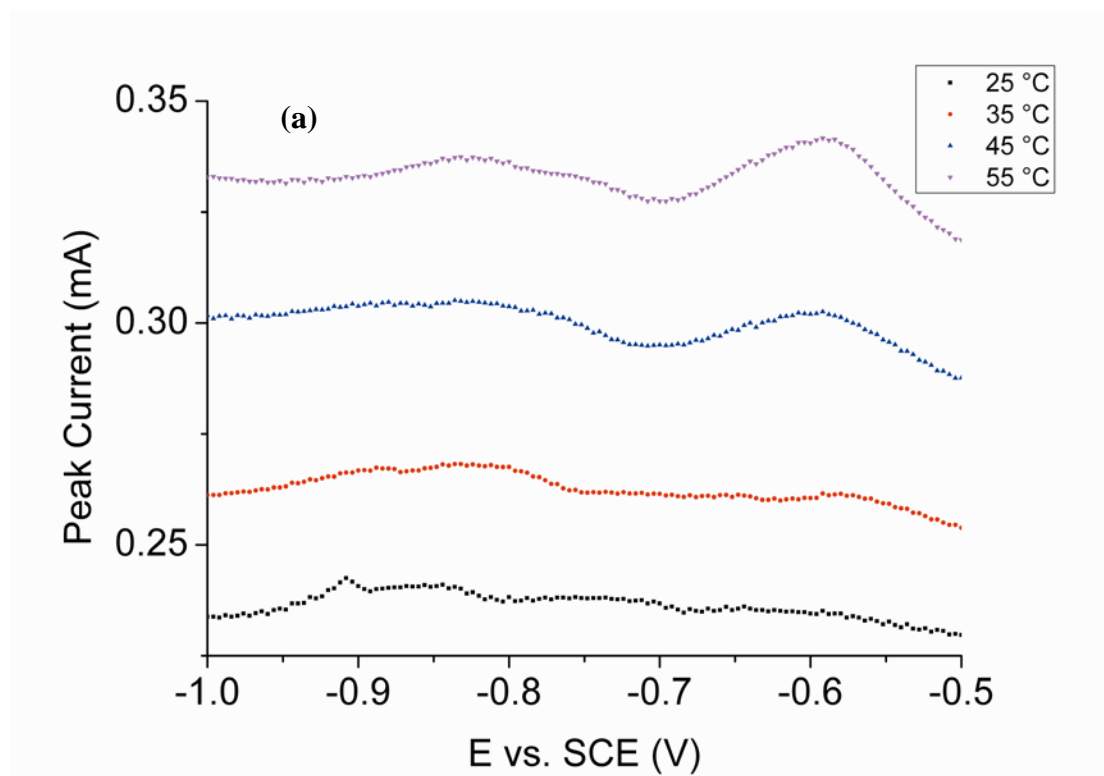


Fig. S3 SWV of AQ-Fc modified AuNAE in pH 7 at various temperatures (25, 35, 45 and 55 °C) (a) potential range from -1.0 to -0.50 V; (b) potential range from 0.0 to +0.6 V.

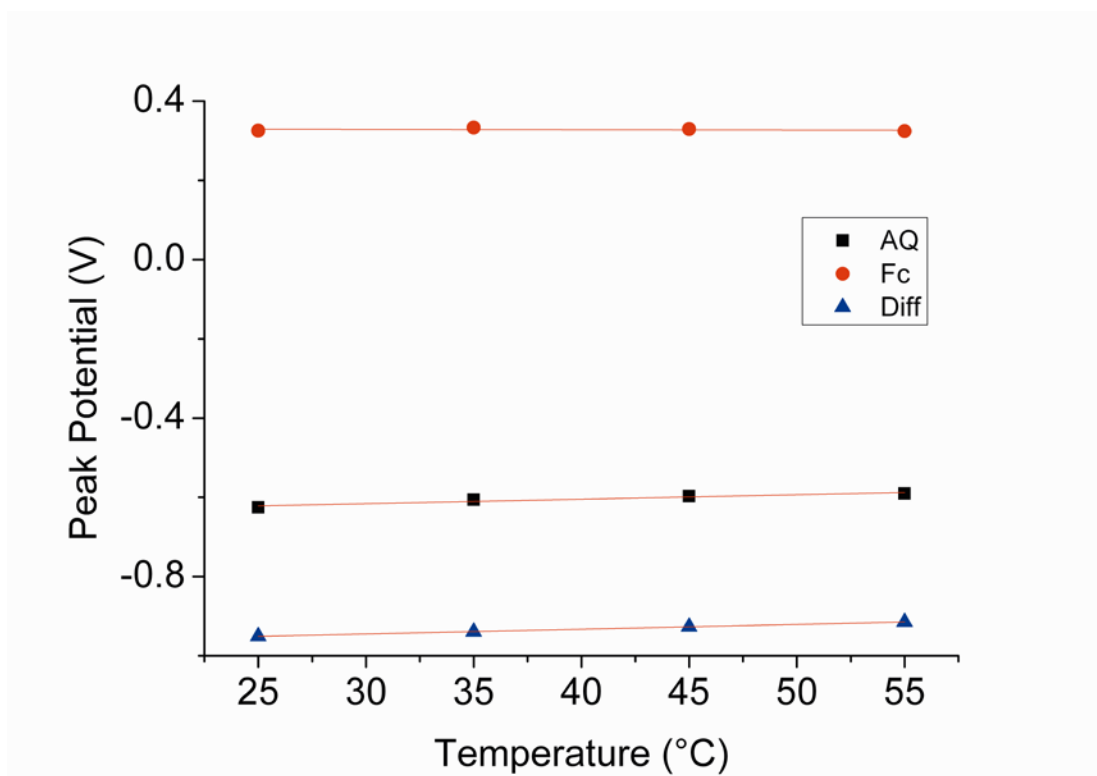


Fig. S4 A plot of peak potential against temperature for the Fc, the AQ and the difference in peak potential between both Fc and AQ, at pH 7.