

Supporting Information

A Novel Method for Cetylpyridinium Bromide Determination in Aqueous Solution Based on Fluorescence Quenching of Dye

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Content

- 1. Fluorescence behavior of CXT vs. pH**
- 2. Fluorescence decay profile of CXT**
- 3. Fluorescence decay profile of CXT/CPB**

1. Fluorescence behavior of CXT vs. pH

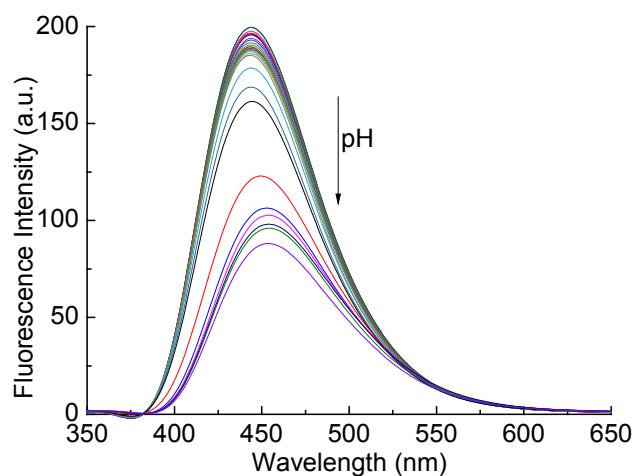


Fig. SI1 Fluorescence spectra of the CXT ($5.0 \mu\text{M}$) in aqueous solution of Brij 35 (1.2 mM) at different pH values.

2. Fluorescence decay profile of CXT

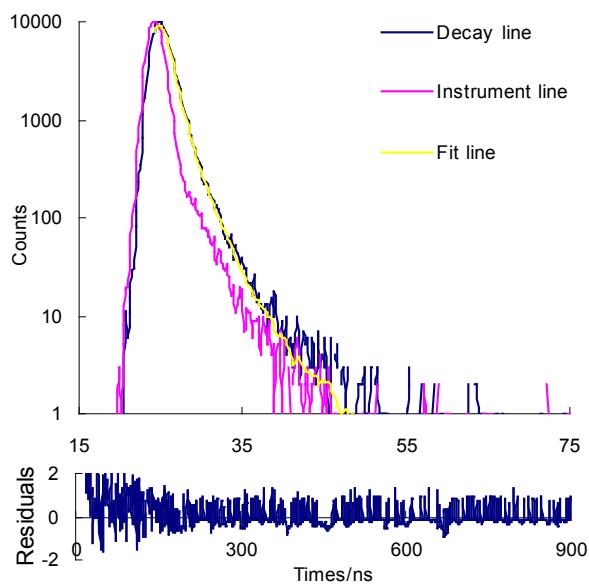


Fig. SI2 Fluorescence decay profile of CXT ($5.0 \mu\text{M}$) in aqueous solution of Brij 35 (1.2 mM) at pH 8.0. $\lambda_{\text{em}} = 444 \text{ nm}$, EX/EM slit = 13.0 / 13.0 nm.

3. Fluorescence decay profile of CXT/CPB

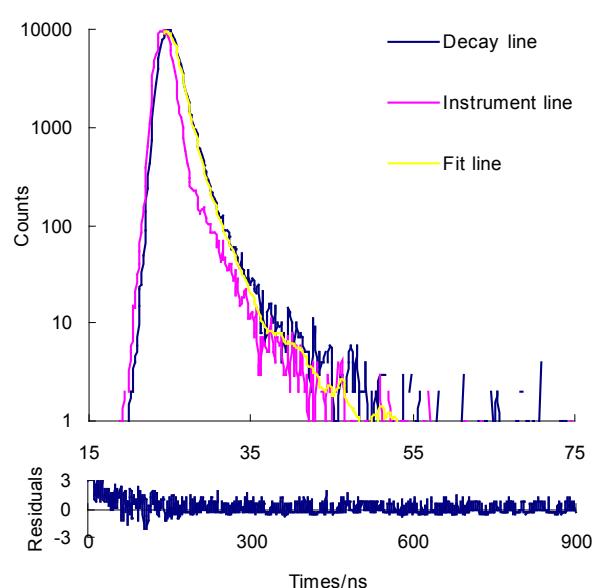


Fig. SI3 Fluorescence decay profile of CXT/CPB (5.0 μ M/ 20 μ M) in aqueous solution of Brij 35 (1.2 mM) at pH 8.0. $\lambda_{\text{em}} = 444$ nm, EX / EM slit = 13.0 / 13.0 nm.