Electronic Supplementary Material (ESI) for Analytical Methods. This journal is © The Royal Society of Chemistry 2014

Supplementary Information

1. The procedure of determination of inorganic magnesium content in coccolithophore calcite

The organically bound magnesium in the sample was initially dissociated and measured. Then nitric acid (0.1 mol·L⁻¹) was used to completely dissolve the sample, resulting to both organically bound and inorganic magnesium dissociation and then the magnesium concentration was remeasured. The inorganic magnesium was quantitated by the subtraction of organically bound magnesium from the total magnesium. This involved firstly weighing the samples, followed by dilution with deionised water. After 20 minutes of ultrasonication, the sample suspension was homogeneous and the alkaline-earth metal ions in sample suspension could be detected. Then Cu²⁺ solution (20 mg mL⁻¹) was added to the sample suspension. After 30 minutes of ultra-sonication, the alkaline-earth metal ions in the sample suspension were quantified again. The pH of the suspension was then adjusted to about 2.0 using nitric acid (0.1 mol·L⁻¹). This was enough to dissolve all of the carbonates (CaCO₃, SrCO₃ and MgCO₃) and to release all the Mg²⁺ from the organic phase. The alkaline-earth metal ions in the sample solution were then quantified. All the measurements were performed by a developed HPCIC method.

For each HPCIC measurement, samples were prepared by the following procedure: after the sample suspension became completely homogeneous by ultra-sonication, a certain amount (1 - 5 mL) of suspension was taken and centrifuged. After centrifugation, the supernatant was filtered (0.22 μ m syringe filter) and adjusted to pH ~ 2.0 using nitric acid (0.1 mol·L⁻¹). The acidic conditions prevented the precipitation of Ca²⁺ or Mg²⁺. Then chromatography was performed on the filtered supernatant.

2. The list of specified impurities in laboratory grade $CaCO_3$ samples from different suppliers

	Content, %					
Component	BDH chemicals (Poole, UK)	AJAX chemicals (Sydney, Australia)	Strem Chemicals (Miami, USA)			
CaCO ₃	> 99.0	> 99.0	> 99.5			
Chloride (Cl)	0.001	0.001	no data			
Sulphate (SO ₄)	0.01	0.01	no data			
Ammonium (NH ₄)	0.05	0.003	no data			
Heavy Metals (Pb)	0.002	0.001	no data			
Magnesium (Mg)	0.02	0.02	no data			
Iron (Fe)	0.001	0.003	no data			
Potassium (K)	0.05	0.01	no data			
Sodium (Na)	(sum of Na and K)	0.1	no data			
Silicate (SiO ₂)	0.01		no data			
Nitrate (NO ₃)	0.02		no data			
Phosphate (PO ₄)	0.002		no data			
Barium (Ba)	no data	0.005	no data			
Strontium (Sr)	no data	0.1	no data			

3. The HR-ICP-MS procedure and data

Samples analysed using magnetic sector ICP-MS with medium spectral resolution employed. Samples analysed after dilution 10x, with Indium added as internal standard (at 100 ng L^{-1}), and nitric acid added to final concentration 1%. Quantitation via comparison to external aqueous calibration standards also prepared in 1% nitric acid. All values in measured samples as ng L^{-1} . Scaled up and blank adjusted samples as $\mu g L^{-1}$.

The analysed solutions

QC1 and QC2: Quality Control, using 100 ng L⁻¹ of Na, Mg, Ca and K standard solution.

Rinse: Rinsed the machine by ultra-pure water

NIST: SRM 1640a - Trace Elements in Natural Water Reference Material

A1: 53.61 mg of standard reference material (limestone)

A2: 49.20 mg of standard reference material (limestone)

A3: 50.58 mg of standard reference material (limestone)

D1: 51.61 mg of calcium carbonate (Strem Chemicals)

F1: 51.03 mg of calcium carbonate (Strem Chemicals)

Samples A1, A2, A3, D1 and F1 were dissolved in 50 mL of 0.03 mol L-1 HNO₃.

Before analysis tests, ng L ⁻¹						
	QC1	Rinse1	Rinse1a	NIST	Rinse1b	Rinse1c
Dilution factor	1	1	1	2	1	1
²³ Na (MR)	103.7	4.6	4.3	1493.5	5.4	5.1
²⁴ Mg (MR)	101.5	0.3	0.3	498.6	0.2	0.3
⁴² Ca (MR)	95.4	3.5	2.7	2688.5	4.3	2.8
³⁹ K (HR)	98.2	1.3	1.1	286.9	1.1	1.0

Analysis of calcium carbonate sample, ng L ⁻¹						
	Blank	A1	A2	A3	D1	F1
Dilution factor	10	10	10	10	10	10
²³ Na (MR)	928.5	117.3	129.3	664.4	8864.3	17114.2
²⁴ Mg (MR)	1.2	159.2	143.3	152.5	529.7	523.4
⁴² Ca (MR)	31.4	37117.1	34427.4	35873.0	33592.6	38418.0
³⁹ K (HR)	5.1	489.8	693.4	260.3	187.5	105.5

After analysis tests, ng L ⁻¹					
	Rinse2	QC2	Rinse2a		
Dilution factor	10	10	10		
²³ Na (MR)	21.3	118.5	16.4		
²⁴ Mg (MR)	0.5	101.3	0.2		
⁴² Ca (MR)	4.3	104.1	2.5		
³⁹ K (HR)	1.4	97.9	1.1		

Suggested target concentrations, μg L ⁻¹						
	A1	A2	A3	D1	F1	
²⁴ Mg (MR)	1.6	1.5	1.5	5.0	5.0	