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Electronic Supporting Information

Investigation of the adsorption behaviour of different types of dyes on MIL-100(Fe) and their removal from natural water

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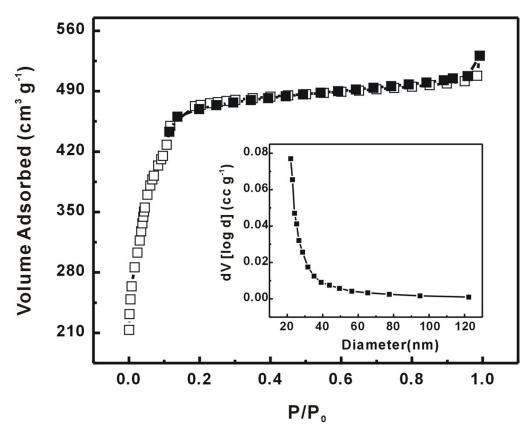


Figure S1. N_2 adsorption-desorption isotherms (left) and the pore size distribution (right) of the as-synthesized MIL-100.

Figure S2. The structures of Methylene Blue (A), Isatin (B) and Methyl Blue (C).

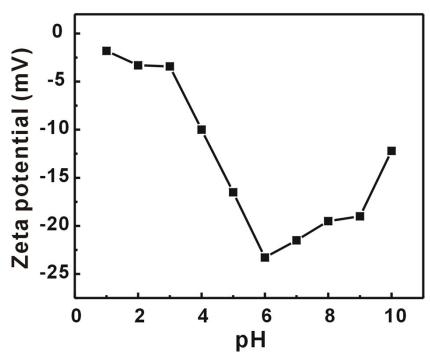


Figure S3. Zeta potentials of MIL-100(Fe) in ultrapure water under various pH at room temperature.

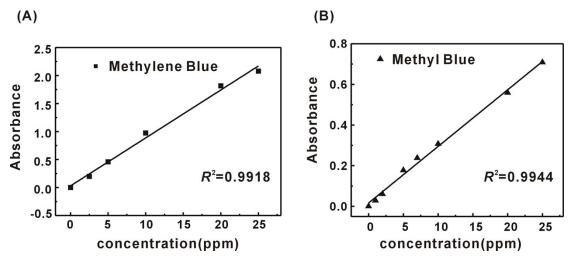


Figure S4. The plots of absorbance *vs* concentration of dyes (A) Methylene Blue and (B) Methyl Blue.

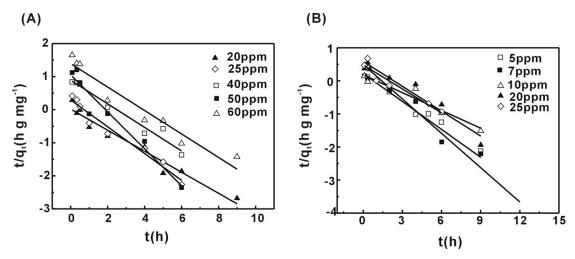


Figure S5. Plots of pseudo-first-order kinetics for the adsorption of Methylene Blue(A) and Methyl Blue(B) at different initial concentrations on MIL-100(Fe) at room temperature and pH 6.0.

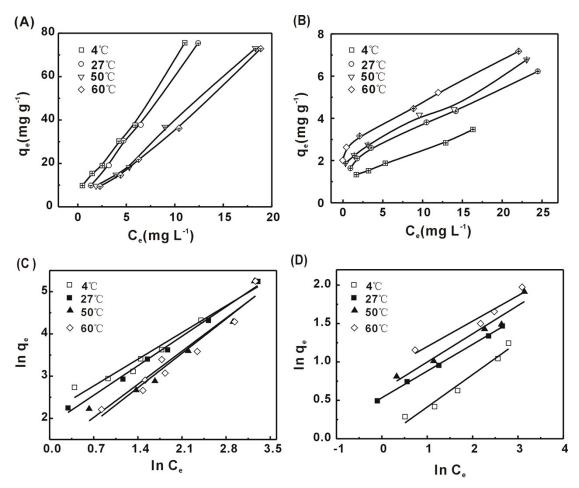


Figure S6. Adsorption isotherms for the adsorption of (A) Methylene Blue and (B) Methyl Blue on MIL-100(Fe) in the temperature range of 4–60°C. And the corresponding Freundlich plots for (C) Methylene Blue and (D) Methyl Blue.

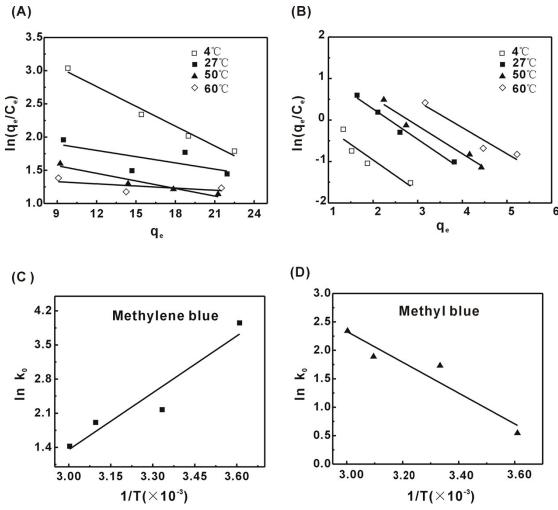


Figure S7. Plots of $ln(q_e/C_e)$ vs q_e at various temperatures and lnK_0 against 1/T for the adsorption of Methylene Blue(A, C) and Methyl Blue(B, D) on MIL-100(Fe).

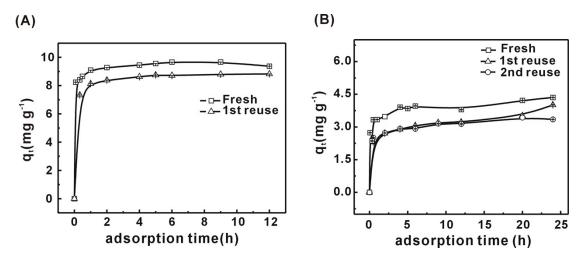


Figure S8. Time-dependent adsorption capacity of Methylene Blue and Methyl Blue on fresh and regenerated MIL-100(Fe).

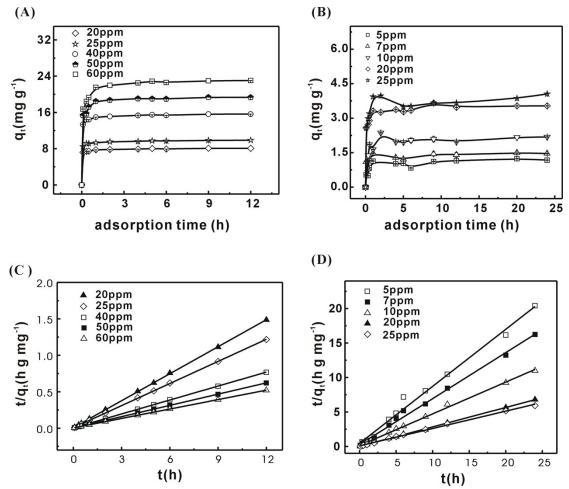


Figure S9. Time-dependent adsorption of Methylene Blue(A) and Methyl Blue(B) on MIL-100(Fe) in rainwater. And the plots of pseudo-second-order kinetics for the adsorption of Methylene Blue (C) and Methyl Blue (D) at different initial concentrations at room temperature.

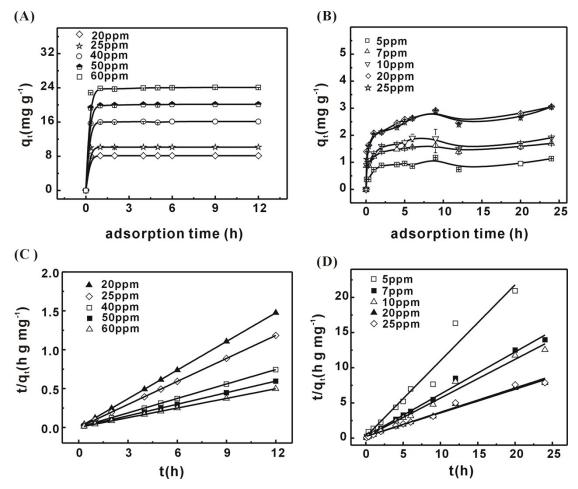


Figure S10. Time-dependent adsorption of Methylene Blue(A) and Methyl Blue(B) on MIL-100(Fe) in river water. And the plots of pseudo-second-order kinetics for the adsorption of Methylene Blue (C) and Methyl Blue (D) at different initial concentrations at room temperature.