Supporting Information

Highly Selective Two-photon Fluorescence Probe for Determination of Mercury Ions

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1. ¹H NMR spectrum of PT in CDCl₃ (Fig. S1)

2. ¹³C NMR spectrum of PT in CDCl₃ (Fig. S2)

3. TOF MS spectrum of PT in CDCl₃ (Fig. S3)

4.¹H NMR spectrum of ATD in DMSO (Fig. S4)

5. ¹³C NMR spectrum of ATD in DMSO (Fig. S5)

6. TOF MS spectrum of ATD in DMSO (Fig. S6)

7. The quadratic relationship of the observed two-photon fluorescence intensity of ATD under excitation at 800 nm (Fig. S7)

8. Quantum yield of ATD (Fig. S8)

9. Optimized reaction time (Fig. S9)

1. ¹H NMR spectrum (400 MHz) of PT in CDCl₃



Fig. S1. ¹H NMR spectrum (400 MHz) of PT in CDCl₃.

2. ¹³C NMR spectrum (400 MHz) of PT in CDCl₃



Fig. S2. ¹³C NMR spectrum (400 MHz) of PT in CDCl₃.

3. TOF MS spectrum (400 MHz) of PT in CDCl₃



Fig. S3. TOF MS spectrum (400 MHz) of PT in CDCl₃.

4.1H NMR spectrum of ATD



Fig. S4. ¹H NM spectrum (400 MHz) of ATD in d₆-DMSO.

5. ¹³C NMR spectrum of ATD



Fig. S5. ¹³C NMR spectrum (400 MHz) of ATD in d₆-DMSO.

6. TOF MS spectrum (400 MHz) of ATD in DMSO



Fig. S6. TOF MS spectrum (400 MHz) of ATD in d₆-DMSO.

7. The quadratic relationship of the observed two-photon fluorescence intensity of ATD under excitation at 800 nm



Fig. S7. The close to quadratic relationship of the observed two-photon fluorescence intensity of ATD under excitation at 800 nm.

8. Quantum yield of ATD.



Fig. S8. Photoluminescence (excited at 400 nm) and absorbance (at 400 nm) of ATD probe and quinine sulfate.

9. Optimized reaction time



Fig. S9. The relationship between fluorescence intensity of ATD (5 μ M) and the reaction time with Hg²⁺. The optimum concentration of Hg²⁺ was selected 100 nM.