

Electronic Supplementary Information

Gold nanoparticle-catalyzed uranine reduction for signal amplification in fluorescent assays for melamine and aflatoxin B1

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Characterization of AuNPs

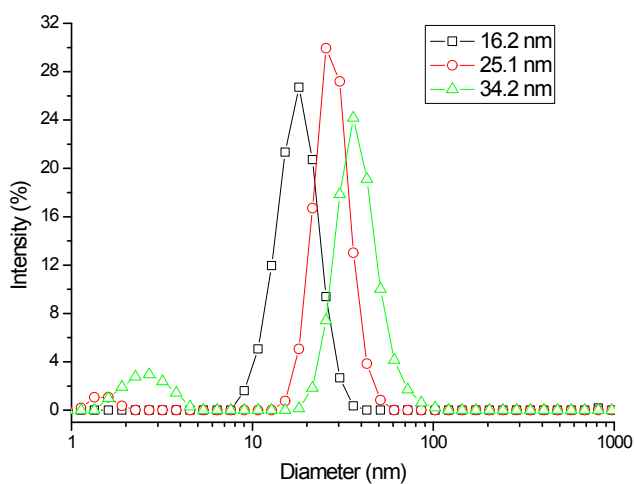


Fig. S1 Size distribution of the prepared AuNPs determined by DLS.

AuNP-catalyzed uranine reduction

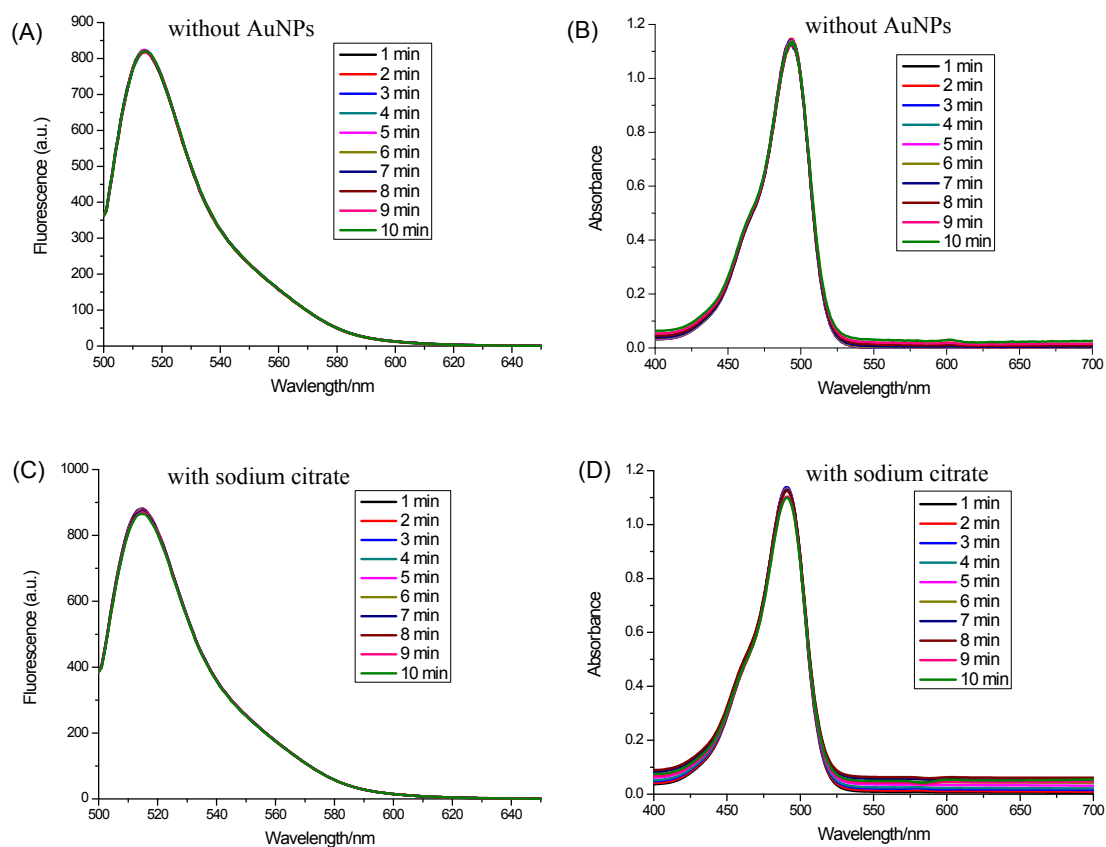


Fig. S2 (A) Time-dependent fluorescence spectra of uranine/NaBH₄ (8 μ M/10 mM) and (B) UV-vis absorption spectra of uranine/NaBH₄ solution (20 μ M/10 mM) in the absence of AuNPs. (C) Time-dependent fluorescence spectra of uranine/NaBH₄ (8 μ M/10 mM) and (D) Absorption spectra of uranine/NaBH₄ solution (20 μ M/10 mM) after addition of 20 μ L 1 wt % sodium citrate.

The effect of surface coating

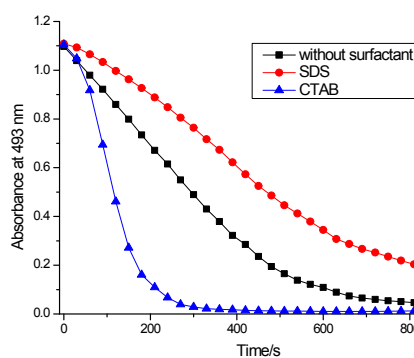


Fig. S3 Time-dependent changes of the absorbance of an uranine/ NaBH_4 solution ($20 \mu\text{M}/10 \text{mM}$) after addition of AuNPs with different surface coatings. The absorbance was recorded at 493nm , i.e., at the absorption maximum of the emissive dianionic form of uranine.

The effect of NaBH_4

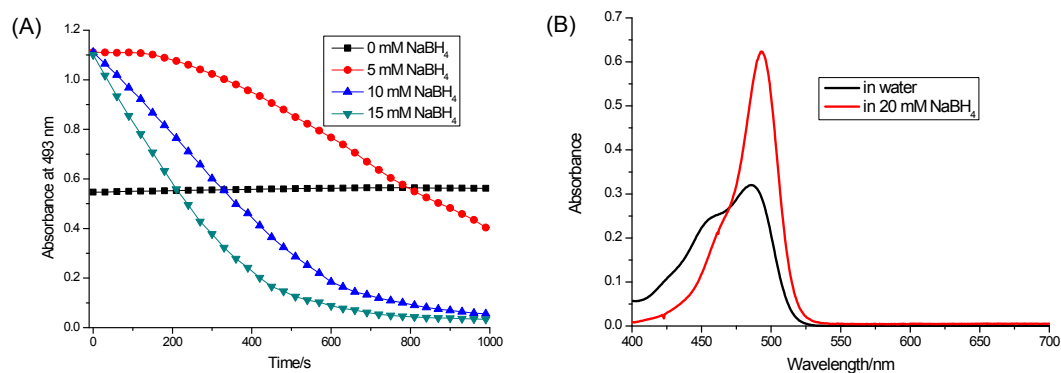


Fig. S4 (A) Effect of different NaBH_4 concentrations on the catalytic reduction of uranine ($20 \mu\text{M}$ uranine in 3 mL water, addition of NaBH_4 at different concentrations followed by addition of $20 \mu\text{L}$ of 16-nm citrate-stabilized AuNPs). (B) Absorption spectra of uranine ($10 \mu\text{M}$) in water and in aqueous NaBH_4 (20mM) solution.

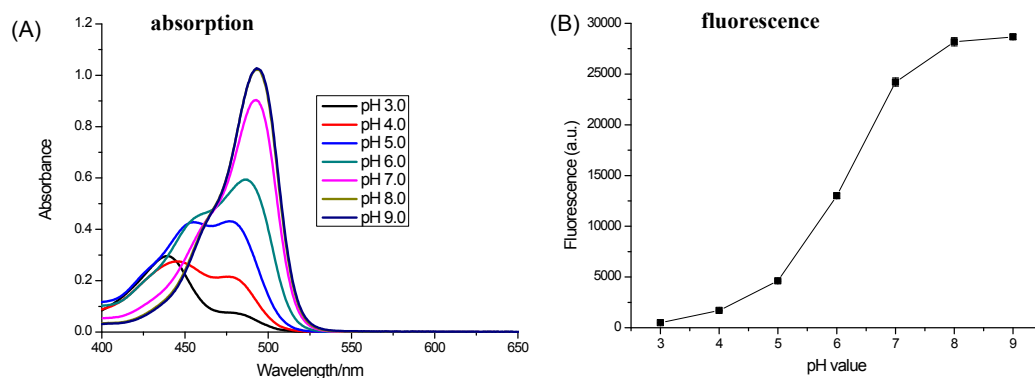


Fig. S5 Effect of pH value on the absorption spectra and fluorescence intensity of uranine. Conditions: $20 \mu\text{M}$ uranine in buffer with different pH value. The fluorescence was excited at 485nm and recorded at 528nm .

The effect of concentration, size and dispersion state of AuNPs

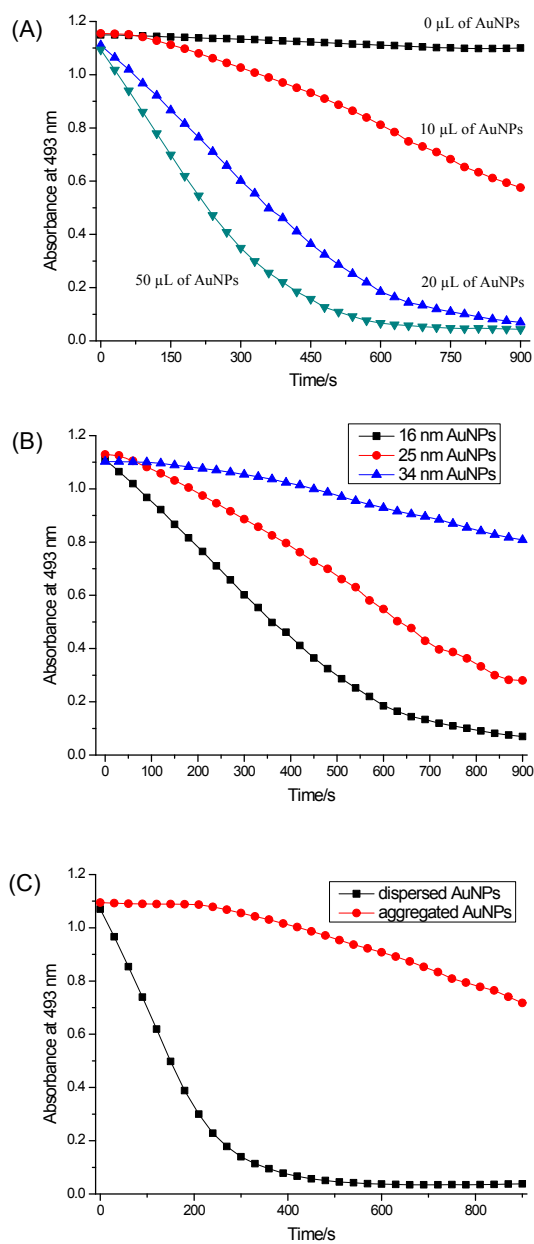


Fig. S6 (A) Effect of AuNP concentration (16-nm AuNPs were used). (B) Influence of AuNP size. $[\text{Au}] = 0.31$ mg/L. (C) Comparison of catalytic activity between dispersed and aggregated AuNPs. Conditions: 20 μM uranine and 10 mM NaBH_4 in 3 mL water, and AuNPs with different volumes, sizes or dispersion states were added.

Detection of melamine

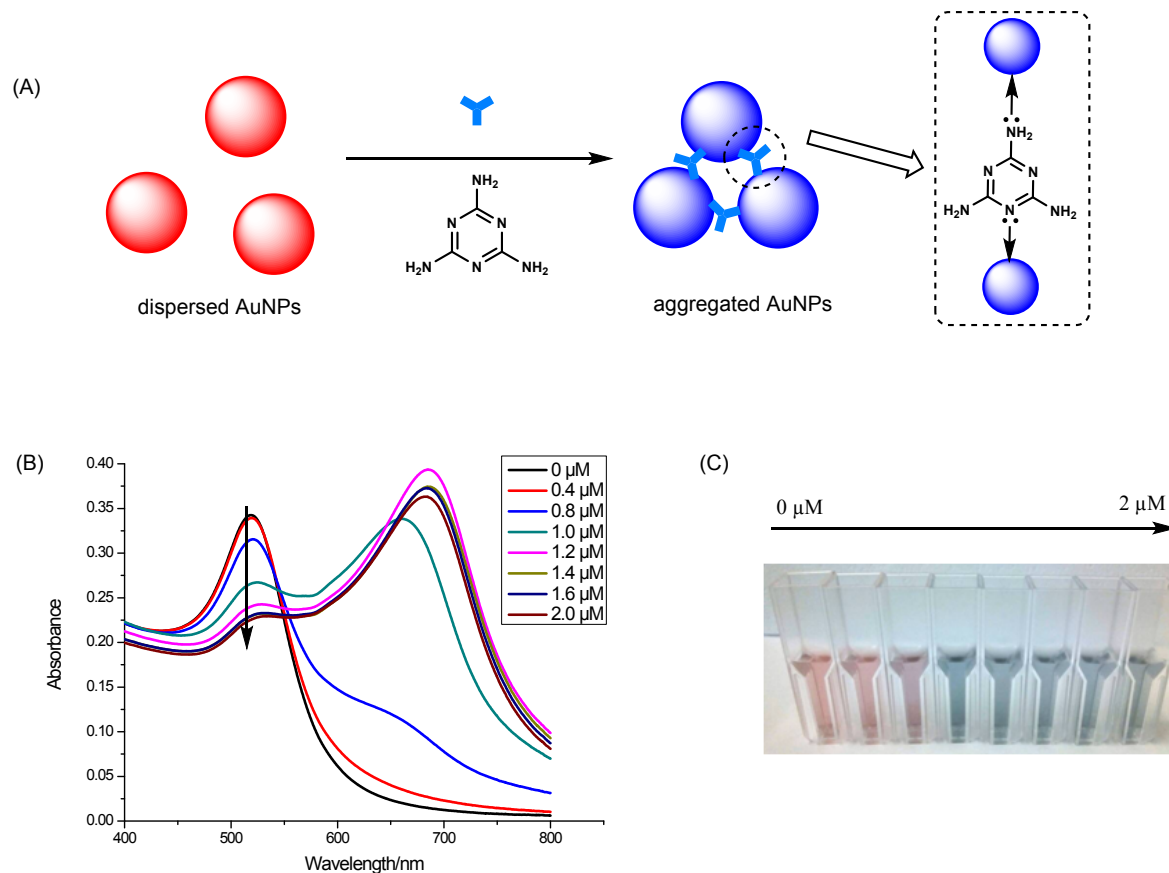


Fig. S7 (A) Schematic illustration of the aggregation of AuNPs induced by melamine. (B) Absorption spectra of AuNPs after addition of melamine with different concentrations and (C) Corresponding image. $[\text{Au}] = 9.46 \text{ mg/L}$.

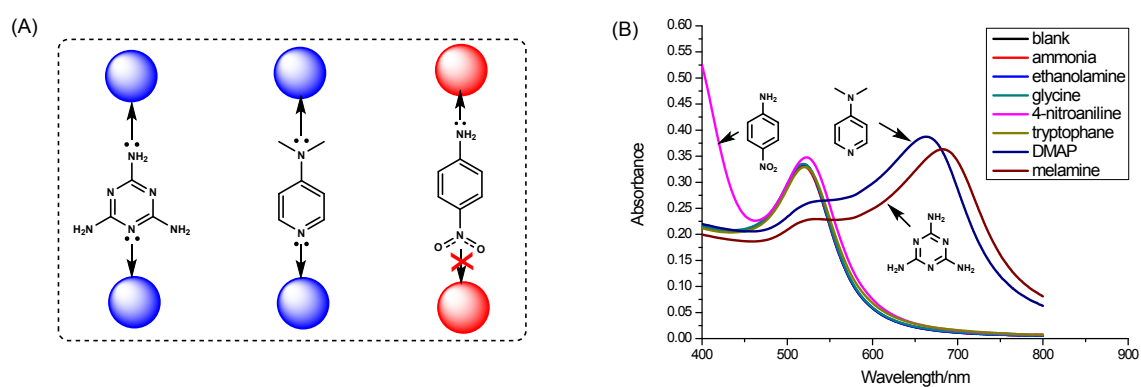


Fig. S8 (A) Schematic illustration of the interactions between AuNPs and amino compounds. (B) Absorption spectra of AuNPs in the presence of different amino compounds. $[\text{Au}] = 9.46 \text{ mg/L}$.

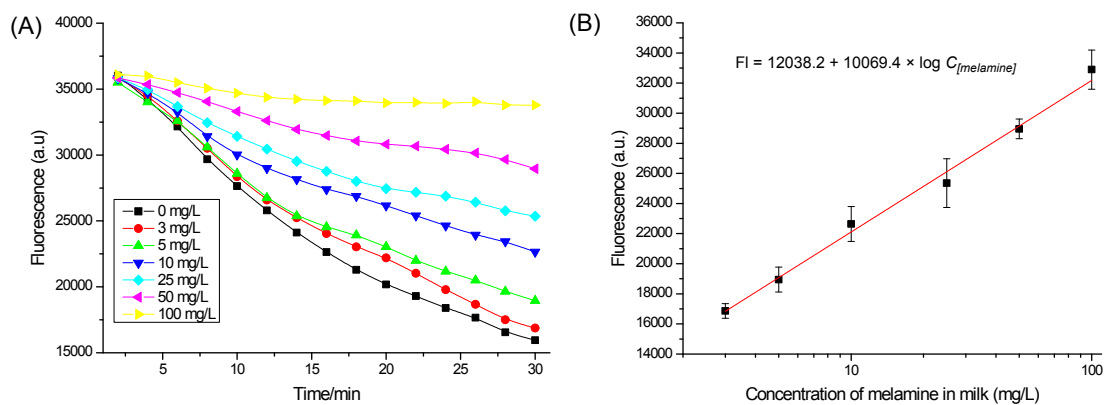


Fig. S9 Kinetic-based fluorescent assay for melamine in milk ($\lambda_{ex}/\lambda_{em} = 485/528$ nm). (A) Time-dependent fluorescence changes, which correspond to different concentrations of melamine. (B) Dependence of the fluorescence intensity at 30 min on melamine concentration in milk. The fluorescence was excited at 485 nm and recorded at 528 nm.

Detection of AFB1

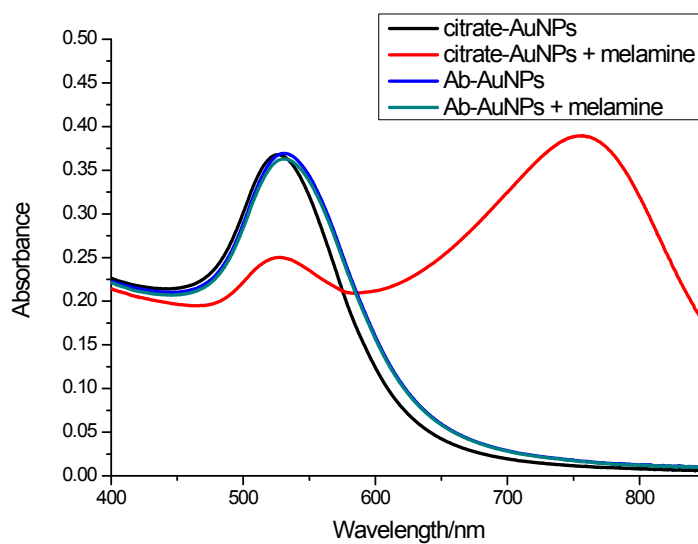


Fig. S10 Absorption spectra of citrate-AuNPs and Ab-AuNPs in the absence/presence of 2 μ M melamine. $[Au] = 9.46$ mg/L.

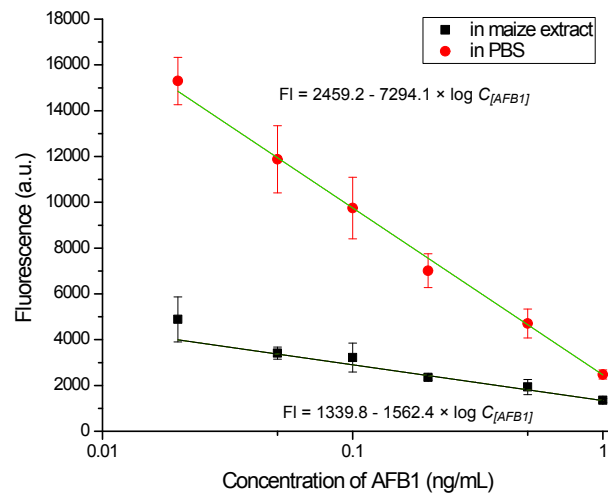


Fig. S11 Dose-response curves of the developed fluorescent immunoassays for the detection of AFB1 in maize extract and in PBS under similar reaction conditions.

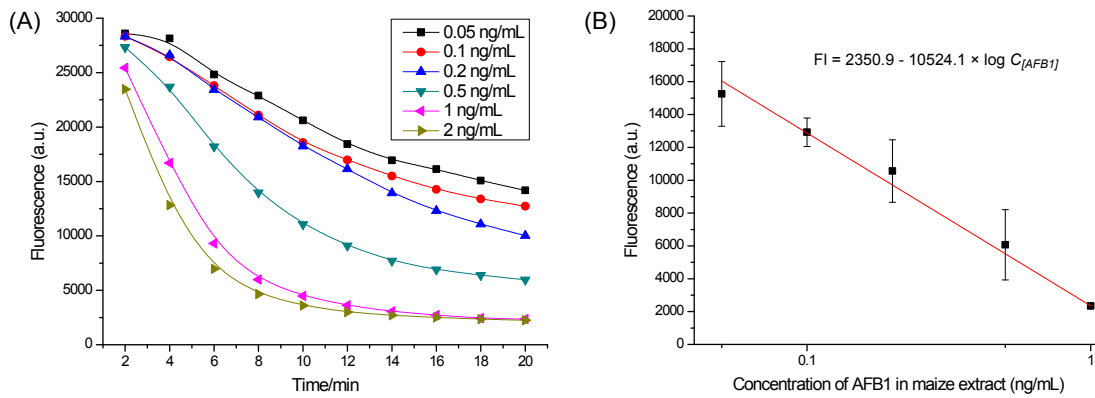


Fig. S12 Kinetic-based fluorescent assay for AFB1 in maize extract ($\lambda_{ex}/\lambda_{em} = 485/528$ nm). (A) Time-dependent fluorescence changes, which correspond to different concentrations of AFB1. (B) The dependence of fluorescence intensity at 20 min on AFB1 concentration in maize extract. The fluorescence was excited at 485 nm and recorded at 528 nm.

Table S1: The detected results of AFB1 in pulverized maize powders and recovery rates.

Spiked concentration (µg/kg)	After extraction and dilution (µg/L)	Detected concentration (µg/L)	Recovery rate (%)
8	0.10	0.104 ± 0.022	104.0
40	0.50	0.597 ± 0.084	119.4
80	1.00	0.847 ± 0.103	84.7