# A water soluble glucopyranosyl conjugate as selective and reactive probe for cysteine in buffer and its application to living cells

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**S01.** Synthesis and characterization of P<sub>2</sub>: P<sub>2</sub> derivative was synthesized by using a literature reported procedure.<sup>1</sup> [1] (a) Bolletta, F.; Fabbri, D.; Lombardo, M.; Prodi, L.; Trombini, C.; Zaccheroni, N. *Organometallics* **1996**, *15*, 2415-2417.; (b) Drillaud, N.; Estelle, B.-L.; Pezron, I.; Len, C.| J. Org. Chem. **2012**, *77*, 9553–9561.

**P**<sub>2</sub> was obtained as a yellow solid (85 %). <sup>1</sup>HNMR (400 MHz, CDCl<sub>3</sub>, δ ppm): 1.9 (t, <sup>1</sup>J=2.2 Hz and <sup>2</sup>J = 2.2 Hz, 1H), 2.8 (s, 6H), 3.8 (d, J= 7.4, 2H), 4.62 (t, <sup>1</sup>J = 9.8 Hz and <sup>2</sup>J = 8.2 Hz, 1H), 7.18 (d, J = 7.8 Hz, 1H), 7.50-7.58 (m, 2H), 8.23-8.29 (m, 2H), 8.53 (d, J = 8.4 Hz, 1H).



Fig. S01: (a) <sup>1</sup>H NMR spectrum (CDCl<sub>3</sub>, 400 MHz) of  $P_2$ .

**S02.** Synthesis and characterization of P<sub>4</sub>: This was synthesized by using a literature reported procedure.<sup>2</sup> [2] Nicolas, D.; Estelle, B.-L.; Isabelle, P.; Christophe, L. J. Org. Chem., 2012, 77, 9553-9561.

The yield is 69 % as white crystals. <sup>1</sup>HNMR (400 MHz, CDCl<sub>3</sub>,  $\delta$  ppm): 2.0 (S, 3H), 2.08 (S, 3H), 2.12 (S, 3H), 2.16 (S, 3H), 3.8-3.9 (m, 1H), 4.14 (d, J= 6.4 Hz, 1H), 4.28-4.3 (dd, <sup>1</sup>J= 2.2 Hz and <sup>2</sup>J= 2.2 Hz 1H), 4.65 (d, J=7.6 Hz, 1H), 4.95 (t, <sup>1</sup>J= 7.2 Hz and <sup>2</sup>J = 6.8 Hz, 1H), 5.15 (t, <sup>1</sup>J= 7.8 Hz and <sup>2</sup>J= 5.6 Hz, 1H) 5.25 (t, J= 8.2 Hz and <sup>2</sup>J= 6.4 Hz 1H).



Fig. S02. (b)  $^{1}$ H NMR (CDCl<sub>3</sub>, 400 MHz) for P<sub>4</sub>.



## **S03.** Characterization of C<sub>1</sub>





**Fig. S03:** (c) <sup>1</sup>H NMR (CDCl<sub>3</sub>, 400 MHz) (d) <sup>13</sup>C NMR (CDCl<sub>3</sub>, 400 MHz) (e) HRMS for  $C_1$ .

**S04.** Characterization of P<sub>5</sub>



Fig. S04. (f)  ${}^{1}$ H NMR (CDCl<sub>3</sub>, 400 MHz) (g)  ${}^{13}$ C NMR (CDCl<sub>3</sub>, 400 MHz) (h) HRMS for P<sub>5</sub>.

**S05.** Characterization of P<sub>6</sub>





Fig. S05. (i)  ${}^{1}$ H NMR (CDCl<sub>3</sub>, 400 MHz) (j)  ${}^{13}$ C NMR (CDCl<sub>3</sub>, 400 MHz) (k) ESI MS for P<sub>6</sub>.

**S06.** Characterization of L



Fig. S06: (1) <sup>1</sup>H NMR (CD3OD, 400 MHz) (m) <sup>13</sup>C NMR (CD3OD, 400 MHz) (o) HRMS for L.

**S07.** Characterization of L<sub>1</sub>



**Fig. S07:** (p) <sup>1</sup>H NMR (CD<sub>3</sub>OD, 400 MHz) (q) <sup>13</sup>C NMR (D<sub>2</sub>O, 400 MHz) (r) HRMS for L<sub>1</sub>.



S08. Fluorescence studies of L with amino acids

Fig. S08 Fluorescence spectra obtained for the titration of L [5  $\mu$ M,  $\lambda_{ex}$  = 360 nm] with different amino acids in HEPES buffer pH at 7.4.

#### S09 Determination of Limit of Detection (LOD) of Cys by L



**Fig. S09** (a) Fluorescence spectral traces of L during titration with Cys to determine LOD. (b) The linear dynamic fluorescence response for the titration of L with Cys to determine the detection limit (LOD). The LOD was derived by using the formula  $3\sigma/k$  where  $\sigma$  = standard deviation of the blank (10 blank samples) and k = is the slope of linear calibration curve.

S10. Comparison of the detection limits of recently developed fluorescent probes for Cys in the literature.

Probe	Detection	Detection	Reference
	medium	Limit (M)	
02N-()-===================================	Tris-HCl buffer	$100 \times 10^{-6}$	Bioorg. Med. Chem.
NÒ2 <sup>O</sup> NO2			Lett. 2008, 18, 2246
	HEPES	$2.13 \times 10^{-5}$	Org. Lett., 2013, 15
	Buffer:DMSO		, 3630–3633
$\bigcirc$	(20:80)		
2	CH <sub>3</sub> CN:H <sub>2</sub> O:	5×10 <sup>-7</sup>	Org. Biomol.
	DMSO		Chem., <b>2012</b> , 10,
- mo <sub>2</sub>	(79:20:1)		1966
N-N C	CH <sub>3</sub> CN:H <sub>2</sub> O	$4.19 \times 10^{-7}$	Analyst, 2013, 138,
			7169–7174
uya -			

0 8 0 0 0 0 0 0 0 0 0 0 0	CH <sub>3</sub> OH/H <sub>2</sub> O	$4 \times 10^{-7}$	Org. Biomol.
	(4:1)		Chem., <b>2011</b> , 9,
$\sum_{k=1}^{n} a_{k}^{k} \neq 0$			3844
	HEPES buffer:	7×10 <sup>-8</sup>	<i>RSC Adv.</i> , <b>2013</b> , <i>3</i> ,
N O O	CH <sub>3</sub> CN (70:30)		11543–11546
	HEPES buffer	$2.5 \times 10^{-7}$	Present work
	(100 %)		
NO <sub>2</sub>			

S11. Histogram for the competitive amino acid titrations of [L] with amino acids



**Fig. S11**. Fluorescence spectra obtained for the competitive titration of L ( $5\mu$ M) with Cys in presence of different amino acids ( $200\mu$ M). (a) Histogram showing the fluorescence response of L at 550 nm band when titrated with different amino acids. (b) Visual fluorescent color change of {L+Amino acids} +Cys} with different amino acids under 365 nm UV-light.



#### S12. Fluorescence spectra for the titration of [L] with different molecular weight thiols

**Fig. S12**: Fluorescence spectra obtained for the titration of L [5  $\mu$ M,  $\lambda_{ex} = 360$  nm] with molecular weight thiols in *HEPES buffer* at pH 7.4; (a) MPA, (b) TAA, (c) Hcy, (d) DTT<sub>red</sub> and (e) GSH<sub>red</sub>.

S13. Fluorescence spectra for the titration of [L] with oxidized state of BSA, HSA and GSH



**Fig. S13**. Fluorescence spectra obtained for the titration of L (5  $\mu$ M,  $\lambda_{ex}$  = 360 nm) in HEPES buffer at pH 7.4 with oxidized state of (a) BSA, (b)HSA and (c) GSH<sub>OX</sub>





**Fig. S14**: Fluorescence spectra obtained for the titration of L (5  $\mu$ M,  $\lambda_{ex}$  = 360 nm) in HEPES buffer at pH 7.4 with reduced state of (a) BSA, (b)HSA and (c) GSH.



#### S15. Absorption spectra obtained during the titration [L] with different amino acids

Fig. S15. UV-Visible spectral traces obtained during the titration of L ( $10\mu$ M) with different amino acids in HEPES buffer at pH 7.4.

## S16. Absorption spectra for the titration of [L] with oxidized forms of GSH



Fig. S16. UV-Visible spectral traces obtained during the titration of L ( $10\mu$ M) with GSH<sub>ox</sub> in HEPES buffer at pH 7.4.



## S17. ESI-MS spectra obtained during the titration of [L] with Cys

**Fig. S17**. ESI-MS spectral titration of **L** with Cys in CD<sub>3</sub>OD-D<sub>2</sub>O (1:1): (a) [**L**] followed by 'n' equivalents of Cys, where, (a) n=0, (b) n= 1, (c) n = 2.5, (d) n = 5 and (e) n = 10.