Supporting Information for:

Colorimetric detection of biological hydrogen sulfide using fluorosurfactant functionalized gold nanorods

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Fig. S1. Etching of gold nanorods. The absorption spectra of AuNRs solution with the increase of etching time. Inset: Corresponding LSPR peak wavelength of the AuNRs during the etching. The solution was a mixture of the as–synthesized CTAB–AuNRs and 0.2% FSN, then subjected 2 hours of ultrasonic vibration at room temperature.



Fig. S2. The zeta potentials of the CTAB–AuNRs, FSN–AuNRs and FSN–AuNRs+H₂S.



Fig. S3. Effects of free FSN (a), NaCl concentration (b) and solution pH (c) on the detection of H_2S (3 μ M). (d) Time–dependent absorbance decrease of FSN–AuNRs in presence of H_2S (5 μ M) at 648 nm.

Table S1

The comparison of this work with some other reported methods for H_2S detection^a

#	Strategy	Probe	Detection limit	Media	Biological H ₂ S	Ref.
2	colorimetry	Au–Ag Core–Shell nanoparticles	50 nM	serum	yes	2
3	colorimetry	gold nanoparticles	30 nM	air	no	3
4	colorimetry	copper nanoparticles	8.1 µM	water samples	no	4
5	colorimetry	Cu@Au nanoparticles	0.3 µM	_	no	5
6	Chemiluminescence	CLSS-1	0.7±0.3	cell lysate	yes	6
7	Chemiluminescence	CHS-1	5.4 µM	vivo	yes	7
8	kinetic spectral	AuNR-Ag nanoprobes	0.1 µM	cell	yes	8
9	fluorimetry	lanthanide coordination polymer	0.3 µM	serum	yes	9
10	fluorimetry	DNS-Az (2)	1 µM	serum	yes	10
11	fluorimetry	benzopyran derivative 1-NH ₂	3.05 µM	cell	yes	11
12	fluorimetry	DNPOCy	1 µM	cell	yes	12
13	colorimetry	FSN–AuNRs	0.2 µM	serum	yes	This study

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