Supplementary Data for

Location-Dependent Sensing of Nitric Oxide and Calcium Ion at Living Rat Kidney Using an Amperometric/Potentiometric Dual

Microsensor

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Fig. S1. Linear sweep voltammetry curves for NO oxidation obtained with WE1 (25 μ m in diameter) of a NO/Ca²⁺ dual microsensor in a deaerated PBS solution (pH = 7.4). Scan rate; 20 mV s⁻¹.



Fig. S2. Dynamic potentiometric response curves to Ca^{2+} from 10⁻⁶ M to 10⁻² M obtained with two different WE2 electrodes of a dual sensor. Blue solid line: pretreated with a solution of TMCS in toluene (1:1, v/v). Black dashed line: without the pretreatment.



Fig. S3. Amperometric responses of WE1 to typical interfering agents: 50 μ M nitrite, 50 μ M AA, 50 μ M UA, 50 μ M AP, 10 μ M DA, 2 μ M H₂S, and 10 μ M H₂O₂. The current response to 0.8 μ M NO was also showed.



Fig. S4. Potentiometric response curves of WE2 to various cations to determine the selectivity coefficient (log $K_{Ca^{2+},x}^{Pot}$, X = Na⁺, K⁺, and Mg²⁺) at calcium detection limit.



Fig. S5. Amperometric and potentiometric response curves of (A) WE1 and (B) WE2 to the L-NAME additions. Four successive injections of L-NAME standard solution (marked with arrows) made the final 10 mM of L-NAME in a tested solution. A NO/Ca²⁺ dual microsensor did not respond to L-NAME.



Fig. S6. Calibration curves of a NO/Ca²⁺ dual microsensor before (full circle symbols) and after (empty circle symbols) kidney tissue experiments. (A) Amperometric responses of WE1. (B) Potentiometric responses of WE2. The obtained sensitivities are also provided. S_{NO} and S_{Ca2+} mean the sensor sensitivities to NO and Ca²⁺, respectively.