## **SUPPORTING INFORMATION**

## for

## Cu<sup>2+</sup>-mediated fluorescence switch of gold nanoclusters for the selective detection of clioquinol

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Figures



Fig. S1 The molecule structure of CQ



Fig. S2 The effect of ethanol on the fluorescence of BSA-AuNCs. Conditions: BSA-AuNCs, 0.192 mg/mL;  $Cu^{2+}$ , 6  $\mu$ M; CQ, 10  $\mu$ M; pH 7.0, Tris-HCl buffer.



**Fig. S3** The circular dichroism spectra of BSA-AuNCs, BSA-AuNCs-Cu<sup>2+</sup> and BSA-AuNCs-Cu<sup>2+</sup>-CQ. Conditions: BSA-AuNCs, 0.192 mg/mL; Cu<sup>2+</sup>, 6  $\mu$ M; CQ, 10  $\mu$ M; pH 7.0, Tris-HCl buffer.



**Fig. S4** Dependence of the fluorescence response of BSA-AuNCs on (A) pH; (B) reaction temperature; (C) incubation time; (D) concentration of NaCl. Conditions: BSA-AuNCs, 0.192 mg/mL;  $Cu^{2+}$ , 6  $\mu$ M; CQ, 10  $\mu$ M.



**Fig. S5** The reusability of AuNCs for sensing CQ. A, the first cycle; B, the second cycle; C, the third cycle; D, the fourth cycle. 1, AuNCs; 2, AuNCs+Cu<sup>2+</sup>+CQ; 3, AuNCs+Cu<sup>2+</sup>+CQ+Cu<sup>2+</sup>+CQ; 4, AuNCs+Cu<sup>2+</sup>+CQ+Cu<sup>2+</sup>+CQ+Cu<sup>2+</sup>+CQ. Conditions: BSA-AuNCs, 0.192 mg/mL; Cu<sup>2+</sup>, 6  $\mu$ M for each cycle; CQ, 10  $\mu$ M for each cycle; pH 7.0, Tris-HCl buffer.



Fig. S6 The selectivity of the fluorescent probe for CQ detection. Conditions: BSA-AuNCs, 0.192 mg/mL; Cu<sup>2+</sup>, 6  $\mu$ M; CQ, 10  $\mu$ M; glucose, sucrose, lactose, and tartaric acid, 100  $\mu$ M; starch and dextrin, 100  $\mu$ g/mL; His, 10  $\mu$ M; other metal ions, 6  $\mu$ M; other amino acids, 100  $\mu$ M; pH 7.0, Tris-HCl buffer.