

## Supporting Information for:

### **Carbon quantum dots directly generated from electrochemical oxidation of graphite electrode in alkaline alcohols and the applications for specific ferric ion detection and cell imaging**

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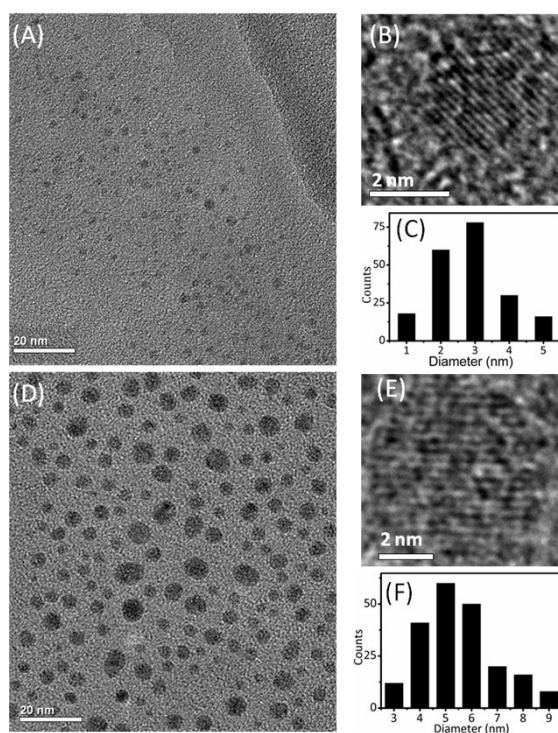
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## Experimental

Atomic absorption spectrophotometer (AAS) equipped with hollow cathode lamp (HCL) was used for iron determination. The optimum conditions for AAS were applied as follow: wavelength 248.3 nm; HCL current 10 mA; acetylene flow rate 0.5 L/min; air flow rate 4.0 L/min; slit width 0.2 nm.

After the acid pretreatment by **hydrochloric** acid, iron ( $\text{Fe}^{3+}$ ) at different concentrations (0, 10, 20, 30 and 40  $\mu\text{M}$ ) was spiked into the tap water, respectively. The spiked water samples were determined by AAS method under the above analytical conditions. Concentrations of  $\text{Fe}^{3+}$  were calculated by the standard curve method [S1].

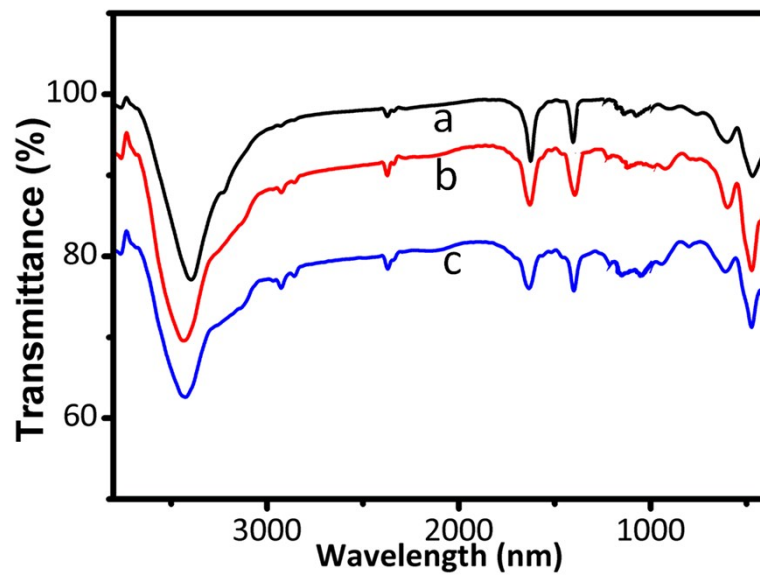
[S1] M. Yaman, G. Kaya, *Anal. Chim. Acta.*,2005, **540**, 77-81.



**Fig. S1** (A, D) TEM and (B, E) HRTEM images as well as (C, F) size distribution of the colourless CQDs obtained from the potentials of 3 V and 7 V, respectively.

**Table S1.** Comparison of the working conditions, quantum yield and characteristics of different EC methods for CQDs generation.

Working electrode	Electrolyte	Voltage	Time	Quantum yield	Advantages and disadvantages	References
Graphite electrode	NaOH/ ethanol/H <sub>2</sub> O	5 V	3 h	10.8 %	Green, low cost, good reproducibility, but it needed relatively longer time.	this work
3D Graphene	Ionic liquid (IL) BMIMPF <sub>6</sub> in acetonitrile	5 V	100 s	/	Shorter time was required for synthesis, but high-cost carbon source was needed.	48
Carbon fibers	acetonitrile containing 0.1 M tetrabutylammonium perchlorate	0.5-2.5 V	2 h	/	Size-selective preparation of C-dots can be achieved only by adjusting the applied potentials, but the used electrolyte is not environment-friendly.	2
Graphite rods	IL [apmim][BF <sub>4</sub> ] and water	12 V	4 h	11.3 %	IL functionalized carbon nanodots were synthesized, but the IL is relatively high-cost	38
Graphite rods	Ultrapure water	15-60 V	120 h	/	Facile, additive-free, but it needed longer time and higher potentials.	7
Carbon paste electrodes with different compositions	0.1M NaH <sub>2</sub> PO <sub>4</sub> aqueous solution	9 V	6 h	/	Shifting and non-shifting fluorescence emissions were observed by changing the compositions of the parent Electrodes, but relatively higher voltage and longer time were needed	33



**Fig. S2** The FTIR spectra for CQDs upon addition with ions of (a)  $\text{Cu}^{2+}$ , (b)  $\text{Cd}^{2+}$  and (c)  $\text{Ni}^{2+}$ .