Electronic Supplementary Material (ESI) for Analytical Methods. This journal is © The Royal Society of Chemistry 2015

# Characterization of anthocyanins in wild *Lycium ruthenicum* Murray by HPLC-DAD/QTOF-MS/MS

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# **Supplementary material**

## ■ Fragmentation pathway of pure anthocyanins

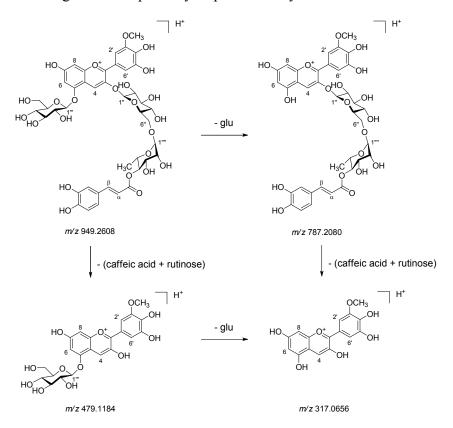


Fig. S1. Proposed fragmentation pathway of A1

Fig. S2. Proposed fragmentation pathway of A5

#### ■ Preparation of anthocyanin from *Lycium ruthenicum* Murray

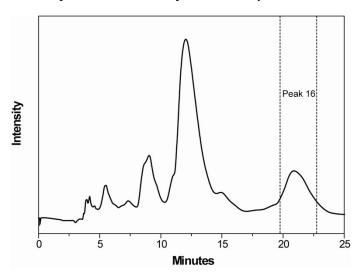


Fig. S3. Preparative chromatogram of the peak 16 from Fr. 12

#### ■ Structure identification

Peak 16 in the extract with high purity was obtained from fraction 12, and identified by mass spectrum,  $^{1}$ H NMR and 2D NMR. It was identified as M alvidin 3-O-[6-O-(4-O-(4-O-(4-O-trans-(β-D-glucopyranoside)-p-coumaroyl)- $\alpha$ -L-rhamno pyranosyl)- $\beta$ -D-glucopyranoside]-5-O-[ $\beta$ -D-glucopyranoside], was isolated from L. ruthenicum for the first time. The specific structural identification results were as follows:

**Peak 16**, dark purple powder, [M+H]<sup>+</sup>: 947.2798, calculated for C<sub>44</sub>H<sub>53</sub>O<sub>23</sub>, 947.2798 with error 2.14 ppm; <sup>1</sup>H NMR see Table S1. Compared with the literature [1], **Peak 16** was identified as Malvidin 3-*O*-[6-*O*-(4-*O*-(4-*O*-trans-(β-D-glucopyranoside)-*p*-coumaroyl)-α-L-rhamnopyranosyl)-β-D-glucopyranoside]-5-*O*-[β-D-glucopyranoside]. The chemical structure was showed in Figure S1.

Table S1  $^{1}$ H NMR data for prepared anthocyanin in CD<sub>3</sub>OD/TFA-d (95:5, v/v)

Н	Peak 16
Anthocyanidin	
4-H	9.05 s
6-H	7.07 s
8-H	7.16 s
2'-H	8.06 s
5'-Н	
6'-H	8.06 s
3'-OCH <sub>3</sub>	4.04 s
5'-OCH <sub>3</sub>	4.04 s
3-O-Glucopyranoside	
1"	5.53 d (7.7)
2"	3.84
3"	3.75
4"	3.63
5''	3.71
6a	3.98
6b	4.05
5-O-Glucopyranoside	
1'''	5.22 d (7.8)
2***	3.82
3'''	3.73
4***	3.61
5'''	3.72
6a	3.85
6b	3.86
6''-O-Rhamnopyranosyl	
1''''	4.74
2****	3.58
3''''	3.56
4''''	4.93 d (9.7)
5''''	3.50
-СН3	1.01 d (6.2)
Hydroxycinnamic acid	
2	7.45 d (8.5)
3	6.83 d (8.4)
5	6.83 d (8.4)
6	7.45 d (8.5)
$\alpha$	6.29 d (15.9)
β	7.60 d (15.9)

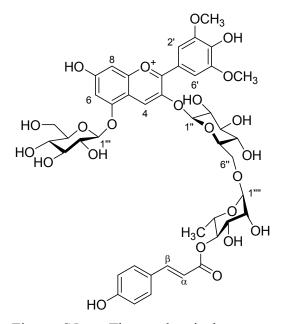


Fig. S5. The chemical structure of the prepared anthocyanin

## Reference

[1] T. Ando, N. Saito, F. Tatsuzawa, T. Kakefuda, K. Yamakage, E. Ohtani, M. Koshi-ishi, Y. Matsusake, H. Kokubun, H. Watanabe, T. Tsukamoto, Y. Ueda, G. Hashimoto, E. Marchesi, K. Asakura, R. Hara, H. Seki, Biochem. Syst. Ecol. 27 (1999) 623.