Supplementary Material for

## Colorimetric detection of hypochlorite in tap water based on

## the oxidation of 3,3',5,5'-tetramethylbenzidine

Yongming Guo\*, Qinge Ma, Fengpu Cao, Qiang Zhao, Xuewei Ji

College of Chemistry and Pharmaceutical Engineering, Nanyang Normal University, Nanyang, 473061, China

\*Corresponding author: Yongming Guo, chinahenangm@163.com



Fig. S1 The plot of  $A_{655}$  of the TMB solution at different pH values (the concentration of TMB is 1 mM, the concentration of ClO<sup>-</sup> is 20  $\mu$ M).



Fig. S2 The plot of  $A_{655}$  of the different concentrations of TMB solution (the concentration of ClO<sup>-</sup> is 20  $\mu$ M, 50 mM pH 4.0 HAc-NaAc buffer).



**Fig. S3** The plot of  $A_{655}$  of the TMB solution in the absence and presence of ClO<sup>-</sup> at different time (the concentration of TMB is 1 mM, the concentration of ClO<sup>-</sup> is 20  $\mu$ M, 50 mM pH 4.0 HAc-NaAc buffer).



**Figure S4.** (A) UV-vis absorption spectra of TMB, TMB with Fe<sup>3+</sup> (200  $\mu$ M) and TMB with Fe<sup>3+</sup> (200  $\mu$ M) and sodium pyrophosphate (1 mM). (B) The plot of A<sub>655</sub> of the TMB solution in the presence of different concentrations of sodium pyrophosphate (the concentration of TMB is 1 mM, 50 mM pH 4.0 HAc-NaAc buffer).



**Fig. S5** UV-vis absorption spectra of TMB and TMB with tap water (the concentration of TMB is 1 mM, 50 mM pH 4.0 HAc-NaAc buffer).