

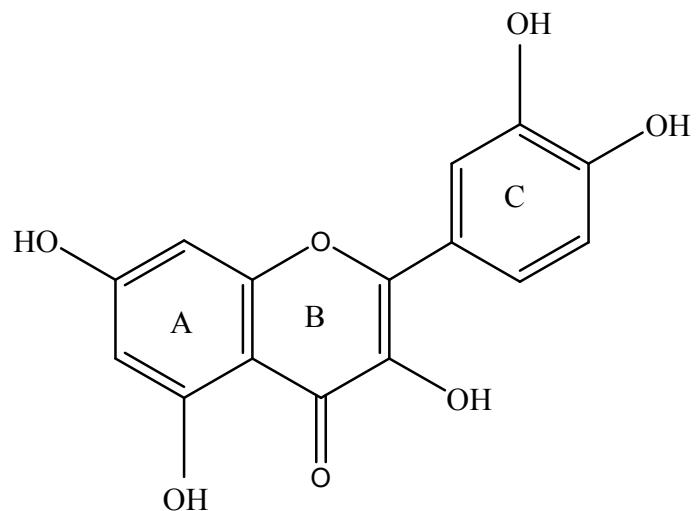
**Electrocatalytic oxidation of N-acetyl-L-cysteine at quercetin multiwall carbon nanotubes modified GCE: Application for simultaneous determination of ascorbic acid, L-DOPA, N-acetyl-L-cysteine, acetaminophen and tryptophan**

Hamid R. Zare,\* Marjan Haji-Dehabadi, Zahra Shekari

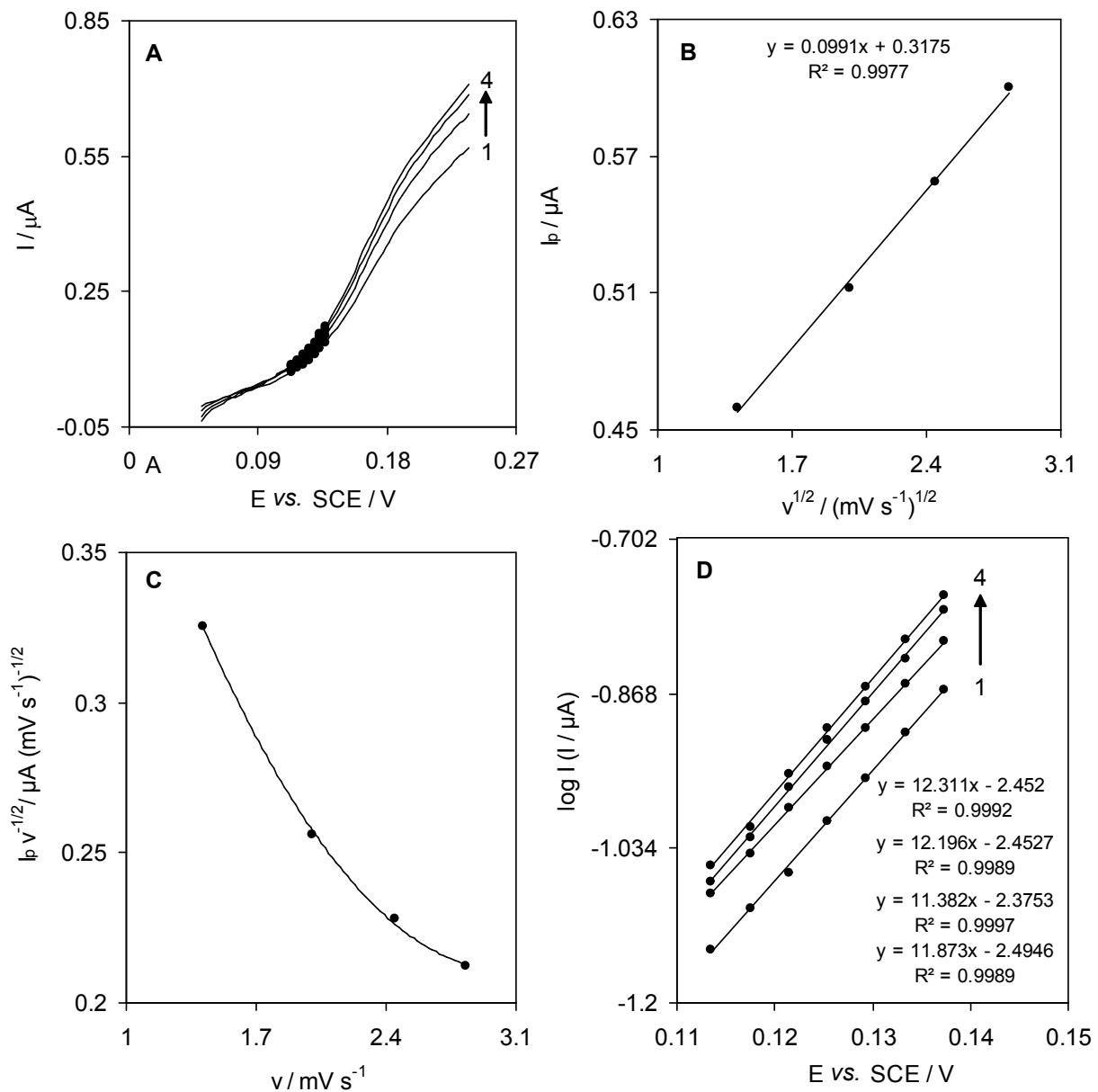
*Department of Chemistry, Yazd University, Yazd, 89195-741, Iran*

*Fax number: +98 353 8210991    Telephon number: +98 353 1232669*

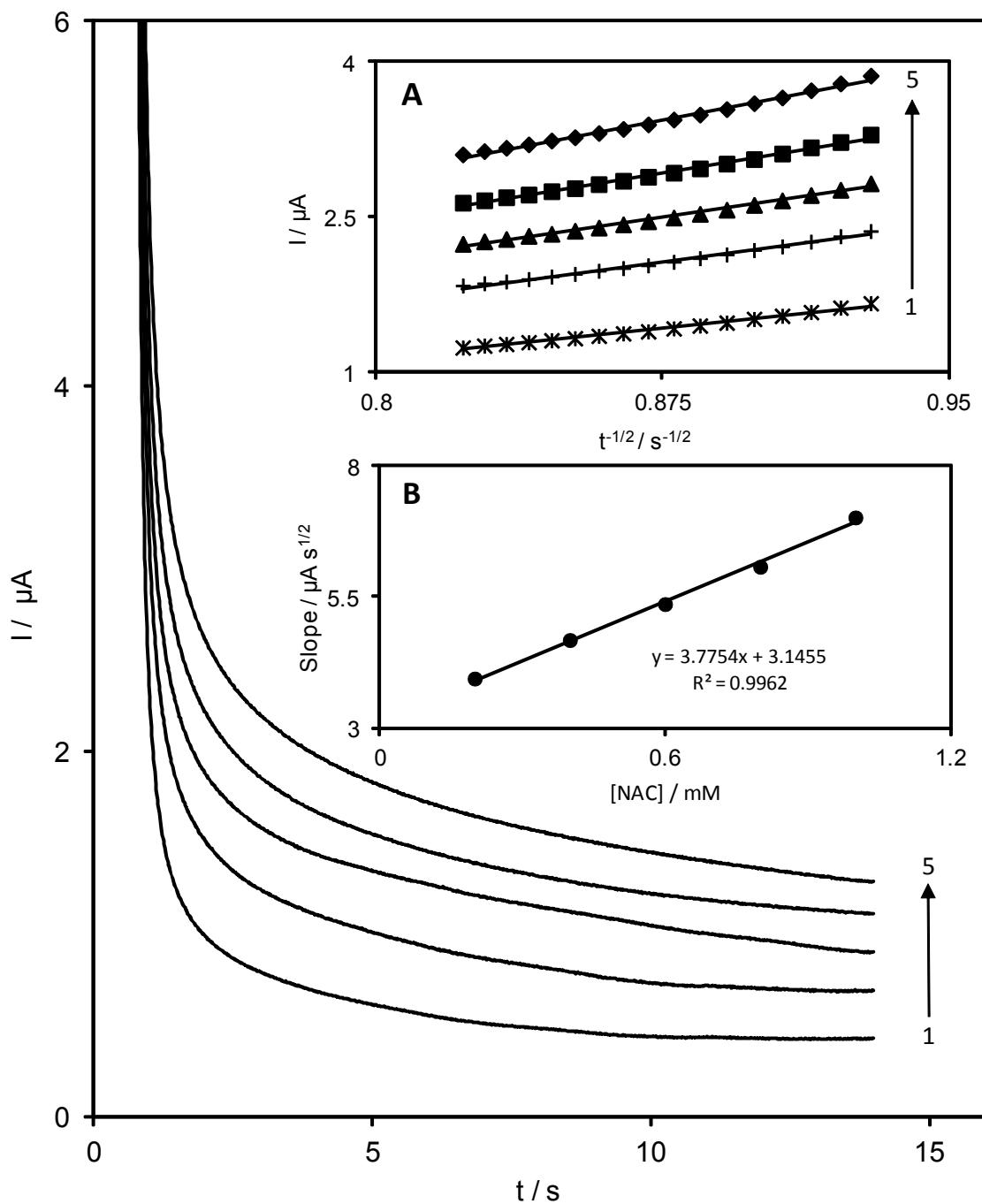
*E-mail address: [hrzare@yazd.ac.ir](mailto:hrzare@yazd.ac.ir)*



**Scheme S1.** The molecular structure of quercetin.



**Fig. S1.** (A) Linear sweep voltammograms of Q-MWCNT-GCE in a 0.10 M phosphate buffer solution (pH 7.0) containing 0.20 mM NAC at different potential scan rates. Numbers of 1–4 correspond to 2, 4, 6 and 8  $\text{mV s}^{-1}$ , respectively. (B) Variation of the electrocatalytic peak current vs. the square root of the potential scan rate. (C) Variation of the sweep rate normalized peak current ( $I_p v^{-1/2}$ ) vs. the potential scan rate. (D) Tafel plots drawn from the linear sweep voltammograms of (A). Equations from bottom to top correspond to plots of 1–4, respectively.



**Fig. S2.** Chronoamperometric responses of Q-MWCNT-GCE in a 0.10 M phosphate buffer solution (pH 7.0) at a potential step 250 mV for different concentrations of NAC. Numbers of 1–5 correspond to 0.2–1.0 mM NAC. Insets: (A) plots of  $I$  vs.  $t^{-1/2}$  obtained from chronoamperograms. (B) Plot of the slope of straight lines against the NAC concentrations.

**Table S1.** Comparison of the analytical parameters of several modified electrodes for NAC electrocatalytic determination

Modified electrode <sup>a</sup>	Method	Linear range ( $\mu\text{M}$ )	Detection limit( $\mu\text{M}$ )	References
AC/RuON/GCE	DPV	0.3–14.0 14.0–1000.0	2.84	[1]
PB/Pd-Al	Amperometry	2.0–40.0	0.54	[15]
CuHCF/CPE	DPV	120.0–830.0	63.0	[40]
Catechol/CPE	CV	80.0–1000.0	60.0	[41]
Nq/NG/PDC/GCE	DPV	4.0–130.0	0.80	[42]
Edge plane pyrolytic graphite electrode	CV	5.0–110.0	1.70	[43]
Fe(II)NP/CPE	Amperometry	960.0–14.000	150.0	[44]
Fe <sub>2</sub> O <sub>3</sub> @CoHCF/CPE	Amperometry	20.0–432.0	0.0209	[45]
Cobalt salophen complexes/CPE	DPV	0.10–100.0	0.05	[46]
N-DHPB/MWCNT/CPE	DPV	0.50–200.0	0.20	[47]
CoNPs–CNFs/GCE	Amperometry	0.12–13.3	25.0	[48]
Q–MWCNT–GCE	DPV	1.1–50.0 50.0–1000.0	0.44	This work

<sup>a</sup>AC–RuON–GCE: Acetaminophene at the surface of a ruthenium oxide nanoparticles–glassy carbon electrode; PB/Pd–Al: Palladized aluminum electrode modified by prussian blue film; CuHCF/CPE: Carbon paste electrode modified with copper (II) hexacyanoferrate (III); Catechol/CPE: Carbon paste electrode modified with catechol; Nq/NG/PDC/GCE: Nanocomposite containing naphthoquinone and nanogold supported on poly (2,6-pyridinedicarboxylic acid) film modified glassy carbon electrode; Fe(II)NP/CPE: Iron nitroprusside modified graphite paste electrode; Cobalt salophen complexes/CPE: Carbon nanotube–paste electrode modified with salophen complexes of cobalt(III) perchlorate; Fe<sub>2</sub>O<sub>3</sub>@CoHCF/CPE: Nanoparticles of iron (III) oxide core–cobalt hexacyanoferrate shell (Fe<sub>2</sub>O<sub>3</sub>@CoHCF)–modified carbon paste electrode; N–DHPB/MWCNT/CPE: N–(3,4-dihydroxyphenethyl)–3,5–dinitrobenzamide modified multiwall carbon nanotubes paste electrode; CoNPs–CNFs/GCE: Carbon nanofibers (CNFs) decorated with cobalt nanoparticles (CoNPs)–glassy carbon electrode; Q–MWCNT–GCE: Quercetin modified multiwall carbon nanotubes–glassy carbon electrode.