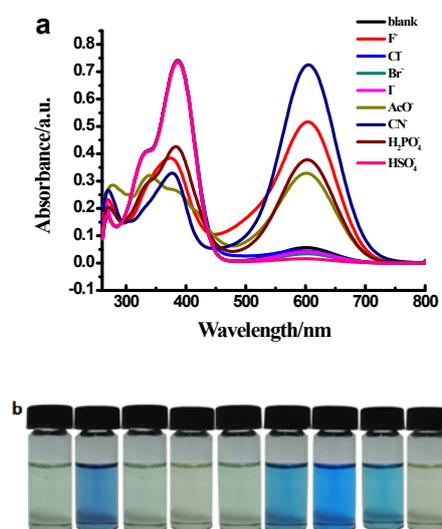
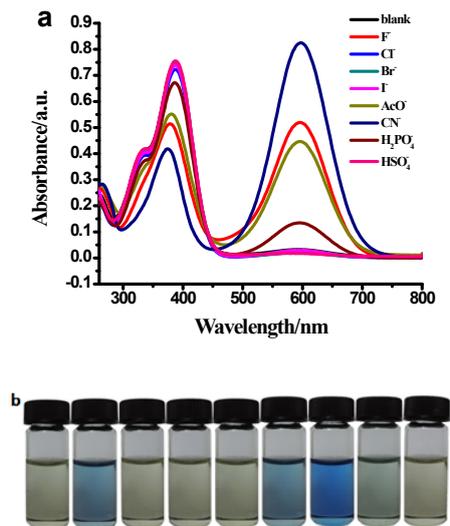


Electronic Supporting Information

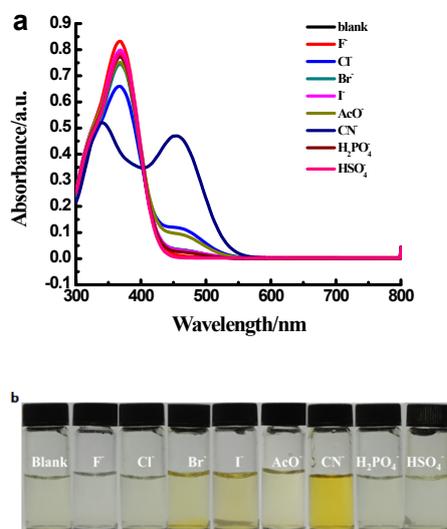
1. Anionic Response of 2-HPEAPB in DMF, DMSO, H₂O, and methanol



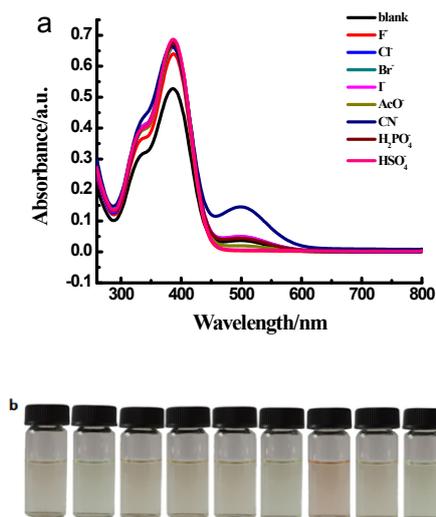
S.Fig. 1 (a) UV-Vis spectra of 2-HPEAP (3.0×10^{-5} mol L⁻¹) in DMF upon the addition of 5 equiv of F⁻, Cl⁻, Br⁻, I⁻, AcO⁻, CN⁻, H₂PO₄⁻, and HSO₄⁻. (b) Color changes observed in 2-HPEAPB (3.0×10^{-5} mol L⁻¹) upon the addition of various anions as tetra-butylammonium salts (1.5×10^{-4} mol L⁻¹).



S.Fig. 2 (a) UV-Vis spectra of 2-HPEAPB ($3.0 \times 10^{-5} \text{ mol L}^{-1}$) in DMSO upon the addition of 5 equiv of F⁻, Cl⁻, Br⁻, I⁻, AcO⁻, CN⁻, H₂PO₄⁻, and HSO₄⁻. (b) Color changes observed in 2-HPEAPB ($3.0 \times 10^{-5} \text{ mol L}^{-1}$) upon the addition of various anions as tetra-butylammonium salts ($1.5 \times 10^{-4} \text{ mol L}^{-1}$).



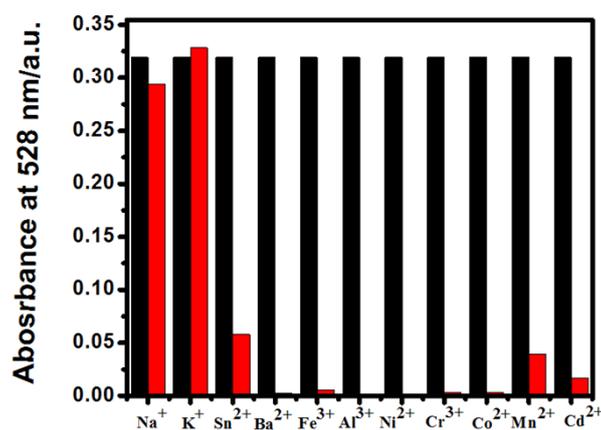
S.Fig. 3 (a) UV-Vis spectra of 2-HPEAPB ($3.0 \times 10^{-5} \text{ mol L}^{-1}$) in H₂O upon the addition of 5 equiv of F⁻, Cl⁻, Br⁻, I⁻, AcO⁻, CN⁻, H₂PO₄⁻, and HSO₄⁻. (b) Color changes observed in 2-HPEAPB ($3.0 \times 10^{-5} \text{ mol L}^{-1}$) upon the addition of various anions as tetra-butylammonium salts ($1.5 \times 10^{-4} \text{ mol L}^{-1}$).



S.Fig. 4 (a) UV–Vis spectra of 2-HPEAPB ($3.0 \times 10^{-5} \text{ mol L}^{-1}$) in methanol upon the addition of 5 equiv of F^- , Cl^- , Br^- , I^- , AcO^- , CN^- , H_2PO_4^- , and HSO_4^- . (b) Color changes observed in 2-HPEAPB ($3.0 \times 10^{-5} \text{ mol L}^{-1}$) upon the addition of various anions as tetra-butylammonium salts ($1.5 \times 10^{-4} \text{ mol L}^{-1}$).

2. The influence of Cations

The anti-jamming ability of 2-HPEAPB to cations including Na^+ , K^+ , Sn^{2+} , Ba^{2+} , Fe^{3+} , Al^{3+} , Ni^{2+} , Cr^{3+} , Co^{2+} , Mn^{2+} , Cd^{2+} was evaluated using UV-Vis competition experiments (S.Fig. 5). These cations (such as K^+ , Na^+) which cannot form complexes or precipitation with CN^- produce slight influence, however, these cations (such as Sn^{2+} , Ba^{2+} , Fe^{3+} , Al^{3+} , Ni^{2+} , Cr^{3+} , Co^{2+} , Mn^{2+} , and Cd^{2+}) which could form precipitation or complexes with CN^- produce great influence.



S.Fig. 5 The UV-Vis intensity of 2-HPEAPB ($3.0 \times 10^{-5} \text{ mol L}^{-1}$) in acetonitrile-water (95:5, v/v) at 528 nm to various metal ions. The black bars represent the UV-Vis intensity of 2-HPEAPB in the presence of 1 equiv of CN^- . The red bars represent the UV-Vis intensity of 2-HPEAPB in the presence of 1 equiv upon subsequent addition of 1 equiv various cations (Na^+ , K^+ , Sn^{2+} , Ba^{2+} , Fe^{3+} , Al^{3+} , Ni^{2+} , Cr^{3+} , Co^{2+} , Mn^{2+} , Cd^{2+}).