

Electronic Supplementary Information

Highly sensitive and selective coumarin probe for hydrogen sulfide imaging in living cells

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Abstract. This Supplementary data include all of the additional information as noted in the manuscript.

S1. ^1H , ^{13}C NMR and Mass Spectra of DNPOCA

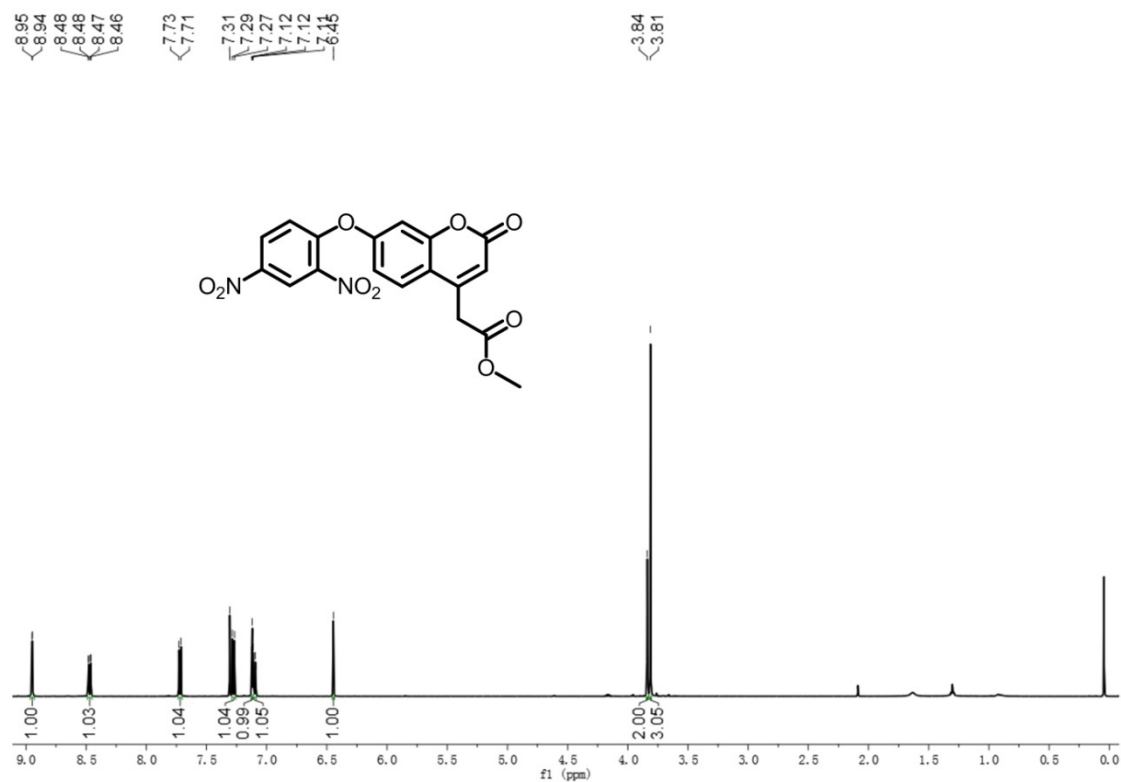


Figure S1. ^1H NMR spectrum of DNPOCA in CDCl_3 .

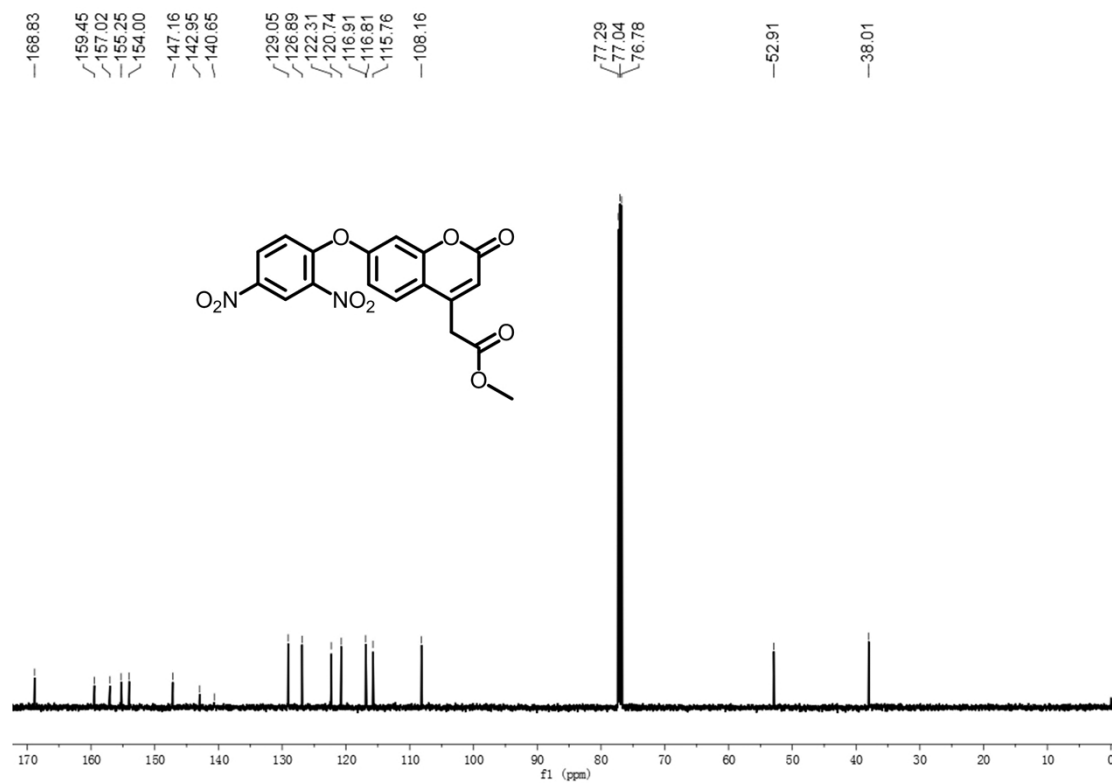


Figure S2. ^{13}C NMR spectrum of DNPOCA in CDCl_3 .

[WJY]-CS-P-1 #75-80 RT: 0.66-0.72 AV: 6 NL: 3.46E2
T: ITMS - c ESI Full ms [50.00-1000.00]

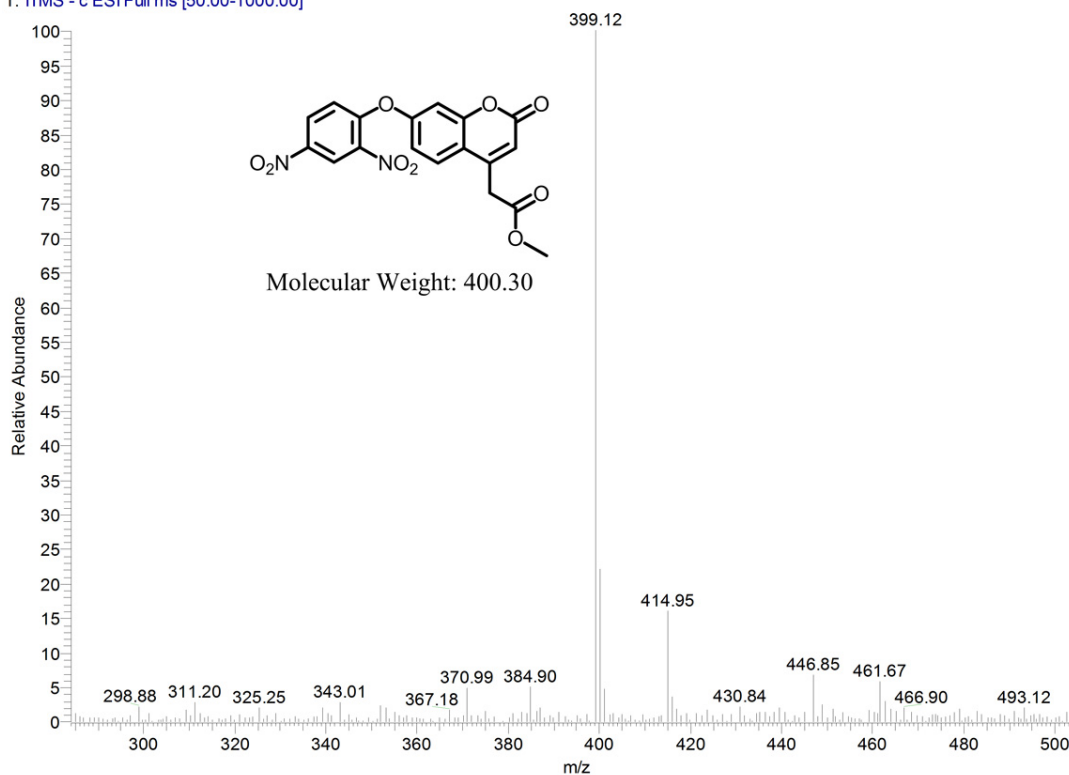


Figure S3. ESI-MS of DNPOCA

S2. The acquisition and analysis of thiolysis product of DNPOCA after incubated with Na₂S

DNPOCA (20 mg, 0.05 mmol) was dissolved in CH₃CN (10 mL), followed by the addition of Na₂S•9H₂O (199.35 mg, 0.83 mmol). The resultant mixture was stirred for 5 hours at room temperature. Subsequently, the solvent was evaporated. The fluorescent product was purified by column chromatography and analyzed by mass spectrometry.

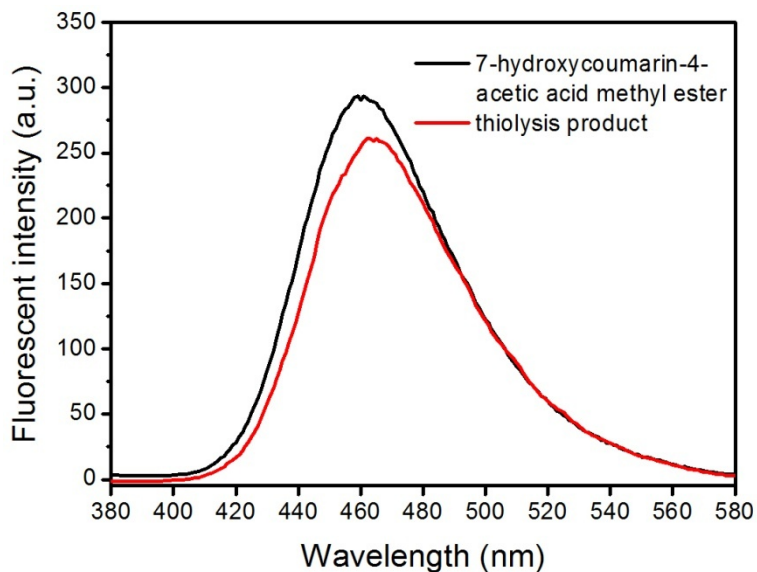


Figure S4. Fluorescence spectra of 7-hydroxycoumarin-4-acetic acid methyl ester and thiolysis product in aqueous solution (20 mM, pH = 7.4, CH₃CN/PBS = 1:9, 3 mM CTAB) at room temperature.

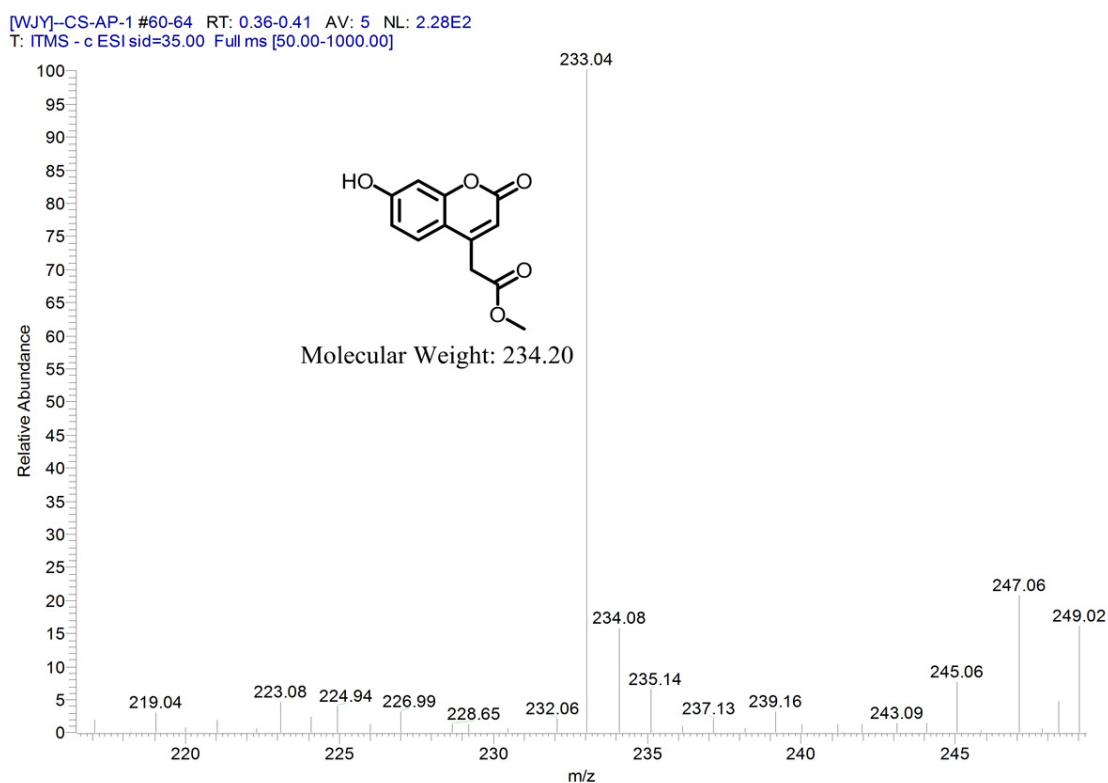


Figure S5. ESI-MS confirmed the formation of 7-hydroxycoumarin-4-acetic acid methyl ester in the reaction of DNPOCA with Na₂S.

S3. Study on the reaction kinetics of DNPOCA

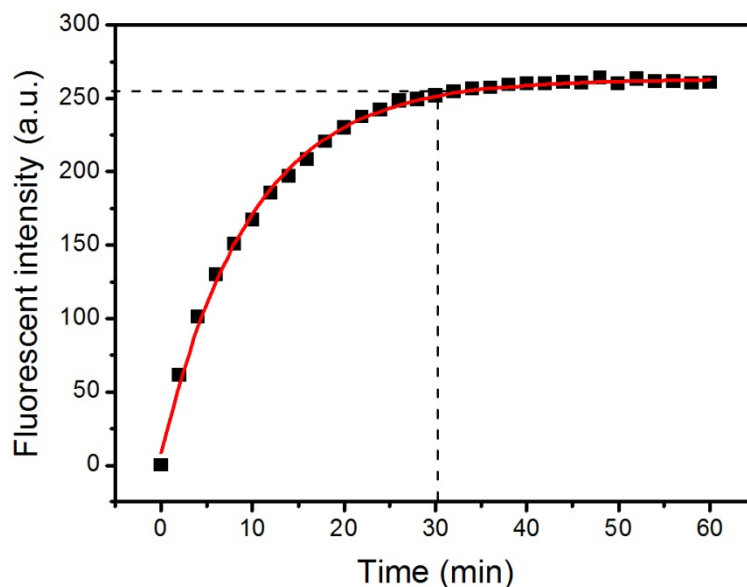


Figure S6. Time-dependent fluorescence intensity changes at 463 nm of the probe DNPOCA (5.0 μM) in the absence of Na_2S (100 μM) in aqueous solution (20 mM, pH = 7.4, $\text{CH}_3\text{CN}/\text{PBS}$ = 1:9, 3 mM CTAB) at room temperature.

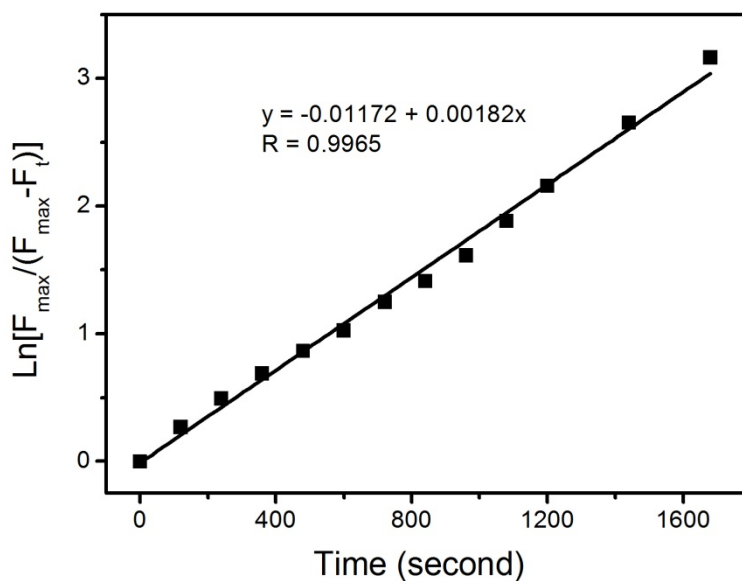


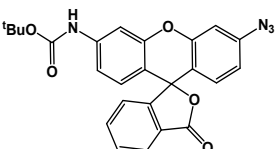
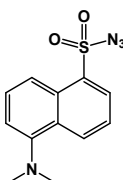
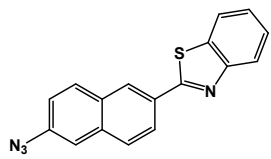
Figure S7. The pseudo-first-order kinetic plot of the reaction of the probe DNPOCA (5.0 μM) with Na_2S (100 μM) in aqueous solution (20 mM, pH = 7.4, $\text{CH}_3\text{CN}/\text{PBS}$ = 1:9, 3 mM CTAB) at room temperature.

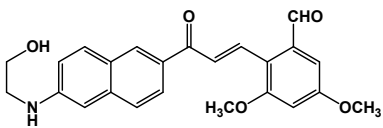
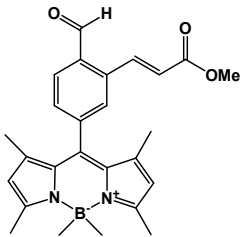
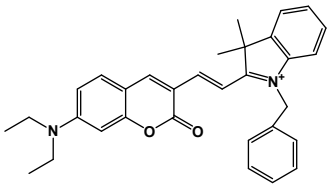
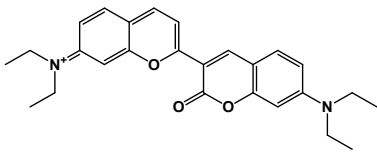
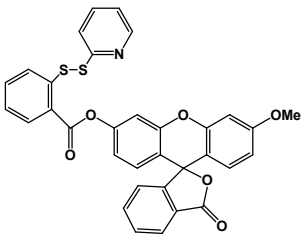
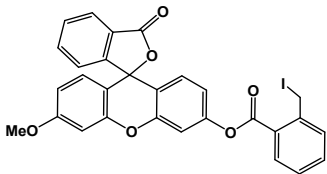
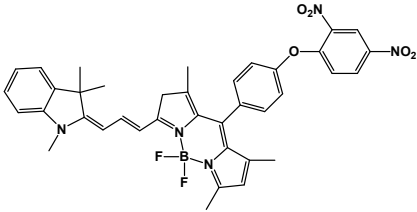
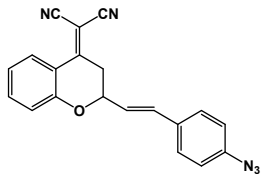
S4. The calculation method for the detection limit.

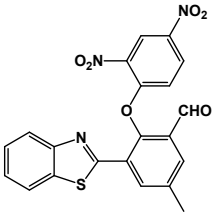
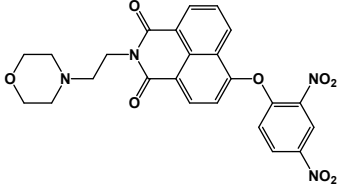
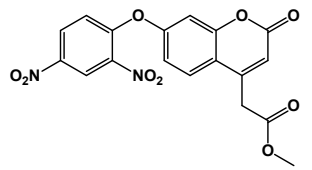
The limit of detection (LOD) for DNPOCA was calculated based on the fluorescence titration. To determine the S/N ratio, the fluorescence emission intensity of DNPOCA without Na₂S was measured by 10 times and the standard deviation of blank measurements was determined. Then, the solution was treated with Na₂S from 0 to 100 μM. A linear regression curve was then achieved according to the fluorescence emission intensity in the range of Na₂S from 0 to 20 μM. The detection limit was calculated with the equation (Eq. 1) at S/N=3, in which σ is the standard deviation of the background and s is the sensitivity.

$$LOD = 3 \times \frac{\sigma}{S} \quad (\text{Eq. 1})$$

Table S1. Comparison of fluorescent probes for hydrogen sulfide

Probe	$\lambda_{\text{ex}}/\lambda_{\text{em}}$ (nm)	Detection medium	Detection limit	Ref.
	490/525	HEPES (pH 7.4), 0.2% DMF	5-10 μM	1
	340/535	PBS (pH 7.4), 0.5% Tween-20	1 μM	2
	360/480	HEPES (pH 7.4), 80% DMF	0.1 μM	3

	360/480	HEPES (pH 7.4), 1% CH ₃ CN	50 nM	4
	465/510	PBS (pH7.0), 1% DMSO	5 μM	5
	475/510 475/652	PBS (pH7.4), 2% DMSO	1 μM	6
	450/485 475/690	PBS (pH7.4), 30% CH ₃ CN	0.14 μM	7
	465/515	PBS (pH 7.4), 10% CH ₃ CN	1-10 μM	8
	455/517	PBS (pH 7.4), 20% CTAB	0.1 μM	9
	650/708	PBS (pH 7.0), 20% ethanol, 3mM CTAB	0.1 μM	10
	520/670	PBS (pH 7.4), 50% DMSO	3.05 μM	11

	440/545	PBS (pH 7.4), 25% CH ₃ CN	48 nM	12
	450/555	PBS (pH 7.4), 10% CH ₃ CN	0.48 μM	13
	365/463	PBS (pH 7.4), 10% CH ₃ CN 3mM CTAB	49.7 nM	This work

S5. Study on the sensitivity of DNPOCA by UV-visible absorption spectra.

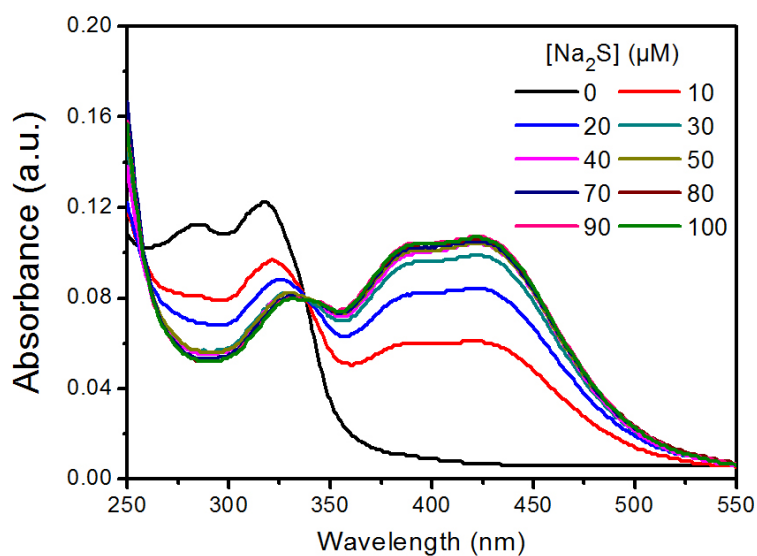


Figure S8. UV-vis absorption spectra changes of the probe DNPOCA (5.0 μM) in the presence of Na₂S (0-20 equiv.) in aqueous solution (20 mM, pH = 7.4, CH₃CN/PBS = 1:9, 3 mM CTAB) at room temperature.



Figure S9. Color changes of the probe DNPOCA (5.0 μM) with different concentrations of Na_2S for 40 min at room temperature (From 1 to 8: 0, 10, 30, 50, 70, 80, 90, 100 μM Na_2S , respectively).

S6. Study on the selectivity of DNPOCA for H_2S

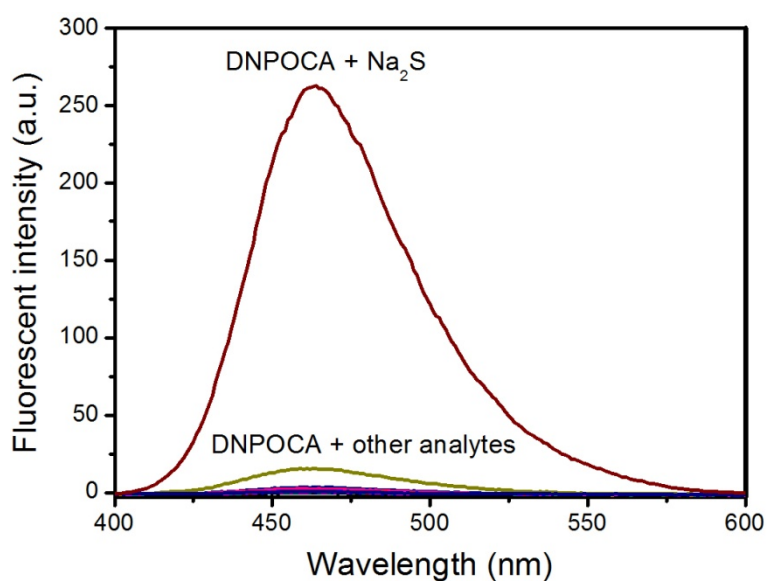


Figure S10. Fluorescent spectra changes of the probe DNPOCA (5.0 μM) in the presence of various species (20 equiv.) in aqueous solution (20 mM, pH = 7.4, $\text{CH}_3\text{CN}/\text{PBS}$ = 1:9, 3 mM CTAB) at room temperature.

S7. Effect of pH for H₂S detection

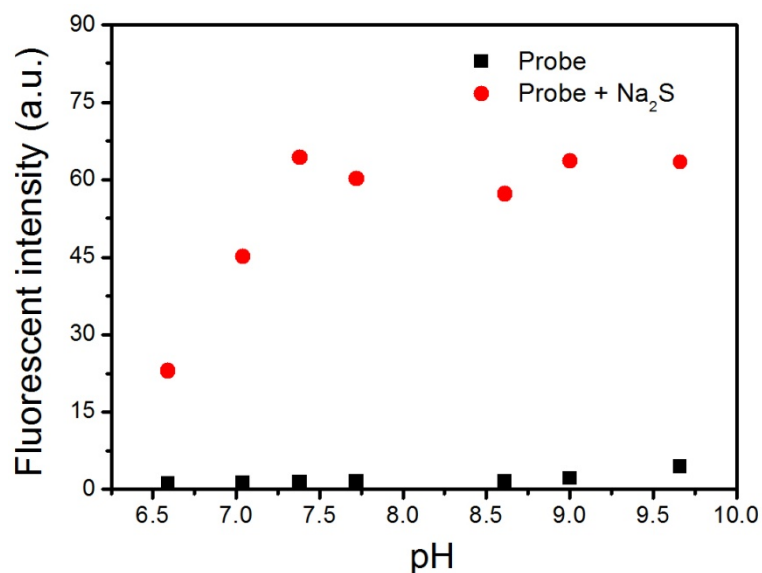


Figure S11. Fluorescence intensity changes of the probe DNPOCA (5 μM) at different pH values in the absence (■) or presence (●) of Na₂S (100 μM) in aqueous solution (20 mM, CH₃CN/PBS = 1:9, 3 mM CTAB) at room temperature. $\lambda_{\text{ex}} = 365$ nm, $d_{\text{ex}} = 1.5$, $d_{\text{em}} = 3$ nm.

S8. Cytotoxicity assay

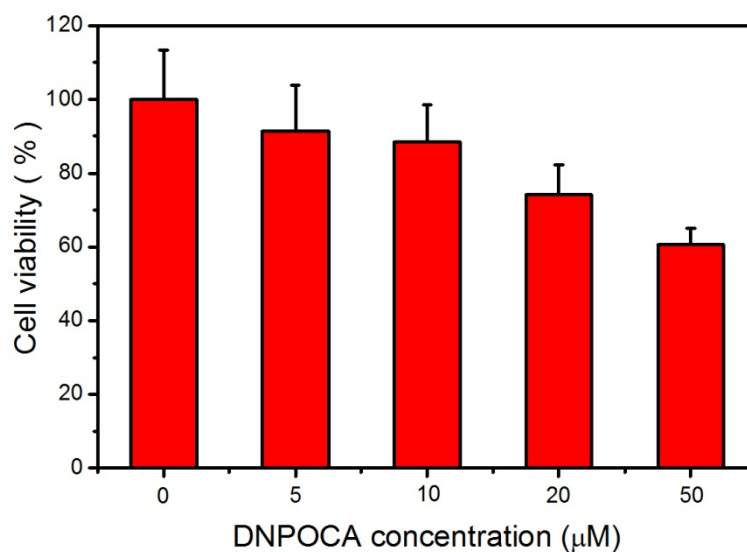


Figure S12. Cytotoxicity assay of the probe DNPOCA for HepG2 cells after 24-h culture.

References

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