

## Supplementary documents

### A highly sensitive reversible fluorescent-colorimetric azino bis-Schiff base sensor for rapid detection of $Pb^{2+}$ in aqueous media

Anupam Ghorai, Jahangir Mondal, Rajat Saha, Sumantra Bhattacharya and Goutam K Patra\*

Department of Chemistry, Guru Ghasidas Vishwavidyalaya, Bilaspur (C.G)

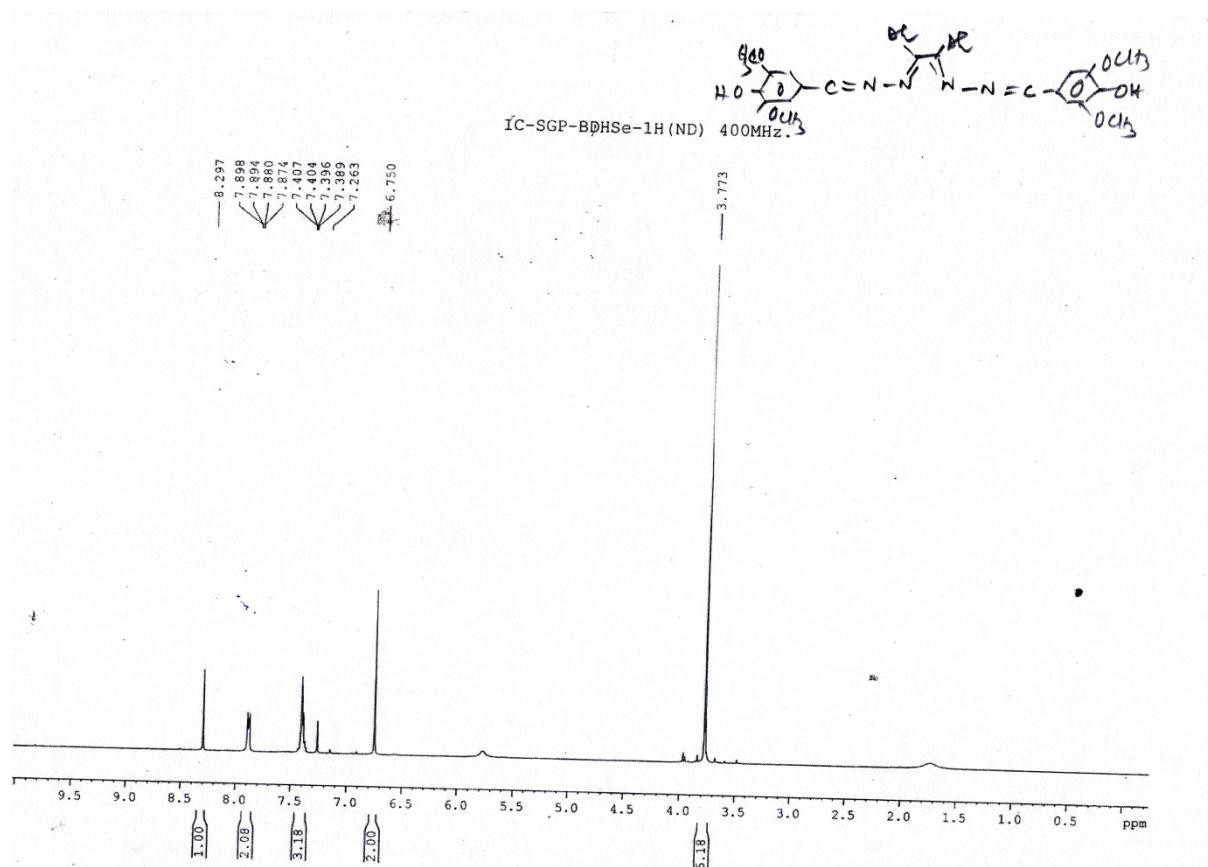
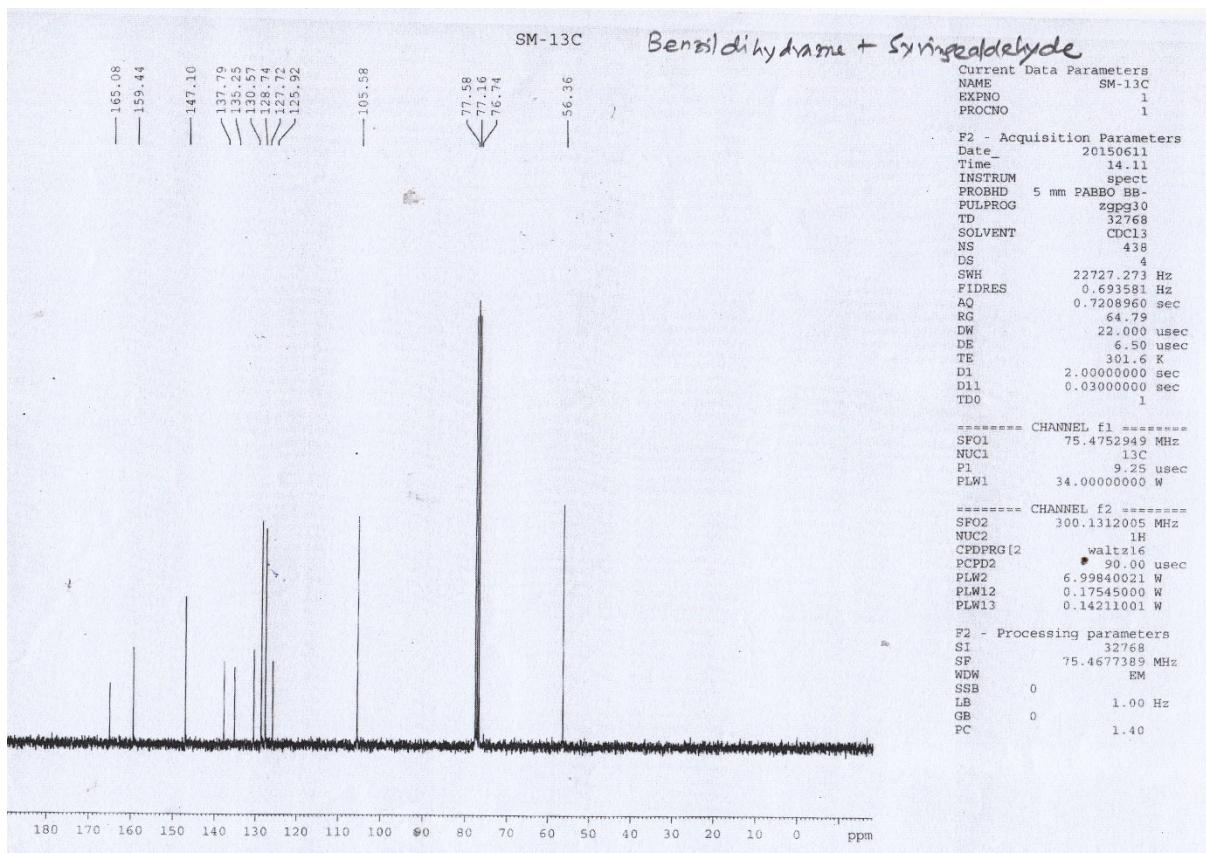
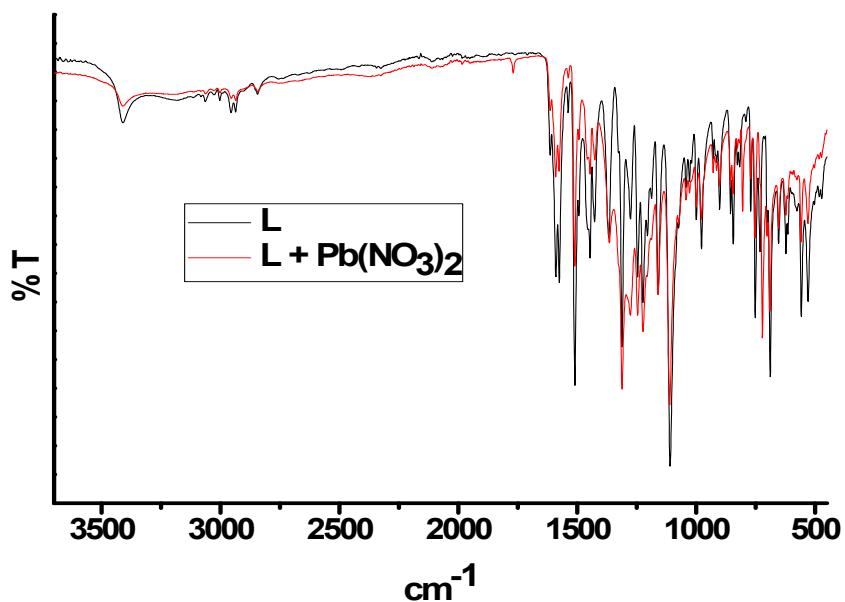


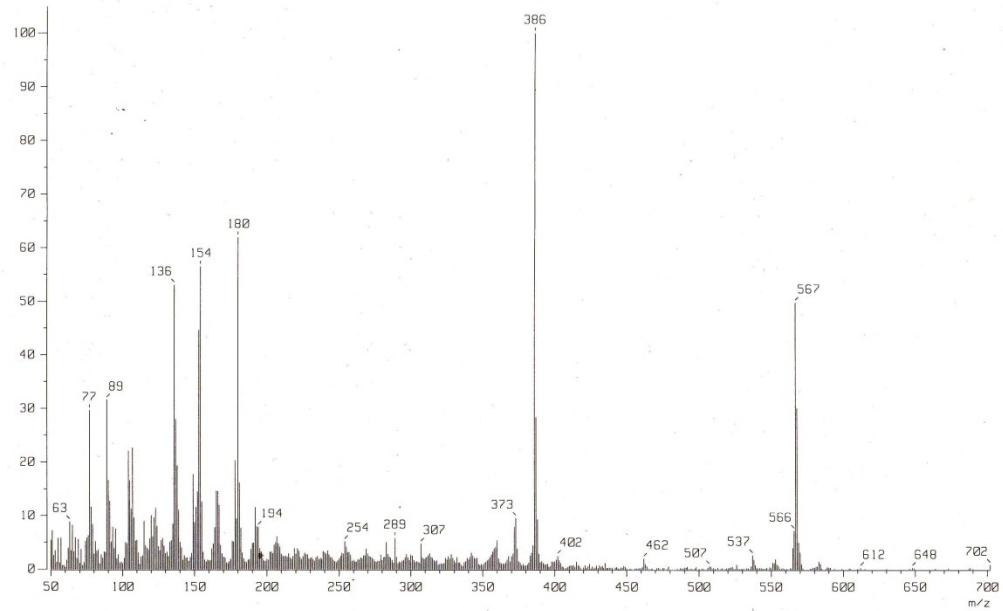
Fig. S1  $^1\text{H}$  NMR spectra of L in  $\text{CDCl}_3$ .



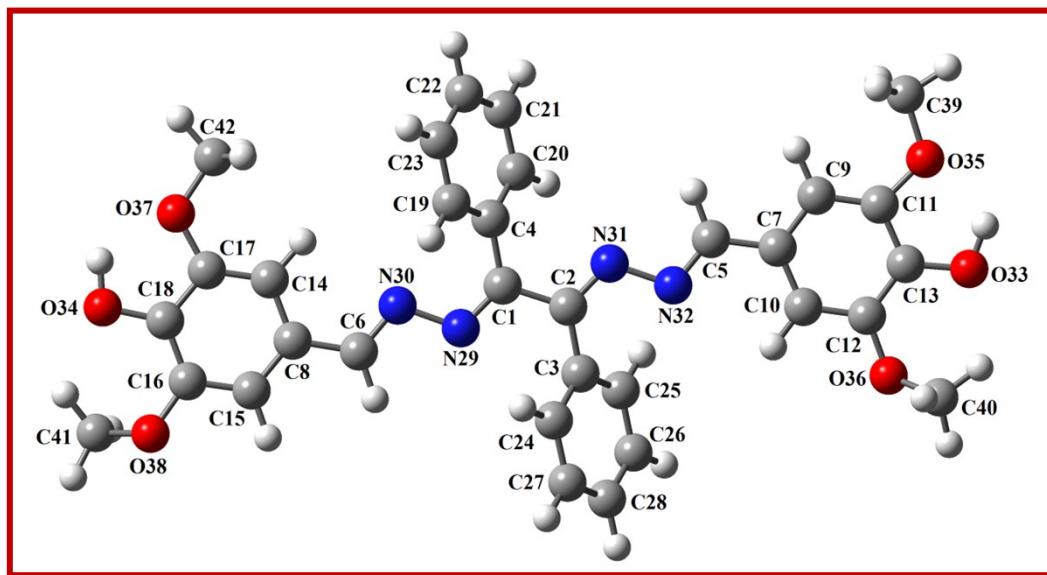
**Fig. S2**  $^{13}\text{C}$  NMR Spectra of **L** in  $\text{CDCl}_3$ .



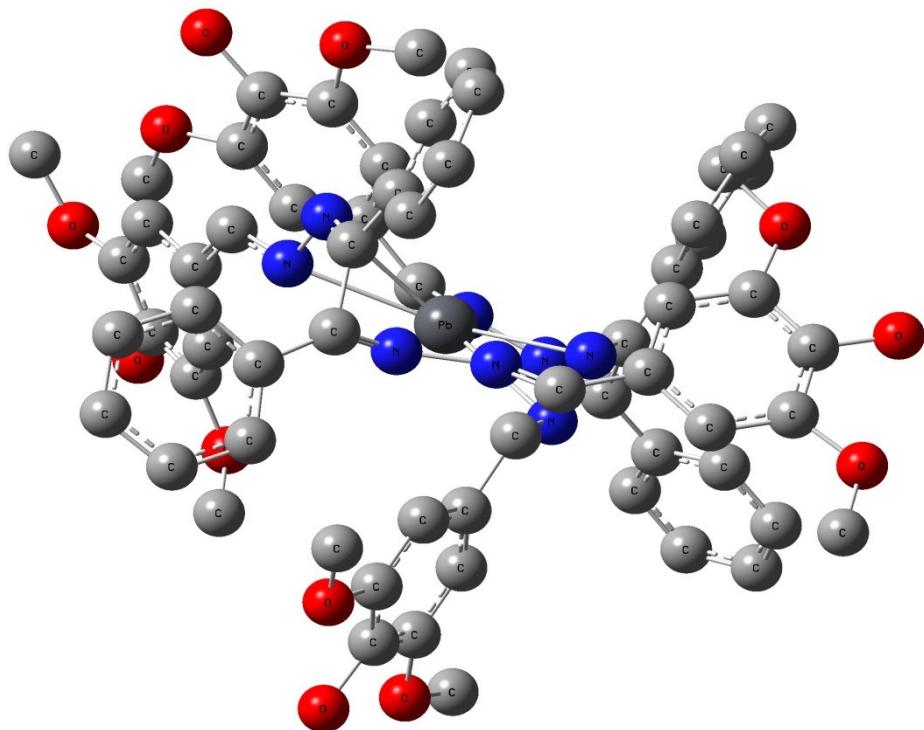
**Fig. S3** FTIR spectra of **L**.



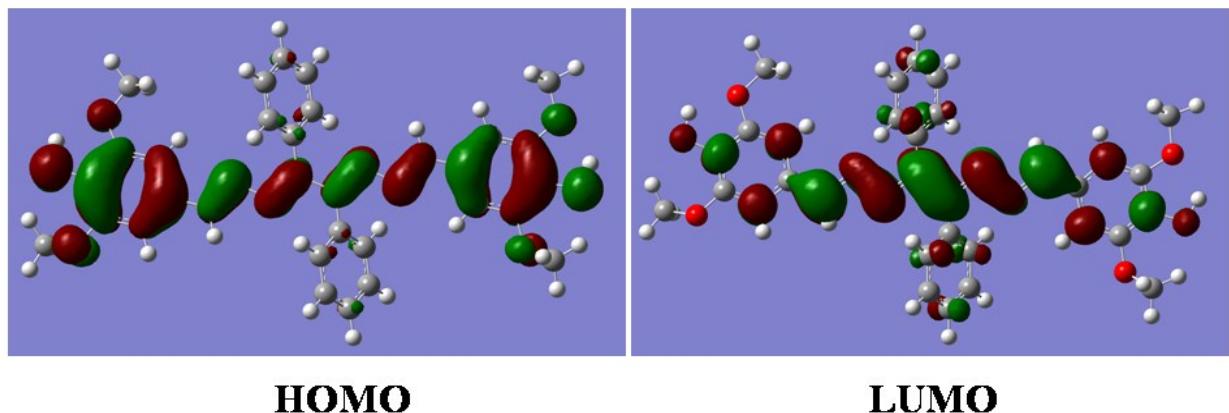
**Fig. S4** Mass spectra of L.



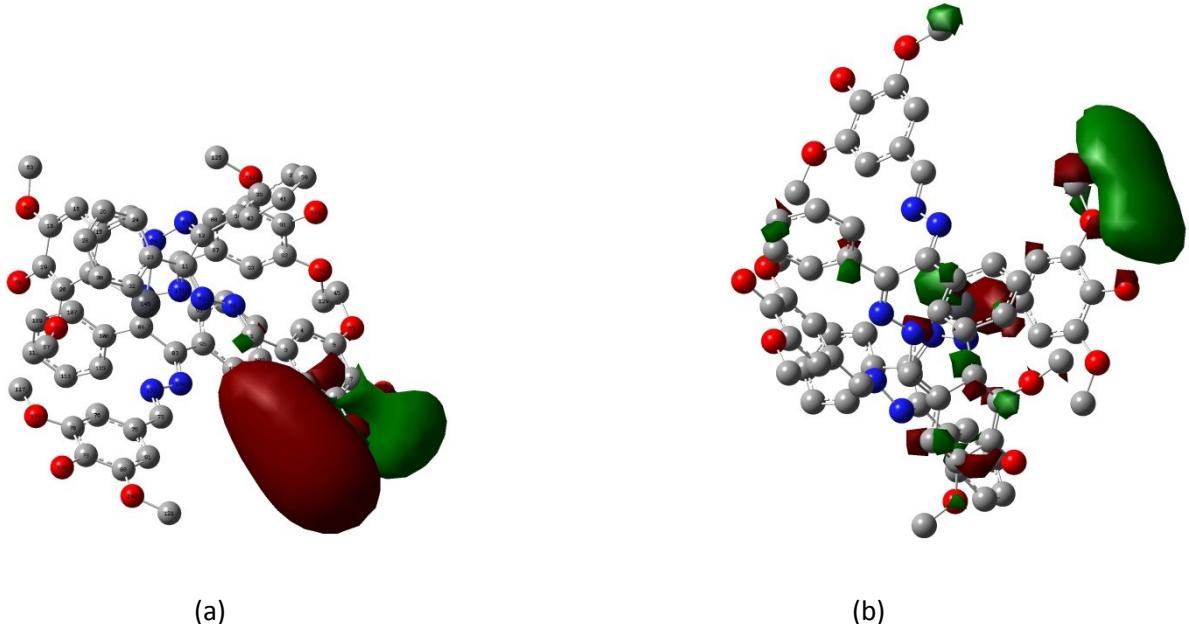
**Figure S5.** Geometry optimized diagram of the molecule L.



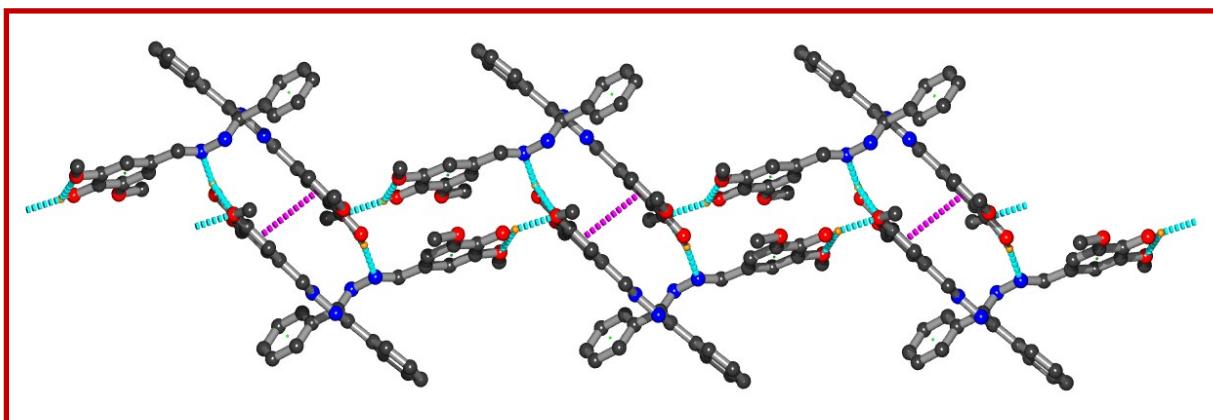
**Figure S6.** Geometry optimized diagram of the molecule  $Pb^{2+}$  complex of **L**.



**Figure S7.** The contour diagrams of HOMO and LUMO of **L**.



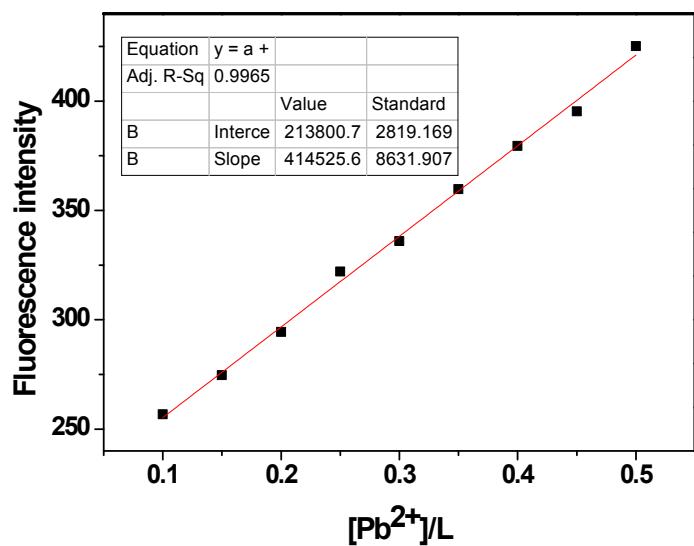
**Figure S8.** Structure of the (a) HOMO and (b) LUMO of  $\text{Pb}^{2+}$  complex of L. Hydrogen atoms are omitted for the sake of simplicity.



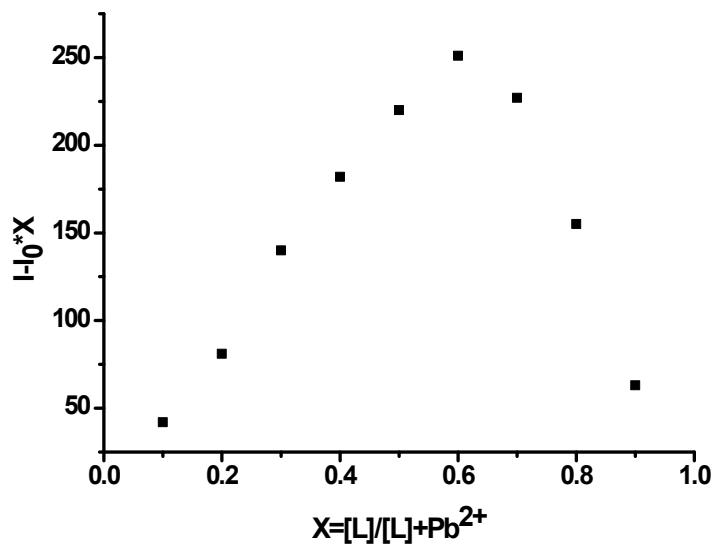
**Fig. S9** 1D supramolecular chain formation through hydrogen bonding and  $\pi \cdots \pi$  interactions in L.



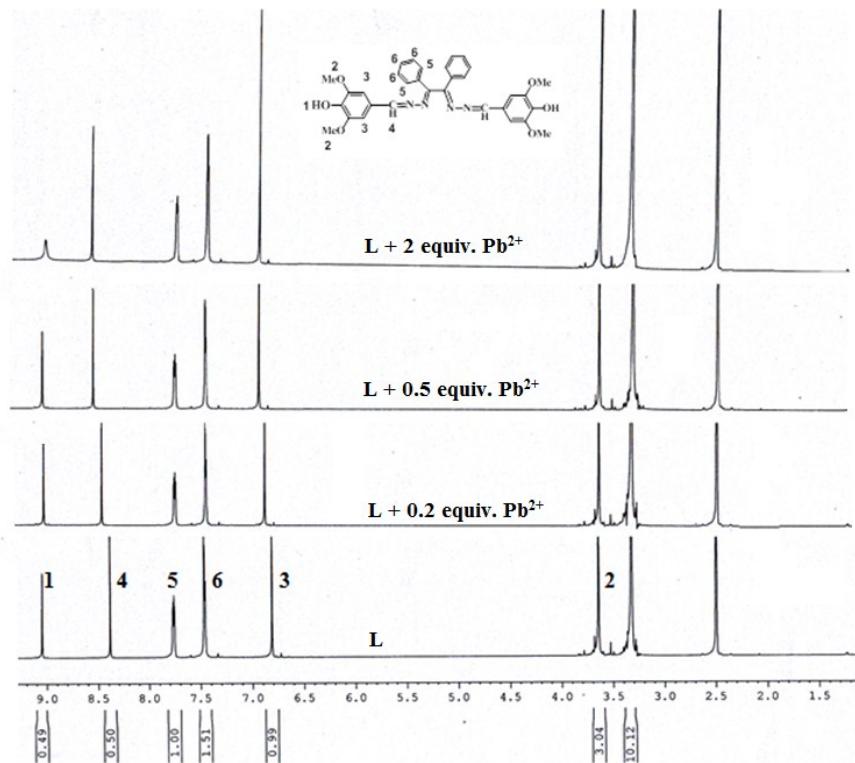
**Figure S10.** The color changes of L ( $10 \mu\text{M}$ ) upon addition of various cations (10 equiv.) in  $\text{MeOH-H}_2\text{O}$  (1:1, v/v) at room temperature.



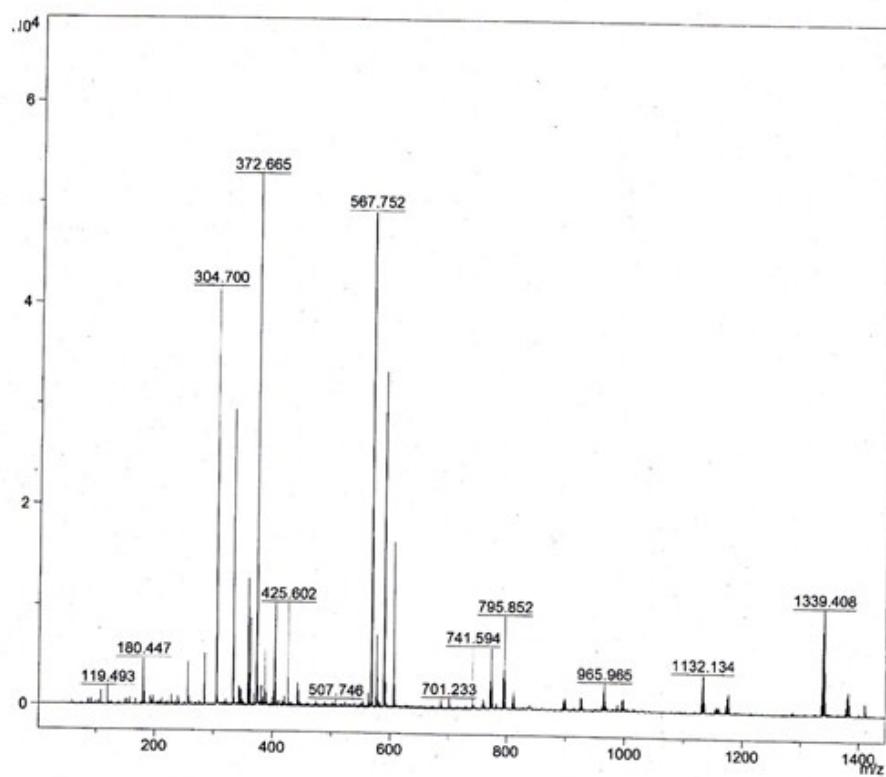
**Fig. S11** Detection limit.



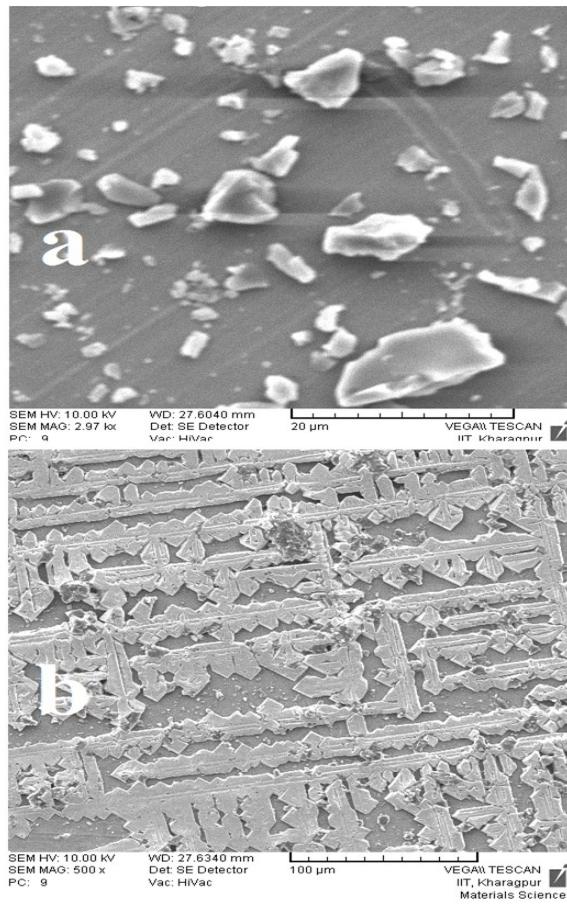
**Fig. S12** Job plot for  $\text{Pb}^{2+}$ .



**Fig. S13** <sup>1</sup>H NMR titration of L with Pb<sup>2+</sup> in <sup>6</sup>-DMSO-D<sub>2</sub>O (9/1, v/v).



**Fig. S14** Mass spectra of L+ Pb<sup>2+</sup>.



**Fig. S15** FTIR spectra of **L** and **L+ Pb(NO<sub>3</sub>)<sub>2</sub>**.

**Table S1** Crystallographic data and structure refinement parameters for receptor **L**

Formula	C <sub>32</sub> H <sub>30</sub> N <sub>4</sub> O <sub>6</sub>
Formula Weight	566.60
Crystal System	Monoclinic
Space group	P21/c (No. 14)
a [Å]	11.840(5)
b [Å]	10.520(5)
c [Å]	23.100(5)
β [°]	96.951(5)
V [Å <sup>3</sup> ]	2856.1(19)
Z	4
D(calc) [g/cm <sup>3</sup> ]	1.318
μ(MoKα) [ /mm ]	0.092
F(000)	1192

Crystal Size [mm]	0.14 x 0.16 x 0.20
Temperature (K)	293
Radiation [Å]	MoK $\alpha$ , 0.71073
$\Theta$ Min-Max [ $^{\circ}$ ]	2.1; 28.8
Dataset	-15: 15 ; -14: 14 ; -30: 30
Total Data	92134
Uniq. Data	7313
R(int)	0.168
Observed data [ $I > 2.0 \text{ sigma}(I)$ ]	3856
$N_{\text{ref}}$	7313
$N_{\text{par}}$	380
R	0.0964
wR <sub>2</sub>	0.2885
S	1.09

**Table S2** Hydrogen bond dimensions of the molecule **L**

D-H $\cdots$ A	D-H/(Å)	H $\cdots$ A/(Å)	D $\cdots$ A/(Å)	$\angle$ D-H $\cdots$ A/( $^{\circ}$ )	Symmetry
O2-H1O2 $\cdots$ O3	0.91	2.34	2.682(3)	102	.
O2-H1O2 $\cdots$ N4	0.91	1.99	2.877(4)	165	-x, -y, -z
O5-H1O5 $\cdots$ O3	0.86	2.07	2.849(3)	151	1+x, -1+y, z
O5-H1O5 $\cdots$ O4	0.86	2.25	2.642(4)	108	.
C1-H1 $\cdots$ O6	0.93	2.55	3.355(5)	146	1-x, -y, -z
C16-H16 $\cdots$ O2	0.93	2.59	3.363(4)	142	-x, -y, -z
C32-H32A $\cdots$ N2	0.96	2.57	3.462(5)	155	1-x, -y, -z
C32-H32B $\cdots$ O1	0.96	2.48	3.196(5)	131	1+x, y, z

**Table S3** All  $\pi\cdots\pi$  interactions of the molecule **L**.

$\pi\cdots\pi$ interactions			
Cgi $\cdots$ Cgj	Distance between ring Centroids/(Å)	Dihedral Angle between Planes I and J/( $^{\circ}$ )	Symmetry
Cg1 $\cdots$ Cg1	3.744(3)	0	-x, -y, -z
Cg2 $\cdots$ Cg2	3.595(3)	0	1-x, -1-y, -z
C-H $\cdots$ $\pi$ interactions			
C-H $\cdots$ Cgi	H $\cdots$ Cg/(Å)	$\angle$ X-H $\cdots$ Cg/( $^{\circ}$ )	Symmetry
C10-H10 $\cdots$ Cg3	2.94	167	1-x, -1/2+y, 1/2-z
C31-H31C $\cdots$ Cg4	2.75	134	1-x, -1/2+y, 1/2-z

\*Cg1: C2->C3->C4->C5->C6->C7->; Cg2: C11->C12->C13->C14->C15->C16->; Cg3: C17->C18->C19->C20->C21->C22->; Cg4: C23->C24->C25->C26->C27->C28->