## SUPPLEMENTARY INFORMATION

# Effective separation of the actinides Am(III) and Cm(III) by electronic modulation of Bis-(1,2,4-triazin-3yl)phenanthrolines 

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### 1.0 Experimental Procedures

## General procedure

NMR spectra were recorded using either a Bruker AMX400 or an Avance DFX400 instrument. Deuterated chloroform $\left(\mathrm{CDCl}_{3}\right)$ and Deuterated DMSO (dimethyl sulfoxide- $\mathrm{d}_{6}$ ) were used as solvents. Chemical shifts ( $\delta$ values) were reported in parts per million (ppm) with the abbreviations $\mathrm{s}, \mathrm{d}, \mathrm{t}, \mathrm{q}, \mathrm{qn}, \mathrm{sx}, \mathrm{dd}$, ddd and br denoting singlet, doublet, triplet, quartet, quintet, sextet, double doublets, doublet of doublets of doublets and broad resonances respectively. Coupling constants $(J)$ are quoted in Hertz. IR spectra were recorded as Nujol $_{\circledR}$ mulls (N) on a Perkin Elmer RX1 FT-IR instrument. All the melting points were determined on a Gallenkamp melting point apparatus. Mass spectra $(m / z)$ were recorded under conditions of electrospray ionisation (ESI). The ions observed were quasimolecular ions created by the addition of a hydrogen ion denoted as $[\mathrm{MH}]^{+}$or $[\mathrm{M}+\mathrm{Na}]$. The instrument used was Xcalibur Tune 2.1 (SP1).

2,9-Bis(5,5,8,8-tetramethyl-5,6,7,8-tetrahydrobenzo[e][1,2,4]triazin-3-yl)-1,10-phenanthroline (3)


To a suspension of 1,10-phenanthroline-2,9-dicarbohydrazonamide ( $0.58 \mathrm{~g}, 2 \mathrm{mmol}$ ) in THF ( 75 mL ) was added 3,3,6,6-tetramethylcyclohexane-1,2-dione ( $0.74 \mathrm{~g}, 4.4 \mathrm{mmol}, 2.2 \mathrm{eq})$. Triethylamine ( $3.5 \mathrm{~mL}, 25.3 \mathrm{mmol}$ ) was added and the mixture was heated under reflux for 3 days. The solution was allowed to cool to room temperature and filtered and the remaining solid residue was washed with $\mathrm{DCM}(25 \mathrm{~mL})$. The filtrate was evaporated and the solid was triturated with $\mathrm{Et}_{2} \mathrm{O}$ $(25 \mathrm{~mL})$. The insoluble solid was filtered and washed with further $\mathrm{Et}_{2} \mathrm{O}(25 \mathrm{~mL})$ and allowed to dry in air to afford the ligand 3 as a yellow solid ( $0.92 \mathrm{~g}, 83 \%$ ); $\mathrm{Mp}\left(247-250{ }^{\circ} \mathrm{C}\right) ;{ }^{1} \mathrm{H}$ NMR (400.1 $\left.\mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta_{\mathrm{H}}(\mathrm{ppm})=1.55(\mathrm{~s}, 12 \mathrm{H}), 1.58(\mathrm{~s}, 12 \mathrm{H}), 1.89(\mathrm{~s}, 8 \mathrm{H}), 7.95(\mathrm{~s}, 2 \mathrm{H}), 8.48(\mathrm{~d}, J=8.4$ $\mathrm{Hz}, 2 \mathrm{H}), 8.90(\mathrm{~d}, J=8.4 \mathrm{~Hz}, 2 \mathrm{H}) ;{ }^{13} \mathrm{C}$ NMR ( $100.6 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta_{\mathrm{C}}(\mathrm{ppm})=29.3,29.9,33.7,36.7$, $37.6,123.5,127.6,129.9,137.4,146.5,153.9,161.4,163.3,165.1 ; \mathrm{C}_{68} \mathrm{H}_{76} \mathrm{~N}_{16}[2 \mathrm{M}+\mathrm{Na}]$ requires $\mathrm{m} / \mathrm{z} 1139.6331$; (FTMS + c ESI) MS found $\mathrm{m} / \mathrm{z} 1139.6345$; IR $v_{\text {max }} / \mathrm{cm}^{-1}=3490$, 2962, 2931, 2866, $2224,1622,1588,1554,1516,1499,1472$.

## 5-Bromo-2,9-bis(5,5,8,8-tetramethyl-5,6,7,8-tetrahydrobenzo[e][1,2,4]triazin-3-yl)-1,10phenanthroline (4)



To a suspension of 5-bromo-1,10-phenanthroline-2,9-dicarbohydrazonamide ( $1.47 \mathrm{~g}, 3.9 \mathrm{mmol}$ ) in 1,4-dioxane ( 200 mL ) was added 3,3,6,6-tetramethylcyclohexane-1,2-dione ( $1.72 \mathrm{~g}, 10.2 \mathrm{mmol}, 2.6$ eq). Triethylamine ( $6 \mathrm{~mL}, 42.7 \mathrm{mmol}$ ) was added and the mixture was heated under reflux for 3 days. After allowing the solution to cool to room temperature, the solvent was evaporated and the
remaining semi-solid residue was triturated with ice-cold $\mathrm{Et}_{2} \mathrm{O}(400 \mathrm{~mL})$. The insoluble solid was filtered and washed with further ice-cold $\mathrm{Et}_{2} \mathrm{O}(400 \mathrm{~mL})$ and allowed to dry in air to afford the $\mathbf{4}$ as a yellow solid ( $1.23 \mathrm{~g}, 49 \%$ ); $\mathrm{Mp}\left(197-200{ }^{\circ} \mathrm{C}\right.$ ); ${ }^{1} \mathrm{H} \operatorname{NMR}\left(400.1 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta_{\mathrm{H}}(\mathrm{ppm})=1.54(\mathrm{~s}$, $12 \mathrm{H}), 1.56(\mathrm{~s}, 12 \mathrm{H}), 1.90(\mathrm{~s}, 8 \mathrm{H}), 8.29(\mathrm{~s}, 1 \mathrm{H}), 8.39(\mathrm{~d}, J=8.0 \mathrm{~Hz}, 1 \mathrm{H}), 8.86(\mathrm{~d}, J=8.4 \mathrm{~Hz}, 1 \mathrm{H})$, $8.88(\mathrm{~d}, J=8.4 \mathrm{~Hz}, 1 \mathrm{H}), 8.95(\mathrm{~d}, J=9.4 \mathrm{~Hz}, 1 \mathrm{H}) ;{ }^{13} \mathrm{C} \operatorname{NMR}\left(100.6 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta_{\mathrm{C}}(\mathrm{ppm})=29.3$, $29.8,33.6,33.8,36.7,37.5,122.0,124.0,124.1,128.9,129.8,130.6,136.3,137.3,146.0,146.9$, 154.4, 154.7, 161.1, 161.3, 163.3, 163.4, 165.0, 165.1; $\mathrm{C}_{34} \mathrm{H}_{37} \mathrm{~N}_{8} \mathrm{Br}[\mathrm{MH}]^{+}$requires $\mathrm{m} / \mathrm{z} 637.2397$ and 639.2377; (FTMS + p ESI) MS found $\mathrm{m} / \mathrm{z} 637.2392$ and 639.2371 ; IR $v_{\max } / \mathrm{cm}^{-1}=3531,3486$, 2959, 2927, 2865, 1644, 1609, 1510, 1475, 1452, 1439.

## 4-(2,9-Bis(5,5,8,8-tetramethyl-5,6,7,8-tetrahydrobenzo[1,2,4]triazin-3-yl)-1,10-phenanthrolin-5-yl)phenol (5)



A suspension of 5-bromo-2,9-bis(5,5,8,8-tetramethyl-5,6,7,8-tetrahydrobenzo[e][1,2,4]triazin-3-yl)-1,10-phenanthroline ( $0.51 \mathrm{~g}, 0.8 \mathrm{mmol}$ ), tetrakis(triphenylphosphane)palladium ( 0 ) ( $0.04 \mathrm{~g}, 0.04$ mmol, 0.05 eq ), (4-hydroxyphenyl)boronic acid ( $0.13 \mathrm{~g}, 0.9 \mathrm{mmol}, 1.1 \mathrm{eq}$ ) and $\mathrm{K}_{2} \mathrm{CO}_{3}(0.15 \mathrm{~g}, 1.1$ mmol, 1.4 eq ) in degassed $\mathrm{EtOH}(75 \mathrm{~mL})$ was heated to reflux for 18 h under nitrogen. The solution was allowed to cool to room temperature and filtered and the remaining solid residue was washed with EtOH ( 20 mL ). The filtrate was evaporated and the solid residue was taken up in DCM ( 150 mL ) and water ( 100 mL ) was added. The organic layer was washed with saturated brine $(100 \mathrm{~mL})$ and dried over $\mathrm{MgSO}_{4}$. The filtrate was evaporated and the solid was triturated with $\mathrm{Et}_{2} \mathrm{O}$ $(100 \mathrm{~mL})$. The insoluble solid was filtered and washed with $\mathrm{Et}_{2} \mathrm{O}(50 \mathrm{~mL})$ and allowed to dry in air to afford 5 as a yellow solid ( $0.31 \mathrm{~g}, 59 \%$ ); $\mathrm{Mp}\left(250-252{ }^{\circ} \mathrm{C}\right) ;{ }^{1} \mathrm{H}$ NMR ( $400.1 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta_{\mathrm{H}}$ $(\mathrm{ppm})=1.58(\mathrm{~s}, 12 \mathrm{H}), 1.61(\mathrm{~s}, 12 \mathrm{H}), 1.93(\mathrm{~s}, 8 \mathrm{H}), 6.46(\mathrm{~d}, J=8.4 \mathrm{~Hz}, 2 \mathrm{H}), 6.63(\mathrm{~d}, J=8.4 \mathrm{~Hz}$, $2 \mathrm{H}), 7.67(\mathrm{~s}, 1 \mathrm{H}), 8.28(\mathrm{~d}, J=8.4 \mathrm{~Hz}, 1 \mathrm{H}), 8.41(\mathrm{~d}, J=8.4 \mathrm{~Hz}, 1 \mathrm{H}), 8.76(\mathrm{~d}, J=8.8 \mathrm{~Hz}, 1 \mathrm{H}), 8.87$ $(\mathrm{d}, J=8.0 \mathrm{~Hz}, 1 \mathrm{H}) ;{ }^{13} \mathrm{C}$ NMR $\left(100.6 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta_{\mathrm{C}}(\mathrm{ppm})=29.3,29.9,33.8,36.7,37.6,115.6$,
$122.9,123.7,127.0,128.2,129.5,129.5,130.0,136.3,137.3,140.2,145.3,146.5,153.2,153.3$, 157.7, 161.0, 161.3, 163.5, 163.6, 165.2, 165.3; $\mathrm{C}_{40} \mathrm{H}_{43} \mathrm{~N}_{8} \mathrm{O}[\mathrm{MH}]^{+}$requires $\mathrm{m}_{\mathrm{z}} 651.3554$; (FTMS + p ESI) MS found $\mathrm{m} / \mathrm{z} 651.3553$; IR $v_{\text {max }} / \mathrm{cm}^{-1}=3399$, 2962, 2931, 2865, 1611, 1587, 1514, 1471, 1456, 1389, 1365.

## ${ }^{1} \mathrm{H}$ and ${ }^{13} \mathrm{C}$ NMR spectra



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Figure S1. ${ }^{1} \mathrm{H}$ and ${ }^{13} \mathrm{C}$ NMR spectra of $\mathrm{CyMe}_{4}$-BTPhen 3.


Figure S2. ${ }^{1} \mathrm{H}$ and ${ }^{13} \mathrm{C}$ NMR spectra of $5-\mathrm{BrCyMe}_{4}$-BTPhen 4.


Figure S3. ${ }^{1} \mathrm{H}$ and ${ }^{13} \mathrm{C}$ NMR spectra of 5-(4-hydroxyphenyl)-CyMe 4 -BTPhen 5.

### 2.0 Extraction Results

## General Procedure

Experiments were performed extracting ${ }^{241} \mathrm{Am}$ (III), ${ }^{244} \mathrm{Cm}$ (III), ${ }^{152} \mathrm{Eu}$ (III) ( $1 \mathrm{kBq} / \mathrm{mL}$ each), all lanthanides (La(III)-Lu(III) except $\mathrm{Pm}(\mathrm{III})$ ), and Y (III) ( $20 \mathrm{mg} / \mathrm{L}$ ) from $\mathrm{HNO}_{3}(500 \mu \mathrm{~L})$ into 30 $\mathrm{mmol} / \mathrm{L}$ BTPhen in 1 -octanol ( $500 \mu \mathrm{~L}$ ). After phase separation, ${ }^{241} \mathrm{Am}$ (III) and ${ }^{152} \mathrm{Eu}(I I I)$ were determined by gamma counting ((Packard Cobra Auto Gamma 5003) in $300 \mu \mathrm{~L}$ aliquots of both phases. ${ }^{241} \mathrm{Am}$ (III) and ${ }^{244} \mathrm{Cm}$ (III) were determined by alpha spectrometry (Canberra 7401); inactive lanthanides were determined by ICP-MS (Perkin-Elmer Elan 6100). An error of $20 \%$ was estimated for the distribution ratios measurements, which were not repeated.

Table S1 Extraction of $\mathrm{Am}(\mathrm{III})$ and $\mathrm{Eu}(\mathrm{III})$ by $\mathrm{CyMe}_{4}$-BTPhen 3 as a function of nitric acid concentration (* gamma measurement).

| $\left[\mathrm{HNO}_{3}\right]$ Initial (mol/L) | $D_{\mathrm{Am}}{ }^{*}$ | $D_{\mathrm{Eu}}{ }^{*}$ | $\mathrm{SF}_{\mathrm{Am} / \mathrm{Eu}}$ |
| :---: | :---: | :---: | :---: |
| 0.1 | 1043.4 | 1.6 | 667.0 |
| 0.3 | 1967.6 | 3.1 | 634.2 |
| 1 | 560 | 4.2 | 133.1 |
| 3 | 335.4 | 4.6 | 72.6 |

Table S2 Extraction of $\mathrm{Am}(\mathrm{III})$ and $\mathrm{Cm}(\mathrm{III})$ by $\mathrm{CyMe}_{4}$ - BTPhen 3 as a function of nitric acid concentration (* alpha measurement).

| $\left[\mathrm{HNO}_{3}\right]$ Initial (mol/L) | $D_{\mathrm{Am}}{ }^{* *}$ | $D_{\mathrm{Cm}}{ }^{* *}$ | $\mathrm{SF}_{\mathrm{Am} / \mathrm{Cm}}$ |
| :---: | :---: | :---: | :---: |
| 0.1 | 377.0 | 300.1 | 1.3 |
| 0.3 | 501.3 | 512.9 | 1.0 |
| 1 | 369.3 | 298.3 | 1.2 |
| 3 | 238.9 | 210.5 | 1.1 |

Table S3 Extraction (reported in $\mathrm{mg} / \mathrm{L}$ ) of $\operatorname{Ln}(\mathrm{III})$ and Y (III) from nitric acid ( $0.1-3 \mathrm{M}$ ) by a 0.03 M solution of $\mathrm{CyMe}_{4}$-BTPhen $\mathbf{3}$ in 1-octanol.

| Metal | $\left[\mathrm{HNO}_{3}\right]$ Initial $(\mathrm{mol} / \mathrm{L})$ |  |  |  |
| :---: | :---: | :---: | :---: | :---: |
|  | 0.1 | 0.3 | 1 | 3 |
|  | 0.0267 | 0.0475 | 0.0309 | 0.0137 |
| Ce | 0.174 | 0.0314 | 0.0189 | 0.0131 |
| Pr | 0.6809 | 1.0217 | 0.1711 | 0.1333 |
| Nd | 1.5877 | 2.4023 | 0.6115 | 0.52 |
| Sm | 2.3121 | 4.2039 | 3.956 | 1.3203 |
| Eu | 1.4755 | 3.0832 | 4.253 | 4.2924 |
| Gd | 0.6238 | 1.323 | 2.1708 | 4.6439 |
| Tb | 0.5455 | 1.1354 | 2.2244 | 1.3628 |
| Dy | 0.3101 | 0.6687 | 0.998 | 0.9335 |
| Ho | 0.1764 | 0.362 | 0.3593 | 0.3773 |
| Er | 0.1123 | 0.1915 | 0.1404 | 0.1262 |
| Tm | 0.0659 | 0.1062 | 0.0613 | 0.0507 |
| Yb | 0.0494 | 0.0692 | 0.0375 | 0.0235 |
| Lu | 0.028 | 0.035 | 0.0215 | 0.02 |

Table S4 Extraction of $\mathrm{Am}(\mathrm{III})$ and $\mathrm{Eu}(\mathrm{III})$ by $5-\mathrm{Br}-\mathrm{CyMe}_{4}-\mathrm{BTPhen} 4$ as a function of nitric acid concentration (* gamma measurement).

| $\left[\mathrm{HNO}_{3}\right]$ Initial $(\mathrm{mol} / \mathrm{L})$ | $D_{\mathrm{Am}}{ }^{*}$ | $D_{\mathrm{Eu}}{ }^{*}$ | $\mathrm{SF}_{\mathrm{Am} / \mathrm{Eu}}$ |
| :---: | :---: | :---: | :---: |
| 0.1 | 63.7 | 0.2 | 317.9 |
| 0.3 | 270.7 | 0.6 | 475.8 |
| 1 | 333.9 | 1.2 | 287.8 |
| 3 | 507.5 | 0.7 | 678.7 |

Table S5 Extraction of $\mathrm{Am}(\mathrm{III})$ and $\mathrm{Cm}(\mathrm{III})$ by $5-\mathrm{Br}^{-\mathrm{CyMe}_{4}-\mathrm{BTPhen}} 4$ as a function of nitric acid concentration (* alpha measurement).

| $\left[\mathrm{HNO}_{3}\right]$ Initial $(\mathrm{mol} / \mathrm{L})$ | $D_{\mathrm{Am}}{ }^{* *}$ | $D_{\mathrm{Cm}}{ }^{* *}$ | $\mathrm{SF}_{\mathrm{Am} / \mathrm{Cm}}$ |
| :---: | :---: | :---: | :---: |
| 0.1 | 61.0 | 9.3 | 6.6 |
| 0.3 | 208.2 | 35.2 | 5.9 |
| 1 | 301.3 | 138.0 | 2.2 |
| 3 | 369.6 | 192.3 | 1.9 |

Table S6 Extraction (reported in $\mathrm{mg} / \mathrm{L}$ ) of $\operatorname{Ln}(\mathrm{III})$ and $\mathrm{Y}(\mathrm{III})$ from nitric acid ( $0.1-3 \mathrm{M}$ ) by a 0.03 M solution of 5-Br-CyMe - -BTPhen 4 in 1-octanol.

|  | $\left[\mathrm{HNO}_{3}\right]$ Initial (mol/L) |  |  |  |
| :---: | :---: | :---: | :---: | :---: |
| Metal | 0.1 | 0.3 | 1 | 3 |
| Y | $1.81 \times 10^{-3}$ | $3.41 \times 10^{-3}$ | $4.87 \times 10^{-3}$ | $6.13 \times 10^{-3}$ |
| La | $2.38 \times 10^{-3}$ | $5.34 \times 10^{-3}$ | $7.39 \times 10^{-3}$ | $7.65 \times 10^{-3}$ |
| Ce | 0.0147 | 0.04 | 0.0549 | 0.0608 |
| Pr | 0.0513 | 0.1359 | 0.1796 | 0.2246 |
| Nd | 0.116 | 0.3228 | 0.4305 | 0.5575 |
| Sm | 0.2619 | 0.7306 | 1.2069 | 1.537 |
| Eu | 0.1929 | 0.5682 | 1.0878 | 0.7184 |
| Gd | 0.0805 | 0.2297 | 0.3686 | 0.189 |
| Tb | 0.0619 | 0.1703 | 0.2673 | 0.1513 |
| Dy | 0.0354 | 0.0876 | 0.1299 | 0.097 |
| Ho | 0.0216 | 0.0478 | 0.0653 | 0.0529 |
| Er | 0.0192 | 0.0341 | 0.0362 | 0.0278 |
| Tm | 0.0155 | 0.0237 | 0.0191 | 0.0171 |
| Yb | 0.0136 | 0.0186 | 0.0147 | 0.0124 |
| Lu | 0.01 | 0.0113 | $8.08 \times 10^{-3}$ | $6.19 \times 10^{-3}$ |

Table S7 Extraction of $\mathrm{Am}(\mathrm{III})$ and $\mathrm{Eu}(\mathrm{III})$ by 5-(4-hydroxyphenyl)-CyMe $\mathrm{Cl}_{4}$-BTPhen 5 as a function of nitric acid concentration (* gamma measurement).

| $\left[\mathrm{HNO}_{3}\right]$ Initial $(\mathrm{mol} / \mathrm{L})$ | $D_{\mathrm{Am}}{ }^{*}$ | $D_{\mathrm{Eu}}{ }^{*}$ | $\mathrm{SF}_{\mathrm{Am} / \mathrm{Eu}}$ |
| :---: | :---: | :---: | :---: |
| 0.1 | 27.9 | 0.5 | 54.6 |
| 0.3 | 70.1 | 1.2 | 59.1 |
| 1 | 1807.3 | 4.6 | 389.6 |
| 3 | 1581.0 | 5.1 | 311.3 |

Table S8 Extraction of Am (III) and Cm (III) by 5-(4-hydroxyphenyl)-CyMe $\mathrm{Cl}_{4}$-BTPhen 5 as a function of nitric acid concentration (* alpha measurement).

| $\left[\mathrm{HNO}_{3}\right]$ Initial (mol/L) | $D_{\mathrm{Am}}{ }^{* *}$ | $D_{\mathrm{Cm}}{ }^{* *}$ | $\mathrm{SF}_{\mathrm{Am} / \mathrm{Cm}}$ |
| :---: | :---: | :---: | :---: |
| 0.1 | 23.7 | 5.7 | 4.2 |
| 0.3 | 63.3 | 13.8 | 4.6 |
| 1 | 831.8 | 153.2 | 5.4 |
| 3 | 2496.3 | 1110.9 | 2.2 |

Table S9 Extraction (reported in $\mathrm{mg} / \mathrm{L}$ ) of $\mathrm{Ln}(\mathrm{III})$ and $\mathrm{Y}(\mathrm{III})$ from nitric acid ( $0.1-3 \mathrm{M}$ ) by a 0.03 M solution of 5-(4-hydroxyphenyl)-CyMe 4 -BTPhen 5 in 1-octanol.

| Metal | $\left[\mathrm{HNO}_{3}\right]$ Initial $(\mathrm{mol} / \mathrm{L})$ |  |  |  |
| :---: | :---: | :---: | :---: | :---: |
|  | 0.1 | 0.3 | 1 | 3 |
| La | 0.0188 | 0.0296 | 0.0158 | $6.61 \times 10^{-3}$ |
| Ce | 0.262 | 0.0372 | 0.025 | 0.0174 |
| Pr | 0.7203 | 0.3674 | 0.2593 | 0.1711 |
| Nd | 0.8555 | 1.2413 | 1.0235 | 0.8784 |
| Sm | 0.6576 | 1.4962 | 2.605 | 2.3609 |
| Eu | 0.4787 | 1.1086 | 6.2924 | 9.1194 |
| Gd | 0.2546 | 0.5899 | 4.7113 | 5.2025 |
| Tb | 0.2219 | 0.5187 | 1.9159 | 1.0156 |
| Dy | 0.1509 | 0.3163 | 1.3779 | 0.5717 |
| Ho | 0.0927 | 0.1546 | 0.4965 | 0.1747 |
| Er | 0.0524 | 0.0651 | 0.0406 | 0.0408 |
| Tm | 0.0282 | 0.0291 | 0.0212 | $7.98 \times 10^{-3}$ |
| Yb | 0.0221 | 0.0248 | 0.0256 | 0.012 |
| Lu | 0.018 | 0.0192 | 0.0299 | 0.0262 |

