Electronic Supplementary Information

Novel pseudo[2]rotaxanes constructed by selfassembly of dibenzyl tetramethylene bis-carbamate derivatives and per-ethylated pillar[5]arene

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1. Materials and methods

All reactions were performed in atmosphere unless noted. All reagents were commercially available and use as supplied without further purification. Solvents were either employed as purchased or dried according ti procedures described in the literature. Compound **EtP[5]A** was prepared by published literature procedures.^{S1} NMR spectra were collected on either a Bruker Avance DMX 300 MHz spectrometer or a Bruker Avance DMX 400 MHz spectrometer with internal standard tetramethylsilane (TMS) and signals as internal references, and the chemical shifts (δ) were expressed in ppm. 2D COSY and NOESY experiments were performed on a Bruker DPX 400 MHz spectrometer. Low-resolution electrospray ionization mass spectra (LR-ESI-MS) were obtained on Finnigan MatTSQ 7000 instruments. High-resolution electrospray ionization mass spectra (HR-ESI-MS) were recorded on an Agilent 6540Q-TOF LCMS equipped with an electrospray ionization (ESI) probe operating in positive-ion mode with direct infusion.

2. Synthesis of Guest





Synthesis of **G1**: 1,4-butanediamine (0.44 g, 5 mmol) and triethylamine (1.02 g, 10 mmol) were added in dichloromethane (50 mL). Then benzyl chloroformate (1.71 g, 10 mmol) was dropped to the mixture in ice-bath in 10 minutes. The mixture was stirred at room temperature for 12 hours. The reaction mixture was filtered and washed with dichloromethane. The filtrate was washed by brine and dried by Na₂SO₄. The organic layer was evaporated under vacuum, and the residue was further purified by flash column chromatography on silica gel (dichloromethane/methanol = 160/1, ν/ν) to afford **G1** (1.64 g, 92.0 %), m.p. 146–147 °C. ¹H NMR (300 MHz, chloroform-*d*, 298 K) δ (ppm): 7.35–7.37 (m, 10H, ArH), 5.10 (s, 4H, CH₂), 4.79 (s, 2H, NH), 3.20 (s, 4H, CH₂), 1.53 (s, 4H, CH₂). ¹³C NMR (75 MHz, chloroform-*d*, 298 K) δ (ppm): 156.5, 136.6, 128.5, 128.1, 66.7, 40.6, 27.2. LR-ESI-MS is: *m/z* calcd for [M + H]⁺, 357.18, found 357.15; calcd for [M + Na]⁺, 379.16, found 379.15. HR-ESI-MS is: *m/z* calcd for [M + Na]⁺, C₂₀H₂₄N₂O₄Na⁺, 379.1628, found 379.1631.



Fig. S1 ¹H NMR spectrum (300 MHz, chloroform-d, 298 K) of G1



Fig. S2 ¹³C NMR spectrum (75 MHz, chloroform-d, 298 K) of G1



Scheme S2 Synthesis of G2

Synthesis of G2: 1,4-butanediamine (0.44 g, 5 mmol) and triethylamine (1.52 g, 15 mmol) were added in dichloromethane (50 mL). Then hydrocinnamoyl chloride (2.53 g, 15 mmol) was dropped to the mixture in ice-bath in 10 minutes. The mixture was stirred at room temperature for 12 hours. The reaction mixture was filtered and washed with dichloromethane. The filtrate was washed by brine and dried by Na₂SO₄. The organic layer was evaporated under vacuum, and the residue further purified by flash column chromatography was on silica gel (dichloromethane/methanol = 60/1, v/v) to afford G2 (1.28 g, 72.7 %), m.p. 176–178 °C. ¹H NMR (300 MHz, chloroform-d, 298 K) δ (ppm): 7.31–7.19 (m, 10H, ArH), 5.78 (s, 2H, NH), 3.18 (d, J = 5.66 Hz, 4H, CH₂), 2.97 (t, J = 7.59 Hz, 4H, CH₂), 2.50 (t, J = 7.62 Hz 4H, CH₂), 1.35 (t, J = 6.24 Hz, 4H, CH₂). ¹³C NMR (75 MHz, chloroform-d, 298 K) δ (ppm): 172.4, 140.9, 128.5, 128.4, 126.2, 38.9, 38.5, 31.8, 26.6. LR-ESI-MS is: *m/z* calcd for [M + H]⁺, 353.22, found 353.15; calcd

for $[M + Na]^+$, 375.20, found 375.15. HR-ESI-MS is: m/z calcd for $[M + Na]^+$, $C_{22}H_{28}N_2O_2Na^+$, 375.2043, found 375.2044.



Fig. S3 ¹H NMR spectrum (300 MHz, chloroform-d, 298 K) of G2



Fig. S4 ¹³C NMR spectrum (75 MHz, chloroform-d, 298 K) of G2



Scheme S3 Synthetic route to G3

Synthesis of **G3**: 1,4-butanediamine (0.44 g, 5 mmol) and triethylamine (1.01 g, 10 mmol) were added in dichloromethane (50 mL). Then di-tert-butyl dicarbonate (2.18 g, 10 mmol) was dropped to the mixture in ice-bath in 10 minutes. The mixture was stirred at room temperature for 12 hours. The reaction mixture was filtered and washed with dichloromethane. The filtrate was washed by brine and dried by Na₂SO₄. The organic layer was evaporated under vacuum, and the residue was further purified by flash column chromatography on silica gel (dichloromethane/methanol = 80/1, ν/ν) to afford **G3** (1.34 g, 92.9 %), m.p. 139–140 °C. ¹H NMR (300 MHz, chloroform-*d*, 298 K) δ (ppm): 4.53 (s, 2H, NH), 3.12 (s, 4H, CH₂), 1.50 (m, 4H, CH₂), 1.44 (s, 18H, CH₃). ¹³C NMR (75 MHz, chloroform-*d*, 298 K) δ (ppm): 156.0, 79.2, 40.3, 28.4, 27.4. LR-ESI-MS is: *m/z* calcd for [M + H]⁺, 289.21, found 289.15; calcd for [M + Na]⁺, 311.19, found 311.15. HR-ESI-MS is: *m/z* calcd for [M + Na]⁺, C₁₄H₂₈N₂O₄Na⁺, 311.1941, found 311.1943.



Fig. S5 ¹H NMR spectrum (300 MHz, chloroform-d, 298 K) of G3



Fig. S6 ¹³C NMR spectrum (75 MHz, chloroform-d, 298 K) of G3



Scheme S4. Synthetic route to G4

Synthesis of **G4**: 1,4-butanediamine (0.44 g, 5 mmol) and triethylamine (1.52 g, 15 mmol) were added in dichloromethane (50 mL). Then butyryl chloride (1.60 g, 15 mmol) was dropped to the mixture in ice-bath in 10 minutes. The mixture was stirred at room temperature for 12 hours. The reaction mixture was filtered and washed with dichloromethane. The filtrate was washed by brine and dried by Na₂SO₄. The organic layer was evaporated under vacuum, and the residue was further purified by flash column chromatography on silica gel (dichloromethane/methanol = 40/1, ν/ν) to afford **G4** (0.46 g, 40.3 %), m.p. 164–165 °C. ¹H NMR (300 MHz, chloroform-*d*, 298 K) δ (ppm): 5.85 (s, 2H, NH), 3.28 (d, J = 5.95 Hz, 4H, CH₂), 2.16 (t, J = 7.48 Hz, 4H, CH₂), 1.73–1.63 (m, 4H, CH₂), 1.56–1.52 (m, 4H, CH₂), 0.95 (t, 6H, CH₃). ¹³C NMR (75 MHz, chloroform-*d*, 298 K) δ (ppm): 173.4, 39.0, 38.6, 26.9, 19.2, 13.8. LR-ESI-MS is: *m/z* calcd for [M + H]⁺, 229.19, found 229.15; calcd for [M + Na]⁺, 251.17, found 251.15. HR-ESI-MS is: *m/z* calcd for [M + Na]⁺, C₁₂H₂₄N₂O₂Na⁺, 251.1730, found 251.1729.



Fig. S7 ¹H NMR spectrum (300 MHz, chloroform-d, 298 K) of G4



Fig. S8 ¹³C NMR spectrum (75 MHz, chloroform-d, 298 K) of G4



Scheme S5. Synthetic route to G5

Synthesis of **G4**: 1,4-butanediamine (0.44 g, 5 mmol) were added in dichloromethane (50 mL). Then *p*-tolyl isocyanate (1.33 g, 10 mmol) was dropped to the mixture in ice-bath in 10 minutes. The mixture was stirred at room temperature for 2 hours. The reaction mixture was filtrated. The filter cake was washed by dichloromethane and water. Dry the cake to afford **G5** (1.56 g, 88.0 %), m.p. 277–279 °C. ¹H NMR (300 MHz, DMSO-*d*₆, 298 K) δ (ppm): 8.26 (s, 2H, NH), 7.25 (d, *J* =8.47 Hz, 4H, ArH), 7.00 (d, *J*=8.45 Hz, 4H, ArH), 6.07 (t, *J*=5.62 Hz, 2H, NH), 3.07 (d, *J*=5.61 Hz, 4H, CH₂), 1.30–1.23 (m, 4H, CH₂), 0.87 (t, *J*=7.29 Hz, 6H, CH₃). The clear ¹³C NMR spectrum of **G5** could not be obtained, because it is not well soluble in DMSO-*d*₆. LR-ESI-MS is: *m/z* calcd for [M + H]⁺, 355.21, found 355.15; calcd for [M + Na]⁺, 377.19, found 377.15. HR-ESI-MS is: *m/z* calcd for [M + Na]⁺, C₂₀H₂₆N₄O₂Na⁺, 377.1948, found 377.1951.



Fig. S9 ¹H NMR spectrum (300 MHz, DMSO-d₆, 298 K) of G5



Scheme S6. Synthetic route to G7_P

Synthesis of $G7_p$: 1,4-butanediamine (0.88 g, 10 mmol) and triethylamine (1.21 g, 12 mmol) were added in dichloromethane (170 mL). Then benzyl chloroformate (1.70 g, 10 mmol) was dropped to the mixture in ice-bath in 10 minutes. The mixture was stirred at room temperature for 12 hours. Some precipitates were observed from the reaction solution, and the resulting reaction mixture was extracted by water (20 mL), ethyl acetate (40 mL), and methanol (1 mL). The organic layer was evaporated under vacuum, and the residue was further purified by flash column chromatography on silica gel (dichloromethane/methanol = 40/1, *v*/*v*) to afford $G7_P$ (0.54 g, 24.3 %). ¹H NMR (400 MHz, DMSO-*d*₆, 298 K) δ (ppm): 7.77 (s, 2H, NH), 7.37–7.31 (m, 5H, ArH), 5.01 (s, 2H, CH₂), 3.03–2.98 (m, 2H, CH₂), 2.79–2.74 (m, 2H, CH₂), 1.55–1.42 (m, 4H, CH₂).



Fig. S10 ¹H NMR spectrum (400 MHz, DMSO- d_6 , 298 K) of G7_p



Scheme S7. Synthetic route to Gs

Synthesis of **Gs**: 2-Nitrobenzyl alcohol (0.766 g, 5 mmol) and 1,1-CarbonyldiiMidazole (1.62 g, 10 mmol) were added in dichloromethane (100 mL). The mixture was stirred at room temperature for 5 hours. The reaction mixture was washed by brine and dried by Na₂SO₄. The organic layer was evaporated under vacuum to afford **Gs** (1.13 g, 91.8 %). ¹H NMR (300 MHz, DMSO- d_6 , 298 K) δ (ppm): 8.36 (s, 1H, CH), 8.19 (d, J = 8.12 Hz, 1H, ArH), 7.91–7.83 (m, 2H, CH), 7.71–7.68 (m, 2H, ArH), 7.12 (s, 1H, ArH), 5.79 (s, 2H, CH₂).



Fig. S11 ¹H NMR spectrum (400 MHz, DMSO-d₆, 298 K) of Gs



Scheme S8. Synthetic route to G6

Synthesis of **G6**: 1,4-butanediamine (0.088 g, 1 mmol) was added in dichloromethane (30 mL). Then **Gs** (0.49 g, 2 mmol) was dropped to the mixture. The mixture was stirred at room temperature for 12 hours. The reaction mixture was washed by brine and dried by Na₂SO₄. The organic layer was evaporated under vacuum, and the residue was further purified by flash column chromatography on silica gel (dichloromethane/methanol = 60/1, *v/v*) to afford **G6** (0.21 g, 47.0%), m.p. 174–175 °C. ¹H NMR (300 MHz, chloroform-*d*, 298 K) δ (ppm): 8.08 (d, *J* =8.10 Hz, 2H, ArH), 7.65–7.60 (m, 4H, ArH), 7.48 (t, *J* =7.54 Hz, 2H, ArH), 5.51 (s, 4H, CH₂), 4.94 (s, 2H, NH), 3.23(s, 4H, CH₂), 1.58 (s, 4H, CH₂, overlapped with the solvent peak of water). ¹³C NMR (75 MHz, DMSO-*d*₆, 298 K) δ (ppm): 156.1, 134.5, 129.4, 125.2, 62.3, 27.1. LR-ESI-MS is: *m/z* calcd for [M + H]⁺, 447.15, found 447.05; calcd for [M + Na]⁺, 469.13, found 469.05. HR-ESI-MS is: *m/z* calcd for [M + Na]⁺, C₂₀H₂₂N₄O₈Na⁺, 469.1330, found 469.1332.



Fig. S12 ¹H NMR spectrum (300 MHz, chloroform-d, 298 K) of G6



Fig. S13 ¹³C NMR spectrum (75 MHz, DMSO-d₆, 298 K) of G6



Scheme S9. Synthetic route to G7

Synthesis of G7: G7_p (0.20g, 0.89 mmol) and Gs (0.22 g, 0.89 mmol) were added in DMSO (20 mL). The mixture was stirred at room temperature for 12 hours. The reaction mixture was washed by brine and extracted by ethyl acetate. The organic layer was dried by Na₂SO₄ and evaporated under vacuum, and the residue was further purified by flash column chromatography on silica gel (dichloromethane/methanol = 40/1, v/v) to afford G7 (0.20 g, 55.9 %). m.p. 117–119 °C. ¹H NMR (300 MHz, chloroform-*d*, 298 K) δ (ppm): 8.09 (d, *J* =8.06 Hz, 1H, ArH), 7.67–7.57 (m, 2H, ArH), 7.47 (t, *J* =7.56 Hz, 1H, ArH), 7.36–7.31 (m, 5H, ArH), 5.51 (s, 2H, CH₂), 5.10 (s, 2H, CH₂), 3.22 (s, 4H, CH₂), 1.56 (s, 4H, CH₂, overlapped with the solvent peak of water). ¹³C NMR (75 MHz, chloroform-*d*, 298 K) δ (ppm): 156.5, 155.9, 133.7, 128.5, 128.1, 125.0, 667, 63.2, 40.7, 40.6, 27.2. LR-ESI-MS is: *m/z* calcd for [M + H]⁺, 402.16, found 402.10; calcd for [M + Na]⁺,

424.14, found 424.05. HR-ESI-MS is: m/z calcd for $[M + Na]^+$, $C_{20}H_{22}N_3O_6Na^+$, 424.1479, found 424.1479.



Fig. S14 ¹H NMR spectrum (300 MHz, chloroform-d, 298 K) of G7



Fig. S15 ¹³C NMR spectrum (75 MHz, chloroform-d, 298 K) of G7

3. Investigation of the interactions between EtP[5]A and Guest by ¹H





Fig. S16 ¹H NMR spectra (300 MHz, CDCl₃, 298 K): (a) **EtP[5]A** (2.5 mM); (b) equimolar mixture of **G1**(2.5 mM) and **EtP[5]A** (2.5 mM); (c) **G1** (2.5 mM); (d) 0.2 mL methanol- d_4 was added into the NMR tube with the mixture of **G1** (2.5 mM) and **EtP[5]A** (2.5 mM) in 0.5 mL chloroform-*d*. The association constant $K_{a,G1} \cdot \text{EtP[5]A}$ value calculated from integrations of complexed and uncomplexed peaks of H₁ of **EtP[5]A** is [(2.09/3.63) × 1.00 × 10⁻³]² = 1279 M⁻¹. The association constant $K_{a,G1} \cdot \text{EtP[5]A}$ value calculated from integrations of complexed peaks of H_b of **G1** is [(1.35/2.35) × 2.50 × 10⁻³]/[(1 - 1.35/2.35) × 2.50 × 10⁻³]/[(1 - 1.35/2



Fig. S17 ¹H NMR spectra (300 MHz, CDCl₃, 298 K): (a) **EtP[5]A** (2.5 mM); (b) equimolar mixture of **G2**(2.5 mM) and **EtP[5]A** (2.5 mM); (c) **G2** (2.5 mM) The association constant $K_{a,G2}$ -_{EtP[5]A} value calculated from integrations of complexed and uncomplexed peaks of H₁ of **EtP[5]A** is $[(0.16/1.16) \times 2.50 \times 10^{-3}]/[(1 - 0.16/1.16) \times 2.50 \times 10^{-3}]^2 = 74 \text{ M}^{-1}$. The association constant $K_{a,G2}$ -_{EtP[5]A} value calculated from integrations of complexed and uncomplexed peaks of H_c of **G2** is $[(0.08/0.61) \times 2.50 \times 10^{-3}]/[(1 - 0.08/0.61) \times 2.50 \times 10^{-3}]^2 = 69 \text{ M}^{-1}$. Therefore, $K_{a,G2}$ -_{EtP[5]A} = $(74 + 69)/2 = (71.5 \pm 2.5) \text{ M}^{-1}$



Fig. S18 ¹H NMR spectra (300 MHz, CDCl₃, 298 K): (a) **EtP[5]A** (2.5 mM); (b) equimolar mixture of **G3**(2.5 mM) and **EtP[5]A** (2.5 mM); (c) **G3** (2.5 mM)



Fig. S19 ¹H NMR spectra (300 MHz, CDCl₃, 298 K): (a) **EtP[5]A** (2.5 mM); (b) equimolar mixture of **G4**(2.5 mM) and **EtP[5]A** (2.5 mM); (c) **G4** (2.5 mM)



Fig. S20 ¹H NMR spectra (300 MHz, CDCl₃, 298 K): (a) **EtP[5]A** (2.5 mM); (b) equimolar mixture of **G6**(2.5 mM) and **EtP[5]A** (2.5 mM); (c) **G6** (2.5 mM)



Fig. S21 ¹H NMR spectra (300 MHz, CDCl₃, 298 K) of (a) **EtP[5]A** (2.5 mM) ; (b) equimolar mixture of **G7**(2.5 mM) and **EtP[5]A** (2.5 mM); (c) **G7** (2.5 mM). The association constant $K_{a,G7} \cdot EtP[5]A$ value calculated from integrations of complexed and uncomplexed peaks of H₁ of **EtP[5]A** is $[(1.05/2.05) \times 2.50 \times 10^{-3}]/[(1 - 1.05/2.05) \times 2.50 \times 10^{-3}]^2 = 861 \text{ M}^{-1}$. The association constant $K_{a,G7} \cdot EtP[5]A$ value calculated from integrations of complexed and uncomplexed peaks of H₂ of **G7** is $[(0.49/0.96) \times 2.50 \times 10^{-3}]/[(1 - 0.49/0.96) \times 2.50 \times 10^{-3}]^2 = 851 \text{ M}^{-1}$. Therefore, $K_{a,G7} \cdot EtP[5]A = (861 + 851)/2 = (856 \pm 5) \text{ M}^{-1}$

4. Partial 2D NOSEY NMR spectra of G1 ⊂EtP[5]A



Fig. S22 2D NOSEY analysis of equimolar mixture G1⊂EtP[5]A in CDCl₃ (20 mM, 400 MHz, 298 K) S18

5. Study of the photocleavage $G7 \subset EtP[5]A$ via UV 365nm by ¹H NMR



Fig. S23 ¹H NMR spectra (300 MHz, CDCl₃, 298 K) of (a) G7⊂ EtP[5]A (1 : 1, 2.5mM) in the absence (b) G7⊂ EtP[5]A (1 : 1, 2.5mM)after UV 365 nm

6. Reference

S1. D. R. Cao, Y. H. Kou, J. Q. Liang, Z. Z. Chen, L. Y. Wang and H. Meier, *Angew. Chem. Int. Ed.*, 2009, **48**, 9721.