Supporting Information

A Simple Access to β-Trifluoromethyl-Substituted Ketones via Copper-Catalyzed Ring-Opening Trifluoromethylation of Substituted Cyclopropanols

Dzmitry G. Kananovich,*a Yulia A. Konik,^b Dzmitry M. Zubrytski,^b Ivar Järving^a and Margus Lopp*a

 ^a Department of Organic Chemistry, Tallinn University of Technology, Akadeemia tee 15, 12618 Tallinn, Estonia
^b Department of Organic Chemistry, Belarusian State University, Leningradskaya 14, 220050, Minsk, Belarus
*Corresponding authors e-mail address: dzmitry.kananovich@ttu.ee

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I. General Methods and Starting Materials

Methanol (>99.6%, Laboratory Reagent grade, Sigma-Aldrich) was used as a solvent. Deuterated solvents were purchased from Armar Chemicals. Other solvents were purified by conventional methods. 3,3-Dimethyl-1-(trifluoromethyl)-1,2-benziodoxole (**3**) and 1-trifluoromethyl-1,2-benziodoxol-3-(1*H*)-one (**4**) used in the optimization experiments were purchased from Sigma-Aldrich. Togni reagent **3** employed in the preparative runs was synthesized according to the published procedure.¹

Purified "white" copper(I) chloride, free of copper(II) salts, was prepared by two methods: (A) commercial "olive-green" reagent was stirred with glacial acetic acid overnight, then filtered under argon and washed subsequently with glacial acetic acid, methanol, diethyl ether and dried; (B) CuCl prepared by the reduction of CuCl₂ with sodium sulfite in aqueous solution was then filtered under argon, washed with glacial acetic acid, methanol, diethyl ether and dried. Both samples of CuCl displayed equal efficiency in the trifluoromethylation reaction and were stored under argon. Copper(I) bromide was prepared from CuBr₂ according to the procedure (B). Other copper salts were used as obtained from commercial suppliers.

Cyclopropanols **2a-f**, **2h-m** were prepared by the Kulinkovich cyclopropanation of the corresponding carboxylic esters.² Cyclopropanol **2g** and *cis*-1,2-disubstituted cyclopropanols **2n**, **2o** were prepared according to the general procedure described in ref.³ Enantiomerically enriched (66% *ee*) cyclopropanol (1*S*,2*S*)-**2n** was synthesized as described in ref.⁴

Silica gel 40 – 100 µm was used for column chromatography; silica gel 60 F₂₅₄ plates were used for TLC. ¹H NMR (400 MHz), ¹⁹F NMR (376 MHz) and ¹³C NMR (100.6 MHz) spectra were taken on a Bruker Avance III spectrometer. Chemical shifts were given in δ value with CHCl₃ (δ = 7.26) and CDCl₃ (δ = 77.16) as internal standards for ¹H NMR and ¹³C NMR spectra, respectively. FT-IR spectra were recorded on a Bruker Tensor 27 FT spectrometer. HPLC determination of *ee* was done on an Agilent Technologies 1200 series chromatograph. Specific rotation was measured using an Anton Paar MCP 500 polarimeter. HRMS data was obtained on an Agilent HPLC/Q-TOF G6540A Mass Spectrometer using APPI or APCI methods in positive or negative ion detection modes. GC-MS analysis was performed on a Shimadzu GCMS-QP2010 instrument.

II. Optimization of Reaction Conditions

The following procedure was used for the optimization runs: 1-propylcyclopropanol (**2a**, 5.0 mg, 0.05 mmol), Togni Reagent **3** or **4** (0.05 mmol), a copper catalyst and an additive were combined together in 0.5 mL of the corresponding solvent in a 1.5 mL vial and stirred overnight (ca. 20 h). The reaction mixture was diluted with water (~ 1-2 mL), extracted with CDCl₃ (~2 × 0.5 mL) and the extract was transferred into a NMR tube through a thin layer of MgSO₄ placed in a Pasteur pipette. Experiments with 5-10% of the catalyst load were run in 0.1 mmol scale, and the lower loads (1-3%) of the catalyst were made in 0.5 mmol scale. A number of the optimization experiments were also performed/reproduced in deuterated solvents and monitored directly without any work-up procedures. Experiments in the entries 13-28 were carried out under argon. The composition of the reaction mixtures was determined by ¹H NMR spectroscopy using an internal standard (for this purpose, a minor inert impurity in cyclopropanol **2a** was used) and for the selected runs (Entries 13, 20, 27-29, Table S1) yields were also confirmed by ¹⁹F NMR spectroscopy with 1,4-bis(trifluoromethyl)benzene as an internal standard. NMR spectra were acquired and processed using the parameters adequate for quantitative measurements. Typically 15-20 s interscan delays were used.

OH n-Pr		Reagent 3 or 4, Cat., Solvent		0		O U	(C III ~
		r.t.		<i>n</i> -Pr	n-Pr	+ <i>n</i> -Pr	√ `x	
	2a			1a		5a		
T I D			0.1	A 1114	Yields of products, %			
Entry	Reagent	Catalyst	Solvent	Additive	1a	5a	X = Hal	X = OMe
1	3	-	CD ₃ OD	-	no reaction			
2	4	-	CD ₃ OD	-	no reaction			
3	3	100% Zn(OTf) ₂	CDCl ₃	-	0	0 mixture of unidentified ketone		
							products	
4	3	20% CuCl ^[a]	CD ₃ OD	-	55	30	6	-
6	4	20% CuCl ^[a]	CD ₃ OD	-	23	60	4	-
7	3	20% CuCl ₂ ·2H ₂ O	CH ₃ OH	-	68	7	20	-
8	4	20% CuCl ₂ ·2H ₂ O	CH ₃ OH	-	12	40	10	-
9	3	20% CuSO ₄ ·5H ₂ O	CH ₃ OH	-	0	8	0	50
10	3	20% Cu(OAc) ₂ ·2H ₂ O	CH ₃ OH	-	0	10	0	45
11	3	20% Cu(OTf) ₂	CH ₃ OH	-	n.d. ^[b]	0	-	25
12	3	100% CuCl ₂ ·2H ₂ O	CH ₃ OH	-	25	15	60	-
13	3	20% CuCl ^[d]	CD ₃ OD	-	80 ^[c]	7	3	-
14	3	20% CuBr	CH ₃ OH	-	63 ^[e]	8	7	-
15	3	20% CuI	CH ₃ OH	-	30 ^[f]	5	-	-
16	3	20% CuCl ^[d]	CH ₃ OH	Xantphos	0	16	-	-
17	3	25% CuCl ^[d]	CH ₃ OH	1,10-Phen	35	30	-	-
18	3	20% CuCl ^[d]	CH ₃ OH	2 eq. LiCl	80	4	10	-
19	3	20% CuBr	CH ₃ OH	2 eq. LiBr	50	7	28	-
				53				

Table S1. Results of the optimization experiments

20	3	10% CuCl ^[d]	CD ₃ OD	1 eq. LiCl	80	4	7	-
21	3	10% CuCl ^[d]	CH ₃ OH	2 eq. LiCl	82	4	8	-
22	3	5% CuCl ^[d]	CH ₃ OH	1 eq. LiCl	83	4	5	-
23	3	10% CuCl ^[d]	CDCl ₃	1 eq. LiCl	70	0	4	-
24	3	10% CuCl ^[d]	CD ₃ CN	1 eq. LiCl	60	6	20	-
25	3	1% CuCl ^[d]	CH ₃ OH	0.5 eq. LiCl	78	4	4	-
26	3	1% CuCl ₂ ·2H ₂ O	CD ₃ OD	0.5 eq. LiCl	74	3	2	-
27	3	100% CuCl ^[d]	CD ₃ OD	5 eq. LiCl	67	7	14	-
28 ^[g]	3	3% CuCl ^[d]	CD ₃ OD	1 eq. LiCl	81	3	8	
29 ^[g]	3	3% CuCl ^[d]	CD ₃ OD	1 eq. LiCl	47	13	8	

[a] Unpurified "olive-green" CuCl was used. [b] Not determined due to the signal overlap in ¹H NMR. [c] ¹⁹F NMR spectrum of the reaction mixture also demonstrates the presence of some fluorine-containing impurities (~6%), $\delta = -64.9$ (t, J = 10.6 Hz), -67.2 (d, J = 8.8 Hz). [d] Purified "white" CuCl was employed. [e] 20% of the starting material **2a** remained. [f] 50% of the starting material **2a** remained. [g] The reactions were performed with cyclopropanol **2h** as a substrate. The experiment in the entry 28 was carried out under argon, while entry 29 was carried out under air.

The reaction given in the entry 26 was performed employing 55 mg (0.55 mmol) of cyclopropanol **2a**, 200 mg (0.61 mmol) of the reagent **3**, 1.0 mg (1 mol.%) of CuCl₂·2H₂O and 13 mg (0.30 mmol) of LiCl in 2 mL of methanol as described above. Volatile 1,1,1-trifluoroheptan-4-on (**1a**) was isolated in 52% yield (48 mg) according to the following procedure: the reaction mixture was diluted with water (2 mL) and extracted with pentane (4 × 2 mL). After the solvent evaporation, the residue was subjected to silica gel column chromatography (eluent – pentane/diethyl ether, 20:1). A colorless liquid. ¹H NMR (400 MHz, CDCl₃): δ = 2.67 (m, 2H), 2.47–2.37 (m, 4H), 1.63 (hex, *J* = 7.4 Hz, 2H), 0.93 (t, *J* = 7.4 Hz, 3H). ¹³C NMR (100.6 MHz, CDCl₃): δ = 207.26, 127.12 (q, *J* = 275.7 Hz), 44.84, 35.02 (q, *J* = 2.5 Hz), 28.03 (q, *J* = 29.8 Hz), 17.37, 13.79. ¹⁹F NMR (376 MHz, CDCl₃): δ = -66.63 (t, *J* = 10.9 Hz). IR (film): v = 2968, 1722, 1143. HRMS (APPI) calcd. for C₇H₁₂F₃O [M+H]⁺ 169.0835, found m/z 169.0834.

III. Synthesis and Characterization of β-Trifluoromethyl Ketones

General procedure: A 25 mL round-bottomed flask was charged with CuCl (3 mg, 0.03 mmol, 3 mol.%), LiCl (42.5 mg, 1 mmol), closed with a rubber septum and flushed with argon. Methanol (1 mL) was added and the mixture was stirred for 10 min to produce a clear colourless solution. Then a solution of cyclopropanol **2** (1 mmol) and reagent **3** (363 mg, 1.1 mmol) in methanol (3 mL) was added via syringe. At this point the reaction mixture usually warms up and attains paleblue color which disappears within 10-15 min. The solution was stirred until the reaction is completed (usually within 1 h; if necessary, the reaction mixture could be kept overnight without the change in yields) and then diluted with water (10 mL). The water phase was extracted with CH₂Cl₂ (4 × 5 mL), washed with brine and dried (MgSO₄). The solvent was evaporated and β-trifluoromethyl ketones **1** were isolated by silica gel column chromatography.



5,5,5-Trifluoro-1-phenylpentan-2-one (1b). Column chromatography eluent – petroleum ether/diethyl ether, 20:1. A colourless solid. Yield 67%. ¹H NMR (400 MHz, CDCl₃): $\delta = 7.38$ –

7.26 (m, 3H), 7.22–7.18 (m, 2H), 3.73 (s, 2H), 2.71 (m, 2H), 2.43–2.31 (m, 2H). ¹³C NMR (100.6 MHz, CDCl₃): δ = 204.72, 133.61, 129.47, 129.07, 127.50, 127.00 (q, *J* = 275.7 Hz), 50.19, 34.33 (q, *J* = 2.6 Hz), 28.09 (q, *J* = 29.8 Hz). ¹⁹F NMR (376 MHz, CDCl₃): δ = -66.58 (t, *J* = 10.9 Hz). IR (neat): v = 1721, 1257, 1143, 701. HRMS (APPI) calcd. for C₁₁H₁₀F₃O [M–H][–] 215.0689, found m/z 215.0685.



5,5,5-Trifluoro-1-(2-methyl-1,3-dioxolan-2-yl)pentan-2-one (1c). Column chromatography eluent – petroleum ether/ethyl acetate, 10:1. A colourless liquid. Yield 73%. ¹H NMR (400 MHz, CDCl₃): $\delta = 4.02$ –

3.94 (m, 4H), 2.82–2.78 (m, 4H), 2.46–2.34 (m, 2H), 1.40 (s, 3H). ¹³C NMR (100.6 MHz, CDCl₃): $\delta = 204.07$, 127.12 (q, J = 275.6 Hz), 107.88, 64.80, 51.92, 36.71 (q, J = 2.6 Hz), 27.95 (q, J = 29.8 Hz), 24.55. ¹⁹F NMR (376 MHz, CDCl₃): $\delta = -66.55$ (t, J = 10.9 Hz). IR (neat): v = 2988, 1719, 1259, 1141. HRMS (APCI) calcd. for C₉H₁₂F₃O₃ [M–H]⁻ 225.0744, found m/z 225.0744.



1-((tert-Butyldiphenylsilyl)oxy)-5,5,5-trifluoropentan-2-one

(1d). Column chromatography eluent – petroleum ether/diethyl ether, 20:1. A colourless liquid. Yield 67%. ¹H NMR (400 MHz, CDCl₃): δ = 7.65–7.62 (m, 4H), 7.48–7.38 (m, 6H), 4.20 (s, 2H), 2.83 (m, 2H), 2.44–2.32 (m, 2H), 1.11 (m, 9H). ¹³C NMR (100.6

MHz, CDCl₃): δ = 207.33, 135.64, 132.44, 130.28, 128.10, 127.08 (q, *J* = 275.7 Hz), 69.70, 31.51 (q, *J* = 2.8 Hz), 27.65 (q, *J* = 30.0 Hz), 26.88, 19.34. ¹⁹F NMR (376 MHz, CDCl₃): δ = -66.68 (t,

J = 10.8 Hz). IR (neat): v = 1730, 1258, 1113, 703. HRMS (APCI) calcd. for C₂₁H₂₄F₃O₂Si [M–H]⁻ 393.1503, found m/z 393.1500.

Cl 7-Chloro-1,1,1-trifluoroheptan-4-one (1e). Column **Cl CF**₃ chromatography eluent – petroleum ether/diethyl ether, 20:1. A colourless liquid. Yield 70%. ¹H NMR (400 MHz, CDCl₃): $\delta = 3.58$ (t, J = 6.2 Hz, 2H), 2.71 (m, 2H), 2.66 (t, J = 7.0 Hz, 2H), 2.48–2.36 (m, 2H), 2.07 (quint, J = 6.8 Hz, 2H). ¹³C NMR (100.6 MHz, CDCl₃): $\delta = 205.93$, 127.00 (q, J = 275.7 Hz), 44.35, 39.45, 35.22 (q, J = 2.6 Hz), 28.00 (q, J = 29.9 Hz), 26.24. ¹⁹F NMR (376 MHz, CDCl₃): $\delta = -66.65$ (t, J = 10.9 Hz). IR (neat): v = 1722, 1147. Anal. calcd. for C₇H₁₀ClF₃O: C 41.50, H 4.97%; found: C 41.19, H, 5.01%. GC-MS m/z 202 (M⁺), 204 (M⁺).



1-(5,5-Dimethyl-4,6-dioxaspiro[2.5]octan-7-yl)-4,4,4-

trifluorobutan-1-one (1f). Column chromatography eluent – petroleum ether/diethyl ether, 20:1. A colourless solid. Yield 68%. ¹H NMR (400 MHz, CDCl₃): $\delta = 4.51$ (dd, J = 12.0, 3.1 Hz, 1H), 2.98

(dt, J = 19.3, 7.8 Hz, 1H), 2.87 (dt, J = 19.3, 7.3 Hz, 1H), 2.47–2.33 (m, 2H), 2.15 (ddd, J = 13.5, 12.0, 1.7 Hz, 1H), 1.55 (s, 3H), 1.45 (s, 3H), 1.27 (dd, J = 13.5, 3.1 Hz, 1H), 0.87 (dddd, J = 10.8, 6.5, 4.7, 1.7 Hz, 1H), 0.78 (ddd, J = 10.8, 6.5, 5.4 Hz, 1H), 0.66 (ddd, J = 10.4, 6.5, 4.7 Hz, 1H), 0.46 (ddd, J = 10.4, 6.5, 5.4 Hz, 1H). ¹³C NMR (100.6 MHz, CDCl₃): $\delta = 207.78$, 127.16 (q, J = 275.7 Hz), 100.74, 74.25, 53.64, 33.47, 30.71 (q, J = 2.7 Hz), 29.77, 27.56 (q, J = 30.0 Hz), 21.10, 14.60, 10.18. ¹⁹F NMR (376 MHz, CDCl₃): $\delta = -66.54$ (t, J = 10.9 Hz). IR (KBr): v = 3095, 3013, 1725, 1199, 1155, 1122, 971. HRMS (APCI) calcd. for C₁₂H₁₆F₃O₃ [M–H]⁻ 265.1057, found m/z 265.1066.



4,4,4-Trifluoro-1-phenylbutan-1-one (**1g**). Column chromatography eluent – petroleum ether/diethyl ether, 20:1. A colourless solid. Yield 70%. ¹H NMR (400 MHz, CDCl₃): $\delta = 8.00-7.95$ (m, 2H), 7.63–7.58 (m,

1H), 7.51–7.47 (m, 2H), 3.26 (m, 2H), 2.66–2.54 (m, 2H). ¹³C NMR (100.6 MHz, CDCl₃): δ = 196.46, 136.24, 133.75, 128.92, 128.15, 127.30 (q, *J* = 275.7 Hz), 31.37 (q, *J* = 2.7 Hz), 28.50 (q, *J* = 29.7 Hz). ¹⁹F NMR (376 MHz, CDCl₃): δ = -66.43 (t, *J* = 10.8 Hz). IR (KBr): v = 1686, 1335, 1140, 750. HRMS (APPI) calcd. for C₁₀H₈F₃O [M–H]⁻ 201.0533, found m/z 201.0532.



6,6,6-Trifluoro-1-phenylhexan-3-one(1h).Columnchromatography eluent – petroleum ether/diethyl ether, 20:1. Acolourless liquid. Yield 72%. ¹H NMR (400 MHz, CDCl₃): δ =

7.31–7.26 (m, 2H), 7.22–7.16 (m, 3H), 2.92 (t, J = 7.5 Hz, 2H), 2.78 (t, J = 7.5 Hz, 2H), 2.64 (m,

2H), 2.45–2.33 (m, 2H). ¹³C NMR (100.6 MHz, CDCl₃): δ = 206.23, 140.68, 128.72, 128.40, 127.03 (q, *J* = 275.7 Hz), 126.44, 44.39, 35.31 (q, *J* = 2.6 Hz), 29.82, 27.99 (q, *J* = 29.9 Hz). ¹⁹F NMR (376 MHz, CDCl₃): δ = -66.63 (t, *J* = 10.9 Hz). IR (neat): v = 1721, 1256, 1145, 701. HRMS (APPI) calcd. for C₁₂H₁₂F₃O [M–H]⁻ 229.0846, found m/z 229.0846.



1-(Cyclohex-1-en-1-yl)-4,4,4-trifluorobutan-1-one (1i). Column chromatography eluent – petroleum ether/diethyl ether, 20:1. A colourless liquid. Yield 68%. ¹H NMR (400 MHz, CDCl₃): $\delta = 6.95$ (m, 1H), 2.91

(m, 2H), 2.50–2.38 (m, 2H), 2.31–2.20 (m, 4H), 1.69–1.58 (m, 4H). ¹³C NMR (100.6 MHz, CDCl₃): δ = 197.36, 140.90, 138.95, 127.40 (q, *J* = 275.7 Hz), 29.70 (q, *J* = 2.6 Hz), 28.67 (q, *J* = 29.5 Hz), 26.24, 23.25, 21.98, 21.62. ¹⁹F NMR (376 MHz, CDCl₃): δ = -66.51 (t, *J* = 11.0 Hz). IR (neat): v = 2937, 1672, 1138. HRMS (APPI) calcd. for C₁₀H₁₄F₃O [M+H]⁺ 207.0991, found m/z 207.0992.

OEt O EtO CF_3 chromatography eluent – petroleum ether/acetone, 20:1. A colourless liquid. Yield 68%. ¹H NMR (400 MHz, CDCl₃): $\delta = 4.87$ (t, J = 5.6 Hz, 1H), 3.71–3.63 (m, 2H), 3.56–3.49 (m, 2H), 2.77–2.74 (m, 4H), 2.46–2.34 (m, 2H), 1.19 (t, J = 7.0 Hz, 6H). ¹³C NMR (100.6 MHz, CDCl₃): $\delta = 204.32$, 127.05 (q, J = 275.6 Hz), 99.96, 62.64, 47.69, 36.47 (q, J = 2.6Hz), 27.80 (q, J = 29.9 Hz), 15.31. ¹⁹F NMR (376 MHz, CDCl₃): $\delta = -66.65$ (t, J = 10.9 Hz). IR (neat): v = 2980, 1723, 1257, 1137, 1062. HRMS (APPI) calcd. for C₁₀H₁₆F₃O₃ [M–H]⁻ 241.1057, found m/z 241.1051.



1,1,1,16,16,16-Hexafluorohexadecane-

4,13-dione (1k). Column chromatography eluent – petroleum ether/diethyl ether,

20:1. A colourless solid. Yield 65%. ¹H NMR (400 MHz, CDCl₃): $\delta = 2.67$ (m, 4H), 2.49–2.33 (m, 8H), 1.65–1.53 (m, 4H), 1.33–1.20 (m, 8H). ¹³C NMR (100.6 MHz, CDCl₃): $\delta = 207.33$, 127.12 (q, J = 275.7 Hz), 42.91, 35.01 (q, J = 2.6 Hz), 29.26, 29.17, 28.05 (q, J = 29.8 Hz), 23.81. ¹⁹F NMR (376 MHz, CDCl₃): $\delta = -66.63$ (t, J = 10.8 Hz). IR (KBr): v = 2933, 2852, 1706, 1327, 1250, 1152. HRMS (APCI) calcd. for C₁₆H₂₃F₆O₂ [M–H]⁻ 361.1608, found m/z 361.1604.

1,1,1-Trifluorodecan-4-one (**1l**). Column chromatography CF_3 eluent – petroleum ether/diethyl ether, 20:1. A colourless liquid. Yield 71%. ¹H NMR (400 MHz, CDCl₃): δ = 2.67 (m, 2H), 2.46–2.35 (m, 4H), 1.66–1.53 (m, 2H), 1.37–1.23 (m, 6H), 0.88 (t, *J* = 7.0 Hz, 3H). ¹³C NMR (100.6 MHz, CDCl₃): δ = 207.41, 127.13 (q, *J* = 275.7 Hz), 42.99, 34.99 (q, *J* = 2.6 Hz), 31.68, 28.96, 28.06 (q, *J* = 29.7 Hz), 23.88, 22.61, 14.15. ¹⁹F NMR (376 MHz, CDCl₃): δ = -66.65 (t, *J* = 10.9 Hz). IR (neat): v = 2933, 1723, 1144. HRMS (APPI) calcd. for C₁₀H₁₈F₃O [M+H]⁺ 211.1304, found m/z 211.1308.

O CF₃ **5,5,5-Trifluoro-4-phenylpentan-2-one** (**1m**). Column chromatography eluent Ph – petroleum ether/diethyl ether, 20:1. A colourless solid. Yield 65%. ¹H NMR (400 MHz, CDCl₃): $\delta = 7.37-7.29$ (m, 5H), 4.01 (m, 1H), 3.14–3.02 (m, 2H), 2.12 (s, 3H). ¹³C NMR (100.6 MHz, CDCl₃): $\delta = 203.81$, 134.54 (q, J = 1.8 Hz), 129.08, 128.86, 128.50, 126.83 (q, J = 279.4 Hz), 44.72 (q, J = 27.6 Hz), 43.13 (q, J = 1.8 Hz), 30.55. ¹⁹F NMR (376 MHz, CDCl₃): $\delta = -69.97$ (d, J = 9.6 Hz). IR (film): v = 1717, 1252, 1105, 701. HRMS (APPI) calcd. for C₁₁H₁₀F₃O [M–H]⁻ 215.0689, found m/z 215.0695.



1-(Benzyloxy)-3-ethyl-5,5,5-trifluoropentan-2-one (**1na**) **and 1-** (**benzyloxy)-4-(trifluoromethyl)hexan-2-one** (**1nb**). The reaction was performed twice: with the racemic *cis*-**2n** and the enantiomerically enriched (66% ee) cyclopropanol (1*S*,2*S*)-**2n**. 1.35 equiv. of Togni reagent **3** was

^{BnO} CF₃ used to achieve full conversion of the starting material. The ratio of the products is **1na**:**1nb** = 55:45 according to ¹⁹F and ¹H NMR spectroscopy. The mixture of ketones **1na**, **1nb** was isolated by column chromatography (eluent – petroleum ether/diethyl ether, 20:1). A colourless liquid. Yield 50%. The analysis of the enantiomeric composition was made by HPLC using Phenomenex Lux 3µ Amylose-2 column (95:5 *n*-hexane/2-propanol, flow rate 1.0 mL·min⁻¹, detection at 210 nm). $[\alpha]_D^{25} = -20.5$ (*c* 0.57, hexane). ¹H NMR (400 MHz, CDCl₃): $\delta = 7.40-7.28$ (m, 5H), 4.64–4.57 (m, 2H), 4.14 (s, 1.1H), 4.07 (s, 0.9H), 2.99 (m, 0.55H), 2.86–2.52 (m, 1.9H), 2.11 (m, 0.55H), 1.75–1.65 (m, 1H), 1.57–1.37 (m, 1H), 0.95 (t, *J* = 7.5 Hz, 1.3H), 0.90 (t, *J* = 7.5 Hz, 1.7H). ¹³C NMR (100.6 MHz, CDCl₃): $\delta = 209.14$, 205.72, 137.16, 137.03, 128.74, 128.69, 128.32, 128.22, 128.09, 128.03, 126.66 (q, *J* = 276.6 Hz), 75.15, 74.80, 73.66, 73.50, 42.28 (q, *J* = 2.3 Hz), 38.96 (q, *J* = 26.1 Hz), 37.30 (q, *J* = 2.3 Hz), 34.05 (q, *J* = 28.6 Hz), 25.13, 21.70, 21.67, 11.32, 11.09. ¹⁹F NMR (376 MHz, CDCl₃): $\delta = -64.92$ (t, *J* = 11.0 Hz), -70.86 (d, *J* = 9.0 Hz). IR (neat): v = 2971, 1734, 1264, 1122, 741. HRMS (APPI) calcd. for C₁₄H₁₆F₃O₂ [M–H]⁻ 273.1108, found m/z 273.1103.



1-Chloro-5-(2,2,2-trifluoroethyl)nonan-4-one (10a). Column chromatography eluent – petroleum ether/diethyl ether, 20:1. A colourless liquid. Yield 33%. ¹H NMR (400 MHz, CDCl₃): δ = 3.58 (t, *J* = 6.3 Hz, 2H), 2.88–2.58 (m, 4H), 2.16–2.02 (m, 3H),

1.68–1.59 (m, 1H), 1.49–1.37 (m, 1H), 1.32–1.17 (m, 4H), 0.90 (t, J = 7.1 Hz, 3H). ¹³C NMR (100.6 MHz, CDCl₃): $\delta = 210.51$, 126.66 (q, J = 276.8 Hz), 45.25 (q, J = 2.2 Hz), 44.36, 39.54, 34.80 (q, J = 28.6 Hz), 32.03, 28.95, 26.14, 22.67, 13.92. ¹⁹F NMR (376 MHz, CDCl₃): $\delta = -65.00$

(t, J = 10.9 Hz). IR (neat): v = 2961, 1719, 1257, 1155, 1131. HRMS (APPI) calcd. for $C_{11}H_{19}ClF_{3}O[M+H]^+ 259.1071$, found m/z 259.1066.

C F_3 **1-Chloro-6-(trifluoromethyl)decan-4-one (1ob).** Column chromatography eluent – petroleum ether/diethyl ether, 20:1. A colourless liquid. Yield 12%. ¹H NMR (400 MHz, CDCl₃): $\delta = 3.58$ (t, J = 6.2 Hz, 2H), 2.84 (m, 1H), 2.73 (dd, J = 17.7, 5.4 Hz, 1H), 2.65 (t, J = 7.0 Hz, 2H), 2.47 (dd, J = 17.7, 6.7 Hz, 1H), 2.07 (m, 2H), 1.65 (m, 1H), 1.39–1.19 (m, 5H), 0.90 (t, J = 6.8 Hz, 3H). ¹³C NMR (100.6 MHz, CDCl₃): $\delta = 206.18, 128.22$ (q, J = 279.6 Hz), 44.37, 41.45 (q, J = 2.3 Hz), 39.87, 38.09 (q, J = 26.1 Hz), 28.97, 28.37 (q, J = 2.1 Hz), 26.25, 22.73, 13.94. ¹⁹F NMR (376 MHz, CDCl₃): $\delta = -71.11$ (d, J = 9.4 Hz). IR (neat): v = 2960, 1722, 1260, 1167, 1131. HRMS (APPI) calcd. for C₁₁H₁₉ClF₃O [M+H]⁺ 259.1071, found m/z 259.1072.



IV. Selected ¹H NMR and ¹⁹F NMR Spectra of the Reaction Mixtures

¹⁹F NMR, entry 20, CD₃OD



BHT inhibition experiments





V. Copies of ¹H, ¹³C and ¹⁹F NMR Spectra of β-Trifluoromethyl Ketones









S29

S30

VI. HPLC Chromatograms of the Compound 1nb

The analysis of the enantiomeric composition for the compound **1nb** was made by HPLC using Phenomenex Lux 3μ Amylose-2 column (95:5 *n*-hexane/2-propanol, flow rate 1.0 mL·min⁻¹, detection at 210 nm).

VII. Copies of ¹H and ¹³C NMR Spectra of Cyclopropanols

.0 4.5 f1 (ppm) 8.5 8.0 7.0 6.0 5.0 3.5 3.0 2.5 1.5 7.5 6.5 5.5 4.0 2.0

0.5

0.

VIII. References

- 1. V. Matoušek, E. Pietrasiak, R. Schwenk, A. Togni, J. Org. Chem. 2013, 78, 6763-6768.
- 2. J. K. Cha, O. G. Kulinkovich, Org. React. 2012, 77, 1–160.
- 3. O. G. Kulinkovich, D. G. Kananovich, Eur. J. Org. Chem. 2007, 2121-2132.
- 4. O. G. Kulinkovich, D. G. Kananovich, M. Lopp, V. Snieckus, *Adv. Synth. Catal.* **2014**, *356*, 3615-3626.