## Electronic Supplementary Information (ESI) for

# Self-assembly Induced Solubilization of Drug-like Molecules in Nanostructured Ionic Liquids

#### 1. Materials

Tetrabutylphonium hydroxide (40% in water) was product of TCI, octanoic acid (99%), decanoic acid (99%), dodecylic acid (98%), myristic acid (98%), palmitic acid (97%), stearic acid (98%) and cholesterol ( $\geq$ 95%, GC), folic acid( $\geq$ 97%, HPLC), indomethacin (99%), hydrocortisone (98%) were purchased from Aladdin Reagent, 98% stigmasterol was from Xi'an Vita-Solar Biotechnology Co., Ltd, cholesterol (≥99%, GC) and DL-naproxen(≥99%) were purchased from Sigma and Xiva Reagent respectively, Vitamin  $D_3$  ( $\geq$ 99%) was kindly supplied by Zhejiang Garden Biochemical High-tech Co., Ltd., China and used as received. Methanol (99.7%, HPLC), ethanol (99.7%, AR), ethyl acetate (99.5%, AR), dimethyl sulfoxide (99.0%, AR), 1-octanol (99.0%, AR), acetonitrile (99.0%, AR), sodium chloride (AR), potassium chloride (AR), disodium hydrogen phosphate (AR) and potassium dihydrogen phosphate (AR) were obtained from Sinopharm Chemical Reagent Group Co. Ltd and used without purification. The deionized water was obtained from the Wahaha Group Co. Ltd. The conventional ILs used in this study were purchased from Lanzhou Green-chem ILS, LICP, CAS, China, including 1-butyl-3methylimidazolium tetrafluoroborate  $([BMIm][BF_4]],$ 99%), 1-butyl-3-methylimidazolium hexafluorophosphate  $([BMIm][PF_6]],$ 99%), 1-butyl-3-methylimidazolium bis(trifluoromethylsulfonyl)imide ([BMIm]Tf<sub>2</sub>N, 99%), N-butyl pyridinium bis(trifluoromethyl sulfonyl)imide ([Bpy]Tf<sub>2</sub>N, 99%), 1-ethyl-3- methylimidazolium ethylsulfate ([EMIm]EtOSO<sub>3</sub>, 99%), 1butyl-3-methylimidazolium hydrogen sulfate ([BMIm]HSO<sub>4</sub>, 99%), 1-butyl-3-methylimidazolium trifluoromethansulfonate  $([BMIm]CF_3SO_3,$ 99%), 1-butyl-3-methylimidazolium acetate ([BMIm][CH<sub>3</sub>COO], 99%), 1-octyl-3-methylimidazolium bromide ([OMIm]Br, 99%), 1-octyl -3methylimidazolium hexafluorophosphate ([OMIm]PF<sub>6</sub>, 99%), with water contents of these ILs below 0.6% (mass fraction).

#### 2. Synthesis of LCC-ILs

Ionic liquids of ( $[P_{4444}][C_nH_{2n+1}COO]$ , n= 7, 9, 11, 13, 15) were synthesized according to the similar procedures reported in the literature.<sup>1,2</sup> Tetrabutylphonium hydroxide and carboxylic acid were mixed in a 100mL round-bottom flask with a molar ratio of 1:1 and then reacted at 40°C for 30 h. Then the produced ILs were distilled under a high vacuum at 55°C for 12 h to remove the water. The purity of carboxylates was ascertained by the <sup>1</sup>H NMR spectrum in CDCl<sub>3</sub>.

#### 3. Solvatochromic measurements

The Kamlet-Taft hydrogen-bond basicity  $\beta$  were measured by solvatochromic experiments,<sup>3,4</sup> using 4nitroaniline and N,N-diethyl-4-nitroaniline as probe. Appropriate amounts of probes molecule was added into the LCC-IL sample and the IL was mixed throughly. Then the IL sample was transferred into a quartz colorimetric cell with 2mm light-path length. The maximum absorption wavelength ( $\lambda_{max}$ ) was recorded at 25°C by UV-vis absorbtion measurements. Every sample was repeated at least five times and the average value was taken as the final one. The Kamlet-Taft parameters dipolarity/ polarizability ( $\pi^*$ ) and the hydrogen bonding basicity ( $\beta$ ) were calculated by using equations (1) and (2),

$$\pi^* = 8.649 - 0.314\nu(1)_{\text{max}}$$
(1)  
$$\beta = [1.035\nu(2)_{\text{max}} - \nu(1)_{\text{max}} + 2.64] / 2.80$$
(2)



Fig. S1 Molecular structures of LCC-ILs.



1-butyl-3-methylimidazolium tetrafluoroborate [BMIm]BF<sub>4</sub>



1-butyl-3-methylimidazolium bis(trifluoromethylsulfonyl)imide  $[BMIm]Tf_2N$ 



1-ethyl-3-methylimidazolium bis(trifluoromethylsulfonyl)imide [EMIm]EtOSO<sub>3</sub>



1-butyl-3-methylimidazolium trifluoromethansulfonate [BMIm]CF<sub>3</sub>SO<sub>3</sub>



1-octyl-3-methylimidazolium bromide [OMIm]Br



1-butyl-3-methylimidazolium hexafluorophosphate [BMIm]PF<sub>6</sub>



N-butyl pyridinium bis(trifluoromethylsulfonyl)imide [BPy]Tf<sub>2</sub>N



1-butyl-3-methylimidazolium hydrogen sulfate [BMIm]HSO<sub>4</sub>



1-butyl-3-methylimidazolium acetate [BMIm][CH<sub>3</sub>COO]



1-octyl-3-methylimidazolium hexafluorophosphate [OMIm]PF<sub>6</sub>





**Fig. S3** Mole solubility of six poorly soluble H-bond donor drugs in  $[P_{4444}][C_{15}H_{31}COO]$  as a function of temperature.

Solute	Solvent				
	Water	Ethanol	DMSO	1-Octanol	
Cholesterol	<0.000001 <sup>[a]</sup>	0.0038 <sup>[e]</sup>	0.0012 <sup>[d]</sup>	0.047 <sup>[i]</sup>	
Naproxen	<0.00003 <sup>[b]</sup>	0.015 <sup>[f]</sup>	0.079 <sup>[d]</sup>	0.020 <sup>[i]</sup>	
Folic acid	<0.000001 <sup>[c]</sup>	0.00007 <sup>[d]</sup>	1.26 <sup>[d]</sup>	0.0095 <sup>[d]</sup>	
Hydrocortisone	<0.00002 <sup>[c]</sup>	0.0024 <sup>[g]</sup>	0.0078 <sup>[d]</sup>	0.0025 <sup>[j]</sup>	
Indomethacin	<0.000001 <sup>[c]</sup>	0.0034 <sup>[h]</sup>	0.10 <sup>[d]</sup>	0.0053 <sup>[i]</sup>	
Vitamin D <sub>3</sub>	<0.00001 <sup>[d]</sup>	0.23 <sup>[d]</sup>	0.33 <sup>[d]</sup>	38.18 <sup>[d]</sup>	
Stigmasterol	<0.00001 <sup>[d]</sup>	0.0096 <sup>[d]</sup>	0.0037 <sup>[d]</sup>	1.89 <sup>[d]</sup>	

Table S1 Molar solubility of six drug-like molecules in water and organic solvents at 25°C.

[a] Ref. 5; [b] Ref. 6; [c] Ref. 7; [d] this work; [e] Ref. 8; [f] Ref. 9; [g] Ref. 10; [h] Ref. 11; [i] Ref. 12; [j] Ref. 13



**Fig. S4** Solubility (mg/ml) of drug-like molecules: a. cholesterol, b. hydrocortisone, c. naproxen, d. indomethacin, e. stigmasterol, f. vitamin D<sub>3</sub>, g. folic acid, in water/ $[P_{4444}][C_{15}H_{31}COO]$  and PBS/ $[P_{4444}][C_{15}H_{31}COO]$  mixtures as a function of mass percent of  $[P_{4444}][C_{15}H_{31}COO]$  at 35°C.

Common surfactant		LCC IL		
Туре	Solubility	Ref.	[P <sub>4444</sub> ][C <sub>15</sub> H <sub>31</sub> COO]	Solubility
quillaja saponin (2.5 wt%)	0.004 wt%	14	2.5wt%	0.141 wt%
Tween 60+ethanol R-(+)-				
limonene+propylene glycol	0.6 wt%	15	30wt%	17.04 wt%
(30 wt%)				
glyceryl-1-monooctanoate	18 20 xx49/	16	94wt%	43.03 wt%
(94 wt%)	18-20 wt%			
Sodium taurodeoxycholate	0.5.0.0	17	15-25mM 6	( 04 7 40m) M
(15-25mM)	0.3-0.911111			0.94-7.40IIIM
sodium cholate (60mM)	3mM	18	60mM	11.77mM

**Table S2** Solubility of cholesterol in common microemulsion/micelle and  $[P_{4444}][C_{15}H_{31}COO]$  aqueoussolution.



Fig. S5 WXRD pattern of  $[P_{4444}][C_{15}H_{31}COO]/Cholesterol with a molar ratio of 9 : 1 at 30°C.$ 



**Fig. S6** POM image (a) and WXRD (b) pattern of  $[P_{4444}][C_{11}H_{23}COO]$ /cholesterol with a molar ratio of 1 : 4 at 40°C.



Fig. S7 IR spectrum of cholesterol and cholesterol dissolved in  $[P_{4444}][C_nH_{2n+1}COO]$  (n= 7, 9, 11, 13, 15)



Fig. S8 <sup>1</sup>H NMR spectrum of cholesterol and  $[P_{4444}][C_{15}H_{31}COO]$ /cholesterol system.



**Fig. S9** POM image (a) and WXRD pattern (b) of  $[P_{4444}][C_{15}H_{31}COO]/stigmasterol with a molar ratio of 2 : 1 at 35°C. The scale bar is 150 <math>\mu$ m.



**Fig. S10** POM image (a) and WXRD (b) pattern of  $[P_{4444}][C_{15}H_{31}COO]/vitamin D_3$  with a molar ratio of 3 : 5 at 25°C. The scale bar is 150 µm.



**Fig. S11** Visual depiction of anti-solvents induced reversible transitions. Photograph of the  $[P_{4444}][C_{15}H_{31}COO]/cholesterol sample (a) before adding acetonitrile and (b) after adding acetonitrile.$ 

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