

Imidazolium salts modified porous hypercrosslinked polymers for a synergistic CO₂ capture and conversion

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1. General information

The 1-Methylimidazole, 1, 4-bis (chloromethyl) benzene, 1, 2-bis (chloromethyl) benzene and iron chloride were provided by Sigma-Aldrich. The chloromethyl polystyrene was purchased from Fluka, and the epoxides were purchased from the VWR international. GC-MS were measured on SHIMADZU-QP2010. GC analyses were performed on an Agilent GC-6890 using a flame ionization detector. NMR spectra were recorded on a Bruker 400. N₂ sorption analysis and CO₂ sorption analysis were performed on a Micromeritics Tristar 3000 (77 and 273 K, respectively). TEM experiments were conducted on a FEI Tecnai G² F20 electron microscope (200 kV). TGA was performed on a Perkin–Elmer Pyris-1 thermogravimetric analyzer. Elemental analysis (CHNS) was performed on an Elementarvario MICRO cube. FT-IR experiments were performed on a Perkin Elmer Spectrum 100. Solid-¹³C NMR experiments were conducted at a Bruker Avance 400 (DRX400) with CP/MAS.

The calculations were carried out by performing DFT by use of the B3PW91functional with the 6-31++G (d, p) basis set as implemented in Gaussian 03 program package. The solvent effect uses the Conductor Polarizable Continuum Model (CPCM) in each case. Vibrational frequency calculations, from which the zero-point energies were derived, have been performed for each optimized structure at the same level to identify the natures of all the stationary points. All the bond lengths are in angstroms (Å). Structures were generated using CYLview.¹

1. CYLview, 1.0b; C. Y. Legault, Université de Sherbrooke, 2009 (<http://www.cylview.org>).

The CO₂ experiments were performed on a Belsorp-mimill at 273 and 298 K. Before each measurement, the samples were heated at 150 °C in vacuum for 24 h. TGA gas capture experiments were conducted on a Perkin-Elmer Pyris-1 thermogravimetric analyzer. Porous carbons (5 mg) were subjected to the following gas capture and cycling experiment at 25 °C: CO₂ (99.8%) gas flow at 20 mL min⁻¹ for 30 min, followed by N₂ (99.9995%) gas flow at 20 mL min⁻¹ for 45 min. Changes in weight were recorded by using TGA. Prior to the cyclic treatment, the sample was first purged under N₂ gas flow at 200 °C for 60 min, followed by cooling to room temperature. Change in buoyancy effects arising from the switching of gases was recorded by using an empty sample pan, and the buoyancy effects were corrected for in the TGA results. For the adsorption kinetics analysis, the porous carbon was first purged under Ar gas flow (20 mL min⁻¹) at 200 °C for 60 min, followed by cooling to room temperature. The gas was then switched from Ar to CO₂ or N₂ (20 mL min⁻¹). The selectivity of CO₂ over N₂ is calculated by the saturated absorption according to reported work by Fuertes.²

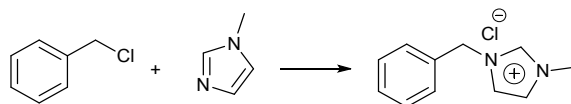
2. M. Sevilla and A. B. Fuertes, *Energy Environ. Sci.*, 2011, 4, 1765;

2. Synthesis

2.1 Material synthesis

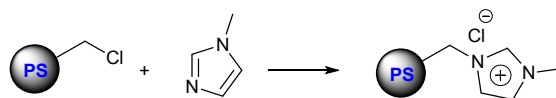
The benzyl halide-functionalized organic polymers were synthesized according to the reported method.¹⁰ Typically, iron (III) chloride (120 mmol) was added to a solution of benzyl halide compound (60 mmol) in anhydrous dichloroethane (80 ml). The resulting mixture was heated at 80 °C for 24 h. When the reaction was completed, the solid product was centrifuged and washed with methanol (3×20 ml). The product was further purified by Soxhlet extraction in methanol for 20 h and dried in vacuum at 60 °C for 24 h. The polymers were obtained in quantitative yields. The content of chloride or bromide in the obtained polymers was determined by elemental analysis (Table S2). The polymers were further reacted with N-methylimidazole (molar ratio of Cl : N-methylimidazole = 1 : 2) in 20 ml toluene at 80 °C for 24 h, the resultant supported imidazolium salts were washed with methanol (3×20 ml) and dried in *vacuum* at 60 °C for 24 h. From elemental analysis results, the modification was not completed for most of samples (ESI).

2.2 Synthesis imidazolium salt



a mixture of benzyl chloride (12 mmol), 1-methylimidazole (10 mmol) and toluene (10 mL) was heated at 80 °C for 24 h in a 25 mL flask with vigorous stirring. After cooled down to room temperature, the solid residue washed with benzene (3×5 mL) and ethyl acetate (3×5 mL). Then, the solid was dried under vacuum at 60 °C for 12 h and the imidazolium salt was obtained. ¹H NMR (CDCl₃, TMS, 400 MHz): δ 10.05 (s, 1H), 7.46 (s, 1H), 7.29 (m, 3H), 7.09 (m, 3H), 5.35 (s, 2H), 3.70 (s, 3H); ¹³C NMR (CDCl₃, TMS, 100.4 MHz): δ 137.36, 133.58, 129.17, 128.79, 124.00, 122.22, 52.76, 36.29.

2.3 Synthesis of polystyrene resin supported imidazolium salt



Polystyrene resin supported imidazolium salt was according to reported method²: a mixture of chloromethyl polystyrene (1.0 g, 5.5 mmol Cl content), 1-methylimidazole (16.5 mmol) and toluene (10 mL) was heated at 80 °C for 24 h in a 25 mL flask with vigorous stirring. After cooled down to room temperature, the solid residue was collected by filtration and washed with methanol (3×5 mL). Then, the solid was dried under vacuum at 60 °C for 12 h and polystyrene resin supported imidazolium salt was obtained. The loading of imidazolium salt attached on the PS was 3.6 mmol/g determined by nitrogen content from elementary analysis.

2. J. Sun, W. G. Cheng, W. Fan, Y. H. Wang, Z. Y. Meng, and S. J. Zhang, *Catal. Today*, 2009, **148**, 361-367.

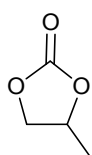
2.4 CO₂ capture

Imidazolium salt-modified porous hypercrosslinked polymers were subjected to the following gas capture and cycling experiment at 25 °C: CO₂ (99.8%) gas flow at 20 ml/min for 30 min, followed by N₂ (99.9995%) gas flow at 20 ml/min for 45 min. Changes in weight were recorded by TGA. Prior to the cyclic treatment, the sample was first purged under N₂ gas flow at 100 °C for 60 min, followed by cooling to room temperature. Change in buoyancy effects arising from the switching of gases was recorded by using an empty sample pan, and the buoyancy effects were corrected. For the adsorption kinetics analysis of CO₂ and N₂, the porous supported imidazolium salt was first purged under Ar gas flow (20 ml/min) at 100 °C for 60 min, followed by cooling to room temperature. The gas was then switched from Ar to CO₂ or N₂ (20 ml/min).

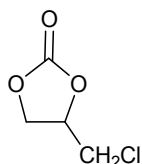
2.5 CO₂ conversion

CO₂ conversion reactions were conducted in a 50 ml stainless steel reactor equipped with a magnetic stirrer and automatic temperature control system. Typically, an appropriate volume of CO₂ (1.0 MPa) was added to a mixture of PO (0.1 ml), ethanol (2 ml), porous supported imidazolium salt (5 mmol% based on contents of the imidazolium salt) in the reactor at room temperature. The temperature was then raised to 120 °C. After the reaction was preceded for 4 h, the reactor was cooled to 0 °C in an ice water bath, and the remaining CO₂ was slowly removed. The product was then analyzed by GC and NMR. The porous supported imidazolium salts could be easily separated by centrifugation, and used in the next run without further purification.

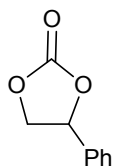
3. Characterization of cyclic carbonates



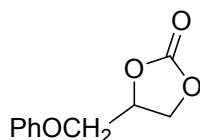
¹H NMR (CDCl₃, TMS, 400 MHz): δ 4.86-4.94 (m, 1H), 4.60 (t, *J*=8.0 Hz, 1H), 4.05 (t, *J*=8.8 Hz, 1H), 1.49 (d, *J*=6.0 Hz, 3H); ¹³C NMR (CDCl₃, TMS, 100.4 MHz): δ 154.95 (C=O), 73.51, 70.46, 18.95.



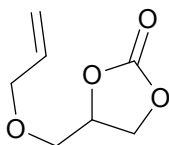
¹H NMR (CDCl₃, TMS, 400 MHz): δ 4.98 (m, 1H), 4.58 (t, 1H, *J*=8.4 Hz), 4.39 (dd, 1H, *J*=6.0 Hz), 3.80 (dd, 1H, *J*=5.2 Hz), 3.71 (dd, 1H, *J*=3.2 Hz); ¹³C NMR (CDCl₃, TMS, 100.4 MHz): δ 154.95 (C=O), 73.51, 70.46, 43.83.



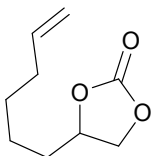
^1H NMR (CDCl_3 , TMS, 400 MHz): δ 7.35-7.44 (m, 5H), 5.68 (t, 1H, $J=8.0$ Hz), 4.80 (t, 1H, $J=8.4$ Hz), 4.34 (t, 1H, $J=8.4$ Hz); ^{13}C NMR (CDCl_3 , TMS, 100.4 MHz): δ 154.81 (C=O), 135.70, 129.63, 129.12, 125.81, 77.92, 71.10.



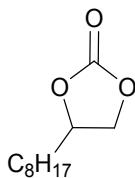
^1H NMR (CDCl_3 , 400 MHz): δ 7.31 (t, $^3J=8.0$ Hz, 2H), 7.02 (t, $^3J=7.4$ Hz, 1H), 6.91 [d, $^3J=8.0$ Hz, 2H], 5.03 [m, 1H], 4.62 [t, $^3J=8.4$ Hz, 1H], 4.55 (dd, $^3J=8.4$ Hz), 4.24 (dd, $^3J=3.6$ Hz, 1H), 4.15 (dd, $^3J=4.4$ Hz, 1H), ^{13}C NMR (CDCl_3 , TMS, 100.4 MHz): δ 157.71, 154.65, 129.62, 121.92, 114.57, 74.11, 68.84, 66.17.



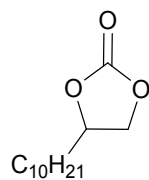
^1H NMR (CDCl_3 , 400 MHz): δ 5.8 – 6.0 (t, 1H), 5.0-5.2 (d, 2H), 4.4-4.7 (3H), 4.0-4.1 (m, 2H), 3.5-3.8 (d, 2H). ^{13}C NMR (CDCl_3 , TMS, 100.4 MHz): δ 155.90, 133.88, 117.47, 72.05, 72.30, 68.92, 66.24.



^1H NMR (CDCl_3 , TMS, 400 MHz): δ 5.8 – 6.0 (t, 1H), 5.0-5.2 (d, 2H), 4.7 (m, 1H), 4.5 (dd, 1H), 4.04 (dd, 1H, $J=8.4$ Hz), 2.4-2.7 (m, 2H), 2.1 (m, 2H), 1.43-1.55 (m, 4H); ^{13}C NMR (CDCl_3 , TMS, 100.4 MHz): δ 155.2 (C=O), 139.2, 115.6, 70.4, 66.3, 36.5, 33.8, 29.6, 25.3.



^1H NMR (CDCl_3 , TMS, 400 MHz): δ 4.6-4.8 (m, 1H), 4.50 (dd, 1H, $J=8.4$ Hz), 4.04 (dd, 1H, $J=8.4$ Hz), 1.6-1.9 (m, 2H), 1.1-1.6 (m, 12H), 0.86 (t, 2H); ^{13}C NMR (CDCl_3 , TMS, 100.4 MHz): δ 155.2 (C=O), 77.1, 69.5, 34.0, 31.9, 29.4, 29.2, 29.2, 24.5, 22.7, 14.2.



^1H NMR (CDCl_3 , TMS, 400 MHz): δ 4.67-4.72 (m, 1H), 4.52 (dd, 1H), 4.07 (dd, 1H), 1.77-1.89 (m, 1H), 1.67-1.69 (m, 1H), 1.42-1.50 (m, 1H), 1.22-1.36 (m, 15H), 0.86 (t, 3H); ^{13}C NMR (CDCl_3 , TMS, 100.4 MHz): δ 155.6 (C=O), 76.7, 69.4, 33.9, 31.8, 29.5, 29.4, 29.3, 29.2, 29.1, 24.3, 22.6, 14.1.

4. Supported Figures

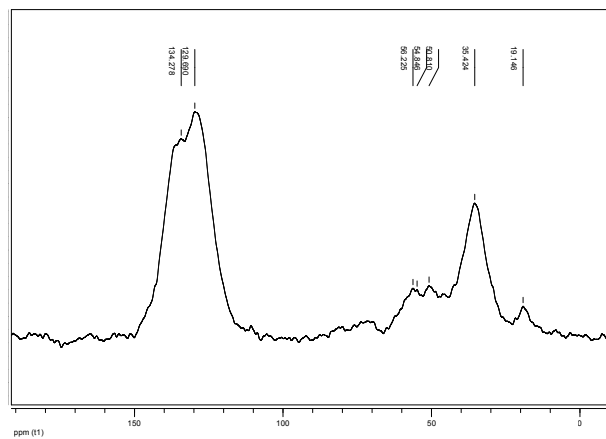


Fig. S1 Solid-¹³C NMR spectrum for POM1-IM

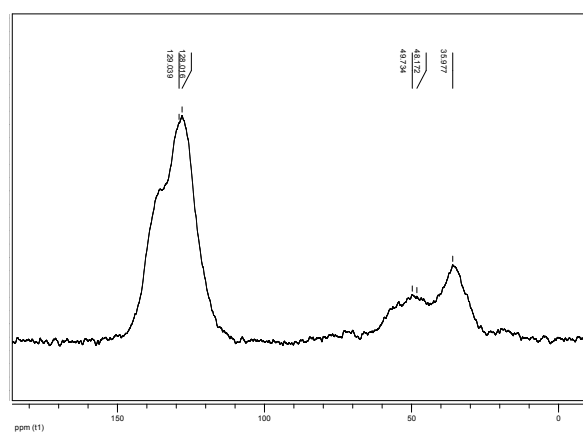


Fig. S2 Solid-¹³C NMR spectrum for POM2-IM

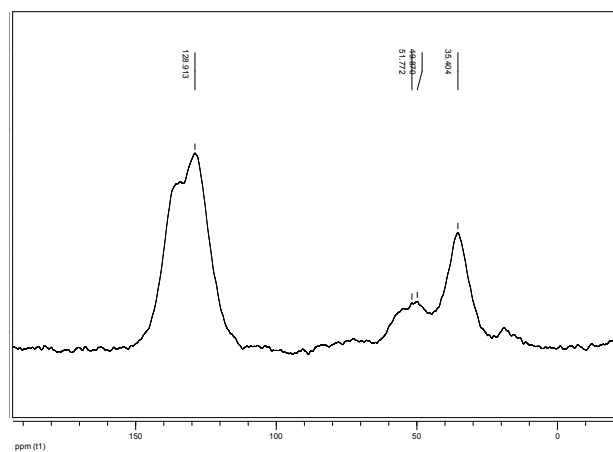


Fig. S3 Solid-¹³C NMR spectrum for POM3-IM

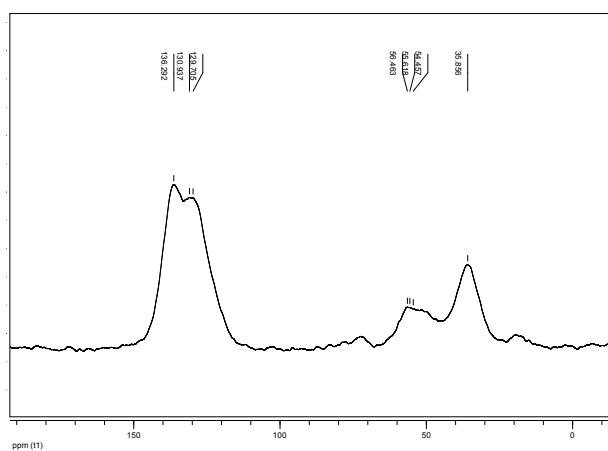


Fig. S4 Solid-¹³C NMR spectrum for POM4-IM

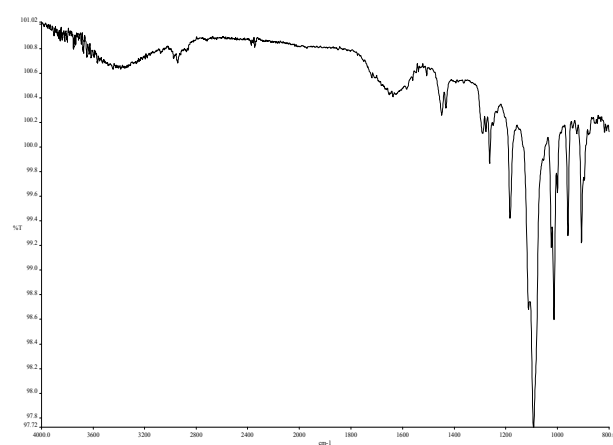


Fig. S5 FT-IR spectrum for POM1

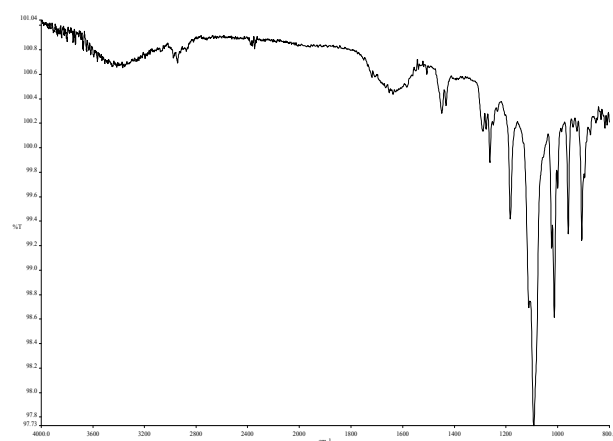


Fig. S6 FT-IR spectrum for POM2

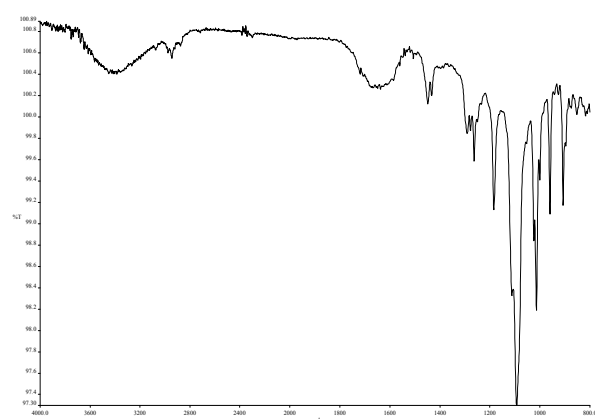


Fig. S7 FT-IR spectrum for POM3

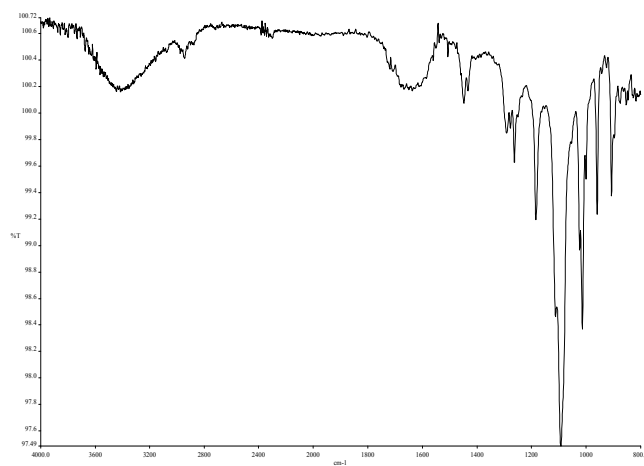


Fig. S8 FT-IR spectrum for POM4

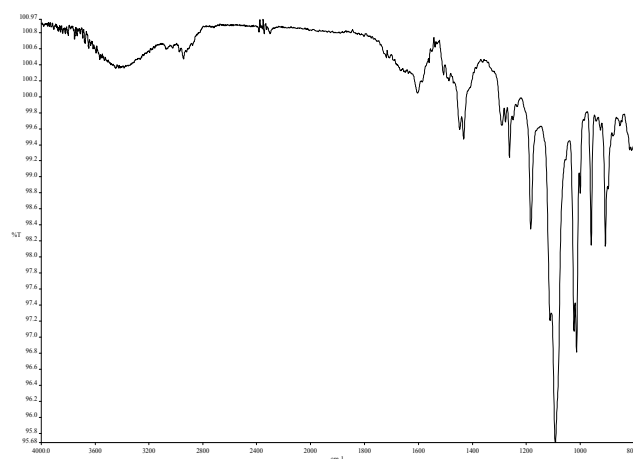


Fig. S9 FT-IR spectrum for POM5

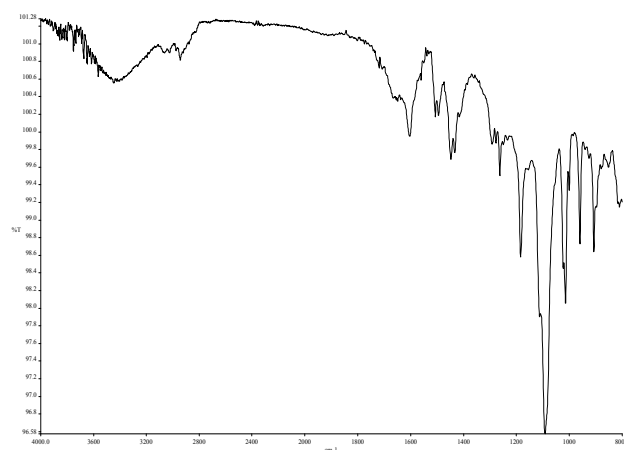


Fig. S10 FT-IR spectrum for POM6

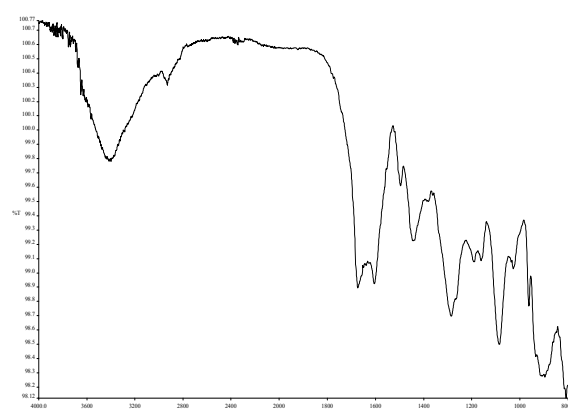


Fig. S11 FT-IR spectrum for POM1-IM

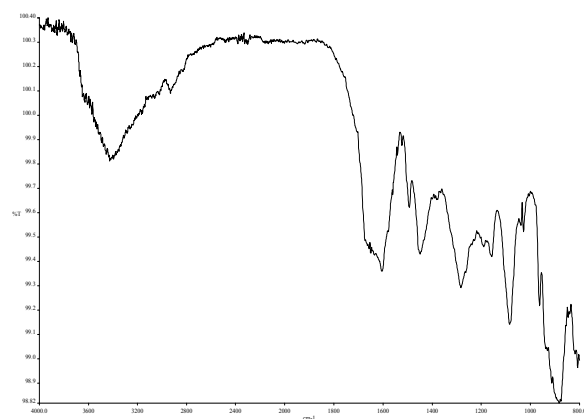


Fig. S12 FT-IR spectrum for POM2-IM

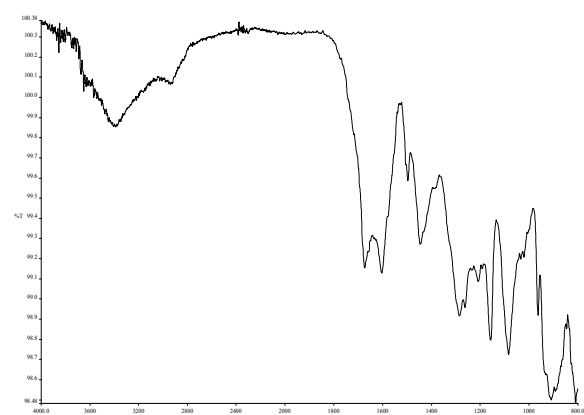


Fig. S13 FT-IR spectrum for POM3-IM

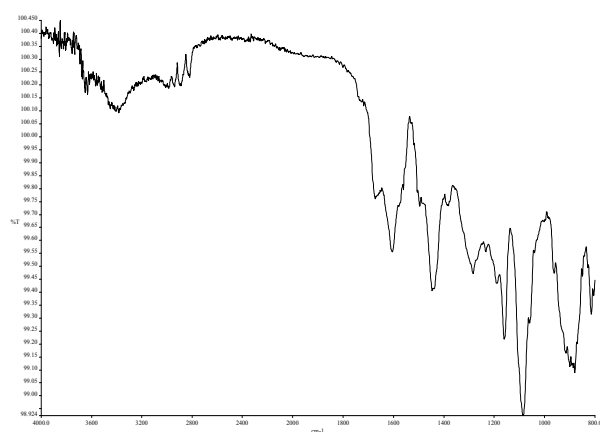


Fig. S14 FT-IR spectrum for POM4-IM

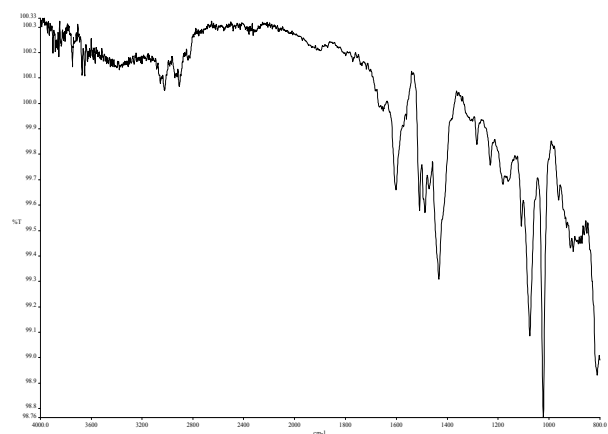


Fig. S15 FT-IR spectrum for POM5-IM

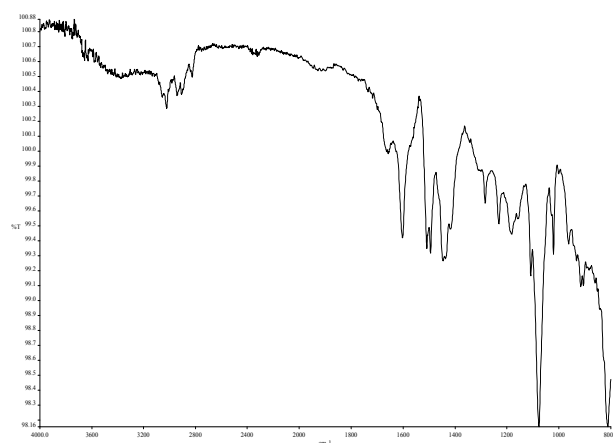


Fig. S16 FT-IR spectrum for POM6-IM

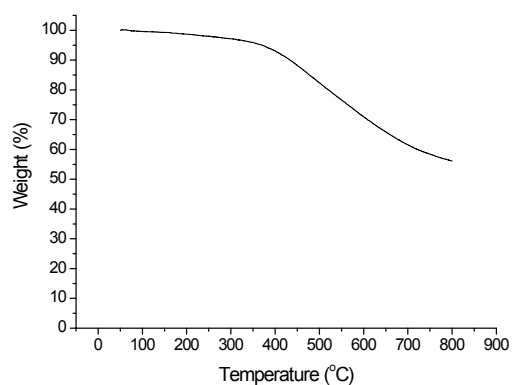


Fig. S17 The thermogram of POM1 by TGA.

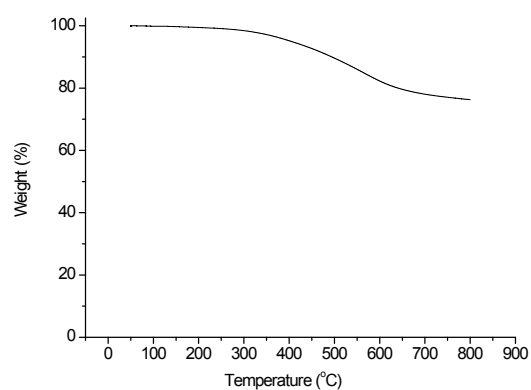


Fig. S18 The thermogram of POM2 by TGA.

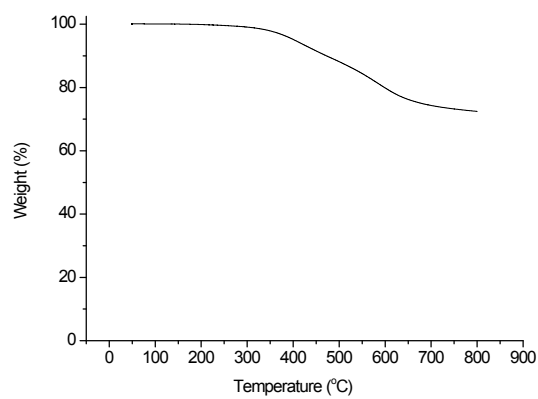


Fig. S19 The thermogram of POM3 by TGA.

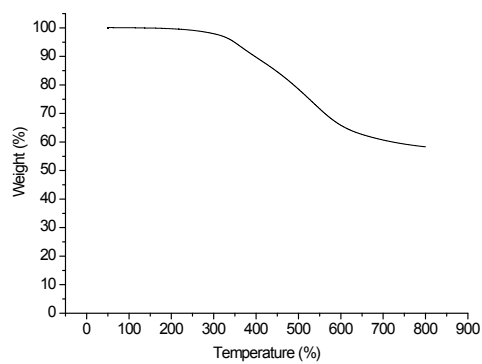


Fig. S20 The thermogram of POM4 by TGA.

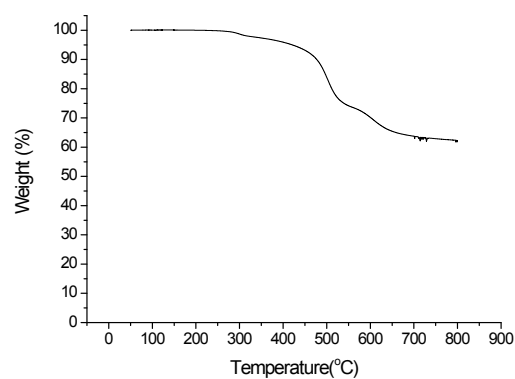


Fig. S21 The thermogram of POM5 by TGA.

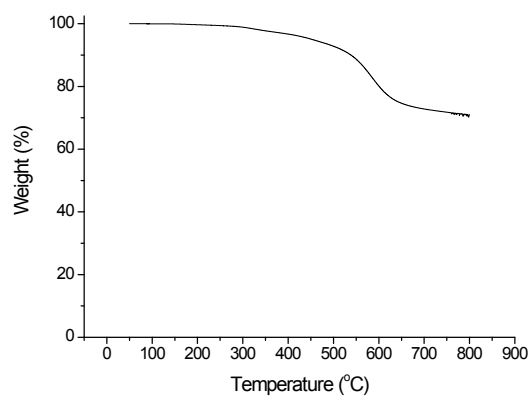


Fig. S22 The thermogram of POM6 by TGA.

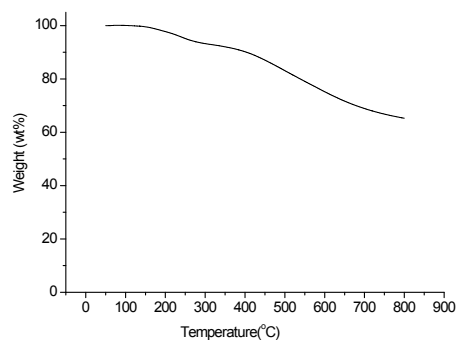


Fig. S23 The thermogram of POM1-IM by TGA.

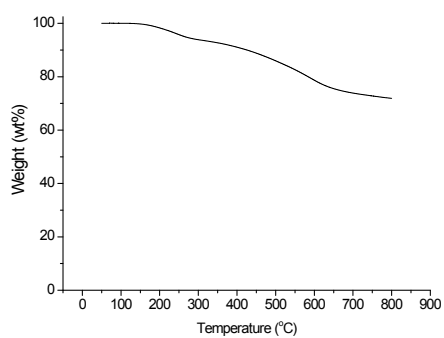


Fig. S24 The thermogram of POM2-IM by TGA.

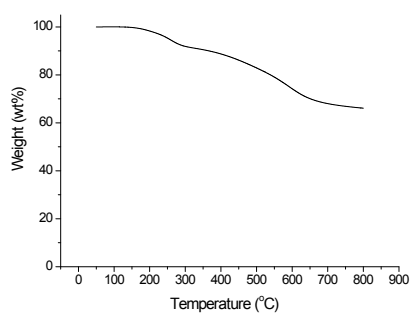


Fig. S25 The thermogram of POM3-IM by TGA.

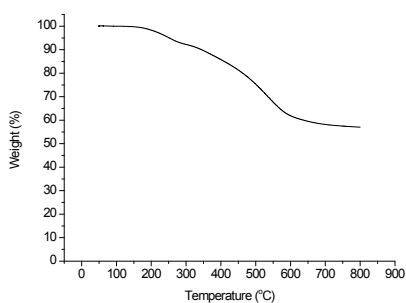


Fig. S26 The thermogram of POM4-IM by TGA.

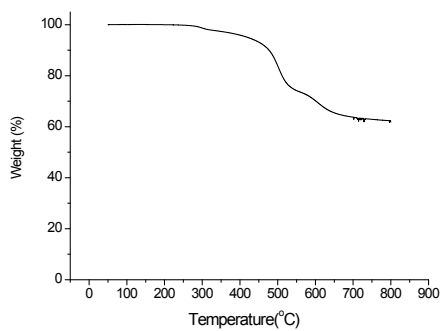


Fig. S27 The thermogram of POM5-IM by TGA.

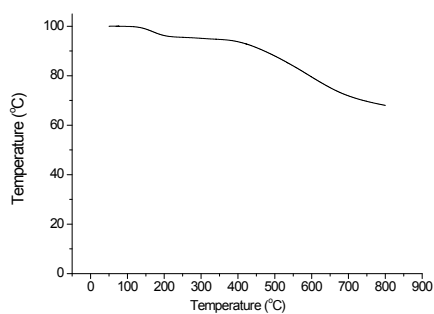


Fig. S28 The thermogram of POM6-IM by TGA.

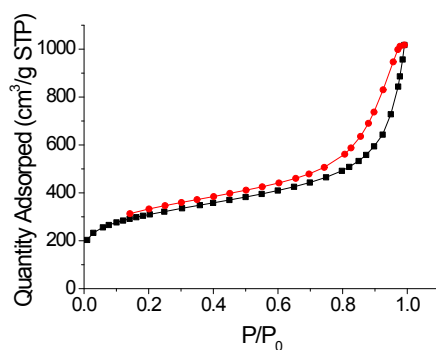


Fig. S29 Nitrogen adsorption isotherms for POM1 measured at 77K.

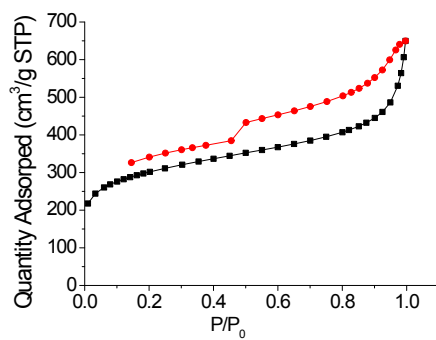


Fig. S30 Nitrogen adsorption isotherms for POM2 measured at 77K.

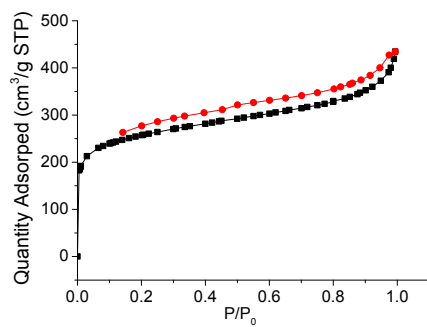


Fig. S31 Nitrogen adsorption isotherms for POM3 measured at 77K.

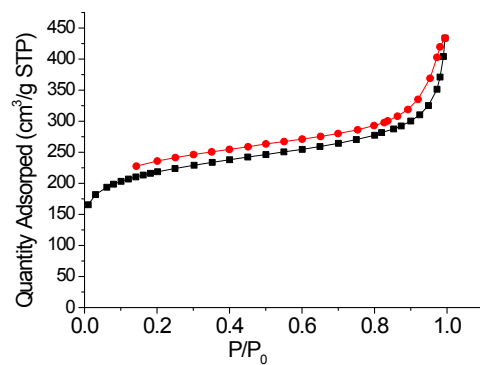


Fig. S32 Nitrogen adsorption isotherms for POM4 measured at 77K.

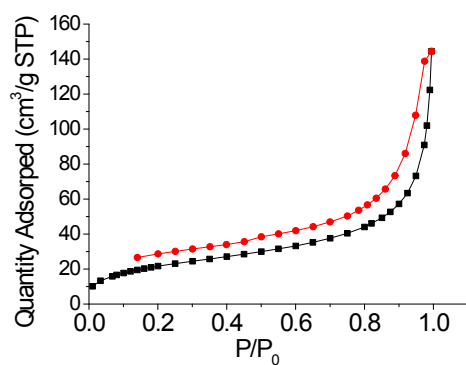


Fig. S33 Nitrogen adsorption isotherms for POM5 measured at 77K.

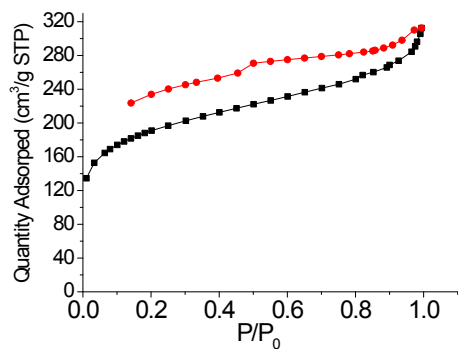


Fig. S34 Nitrogen adsorption isotherms for POM6 measured at 77K.

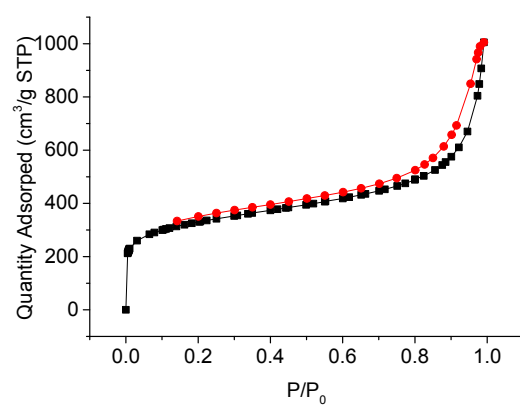


Fig. S35 Nitrogen adsorption isotherms for POM1-IM measured at 77K.

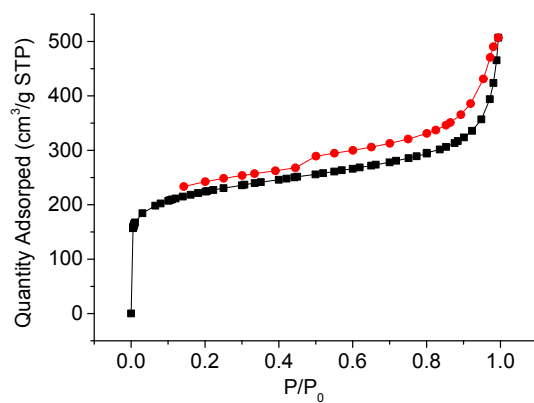


Fig. S36 Nitrogen adsorption isotherms for POM2-IM measured at 77K.

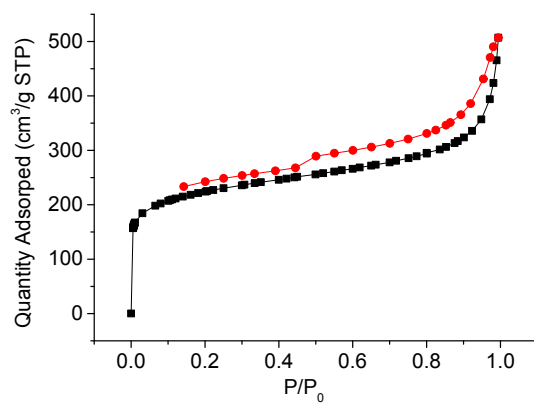


Fig. S37 Nitrogen adsorption isotherms for POM3-IM measured at 77K.

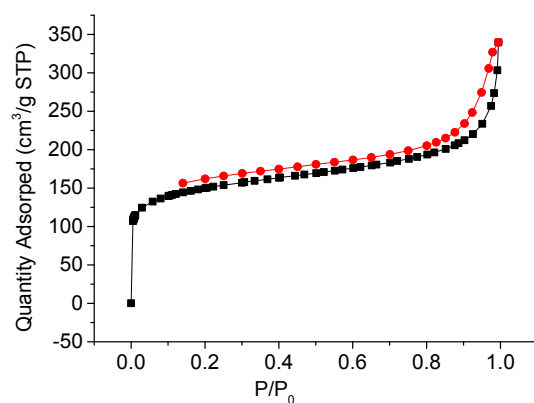


Fig. S38 Nitrogen adsorption isotherms for POM4-IM measured at 77K.

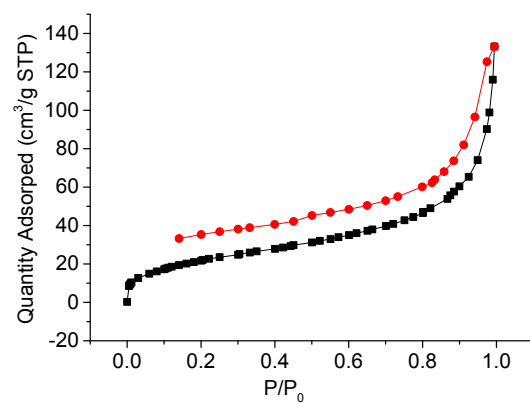


Fig. S39 Nitrogen adsorption isotherms for POM5-IM measured at 77K.

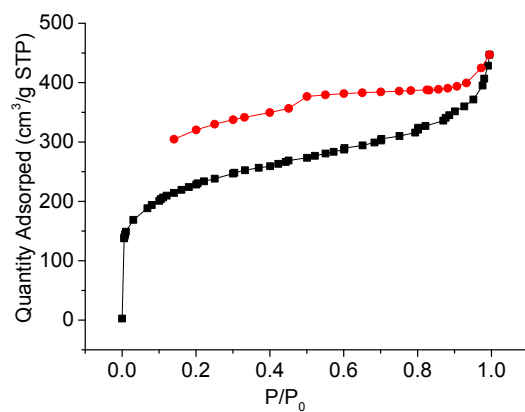


Fig. S40 Nitrogen adsorption isotherms for POM6-IM measured at 77K.

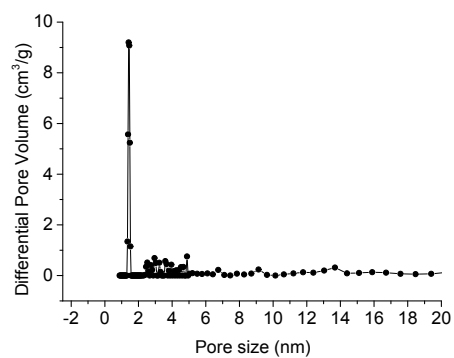


Fig. S41 Pore size distribution for POM1 calculated using NLDFT.

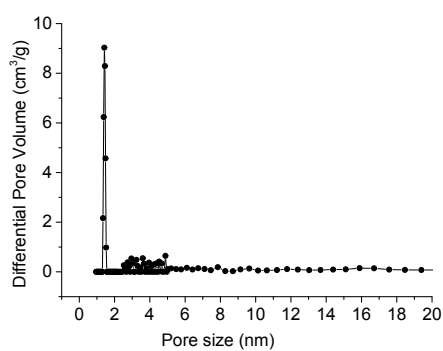


Fig. S42 Pore size distribution for POM2 calculated using NLDFT.

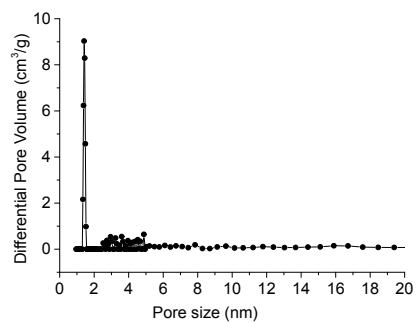


Fig. S43 Pore size distribution for POM3 calculated using NLDFT.

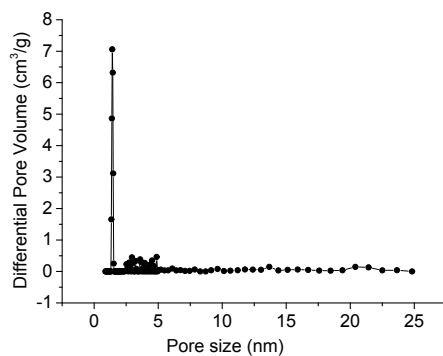


Fig. S44 Pore size distribution for POM4 calculated using NLDFT.

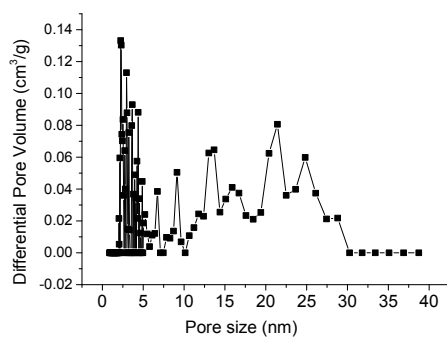


Fig. S45 Pore size distribution for POM5 calculated using NLDFT.

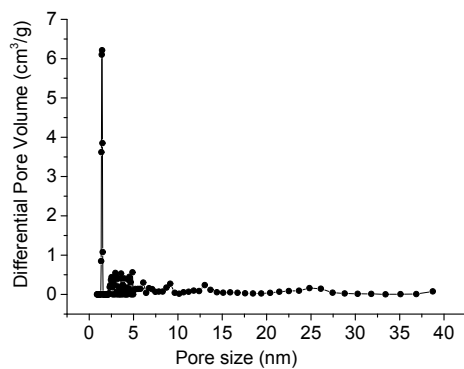


Fig. S46 Pore size distribution for POM6 calculated using NLDFT.

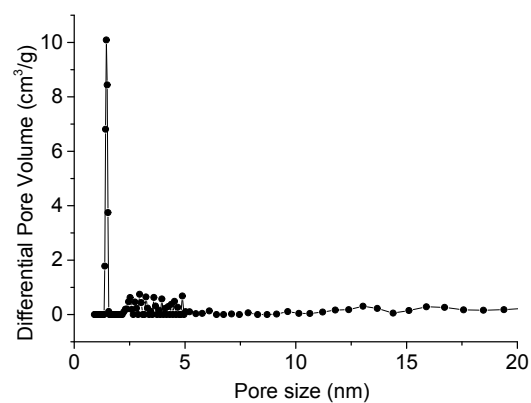


Fig. S47 Pore size distribution for POM1-IM calculated using NLDFT.

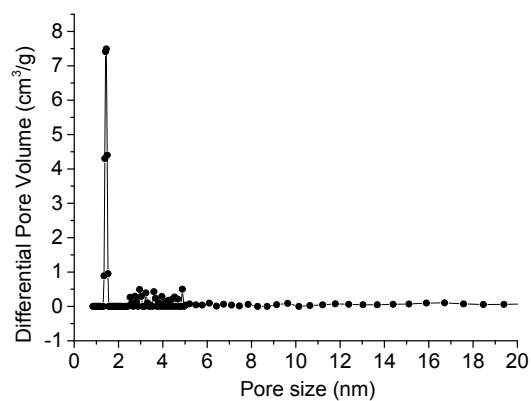


Fig. S48 Pore size distribution for POM2-IM calculated using NLDFT.

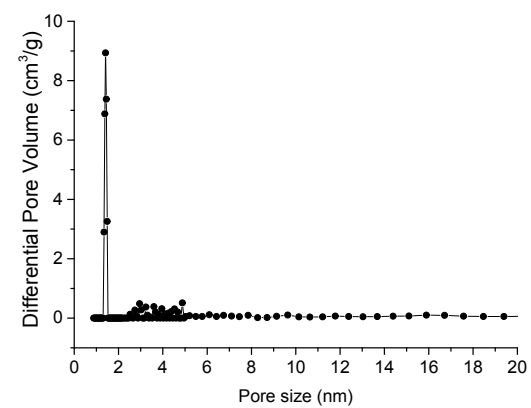


Fig. S49 Pore size distribution for POM3-IM calculated using NLDFT.

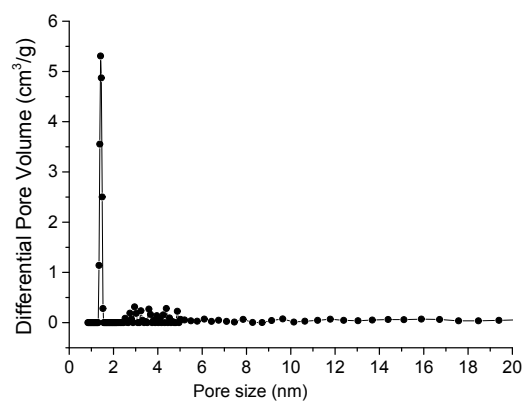


Fig. S50 Pore size distribution for POM4-IM calculated using NLDFT.

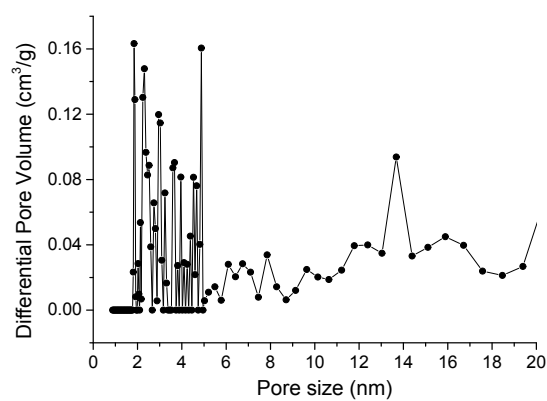


Fig. S51 Pore size distribution for POM5-IM calculated using NLDFT.

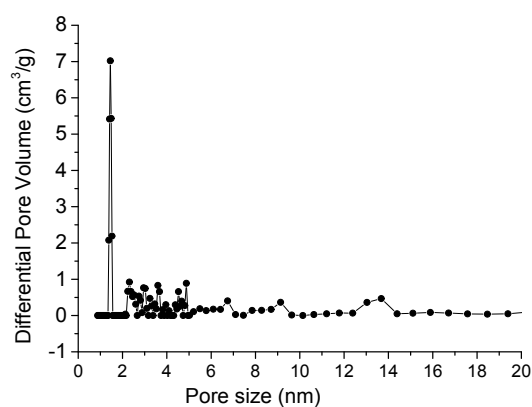


Fig. S52 Pore size distribution for POM6-IM calculated using NLDFT.

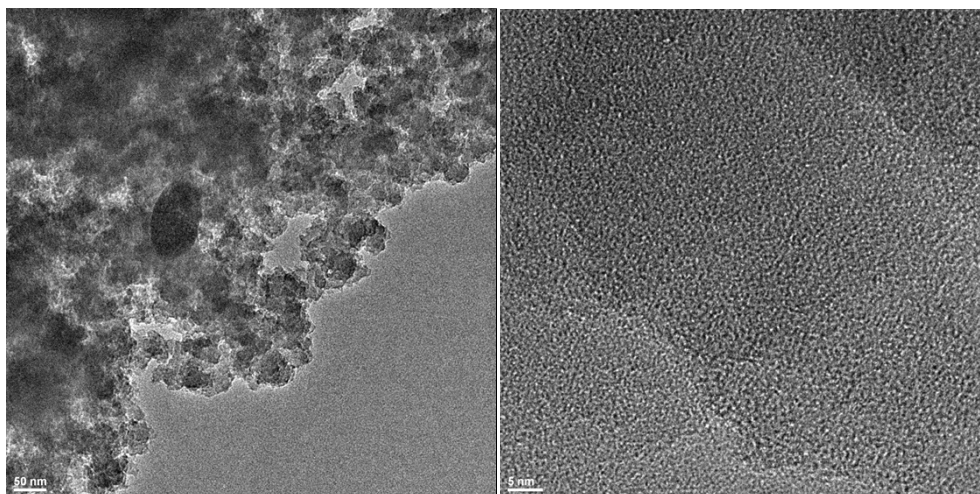


Fig. S53 Morphological structures of POM1-IM

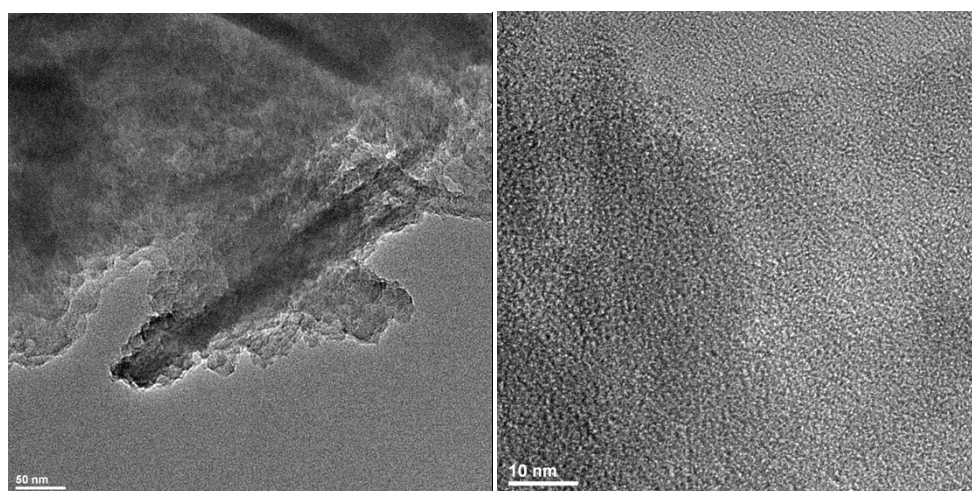


Fig. S54 Morphological structures of POM2-IM

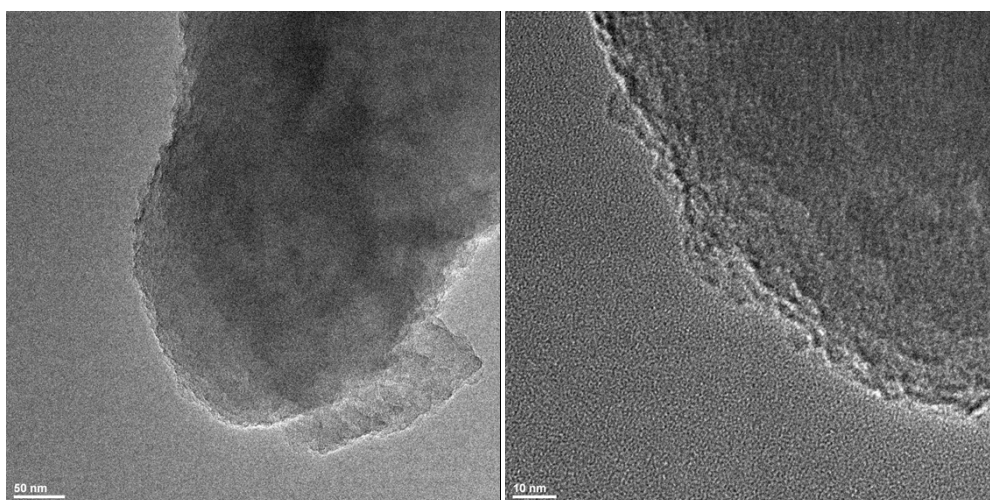


Fig. S55 Morphological structures of POM3-IM

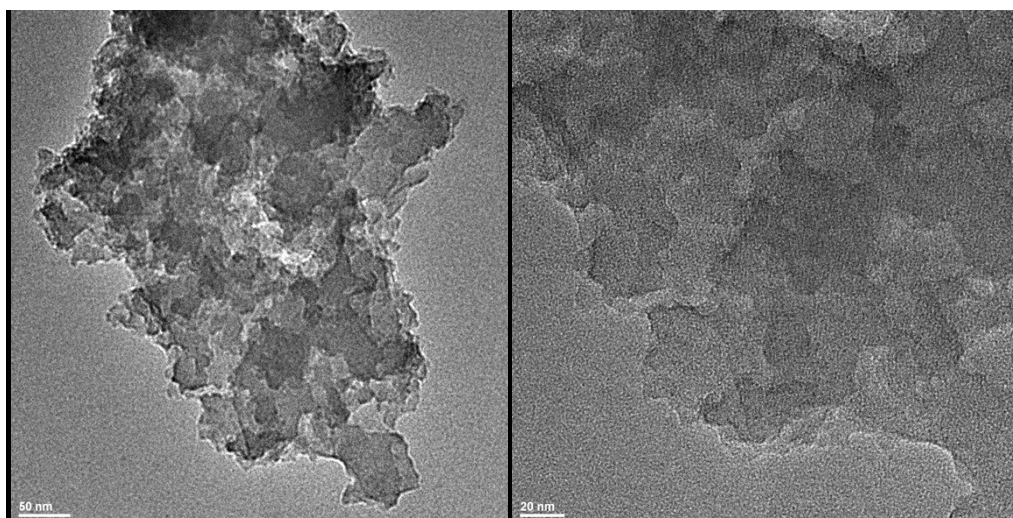


Fig. S56 Morphological structures of POM4-IM

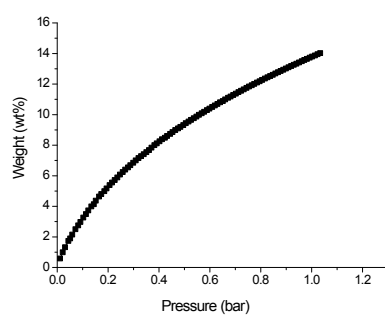


Fig. S57 CO₂ capture isotherms for POM1at 273K.

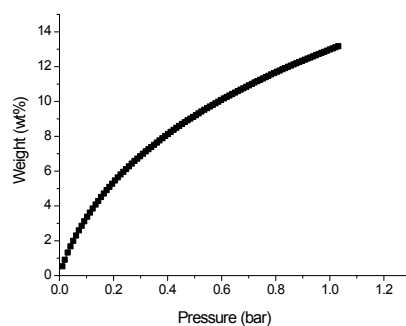


Fig. S58 CO₂ capture isotherms for POM2at 273K.

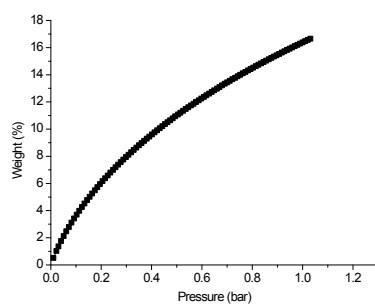


Fig. S59 CO₂ capture isotherms for POM3 at 273K.

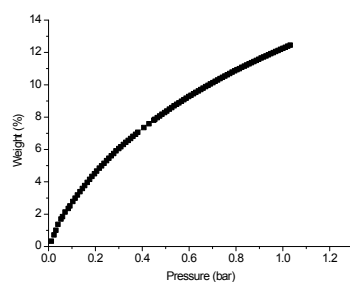


Fig. S60 CO₂ capture isotherms for POM4 at 273K.

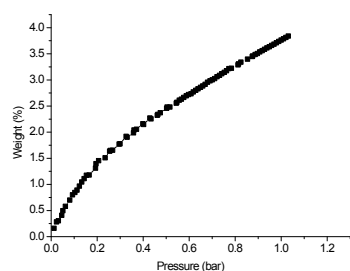


Fig. S61 CO₂ capture isotherms for POM5 at 273K.

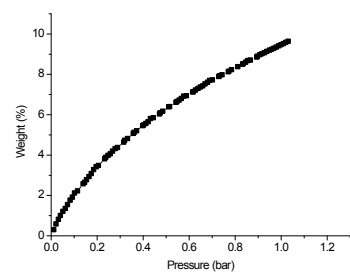


Fig. S62 CO₂ capture isotherms for POM6 at 273K.

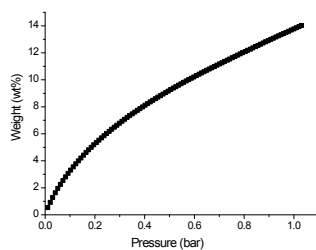


Fig. S63 CO₂ capture isotherms for POM1-IM at 273K.

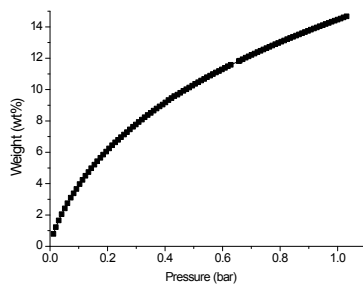


Fig. S64 CO₂ capture isotherms for POM2-IM at 273K.

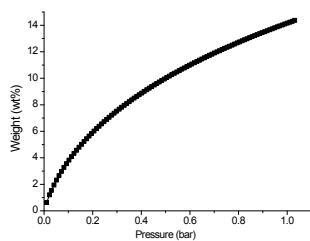


Fig. S65 CO₂ capture isotherms for POM3-IM at 273K.

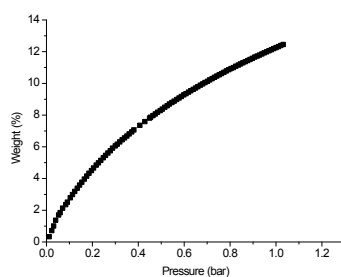


Fig. S66 CO₂ capture isotherms for POM4-IM at 273K.

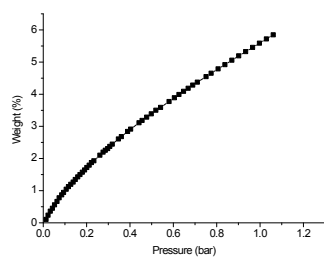


Fig. S67 CO₂ capture isotherms for POM5-IM at 273K.

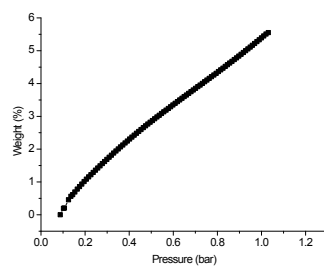


Fig. S68 CO₂ capture isotherms for POM6-IM at 273K.

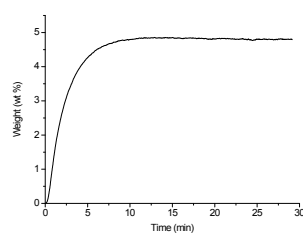


Fig. S69 CO₂ capture by TGA for POM1-IM at 298K.

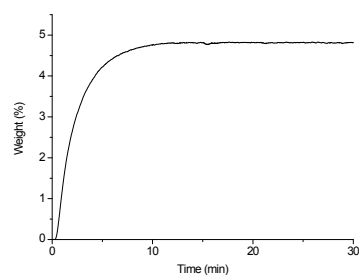
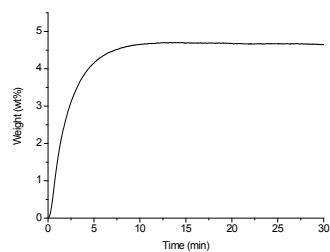
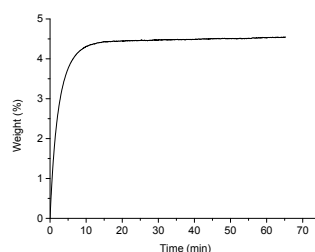


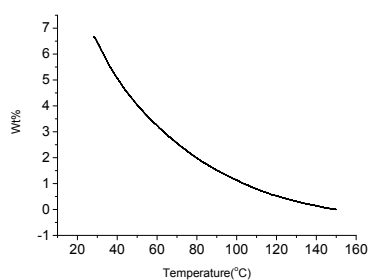
Fig. S70 CO₂ capture by TGA for POM2-IM at 298K.



(a) Fresh polymer



(b) After heated in water at 80 °C and 18 h



(c) CO₂ capture under different temperature

Fig. S71 CO₂ capture by TGA for POM3-IM at 298K (a) and (b); at different temperature (c).

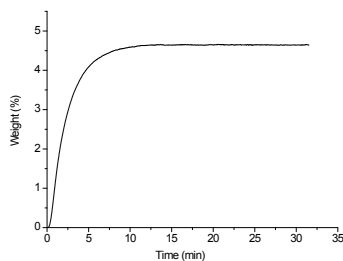


Fig. S72 CO₂ capture by TGA for POM4-IM at 298K.

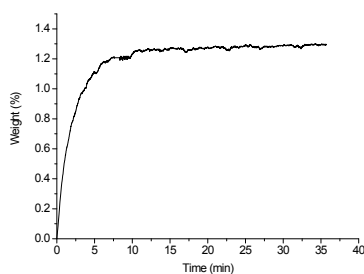


Fig. S73 CO₂ capture by TGA for POM5-IM at 298K.

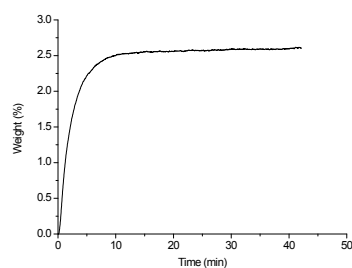


Fig. S74 CO₂ capture by TGA for POM6-IM at 298K.

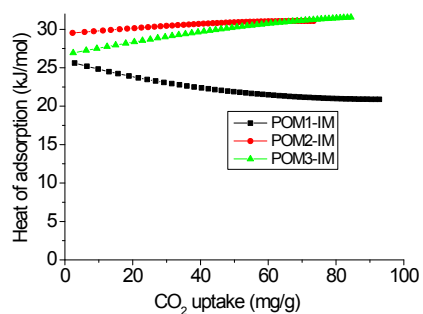


Fig. S75 Heat of adsorption (Q_{st}) for CO₂ capture

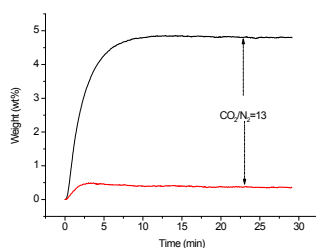


Fig. S76 Adsorption kinetics of CO₂ and N₂ obtained for POM-IM2 at 298K by TGA.

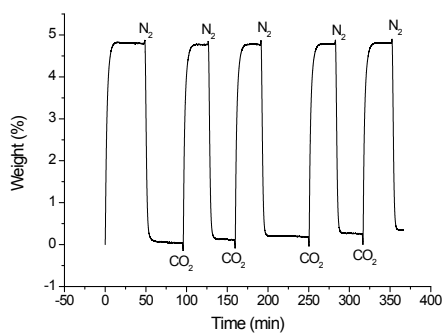


Fig. S77 CO₂ adsorption-desorption cycles obtained for POM-IM2 at 298K by TGA.

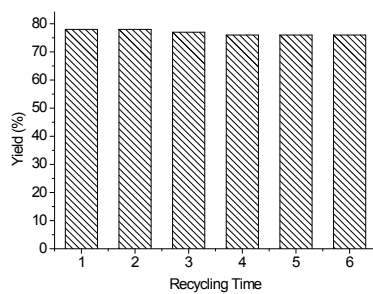


Fig. S78 Catalyst recycling. Reaction conditions: PO (1.43 mmol), POM3-IM (5 mmol% based on the imidazolium salt), Ethanol (2 ml), CO₂ pressure (1 MPa), Temperature (120 °C), Time (4 h).

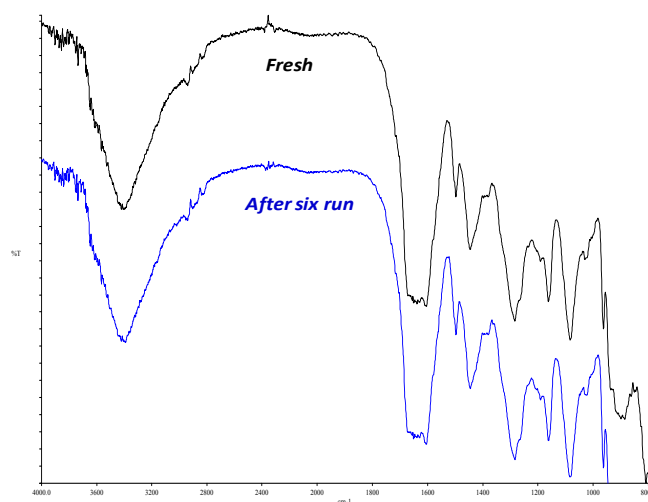


Fig. S79 FT-IR of POM-IM3 in Scheme 1: fresh and after sixth run: Reaction conditions: Reaction condition: PO (0.1 mL), Catalyst (100 mg, 5 mmol% based on the imidazolium salt), CO₂ pressure (1 MPa), reaction time (4.0 h), temperature (120 °C).

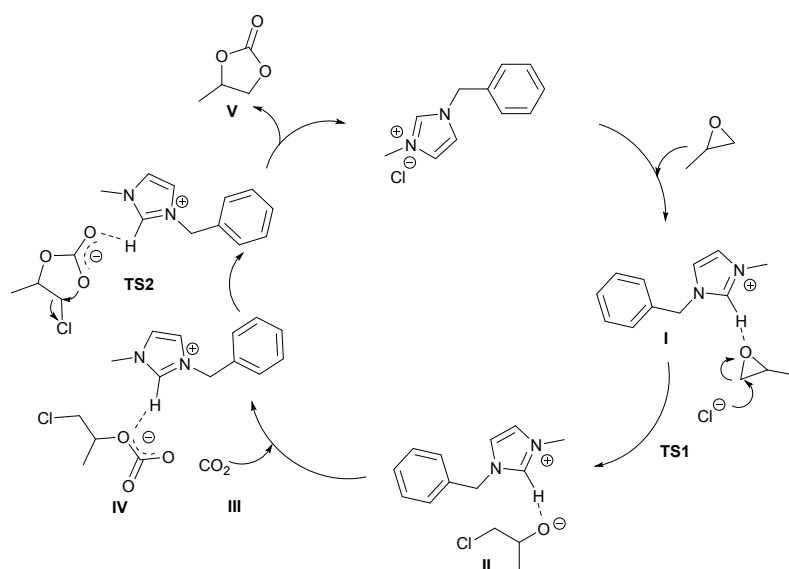


Fig. S80 Mechanistic hypothesis of the catalytic reaction.

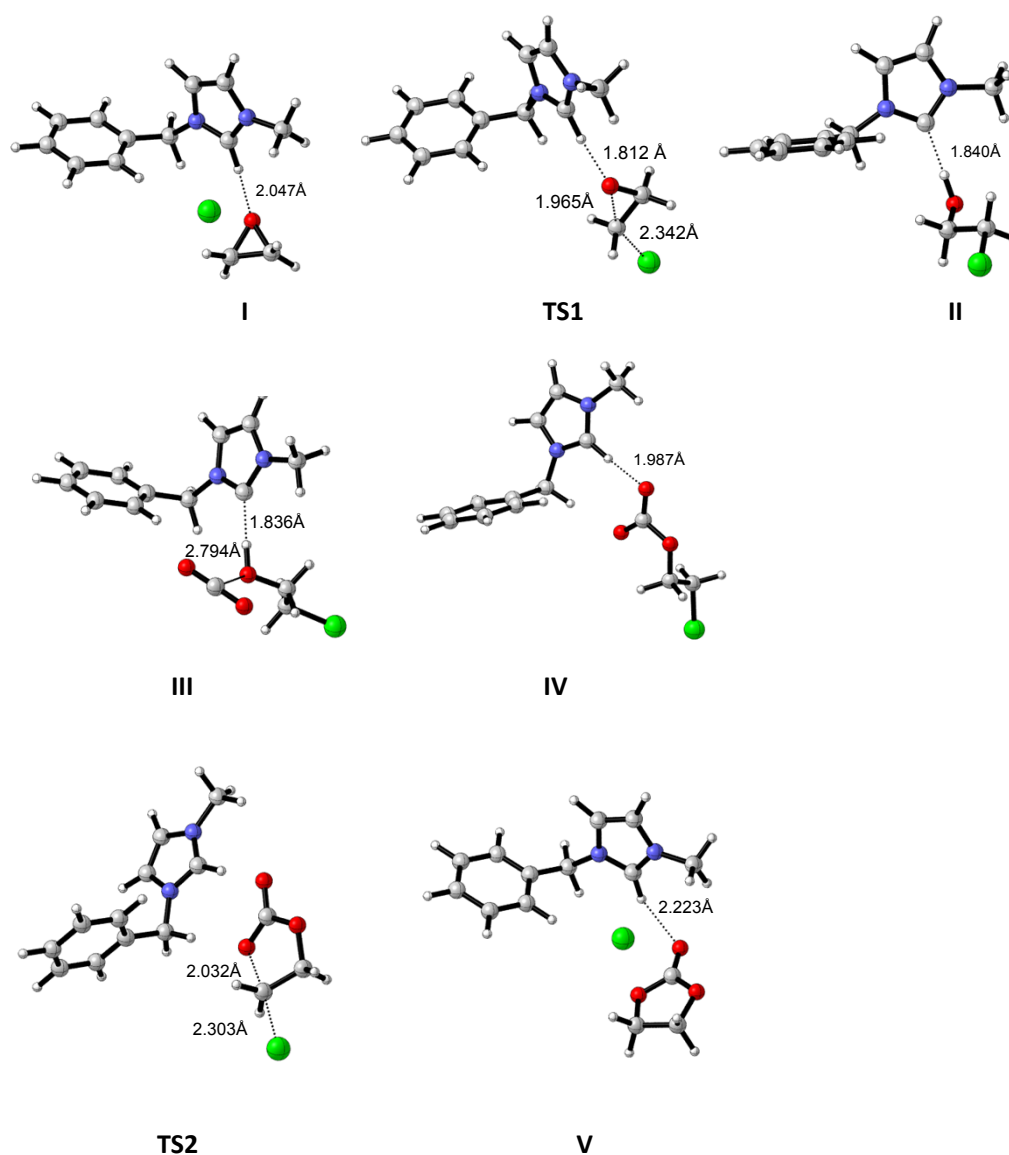


Fig. S81 Optimized geometries for the intermediates and transition states. H: white, C: gray, N: blue, O: red, Cl: green. Bond lengths and distances are in Å.

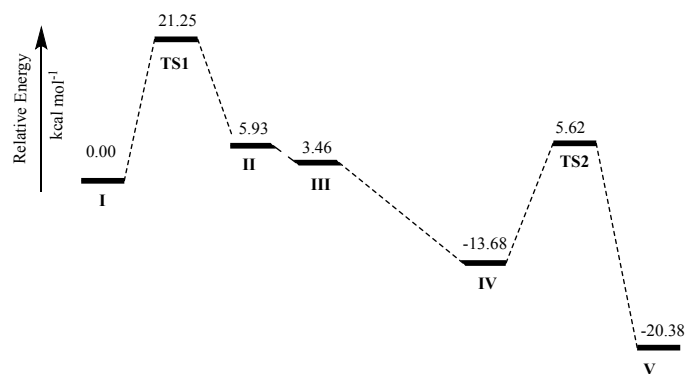


Fig. S82 Relative energies for fixation of CO₂ with PO catalyzed by 1-benzyl-3-methylimidazolium chloride in ethanol, relative energies at the B3PW91/6-311++G (d, p) level.

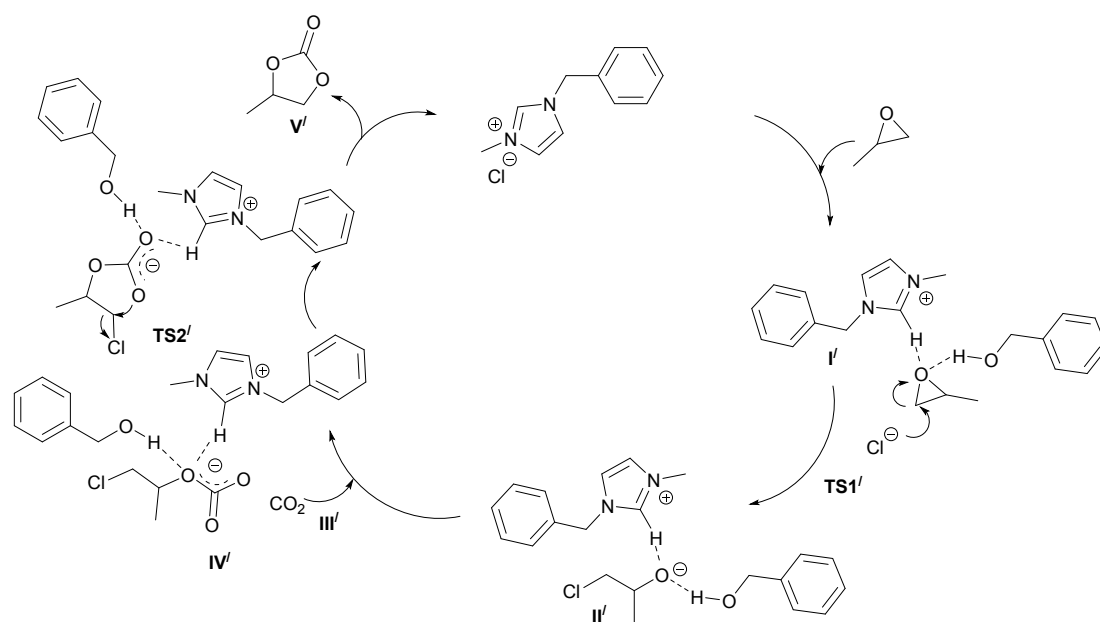


Fig. S83 Mechanistic hypothesis of the catalytic reaction.

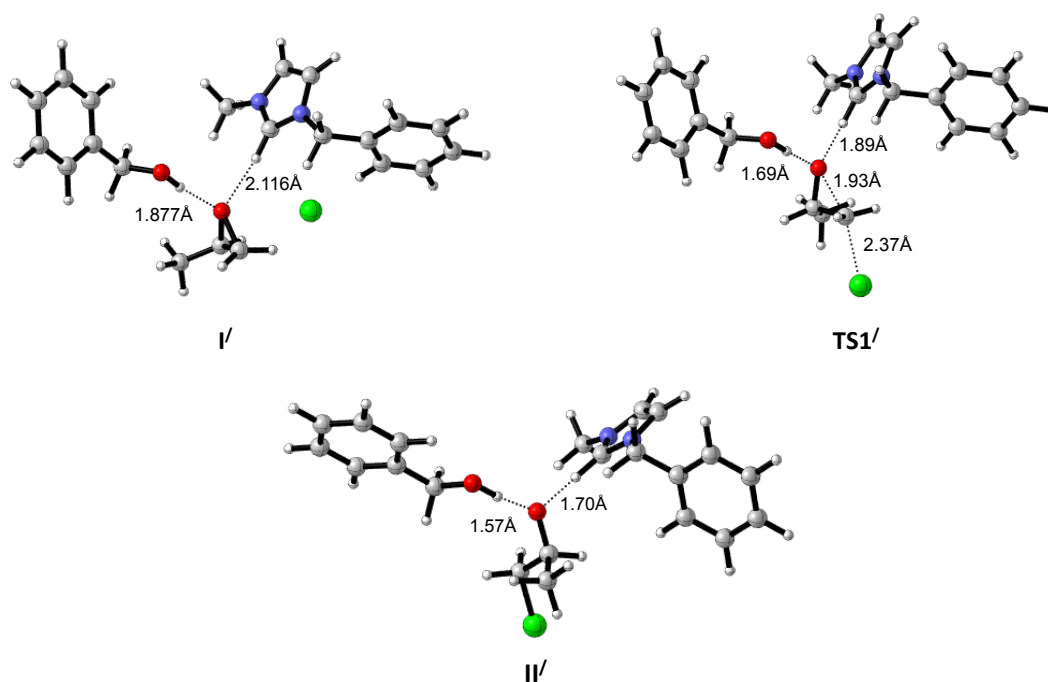


Fig. S84 Optimized geometries for the intermediates and transition states. H: white, C: gray, N: blue, O: red, Cl: green. Bond lengths and distances are in Å.

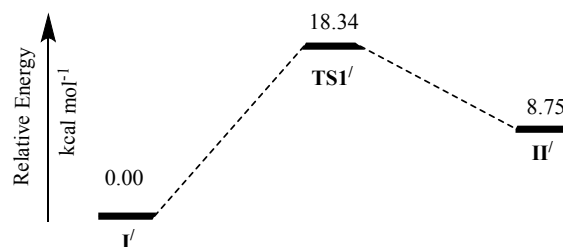


Fig. S85 Relative energies of ring opening step for fixation of CO₂ with PO catalyzed by 1-benzyl-3-methylimidazolium chloride and benzyl alcohol in ethanol, relative energies at the B3PW91/6-311++G (d, p) level.

5. Supporting Tables

Table S1 Elemental analysis results for POM1~6-IM

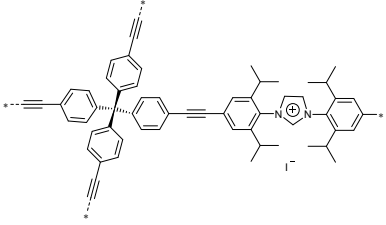
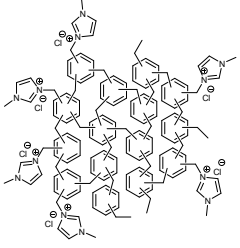
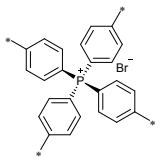
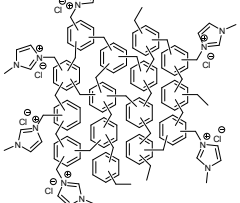
POM	C (wt%)	H (wt%)	N (wt%)	Degree of halogen substitution (%)
POM1-IM	74.36	4.85	2.01	21%
POM2-IM	75.41	4.69	2.01	23%
POM3-IM	72.11	5.06	2.84	57%
POM4-IM	68.04	4.44	1.85	34
POM5-IM	80.74	5.13	1.19	>99%
POM6-IM	71.89	4.63	<0.50	-
POM3-IM ^a	72.22	5.04	2.82	

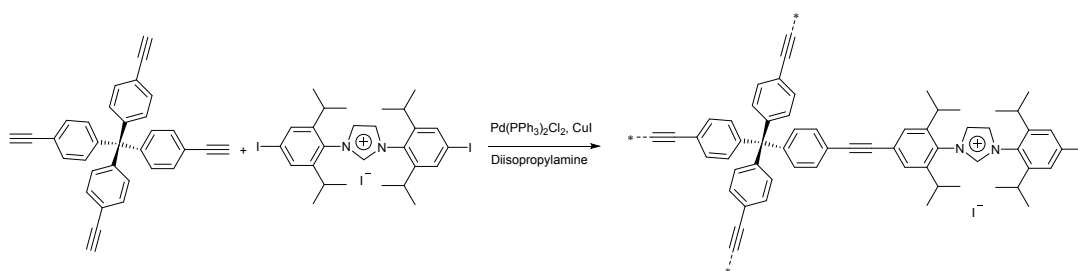
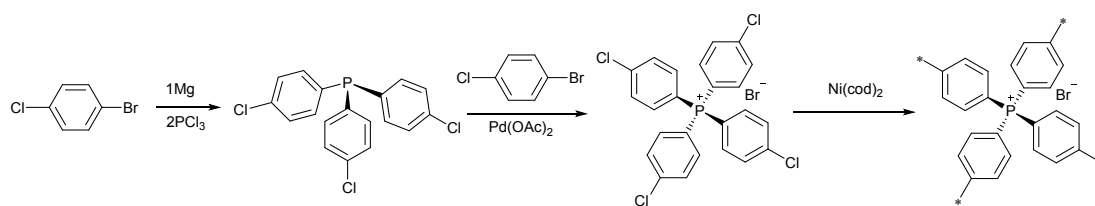
^aAfter six runs

Table S2 Elemental analysis results for POM1~6

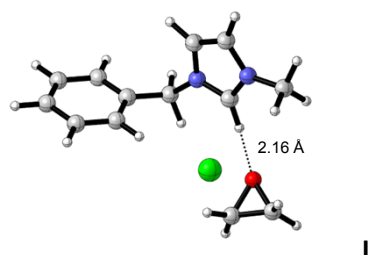
POM	C (wt%)	H (wt%)	Cl (wt%)	Br (wt%)
POM1	74.53	4.56	24.13	/
POM2	74.71	4.42	10.70	/
POM3	76.81	4.76	5.61	/
POM4	71.43	3.92	/	6.8
POM5	71.91	4.35	/	1.5
POM6	87.42	5.05	<0.5	/

Table S3 Comparison with the other reported functionalized porous organic materials

Epoxide	Catalyst	mol%	P(MPa)	Temp. (°C)	Time (h)	Yield (%)	Ref
		0.065	1	150	10	87	3a ^a
		5	1	120	8	92	This work
		2	0.1	140	20	98	3h ^b
		5	1	120	12	91	This work

^a the synthetic method of catalyst:^b the synthetic method of catalyst:

6. Cartesian coordinates for all the optimized geometries



C	3.15234700	-2.20329900	-0.78958600
C	2.58302700	-2.34474800	0.55409700
H	3.87486300	-1.40271500	-0.93457900
H	2.89674400	-1.67181600	1.34689000
H	2.19138400	-3.31058000	0.86093500
O	1.82204600	-1.70310600	-0.50082000
C	2.05688100	3.13556700	-0.33963600
H	2.57538900	2.51343300	0.39043100
H	2.66939800	3.27850400	-1.22983600
H	1.82348800	4.10063100	0.10578500
H	-0.40743000	4.17986100	-1.25191200
C	-0.34781400	3.10975900	-1.15774400
C	-1.26737100	2.14149300	-1.41501900
H	-2.28115700	2.21039900	-1.76827400
C	0.57920800	1.15943600	-0.69703600
N	-0.66827900	0.93046100	-1.12268100
N	0.79603100	2.47743000	-0.70793300
H	-0.50099600	-1.12952700	-1.05839400
C	-1.29785800	-0.40596200	-1.23274300
H	-1.64268900	-0.51882400	-2.26060500
H	1.28155100	0.38801500	-0.41092200
Cl	3.26427600	0.96528000	2.47791000
C	-2.44157400	-0.58948300	-0.25819100
C	-3.75619600	-0.67582200	-0.72317300
C	-2.19558500	-0.68709400	1.11659300
C	-4.81297400	-0.85306800	0.17121700
H	-3.95650700	-0.60783400	-1.78726300
C	-3.24945900	-0.85971200	2.01010700
H	-1.17924200	-0.62918200	1.49142100
C	-4.56127400	-0.94238600	1.53876600
H	-5.82861800	-0.91978500	-0.20165900
H	-3.04779600	-0.93438800	3.07254600
H	-5.38082100	-1.07905700	2.23504500
C	3.24511900	-3.34912300	-1.75839200

H	3.10767900	-2.99946300	-2.78542900
H	4.23326300	-3.81383000	-1.69149300
H	2.48931000	-4.10734300	-1.54186200

1 2 1.0 3 1.0 6 1.0 34 1.0

2 4 1.0 5 1.0 6 1.0

3

4

5

6

7 8 1.0 9 1.0 10 1.0 17 1.0

8

9

10

11 12 1.0

12 13 2.0 17 1.0

13 14 1.0 16 1.0

14

15 16 1.5 17 1.5 21 1.0

16 19 1.0

17

18 19 1.0

19 20 1.0 23 1.0

20

21

22

23 24 1.5 25 1.5

24 26 1.5 27 1.0

25 28 1.5 29 1.0

26 30 1.5 31 1.0

27

28 30 1.5 32 1.0

29

30 33 1.0

31

32

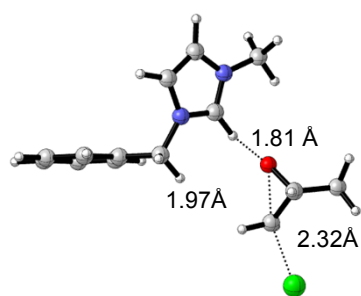
33

34 35 1.0 36 1.0 37 1.0

35

36

37

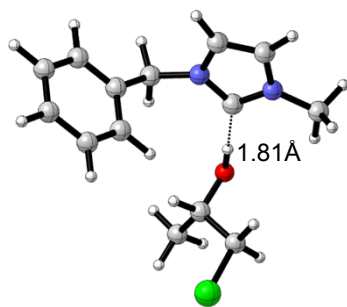


TS1

C	3.24736500	-1.19413000	0.47826000
C	3.43926400	0.16697300	-0.05759400
H	3.61918500	-1.38677200	1.47363600
H	2.37044000	-1.73639000	0.16292400
H	3.16679400	0.24050400	-1.12288200
O	2.41522500	0.56557600	0.80623600
C	0.62021400	3.82479800	0.78305000
H	0.17581200	4.23629400	1.68859700
H	1.50953900	3.25114700	1.03510800
H	0.88036200	4.63110800	0.09831800
H	-1.97343300	4.26909700	-0.26928000
C	-1.59007500	3.26594800	-0.33801700
C	-2.14258200	2.13653900	-0.85396900
H	-3.09999000	1.97055600	-1.31556800
C	-0.12763200	1.61910100	-0.08493900
N	-1.21571300	1.12208200	-0.68663300
N	-0.33690000	2.92160000	0.13528400
H	-0.44379100	-0.78605300	-0.89307200
C	-1.38964600	-0.28719900	-1.10303000
H	-1.54344400	-0.29549400	-2.18227300
H	0.79732500	1.08034400	0.18792700
Cl	4.66145400	-2.75164100	-0.49790900
C	-2.53835200	-0.96629000	-0.38973300
C	-3.70976800	-1.28362200	-1.08178800
C	-2.43717800	-1.29152400	0.96801200
C	-4.76993600	-1.91192000	-0.42658700
H	-3.79462200	-1.04388300	-2.13646700
C	-3.49522200	-1.91596200	1.62331800
H	-1.52838700	-1.05984900	1.51362600
C	-4.66502700	-2.22650700	0.92672600
H	-5.67340800	-2.15471100	-0.97412200
H	-3.40591100	-2.16583700	2.67432400
H	-5.48718400	-2.71524800	1.43693400
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H	5.56213800	0.39312200	-0.45192700

H	4.71656700	1.89860800	-0.04359800
H	5.08556700	0.72001000	1.22994700

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H	-2.95580700	0.18599700	-1.01901900
H	-1.68618500	-1.18190900	0.65705100
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C	-1.36515100	3.58947700	-0.97861600
H	-1.42172300	3.53854100	-2.06759300
H	-2.23160800	3.09226400	-0.54899700
H	-1.36021500	4.63607300	-0.66890900
H	1.33728100	4.16320500	-1.44813900
C	1.12757800	3.31514200	-0.81789800
C	1.96269500	2.45285400	-0.18899200
H	3.03833100	2.40335400	-0.17074300
C	-0.17320300	1.82273200	0.31475900
N	1.15195900	1.55411900	0.49294000
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C	2.44011800	-0.58316400	0.53879900
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C	-3.36322800	-1.62805700	1.93293500

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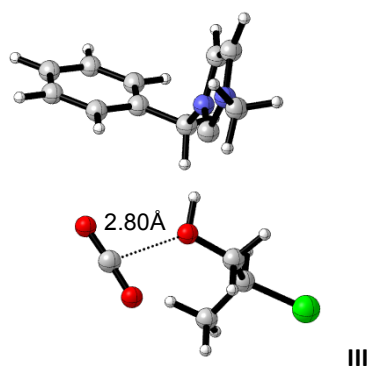
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H	2.91020600	1.12868500	-0.45338300
H	3.90566500	-0.11038300	0.33383600
O	1.88778800	-0.32950200	0.61313600
C	0.62793400	3.37996500	1.54335400
H	0.16876500	3.66690600	2.49095600
H	1.42013900	2.65921600	1.73197800
H	1.05133300	4.26641300	1.06724800
H	-1.79634400	4.38117800	0.55377500
C	-1.53118500	3.38045200	0.25594200
C	-2.17703300	2.49355900	-0.53987300
H	-3.11512200	2.57165300	-1.06309600
C	-0.24432600	1.50085800	0.15841000
N	-1.37729400	1.35850000	-0.58606000
N	-0.36164900	2.75641500	0.66978800
H	-0.82968900	-0.48984300	-1.29970000
C	-1.71595800	0.14449400	-1.33571400
H	-1.88996300	0.41413400	-2.37959500
H	1.11310800	0.29345700	0.44262700
C	2.31979600	-1.29298600	3.20259000
O	1.18271100	-1.29271100	3.43737300
O	3.46674200	-1.31455200	3.02327300
Cl	4.34832900	-0.08289100	-2.57158800
C	-2.92207800	-0.59124600	-0.78318500
C	-4.03184100	-0.83875200	-1.59467300
C	-2.93159200	-1.05518600	0.53786900
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C	-5.13639500	-1.99440900	0.21723700
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H	-5.99152800	-2.53631600	0.60516400
C	3.18467900	-2.24238900	-1.30720100
H	4.13313100	-2.44525700	-0.80359300
H	2.37344300	-2.62211600	-0.68101500
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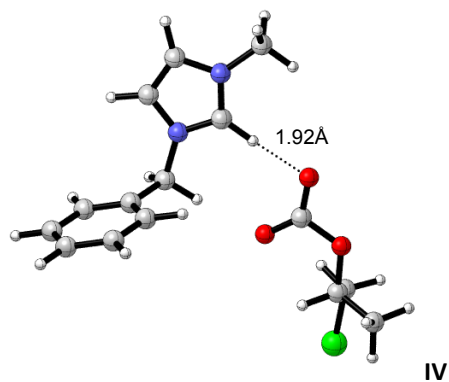
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H	-2.47273300	5.03395700	-0.17736600
H	-4.67287700	3.21905100	-0.54285300
C	-3.82270800	2.55937300	-0.54077700
C	-3.68620300	1.28175100	-0.98416200
H	-4.39743900	0.61719000	-1.44208800
C	-1.73253300	1.92965300	-0.16736500
N	-2.37658100	0.90781000	-0.74279600
N	-2.59404800	2.94438700	-0.03521600
H	-0.70644000	-0.31606800	-0.81607100
C	-1.76616400	-0.40271400	-1.07085600
H	-1.85345700	-0.53957300	-2.14933800
H	-0.68211300	1.93733300	0.14123200
C	4.59213200	0.14208300	-0.71953900
C	3.93944100	-0.14093300	0.63386100
H	3.83703000	0.33777800	-1.47531400
H	5.29862200	0.96835500	-0.66159200
H	3.32216100	-1.03637000	0.54757100
O	3.09338400	0.98403400	0.92967300
C	1.77715100	0.99051400	0.41295900
O	1.14367200	2.01887000	0.73591700
O	1.41125300	0.01611100	-0.26948000
Cl	5.54155100	-1.29913200	-1.33147900
C	-2.42723700	-1.54570600	-0.33220300
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C	-3.83957900	-3.51784400	-0.33800400
H	-3.39267200	-2.34185900	-2.08329900
C	-2.83010500	-2.76805700	1.72198800
H	-1.58988200	-1.01986900	1.58415500
C	-3.63958500	-3.67314800	1.03229300
H	-4.46298700	-4.21921100	-0.88079900
H	-2.66486200	-2.88905500	2.78657400
H	-4.10697600	-4.49615100	1.56090600
C	4.93297100	-0.27345200	1.77804800
H	5.61430600	-1.10816500	1.60162600
H	5.52386300	0.64051400	1.88551200
H	4.40209200	-0.45602100	2.71457800

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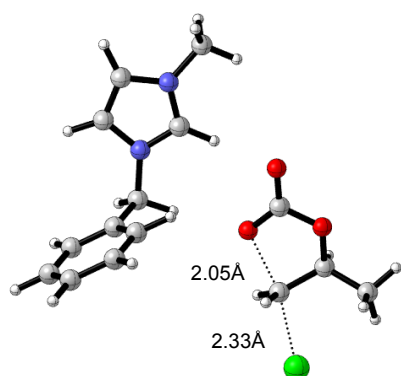
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H	2.00571200	-4.18185100	0.89145400
H	3.35882200	-4.71291700	-0.13410800
H	5.17560600	-2.50261300	-0.50911400
C	4.21209200	-2.02354700	-0.51744200
C	3.82868000	-0.80394200	-0.97938900
H	4.39742400	-0.01890700	-1.44587700
C	2.03888000	-1.81393000	-0.15843100
N	2.46977800	-0.69282300	-0.74735800
N	3.08159200	-2.63718100	-0.00896300
H	0.58745000	0.15988700	-0.91179800
C	1.61939100	0.47272400	-1.08329300
H	1.74422400	0.66720900	-2.14866700
H	1.01280500	-2.01680800	0.14732700
C	-3.36015000	0.31759900	-0.47174100
C	-3.98436000	-1.01375300	-0.06320000
H	-2.94273000	0.96308800	0.28227100
H	-3.18764300	0.55120000	-1.50308700
H	-4.19096600	-1.58234300	-0.97504500
O	-3.00122000	-1.76109400	0.70096400

C	-1.70596000	-1.58684900	0.26947100
O	-0.84676300	-2.33517100	0.74523400
O	-1.55431800	-0.64545200	-0.57187700
Cl	-5.14993000	1.79573000	-0.66945800
C	1.97097700	1.70014000	-0.27046400
C	2.53699300	2.81743100	-0.88945500
C	1.71509200	1.73939500	1.10548400
C	2.84723500	3.95764600	-0.14656900
H	2.73334800	2.79957100	-1.95637700
C	2.02857000	2.87443000	1.84878900
H	1.26650600	0.88281300	1.59773300
C	2.59574200	3.98689300	1.22359500
H	3.28439800	4.81876600	-0.63897300
H	1.82454100	2.89452000	2.91327100
H	2.83614300	4.87142200	1.80237300
C	-5.22376300	-0.97544200	0.81871100
H	-6.07800100	-0.58497000	0.27156000
H	-5.44347800	-1.99431800	1.14461200
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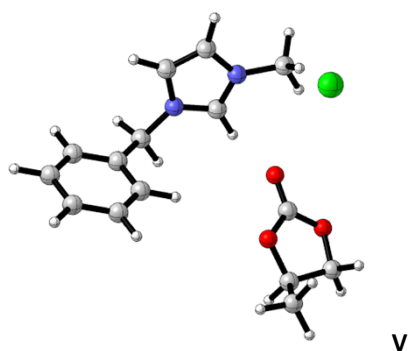
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C	0.59585100	3.39381300	-0.98797000
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H	1.48949500	2.84217800	-1.27111200
H	0.45917300	4.23855900	-1.66170000
H	-2.17846600	3.87298300	-0.74557000
C	-1.88521200	2.85894200	-0.95395600
C	-2.63008300	1.73283100	-1.11608200
H	-3.69452300	1.58226700	-1.07594200
C	-0.50172700	1.18524700	-1.35352400
N	-1.74766900	0.69761000	-1.36508400
N	-0.56067300	2.49586300	-1.10749400
H	-1.18428600	-1.23351300	-1.85536300
C	-2.11233300	-0.71543900	-1.61326700
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C	3.67598600	-2.42047500	0.39841900

C	4.91120800	-1.64539300	-0.07403700
H	3.59909500	-3.38354300	-0.11033300
H	5.70262300	-2.28213700	-0.46118400
H	5.30274800	-0.97394200	0.69194400
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C	-2.80851300	-1.35442900	-0.43132000
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H	-1.07502400	-1.30141500	0.85364700
C	-4.10461200	-2.55653600	1.74081700
H	-5.83584500	-2.62629000	0.46274600
H	-2.22079800	-2.35470600	2.76709800
H	-4.60519800	-3.02145800	2.58243900
C	3.53395100	-2.57728300	1.89444200
H	2.60361300	-3.09107600	2.14155200
H	4.36456200	-3.17722300	2.27533600
H	3.54943200	-1.60458500	2.39088100

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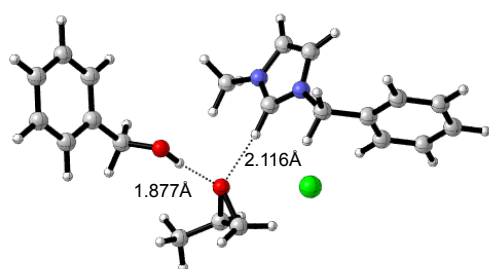
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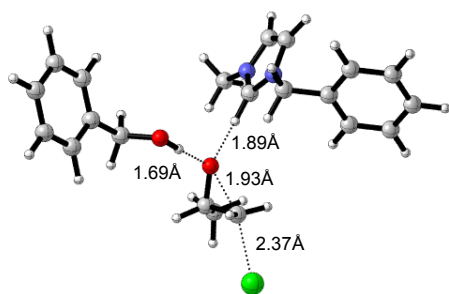


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H	0.92592500	2.13333700	-2.28962300
H	-0.80933700	2.40366500	-2.84382200
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C	0.12760100	0.63005700	3.24873100
H	-0.31689700	1.39149900	2.61320200
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C	2.20586200	-1.95306000	1.77664100
H	2.84674400	-2.81307500	1.69348900
C	1.03837300	-0.20724300	1.08387200
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N	0.90417100	-0.28938000	2.41011400
H	1.72424700	-0.70425700	-1.32913600
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H	1.76496900	-2.43739100	-1.01112900
H	0.57816700	0.53686500	0.44508100
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C	4.32698100	-2.64562800	-1.41767600
C	4.46839100	-0.34605500	-0.69382700
C	5.70390000	-2.66063600	-1.64599400
H	3.73876700	-3.53621500	-1.61242700
C	5.84307600	-0.36414000	-0.91620200
H	3.99803400	0.55999900	-0.32309000
C	6.46377100	-1.52035800	-1.39415600
H	6.17862500	-3.56199500	-2.01646000
H	6.43001300	0.52604600	-0.72066100
H	7.53364900	-1.52996200	-1.56887600
C	-1.55838300	4.11402600	-0.72773800
H	-1.90119800	4.16171300	0.30923100
H	-1.30322700	5.13012600	-1.04260200
H	-2.37649400	3.75238500	-1.35450600

C	-7.23817400	-2.18108200	-0.34537900
C	-7.14320800	-1.12583700	-1.25352600
C	-6.09708400	-0.20941700	-1.15156100
C	-5.13288700	-0.33373700	-0.14506200
C	-5.23886000	-1.39467600	0.76070500
C	-6.28379100	-2.31353200	0.66362600
H	-8.05301500	-2.89244600	-0.42074500
H	-7.88488400	-1.01476700	-2.03677600
H	-6.02950800	0.61223300	-1.85746700
H	-4.50046300	-1.49976200	1.54921900
H	-6.35478000	-3.12872700	1.37519500
C	-3.98174200	0.63671900	-0.06177200
H	-4.30611000	1.62774100	-0.40075700
H	-3.64652700	0.72715600	0.97833500
O	-2.90857400	0.16003400	-0.88709700
H	-2.17154300	0.79548100	-0.83243600

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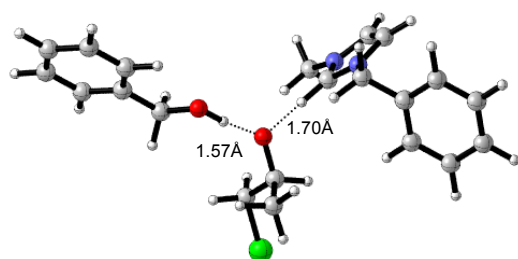
C	0.29382100	2.90310800	-0.27541700
C	-0.45830000	2.79461700	0.98168900
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H	-0.22663400	2.71617200	-1.20019700
H	-1.49355500	3.14896500	0.89645800
O	-0.33307300	1.40636800	0.76539400

C	1.14540900	-0.90977400	4.20240000
H	2.01502200	-0.65763200	4.80801200
H	0.54192000	-0.02014400	4.04022200
H	0.54889000	-1.66870000	4.70673600
H	2.65710700	-3.19219900	3.51028400
C	2.33517900	-2.56671700	2.69596900
C	2.53263400	-2.67922300	1.35588100
H	3.06196600	-3.42022000	0.78329200
C	1.33293000	-0.85225900	1.71772000
N	1.90018200	-1.60054200	0.76444900
N	1.58545000	-1.42192400	2.90050500
H	1.24081800	-0.39899500	-0.78110000
C	1.82573200	-1.31428600	-0.68699700
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H	0.75922200	0.06131600	1.53491200
Cl	0.63253900	5.16933100	-0.89570400
C	3.19151500	-1.16763200	-1.32054000
C	3.65556900	-2.12716500	-2.22372500
C	3.99938500	-0.06416200	-1.02110500
C	4.91106000	-1.99038400	-2.81821200
H	3.03402100	-2.98231800	-2.46752400
C	5.25337700	0.07152200	-1.61074400
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C	5.71208800	-0.89259500	-2.51088300
H	5.25949900	-2.74032200	-3.51906100
H	5.87001900	0.93107400	-1.37410700
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C	0.18778300	3.30896300	2.25419800
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H	-0.34384500	2.91375400	3.12448600
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C	-4.66049700	-0.56302100	-0.20693500
C	-4.89263200	-1.92834100	-0.01013900
C	-5.94547500	-2.57500800	-0.65848900
C	-6.78157800	-1.85983000	-1.51596200
C	-6.56082800	-0.49651200	-1.71880500
H	-5.34226000	1.20567400	-1.22508500
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O	-2.37547400	0.13572400	-0.42027200
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C	-0.81688000	3.21058700	0.09489100
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