

Electronic Supplementary Information

A bisalkynylated 3,6-diiminocyclohexa-1,4-diene-1,4-diamine

Philipp Biegger,^a Manuel Schaffroth,^a Kerstin Broedner,^a Olena Tverskoy,^a Frank Rominger,^a
and Uwe H. F. Bunz^{a,b,*}

^a Organisch-Chemisches Institut, Ruprecht-Karls-Universität Heidelberg, Im Neuenheimer Feld 270, 69120
Heidelberg, Germany.

^b Centre for Advanced Materials, Ruprecht-Karls-Universität Heidelberg, Im Neuenheimer Feld 225, 69120
Heidelberg, Germany.

* Fax: (+49)6221548401; E-mail: uwe.bunz@oci.uni-heidelberg.de

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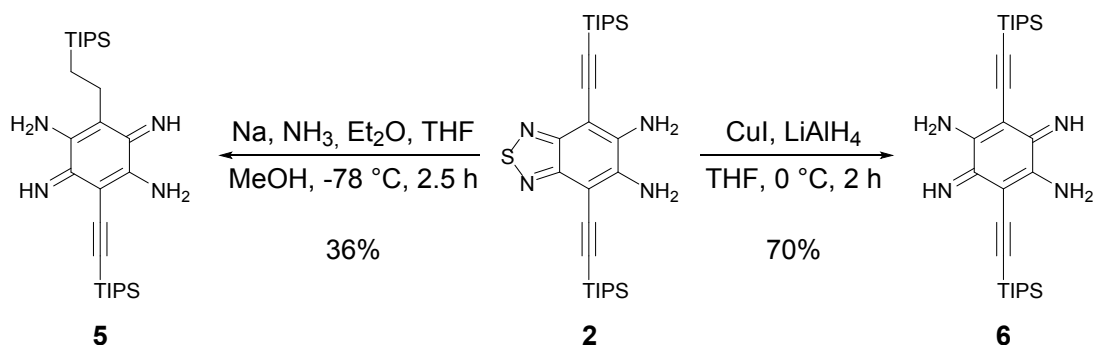
S1. General methods

All reagents and solvents were obtained from Fisher Scientific, ABCR, Sigma-Aldrich or Merck and were used as purchased. Absolute solvents were dried by a MB SPS-800 solvent purification system using drying columns. Thin layer chromatography (TLC) was carried out on Polygram® SIL G/UV₂₅₄ plates from Macherey, Nagel & Co. KG, Düren (Germany) and examined under ultraviolet light irradiation (254 nm and 365 nm). Column chromatography was performed using silica gel from Macherey, Nagel & Co. KG, Düren (Germany) (particle size: 0.032-0.062 mm). Melting points were determined in glass capillaries with a Melting Point Apparatus MEL-TEMP (Electrothermal, Rochford, UK) and were kept uncorrected. Deuterated solvents were purchased from Deutero GmbH, Kastellaun. NMR spectra (¹H, ¹³C) were recorded at room temperature on a Bruker Avance III 400 (400 MHz) at the NMR Spectroscopy Facility of the University of Heidelberg. The spectra were integrated and processed using TopSpin 3.2 (Bruker). Chemical shifts (δ) are given in parts per million (ppm) relative to solvent signals.¹ The following abbreviations describe the signal multiplicity: s = singlet, d = doublet, dd = doublet of doublets, t = triplet, m = multiplet. The ¹³C-NMR signal structure was analyzed by DEPT and is described as follows: + = primary or tertiary C-atom (positive signal), - = secondary C-atom (negative signal), and C_{quart} = quaternary C-atom (no signal). Obvious peak assignments are made by comparison to known substances.² IR spectra were recorded as a solid on a Jasco FT/IR-4100 spectrometer. The intensities are characterized as follows: vs = very strong 0–10% transmission (T), s = strong 10–40% T, m = middle 40–70% T, w = weak 70–90% T, vw = very weak 90–100% T. High resolution mass spectra (HRMS) were obtained by positive matrix-assisted laser desorption/ionization (MALDI), electrospray ionization (ESI) or direct analysis in real time (DART) experiments on a Bruker ApexQe hybrid 9.4 T FT-ICR at the Mass Spectrometry Facility of the University of Heidelberg. Elemental Analysis was accomplished on an Elementar vario MIKRO cube machine by the Microanalytical Laboratory of the University of Heidelberg. Crystal structure analysis was performed on a Bruker Smart CCD or a Bruker APEX-II Quazar diffractometer. Absorption spectra were recorded on a Jasco UV/VIS V-670 spectrophotometer. Emission spectra were recorded on a Jasco FP-6500 spectrofluorometer in solution. Cyclic voltammetry was performed on a VersaSTAT3-200 potentiostat (Princeton Applied Research). Computational studies were carried out using DFT calculations on Gaussian09.³ Geometry optimization was found by B3LYP functional and 6-311+G* basis set and checked for imaginary frequencies. Using this geometry the absolute energy and FMO energies were assigned on single point approach by employing B3LYP/6-311++G**.

S2. Synthetic procedures and analytical data

4,7-bis((triisopropylsilyl)ethynyl)benzo[*c*][1,2,5]thiadiazole-5,6-diamine **2**⁴ and triptycene-2,3-dione **16**⁵ were synthesized according to literature procedures.

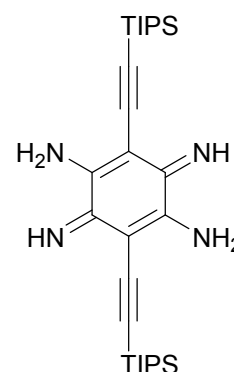
Reduction of 4,7-bis((triisopropylsilyl)ethynyl)benzo[*c*][1,2,5]thiadiazole-5,6-diamine:



Scheme 1 Reduction of benzothiadiazole-diamine **2** leads to different products.

3,6-diimino-2,5-bis((triisopropylsilyl)ethynyl)cyclohexa-1,4-diene-1,4-diamine (**6**)

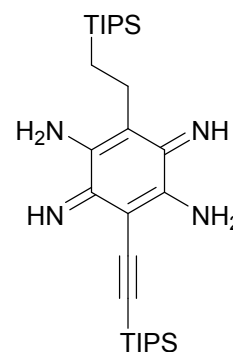
Under an atmosphere of argon, **2** (114 mg, 216 μ mol) was dissolved in dry THF (12 mL). After addition of CuI (103 mg, 541 μ mol, 2.5 eq), the mixture was cooled to 0 °C. LiAlH₄ (82.1 mg, 2.16 mmol, 10 eq) was added in small portions over 10 min. The mixture was stirred for additional 1.5 h at 0 °C. Complete conversion was detected by UPLC/MS analysis. At 0 °C, a saturated solution of sodium potassium tartrate in H₂O (20 mL) was added dropwise, and the mixture was extracted with diethyl ether (3x30 mL). The organic phase was stirred for 15 min over MnO₂ (tip of a spatula). After filtration over Celite, the filtrate was concentrates under reduced pressure. The crude product was purified by column chromatography (SiO₂, petrol ether/ dichloromethane 2:1 \rightarrow 1:1 + 1 % trimethylamine) to yield **6** as a red solid (74.4 mg, 70 %).



R_f-value = 0.30 (dichloromethane) – mp > 238 °C decomp. – $\nu_{\text{max}}/\text{cm}^{-1}$ = 3437 (w), 3247 (w), 3061 (w, br), 2940 (m), 2862 (m), 2134 (m), 1627 (m), 1541 (s), 1459 (w), 1358 (s), 1224 (s), 1065 (w), 994 (w), 961 (w), 876 (s), 707 (m), 674 (m), 587 (w), 473 (w), 449 (w) – ¹H-NMR (THF-d₈, 500 MHz, 298 K): δ [ppm] = 1.14–1.20 (m; 42H), 5.47 (s; br; 2H), 6.75 (s; br; 2H), 9.59 (s; br; 2H) – ¹³C-NMR (250 MHz, THF-d₈, 298 K): δ [ppm] = 12.0, 18.9, 91.6, 100.5, 103.3, 150.2, 159.7 – HRMS (DART (+) FT-ICR) *m/z*: [M+H]⁺ Calc. for C₂₈H₄₉N₄Si₂: 497.3490; Found 497.3488 correct isotope distribution.

3,6-diimino-2-{2-[tri(propan-2-yl)silyl]ethyl}-5-[[tri(propan-2-yl)silyl]ethynyl]cyclohexa-1,4-diene-1,4-diamine (5)

Under an atmosphere of argon, ammonia gas was condensed into a Schlenk tube at -78 °C to give 5 mL of liquid ammonia. Sodium (40 mg, 1.74 mmol, 7.5 eq) was added and stirred for 10 min until everything was dissolved. A solution of **2** (122 mg, 232 μmol) in a mixture of dry diethyl ether (1.5 mL), dry THF (1.0 mL), and dry MeOH (50.0 μL) was added dropwise. After stirring for 2 h at -78 °C, NH₄Cl (2.00 g) was added in portions and ammonia was allowed to volatilize under warming to r.t. H₂O was added and the mixture was extracted with dichloromethane (2x). After drying over Na₂SO₄, filtration and concentration, the crude product was purified by column chromatography (SiO₂, petrol ether/ethyl acetate 100:1 → 4:1). The resulting product was recrystallized from a mixture of diethyl ether and petrol ether to yield **5** as red needles (42 mg, 36 %).

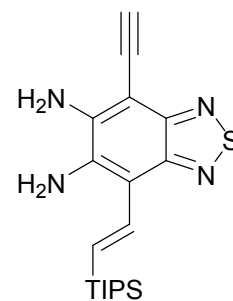


R_F-value = 0.10 (dichloromethane) – mp > 128 °C decomp. – ν_{max}/cm⁻¹ = 3491 (w), 3464 (w), 3281 (m, br), 3244 (w), 2940 (s), 2888 (m), 2863 (s), 2125 (m), 1618 (m), 1530 (s), 1460 (m), 1347 (s), 1216 (m), 997 (m), 876 (s), 838 (w), 740 (w), 699 (s), 671 (s), 657 (s), 590 (w), 574 (w), 442 (w), 427 (s), 417 (w) – ¹H-NMR (CDCl₃, 500 MHz, 298 K): δ [ppm] = 0.68–0.76 (m; 2H), 1.04–1.20 (m; 42H), 2.27–2.35 (m; 2H), 4.88 (s; br; 2H), 5.58 (s; br; 2H), 9.13 (s; br; 1H), 9.47 (s; br; 1H) – ¹³C-NMR (125 MHz, CDCl₃, 298 K): δ [ppm] = 7.7, 11.0, 11.4, 18.9, 19.1, 19.5, 91.8, 99.2, 103.5, 110.7, 140.7, 149.3, 160.7, 160.9 – HRMS (ESI (+) FT-ICR) *m/z*: [M+H]⁺ Calc. for C₂₈H₅₃N₄Si₂: 501.3803; Found 501.3812 correct isotope distribution.

Analytical data of isolated byproducts

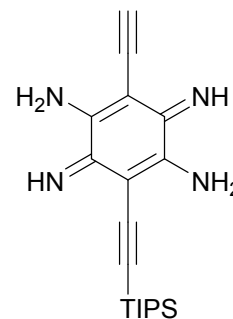
4-ethynyl-7-{(E)-2-[tri(propan-2-yl)silyl]ethenyl}-2,1,3-benzothiadiazole-5,6-diamine (7)

R_F-value = 0.72 (dichloromethane) – mp > 82 °C decomp. – ν_{max}/cm⁻¹ = 3362 (w), 3303 (m), 3235 (w), 2940 (s), 2888 (w), 2862 (s), 2126 (w), 2094 (w), 1615 (m), 1535 (m), 1448 (s), 1364 (m), 1312 (m), 1230 (w), 1007 (w), 995 (w), 881 (m), 819 (w), 790 (w), 707 (w), 654 (m), 561 (w), 514 (w) – ¹H-NMR (THF-d₈, 500 MHz, 298 K): δ [ppm] = 1.14–1.21 (m; 21H), 4.14 (s; 1H), 5.19 (s; 2H), 5.63 (s; 2H), 7.12 (d; *J* = 19.7 Hz; 2H), 7.34 (d; *J* = 19.7 Hz; 2H) – ¹³C-NMR (125 MHz, THF-d₈, 298 K): δ [ppm] = 11.7, 19.0, 78.6, 88.1, 92.2, 110.7, 129.8, 138.7, 138.8, 146.7, 150.7, 152.1 – HRMS (DART (+) FT-ICR) *m/z*: [M+H]⁺ Calc. for C₁₉H₂₉N₄SSi: 373.1877; Found 373.1874 correct isotope distribution.



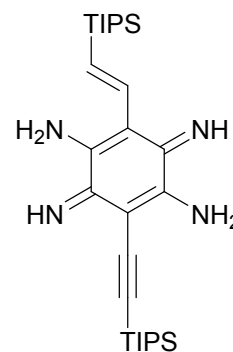
2-ethynyl-3,6-diimino-5-{{tri(propan-2-yl)silyl}ethynyl}cyclohexa-1,4-diene-1,4-diamine (8)

R_f -value = 0.13 (dichloromethane/methanol 95:5, v/v) – mp > 129 °C decomp. – $\nu_{\max}/\text{cm}^{-1}$ = 3453 (w), 3275 (m), 3243 (m), 2940 (m), 2862 (m), 2115 (w), 1625 (m), 1535 (s), 1457 (w), 1538 (w), 1357 (s), 1213 (m), 989 (w), 878 (w), 865 (s), 695 (w), 660 (s), 590 (w), 55 (w), 431 (w), 450 (m), 413 (w), 404 (w) – $^1\text{H-NMR}$ (THF- d_8 , 400 MHz, 298 K): δ [ppm] = 1.14–1.20 (m; 21H), 4.18 (s; 1H), 6.39 (s; br; 4H), 9.56 (s; br; 2H) – $^{13}\text{C-NMR}$ (100 MHz, THF- d_8 , 298 K): δ [ppm] = 12.0, 19.0, 77.0, 90.1, 90.4, 91.6, 100.6, 103.2, 150.3, 150.4, 159.7, 159.8 – HRMS (ESI (+) FT-ICR) m/z : $[\text{M}+\text{H}]^+$ Calc. for $\text{C}_{19}\text{H}_{29}\text{N}_4\text{Si}$: 341.2156; Found 341.2158 correct isotope distribution.



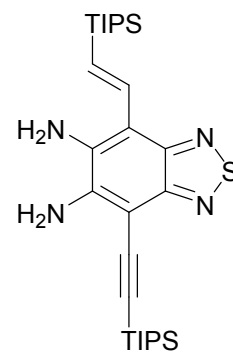
3,6-diimino-2-{{(E)-2-{{tri(propan-2-yl)silyl}ethenyl}}-5-{{tri(propan-2-yl)silyl}ethynyl}cyclohexa-1,4-diene-1,4-diamine (9)

R_f -value = 0.21 (petroleum ether/diethyl ether 4:1, v/v) – mp > 170 °C decomp. – $\nu_{\max}/\text{cm}^{-1}$ = 3443 (w), 34247 (w), 3095 (w, br), 2940 (m), 2863 (m), 2133 (w), 1732 (w), 1625 (m), 1529 (s), 1461 (m), 1355 (s), 1222 (w), 1069 (w), 1039 (w), 1011 (w), 994 (w), 878 (s), 706 (w), 668 (w), 492 (w), 469 (w), 448 (w), 435 (w), 410 (w) – $^1\text{H-NMR}$ (THF- d_8 , 500 MHz, 295 K): δ [ppm] = 1.13–1.16 (m; 21H), 1.16–1.19 (m; 21H), 5.68 (s; br; 4H), 6.00 (d; J = 19.9 Hz; 1H), 6.69 (d; J = 19.9 Hz; 1H), 9.46 (s; br; 1H), 9.56 (s; br; 1H) – $^{13}\text{C-NMR}$ (125 MHz, THF- d_8 , 295 K): δ [ppm] = 11.5, 12.0, 19.0, 91.1, 101.0, 102.4, 107.4, 131.1, 139.5, 142.7, 150.7, 160.3, 160.7 – HRMS (DART (+) FT-ICR) m/z : $[\text{M}+\text{H}]^+$ Calc. for $\text{C}_{28}\text{H}_{51}\text{N}_4\text{Si}_2$: 499.3647; Found 499.3641 correct isotope distribution.

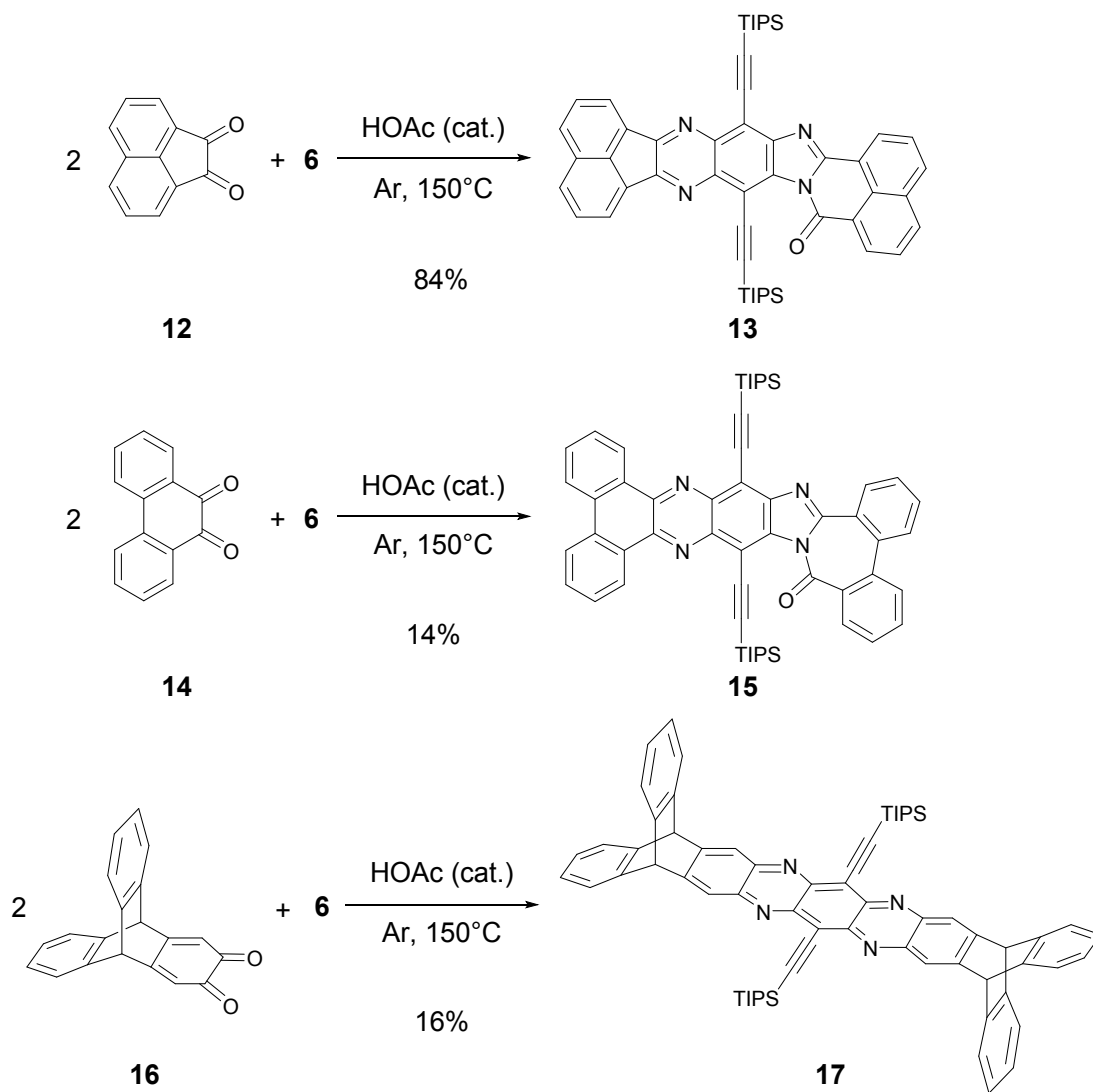


4-{{(E)-2-{{tri(propan-2-yl)silyl}ethenyl}}-7-{{tri(propan-2-yl)silyl}ethynyl}}-2,1,3-benzothiadiazole-5,6-diamine (10)

R_f -value = 0.08 (dichloromethane) – mp > 150 °C decomp. – $\nu_{\max}/\text{cm}^{-1}$ = 3459 (w), 3370 (w), 2940 (s), 2887 (w), 2863 (s), 2137 (w), 1648 (w), 1621 (w), 1586 (w), 1451 (s), 1366 (m), 1314 (w), 1072 (w), 1012 (w), 994 (m), 882 (s), 734 (m), 674 (m), 508 (w), 501 (w), 494 (w), 463 (w), 435 (w), 420 (m), 413 (w) – $^1\text{H-NMR}$ (THF- d_8 , 500 MHz, 295 K): δ [ppm] = 1.14–1.19 (m; 21H), 1.20–1.22 (m; 21H), 5.16 (s; 2H), 5.48 (s; 2H), 7.12 (d; J = 19.5 Hz; 2H), 7.35 (d; J = 19.5 Hz; 2H) – $^{13}\text{C-NMR}$ (125 MHz, THF- d_8 , 295 K): δ [ppm] = 11.7, 12.1, 19.0, 19.1, 93.8, 100.7, 102.3, 110.9, 130.0, 138.6, 138.8, 146.3, 150.7, 151.9 – HRMS (DART (+) FT-ICR) m/z : $[\text{M}+\text{H}]^+$ Calc. for $\text{C}_{28}\text{H}_{49}\text{N}_4\text{SSi}_2$: 529.3211; Found 529.3211 correct isotope distribution.



Condensation reactions of diaminoquinoneimine **6**:



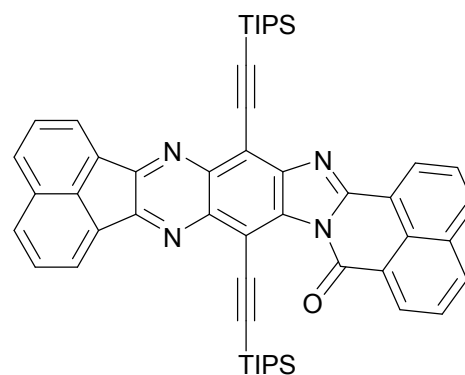
Scheme 2 Condensation reactions of diaminoquinoneimine **6** with commercially available *ortho*-quinones **13** and **15** and triptycene-2,3-dione **17**.

General Procedure (GP):

In a heat-gun dried thin schlenk tube *ortho*-quinone and diaminoquinoneimine were mixed and rinsed down with three drops of acetic acid under an argon atmosphere. The tube was closed with a glass stopper and tightened with a screw clamp and the reaction was heated for 16 h at 150 °C under stirring. The mixture was cooled to room temperature and directly subjected to column chromatography.

8,18-bis((triisopropylsilyl)ethynyl)-16H-acenaphtho[1,2-b]benzo[4',5']isoquinolino[2',1':1,2]imidazo[4,5-g]quinoxalin-16-one (13)

GP was carried out with 26.0 mg (143 μmol , 2.00 eq.) 1,2-acenaphthylenedione and 35.5 mg (71.4 μmol , 1.00 eq.) diaminoquinoneimine **6**. Column chromatography (silica gel; 25 \times 3; petroleum ether/ethyl acetate gradient from 1:0 to 2:1, v/v) to yield an orange fluorescent solid. Yield: 39.4 mg (48.8 μmol , 84%).

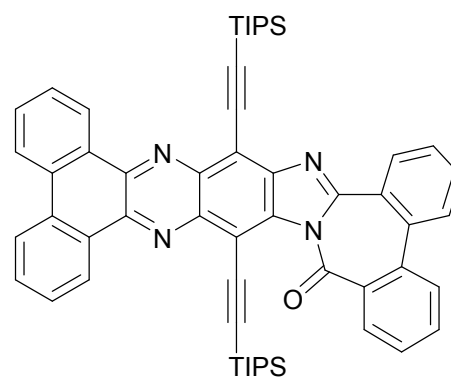


R_f -value = 0.28 (petroleum ether/ethyl acetate 9:1, v/v) – mp

> 335 $^{\circ}\text{C}$ decomp. – $\nu_{\text{max}}/\text{cm}^{-1}$ = 3060 (vw), 2924 (w), 2862 (w), 2162 (vw), 1722 (w), 1615 (vw), 1596 (w), 1543 (vw), 1506 (vw), 1462 (w), 1437 (w), 1422 (vw), 1396 (vw), 1363 (vw), 1339 (w), 1328 (w), 1299 (vw), 1265 (vw), 1206 (vw), 1177 (vw), 1167 (vw), 1129 (w), 1111 (w), 1072 (w), 1032 (w), 1016 (w), 999 (w), 908 (w), 882 (w), 835 (vw), 823 (w), 800 (w), 768 (m), 759 (m), 732 (vw), 716 (vw), 698 (vw), 673 (w), 659 (w), 611 (vw), 596 (w), 579 (w), 557 (w), 531 (vw), 518 (vw), 505 (vw), 488 (vw), 466 (w), 454 (w), 420 (vw), 402 (vw) – $^1\text{H-NMR}$ (CDCl_3 , 500.13 MHz, 295.0 K): δ [ppm] = 1.38–1.40 (m, 21 H, $^i\text{Pr-H}$), 1.40–1.42 (m, 21 H, $^i\text{Pr-H}$), 7.84–7.87 (m, 4 H), 8.11 (dd, 2 H, $J = 2.7, 8.0$ Hz), 8.19 (d, 1 H, $J = 8.0$ Hz), 8.27 (d, 1 H, $J = 8.0$ Hz), 8.42 (t, 2 H, $J = 6.6$ Hz), 8.72 (d, 1 H, $J = 6.6$ Hz), 8.96 (d, 1 H, $J = 6.6$ Hz) – $^{13}\text{C-NMR}$ (150.95 MHz, CDCl_3 , 295.0 K): δ [ppm] = 11.88 (+, CHCH_3), 11.93 (+, CHCH_3), 19.12 (+, CHCH_3), 19.16 (+, CHCH_3), 100.90 (C_{quart} , CCSi), 101.08 (C_{quart} , CCSi), 105.59 (C_{quart} , CCSi), 109.57 (C_{quart} , CCSi), 111.12 (C_{quart}), 114.35 (C_{quart}), 120.84 (C_{quart}), 122.25 (+), 122.27 (+), 124.30 (C_{quart}), 127.31 (+), 127.59 (+), 127.85 (C_{quart}), 128.53 (+), 128.83 (+), 128.88 (+), 129.46 (+), 129.59 (+), 130.16 (C_{quart}), 131.62 (+), 132.25 (C_{quart}), 132.34 (C_{quart}), 132.63 (+), 133.96 (C_{quart}), 134.72 (+), 137.27 (C_{quart}), 132.63 (C_{quart}), 140.99 (C_{quart}), 142.16 (C_{quart}), 148.10 (C_{quart}), 153.81 (C_{quart}), 154.03 (C_{quart}), 154.08 (C_{quart}), 158.73 (C_{quart}) – HRMS (MALDI FT-ICR) m/z : $[\text{M}+\text{H}]^+$ Calc. for $\text{C}_{52}\text{H}_{55}\text{N}_4\text{OSi}_2$: 807.3908; Found 807.3903 correct isotope distribution.

10,22-bis((triisopropylsilyl)ethynyl)-20H-dibenzo[a,c]dibenzo[3',4':5',6']azepino[1',2':1,2]imidazo[4,5-i]phenazin-20-one (15)

GP was carried out with 25.2 mg (121 μmol , 2.05 eq.) 9,10-phenanthrenequinone and 29.3 mg (59.0 μmol , 1.00 eq.) diaminoquinoneimine **6**. Column chromatography (silica gel; 20 \times 2; petroleum ether/dichloromethane gradient from 1:0 to 2:1, v/v)

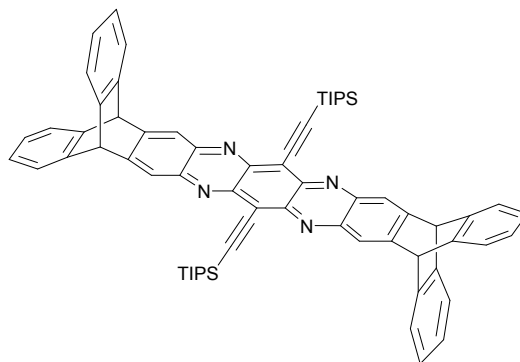


to yield an orange fluorescent solid. Yield: 7 mg (8.30 μmol , 14%).

R_f -value = 0.21 (petroleum ether/dichloromethane 2:1, v/v) – m.p. 243 $^{\circ}\text{C}$ – $\nu_{\text{max}}/\text{cm}^{-1}$ = 3070 (vw), 2939 (w), 2925 (w), 2862 (w), 2360 (w), 2342 (w), 2162 (vw), 2141 (vw), 1731 (w), 1597 (vw), 1516 (vw), 1493 (vw), 1459 (w), 1449 (w), 1382 (vw), 1356 (w), 1333 (w), 1284 (w), 1265 (w), 1248 (w), 1200 (vw), 1187 (vw), 1162 (vw), 1144 (vw), 1118 (w), 1092 (w), 1073 (w), 1042 (vw), 1017 (w), 995 (w), 960 (w), 914 (w), 881 (w), 818 (vw), 797 (vw), 781 (w), 760 (m), 733 (m), 725 (m), 706 (w), 674 (w), 658 (w), 628 (w), 582 (w), 551 (w), 529 (vw), 500 (w), 482 (vw), 471 (vw), 458 (vw), 432 (vw), 418 (vw), 403 (vw) – $^1\text{H-NMR}$ (CDCl_3 , 600.25 MHz, 295.0 K): δ [ppm] = 1.36–1.39 (m, 21 H, $i\text{Pr-H}$), 1.40–1.43 (m, 21 H, $i\text{Pr-H}$), 7.59 (t, J = 7.8 Hz, 1 H), 7.67–7.82 (m, 8 H), 7.86 (d, 1 H, J = 7.8 Hz), 8.19 (d, 1 H, J = 6.8 Hz), 8.57 (dd, 2 H, J = 3.3, 7.8 Hz), 8.88 (d, 1 H, J = 7.8 Hz), 9.61–9.65 (m, 2 H) – $^{13}\text{C-NMR}$ (150.95 MHz, CDCl_3 , 295.0 K): δ [ppm] = 11.83 (+, CHCH_3), 12.27 (+, CHCH_3), 19.09 (+, CHCH_3), 19.17 (+, CHCH_3), 99.68 (C_{quart} , CCSi), 101.07 (C_{quart} , CCSi), 105.91 (C_{quart} , CCSi), 106.86 (C_{quart} , CCSi), 110.57 (C_{quart}), 113.46 (C_{quart}), 122.98 (+), 123.02 (+), 126.12 (C_{quart}), 127.07 (+), 127.40 (+), 127.60 (+), 127.92(+), 128.69 (+), 129.27 (+), 129.70 (+), 129.82 (+), 130.53 (+), 130.55 (+), 130.59 (C_{quart}), 130.66 (C_{quart}), 130.99 (+), 132.34 (C_{quart}), 132.44 (C_{quart}), 132.53 (+), 132.62 (+), 133.30 (+), 135.14 (C_{quart}), 135.68 (C_{quart}), 135.97 (C_{quart}), 137.20 (C_{quart}), 140.39 (C_{quart}), 141.87 (C_{quart}), 141.93 (C_{quart}), 141.98 (C_{quart}), 148.14 (C_{quart}), 154.98 (C_{quart}), 166.01 (C_{quart}) – HRMS (DART FT-ICR) m/z : $[\text{M}+\text{H}]^+$ Calc. for $\text{C}_{56}\text{H}_{59}\text{N}_4\text{OSi}_2$ 859.4222; Found 859.4255 correct isotope distribution.

8,19-Bis(triisopropylsilylethynyl)-9,14-dihydro-5,22[1',2'],11,16[1',2']-dibenzeno-7,9,18,20-tetraazonacene (17)

GP was carried out with 26.4 mg (92.7 μmol , 4.85 eq.) 9,10-phenanthrenequinone and 9.5 mg (19.1 μmol , 1.00 eq.) diaminoquinoneimine **6**. Column chromatography (silica gel; 30 \times 3; petroleum ether/ethyl acetate gradient from 1:0 to 2:1, v/v) to yield a darkish green solid, which could only be characterized by single crystal X-ray structure and mass spectrometry due to the small isolated amount. Yield: 2 mg (not NMR pure).



HRMS (MALDI FT-ICR) m/z : $[\text{M}+\text{H}_2]^+$ Calc. for $\text{C}_{68}\text{H}_{68}\text{N}_4\text{Si}_2$: 996.4977; Found 996.4955 correct isotope distribution.

S3. Full scaled absorption and emission spectra

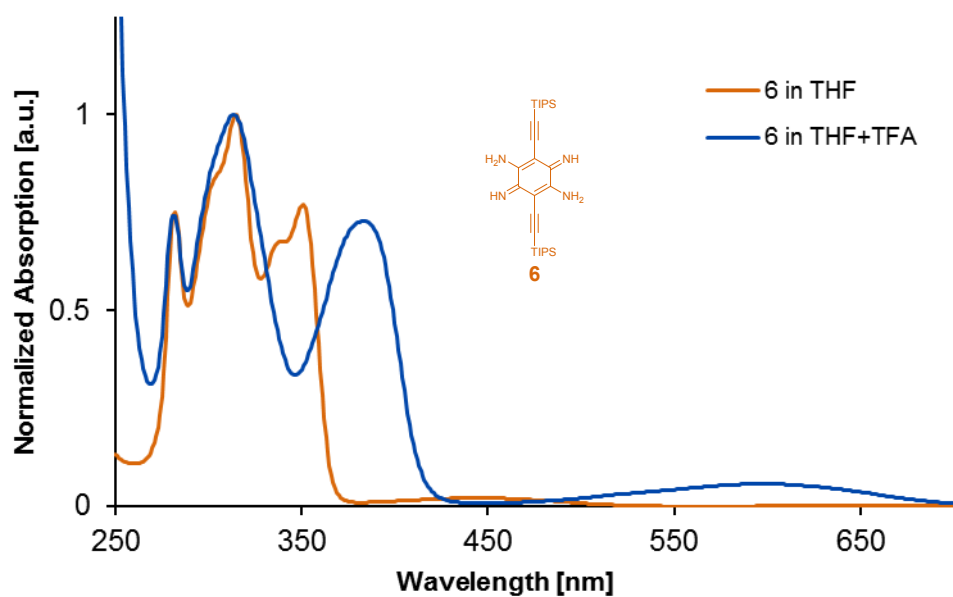


Figure 1 UV/Vis-absorption spectrum of **6** in tetrahydrofuran (orange) and with trifluoroacetic acid (blue).

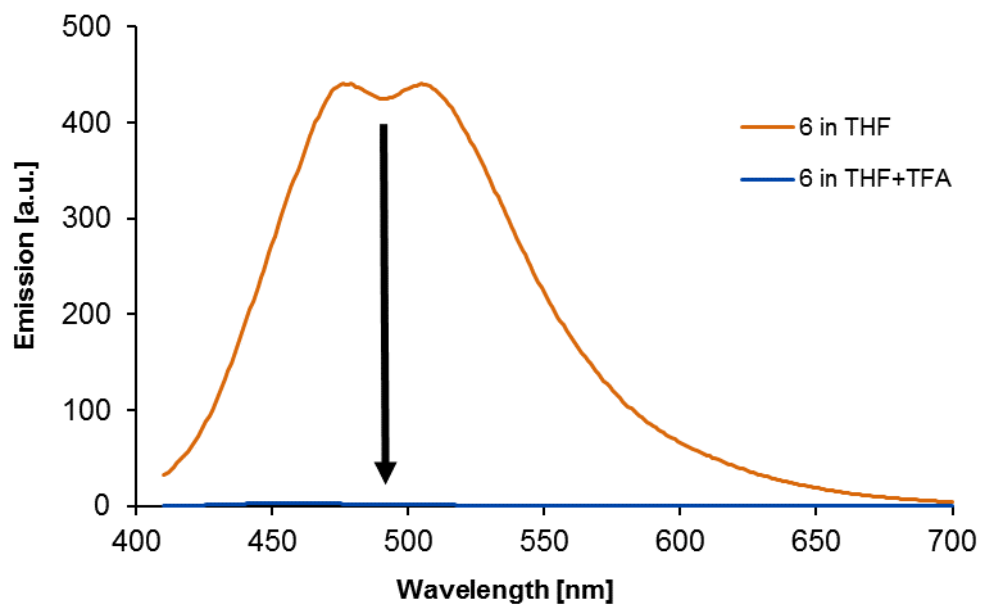


Figure 2 UV/Vis-emission spectrum of **6** in tetrahydrofuran (orange) and with trifluoroacetic acid (blue).

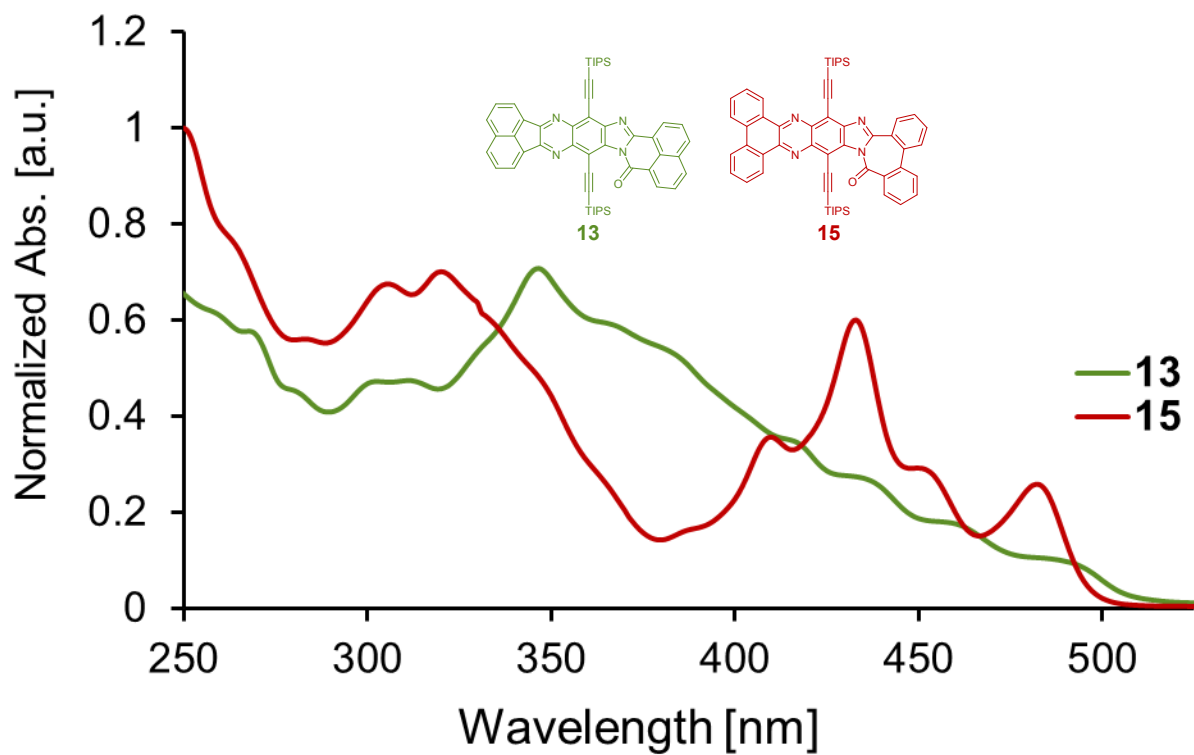


Figure 3 UV/Vis-absorption spectra of **13** and **15** in hexane.

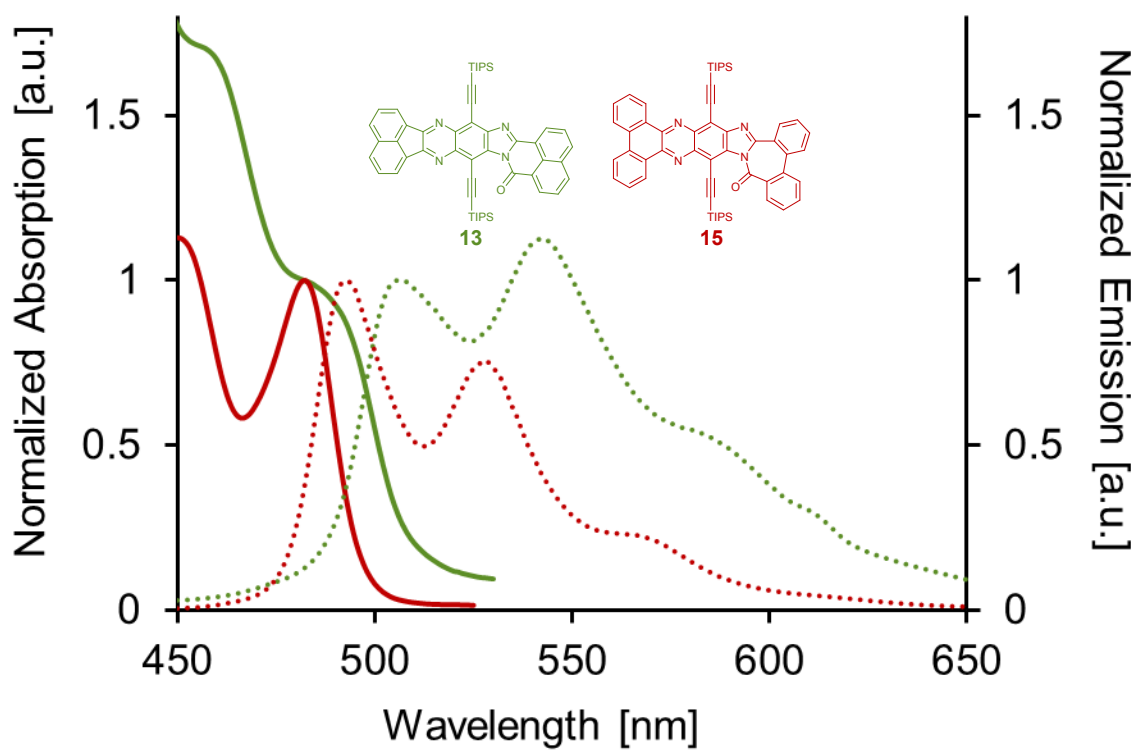


Figure 4 Long-wavelength absorption and emission spectra of **13** and **15** in hexane.

S4. Cyclic voltammograms

The cyclic voltammetry experiments were carried out using a platinum working electrode, a platinum/titanium wire auxiliary electrode, a silver wire pseudo-reference electrode, a 0.1 M NBu₄PF₆ solution in degassed dry tetrahydrofuran and ferrocene/ferrocenium as reference redox system and internal standard (-5.1 eV⁶). For determining the first and second reduction potential $E^{(0/-)}$ of **13** and **15** and the first oxidation potential of ferrocene the half-wave potentials were used.

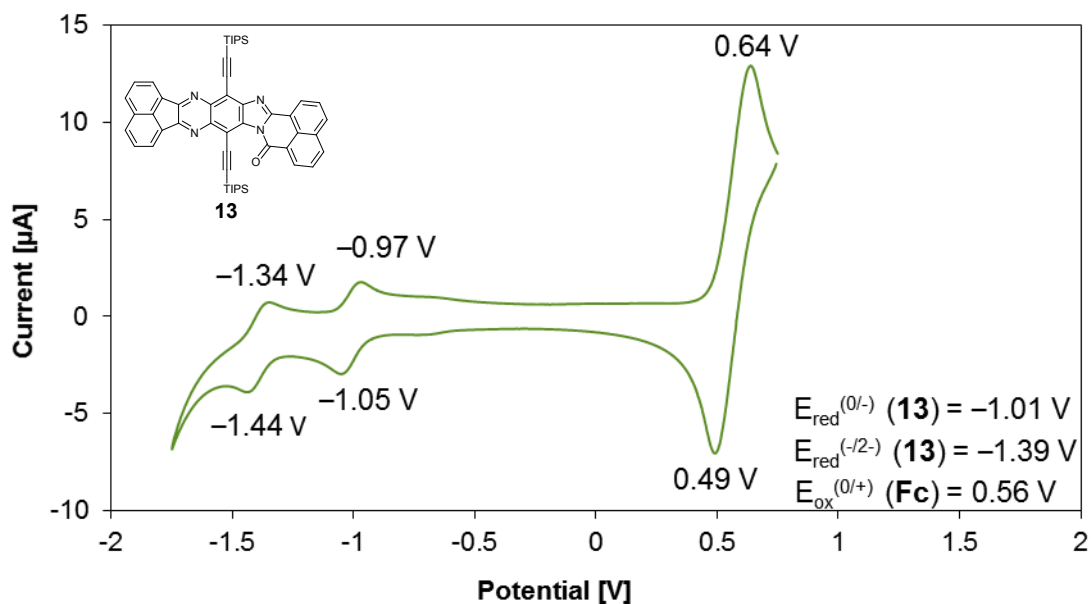
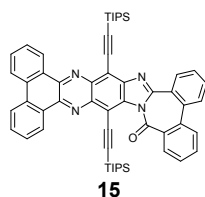


Figure 5 Cyclic voltammogram of **13**.



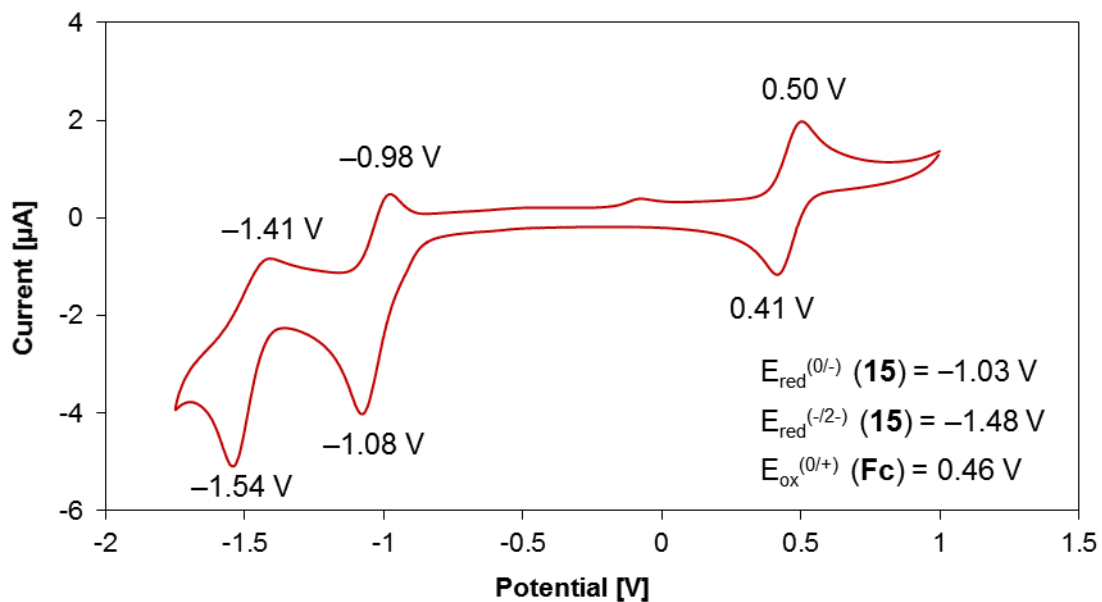
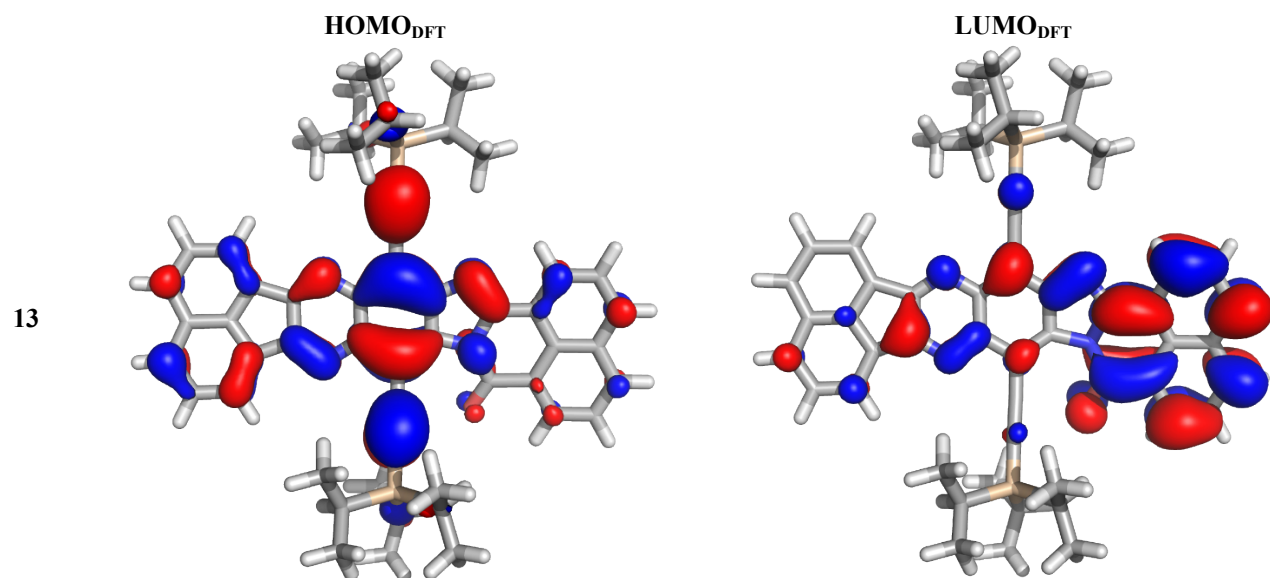
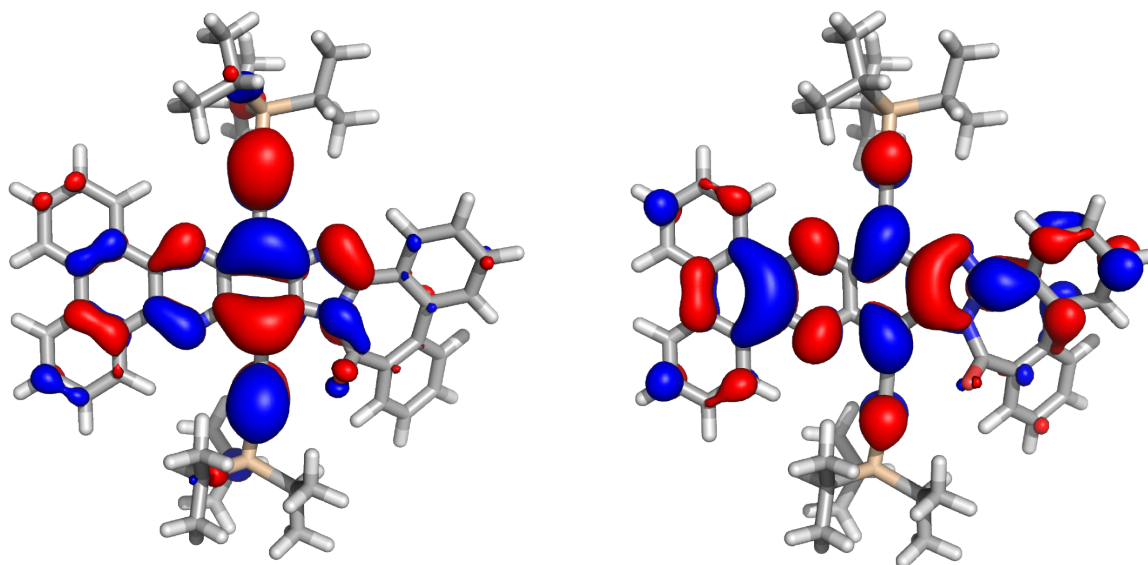


Figure 6 Cyclic voltammogram of **15**.

S5. Calculated frontier molecular orbitals (FMOs)



15



The geometry optimization was performed by Gaussian09 employing B3LYP 6-311+G*, and a single-point energy calculation (Gaussian09, B3LYP 6-311++G**) followed.²

Figure 7 Quantum-chemical calculations (B3LYP 6-311+G*/B3LYP 6-311++G**) of the FMOs for compounds **13** and **15**.²

S6. Crystallographic experiments

Crystal data for **1**: $C_{28}H_{42}N_4S_2Si_2$, $M = 554.95$, purple crystal (plate) obtained by slow evaporation of deuterated chloroform, dimensions $0.130 \times 0.070 \times 0.060 \text{ mm}^3$, monoclinic, $a = 8.5203(3) \text{ \AA}$, $b = 15.4125(6) \text{ \AA}$, $c = 11.7589(5) \text{ \AA}$, $\alpha = 90^\circ$, $\beta = 99.3033(9)^\circ$, $\gamma = 90^\circ$, $V = 1523.86(10) \text{ \AA}^3$, $\rho = 1.209 \text{ g/cm}^3$, $T = 200(2) \text{ K}$, space group $P2_1/n$, $Z = 2$, $\theta_{max} = 26.449^\circ$, radiation Mo K_{α} , $\lambda = 0.71073 \text{ \AA}$, 0.5° omega-scans with CCD area detector, covering the asymmetric unit in reciprocal space with a mean redundancy of 4.48 and a completeness of 98.3% to a resolution of 0.80 \AA , 14317 reflections measured, 3094 unique ($R_{int} = 0.0302$), 2623 observed ($I > 2\sigma(I)$), intensities were corrected for Lorentz and polarization effects, an empirical absorption correction was applied using SADABS (2012/1)⁷ based on the Laue symmetry of the reciprocal space, $\mu = 0.28 \text{ mm}^{-1}$, $T_{min} = 0.87$, $T_{max} = 0.96$, structure refined against F^2 with a Full-matrix least-squares algorithm using the SHELXL-2014/7 (Sheldrick, 2014) software⁸, 163 parameters refined, hydrogen atoms were treated using appropriate riding models, goodness of fit 1.07 for observed reflections, final residual values $R_I(F) = 0.037$, $wR(F^2) = 0.092$ for observed reflections, residual electron density -0.21 to 0.36 e\AA^{-3} . CCDC 1408361 contains the supplementary crystallographic data for this paper. These data can be obtained free of charge from The Cambridge Crystallographic Data Centre via www.ccdc.cam.ac.uk/data_request/cif.

Crystal data for **5**: $C_{28}H_{52}N_4Si_2$, $M = 500.91$, orange crystal (needle) obtained by slow evaporation of a 1:1 mixture of diethyl ether/petroleum ether, dimensions $0.150 \times 0.080 \times 0.070 \text{ mm}^3$, triclinic, $a = 7.8130(18) \text{ \AA}$, $b = 12.886(3) \text{ \AA}$, $c = 15.976(4) \text{ \AA}$, $\alpha = 102.557(5)^\circ$, $\beta = 96.908(6)^\circ$, $\gamma = 93.896(6)^\circ$, $V = 1551.2(6) \text{ \AA}^3$, $\rho = 1.072 \text{ g/cm}^3$, $T = 200(2) \text{ K}$, space group $P\bar{1}$, $Z = 2$, $\theta_{max} = 21.942^\circ$, radiation Mo K_{α} , $\lambda = 0.71073 \text{ \AA}$, 0.5° omega-scans with CCD area detector, covering the asymmetric unit in reciprocal space with a mean redundancy of 2.86 and a completeness of 99.9% to a resolution of 0.95 \AA , 10797 reflections measured, 3777 unique ($R_{int} = 0.0627$), 2334 observed ($I > 2\sigma(I)$), intensities were corrected for Lorentz and polarization effects, an empirical absorption correction was applied using SADABS (2012/1)⁷ based on the Laue symmetry of the reciprocal space, $\mu = 0.14 \text{ mm}^{-1}$, $T_{min} = 0.84$, $T_{max} = 0.96$, structure refined against F^2 with a Full-matrix least-squares algorithm using the SHELXL-2014/7 (Sheldrick, 2014) software⁸, 359 parameters refined, hydrogen atoms were treated using appropriate riding models, except those at the nitrogen atoms, which were refined restrained, goodness of fit 1.04 for observed reflections, final residual values $R_I(F) = 0.059$, $wR(F^2) = 0.122$ for observed reflections, residual electron density -0.20 to 0.22 e\AA^{-3} . CCDC 1408362 contains the supplementary crystallographic data for this paper. These data can be obtained free of charge from The Cambridge Crystallographic Data Centre via www.ccdc.cam.ac.uk/data_request/cif.

Crystal data for **6**: $C_{28}H_{48}N_4Si_2$, $M = 496.88$, orange crystal (block) obtained by slow evaporation of a 1:1 mixture of diethyl ether/petroleum ether, dimensions $0.180 \times 0.130 \times 0.110 \text{ mm}^3$, monoclinic, $a = 19.958(2) \text{ \AA}$, $b = 14.9301(16) \text{ \AA}$, $c = 14.7268(16) \text{ \AA}$, $\alpha = 90^\circ$, $\beta = 132.098(2)^\circ$, $\gamma = 90^\circ$, $V = 3256.1(6) \text{ \AA}^3$, $\rho = 1.014 \text{ g/cm}^3$, $T = 200(2) \text{ K}$, space group $C2/c$, $Z = 4$, $\theta_{max} = 25.037^\circ$, radiation $Mo K\alpha$, $\lambda = 0.71073 \text{ \AA}$, 0.5° omega-scans with CCD area detector, covering the asymmetric unit in reciprocal space with a mean redundancy of 3.51 and a completeness of 99.7% to a resolution of 0.84 \AA , 10051 reflections measured, 2866 unique ($R_{int} = 0.0500$), 1938 observed ($I > 2\sigma(I)$), intensities were corrected for Lorentz and polarization effects, an empirical absorption correction was applied using SADABS (2012/1)⁷ based on the Laue symmetry of the reciprocal space, $\mu = 0.13 \text{ mm}^{-1}$, $T_{min} = 0.67$, $T_{max} = 0.96$, structure refined against F^2 with a Full-matrix least-squares algorithm using the SHELXL-2014/7 (Sheldrick, 2014) software⁸, 248 parameters refined, hydrogen atoms were treated using appropriate riding models, goodness of fit 1.21 for observed reflections, final residual values $R_1(F) = 0.086$, $wR(F^2) = 0.239$ for observed reflections, residual electron density -0.19 to 0.28 e\AA^{-3} . CCDC 1408363 contains the supplementary crystallographic data for this paper. These data can be obtained free of charge from The Cambridge Crystallographic Data Centre via www.ccdc.cam.ac.uk/data_request/cif.

Crystal data for **8**: $C_{19}H_{28}N_4Si$, $M = 340.54$, yellow crystal (needle) obtained by slow evaporation of a 1:1 mixture of diethyl ether/petroleum ether, dimensions $0.080 \times 0.050 \times 0.030 \text{ mm}^3$, monoclinic, $a = 37.446(8) \text{ \AA}$, $b = 7.8246(18) \text{ \AA}$, $c = 30.575(7) \text{ \AA}$, $\alpha = 90^\circ$, $\beta = 116.744(5)^\circ$, $\gamma = 90^\circ$, $V = 8000(3) \text{ \AA}^3$, $\rho = 1.131 \text{ g/cm}^3$, $T = 200(2) \text{ K}$, space group $C2/c$, $Z = 16$, $\theta_{max} = 17.222^\circ$, radiation $Mo K\alpha$, $\lambda = 0.71073 \text{ \AA}$, 0.5° omega-scans with CCD area detector, covering the asymmetric unit in reciprocal space with a mean redundancy of 6.24 and a completeness of 100.0% to a resolution of 1.20 \AA , 15868 reflections measured, 2428 unique ($R_{int} = 0.1858$), 1347 observed ($I > 2\sigma(I)$), intensities were corrected for Lorentz and polarization effects, an empirical absorption correction was applied using SADABS (2012/1)⁷ based on the Laue symmetry of the reciprocal space, $\mu = 0.12 \text{ mm}^{-1}$, $T_{min} = 0.82$, $T_{max} = 0.96$, structure refined against F^2 with a Full-matrix least-squares algorithm using the SHELXL-2014/7 (Sheldrick, 2014) software⁸, 474 parameters refined, hydrogen atoms were treated using appropriate riding models, goodness of fit 1.02 for observed reflections, final residual values $R_1(F) = 0.066$, $wR(F^2) = 0.134$ for observed reflections, residual electron density -0.28 to 0.22 e\AA^{-3} . CCDC 1408364 contains the supplementary crystallographic data for this paper. These data can be obtained free of charge from The Cambridge Crystallographic Data Centre via www.ccdc.cam.ac.uk/data_request/cif.

Crystal data for **10**: $C_{28}H_{48}N_4SSi_2$, $M = 528.94$, yellow crystal (plate) obtained by slow evaporation of deuterated chloroform, dimensions $0.210 \times 0.170 \times 0.040 \text{ mm}^3$, monoclinic, $a = 15.4408(14) \text{ \AA}$, $b =$

13.9268(12) Å, $c = 14.8221(13)$ Å, $\alpha = 90^\circ$, $\beta = 103.112(3)^\circ$, $\gamma = 90^\circ$, $V = 3104.3(5)$ Å³, $\rho = 1.132$ g/cm³, $T = 200(2)$ K, space group P2₁/c, $Z = 4$, $\theta_{max} = 20.816^\circ$, radiation Mo K_{α} , $\lambda = 0.71073$ Å, 0.5° omega-scans with CCD area detector, covering the asymmetric unit in reciprocal space with a mean redundancy of 7.29 and a completeness of 100.0% to a resolution of 1.00 Å, 24289 reflections measured, 3257 unique ($R_{int} = 0.0758$), 2347 observed ($I > 2\sigma(I)$), intensities were corrected for Lorentz and polarization effects, an empirical absorption correction was applied using SADABS (2012/1)⁷ based on the Laue symmetry of the reciprocal space, $\mu = 0.20\text{mm}^{-1}$, $T_{min} = 0.81$, $T_{max} = 0.86$, structure refined against F^2 with a Full-matrix least-squares algorithm using the SHELXL-2014/7 (Sheldrick, 2014) software⁸, 345 parameters refined, hydrogen atoms were treated using appropriate riding models, except those at the NH₂-groups, which were refined isotropically, goodness of fit 1.04 for observed reflections, final residual values $R_I(F) = 0.054$, $wR(F^2) = 0.107$ for observed reflections, residual electron density -0.31 to 0.24 eÅ⁻³. CCDC 1408365 contains the supplementary crystallographic data for this paper. These data can be obtained free of charge from The Cambridge Crystallographic Data Centre via www.ccdc.cam.ac.uk/data_request/cif.

Crystal data for **15**: C₅₆H₅₈N₄OSi₂, $M = 859.24$, orange crystal (needle) obtained by slow evaporation of a 1:1 mixture of dichloromethane/petroleum ether, dimensions 0.170 x 0.030 x 0.030 mm³, monoclinic, $a = 37.073(4)$ Å, $b = 11.0554(13)$ Å, $c = 26.576(3)$ Å, $\alpha = 90^\circ$, $\beta = 118.358(3)^\circ$, $\gamma = 90^\circ$, $V = 9585.2(19)$ Å³, $\rho = 1.191$ g/cm³, $T = 200(2)$ K, space group C2/c, $Z = 8$, $\theta_{max} = 19.588^\circ$, radiation Mo K_{α} , $\lambda = 0.71073$ Å, 0.5° omega-scans with CCD area detector, covering the asymmetric unit in reciprocal space with a mean redundancy of 5.98 and a completeness of 100.0% to a resolution of 1.06 Å, 26024 reflections measured, 4224 unique ($R_{int} = 0.1494$), 2239 observed ($I > 2\sigma(I)$), intensities were corrected for Lorentz and polarization effects, an empirical absorption correction was applied using SADABS1 (2012/1)⁷ based on the Laue symmetry of the reciprocal space, $\mu = 0.12\text{mm}^{-1}$, $T_{min} = 0.82$, $T_{max} = 0.96$, structure refined against F^2 with a Full-matrix least-squares algorithm using the SHELXL-2014/7 (Sheldrick, 2014) software⁸, 650 parameters refined, hydrogen atoms were treated using appropriate riding models, goodness of fit 1.02 for observed reflections, final residual values $R_I(F) = 0.075$, $wR(F^2) = 0.152$ for observed reflections, residual electron density -0.32 to 0.43 eÅ⁻³. CCDC 1408366 contains the supplementary crystallographic data for this paper. These data can be obtained free of charge from The Cambridge Crystallographic Data Centre via www.ccdc.cam.ac.uk/data_request/cif.

Crystal data for **17**: C₆₈H₆₆N₄Si₂, $M = 995.47$, green crystal (plate) obtained by slow evaporation of a 1:1 mixture of dichloromethane/petroleum ether, dimensions 0.140 x 0.100 x 0.030 mm³, monoclinic, $a = 18.0690(12)$ Å, $b = 24.9858(17)$ Å, $c = 29.3954(19)$ Å, $\alpha = 90^\circ$, $\beta = 101.4712(17)^\circ$, $\gamma = 90^\circ$, $V =$

13006.0(15) Å³, $\rho = 1.017$ g/cm³, $T = 200(2)$ K, space group P2₁/n, $Z = 8$, $\theta_{max} = 20.816^\circ$, radiation Mo K_{α} , $\lambda = 0.71073$ Å, 0.5° omega-scans with CCD area detector, covering the asymmetric unit in reciprocal space with a mean redundancy of 5.66 and a completeness of 100.0% to a resolution of 1.00 Å, 78017 reflections measured, 13606 unique ($R_{int} = 0.1000$), 8280 observed ($I > 2\sigma(I)$), intensities were corrected for Lorentz and polarization effects, an empirical absorption correction was applied using SADABS (2012/1)⁷ based on the Laue symmetry of the reciprocal space, $\mu = 0.09$ mm⁻¹, $T_{min} = 0.81$, $T_{max} = 0.96$, structure refined against F^2 with a Full-matrix least-squares algorithm using the SHELXL-2014/7 (Sheldrick, 2014) software⁸, 1443 parameters refined, hydrogen atoms were treated using appropriate riding models, goodness of fit 1.03 for observed reflections, final residual values $R_1(F) = 0.086$, $wR(F^2) = 0.197$ for observed reflections, residual electron density -0.25 to 0.48 eÅ⁻³. CCDC 1408367 contains the supplementary crystallographic data for this paper. These data can be obtained free of charge from The Cambridge Crystallographic Data Centre via www.ccdc.cam.ac.uk/data_request/cif.

S7. Full scaled crystal structures and packing motifs

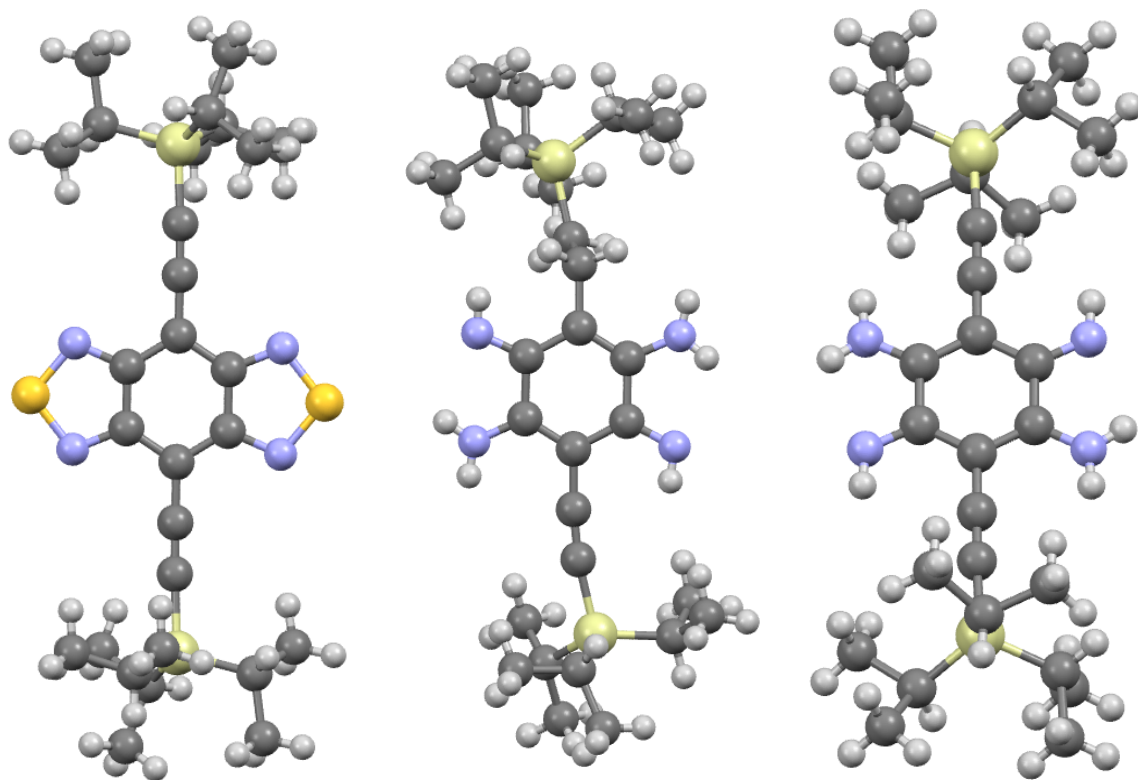


Figure 8 X-Ray crystal structures of **1**, **5** and **6**.

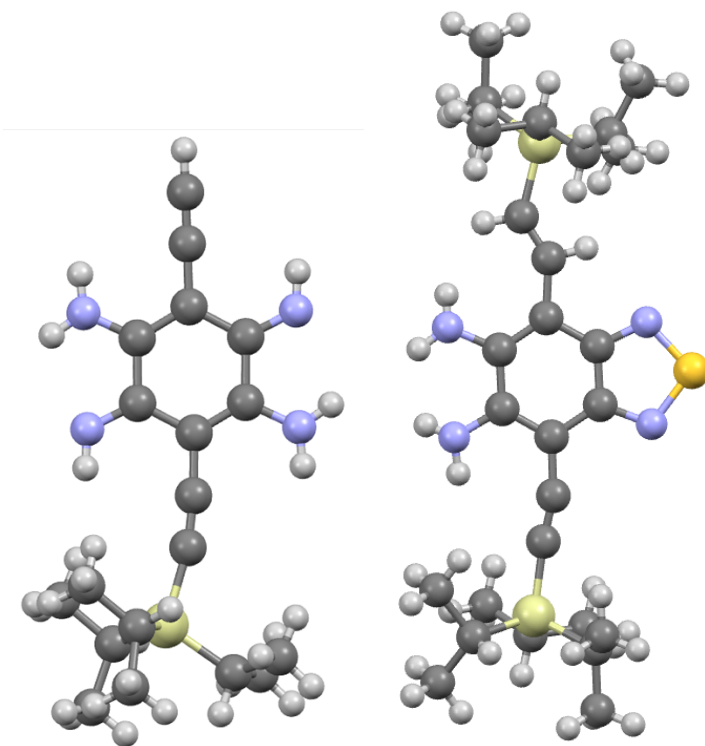


Figure 9 X-Ray crystal structure of **8** and **10**.

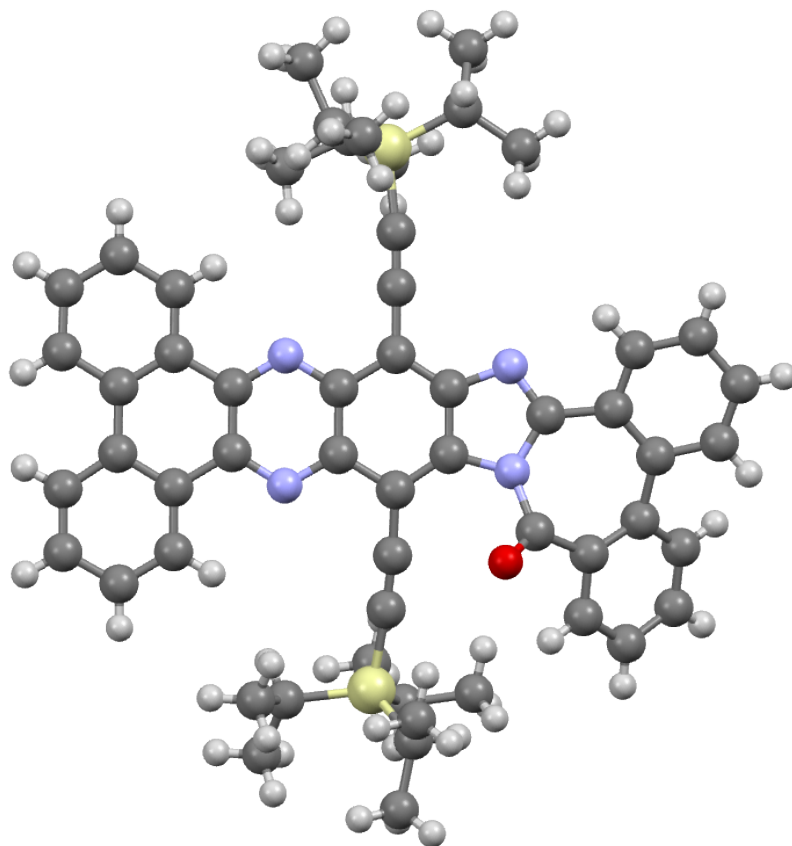


Figure 10 X-Ray crystal structure of **15**.

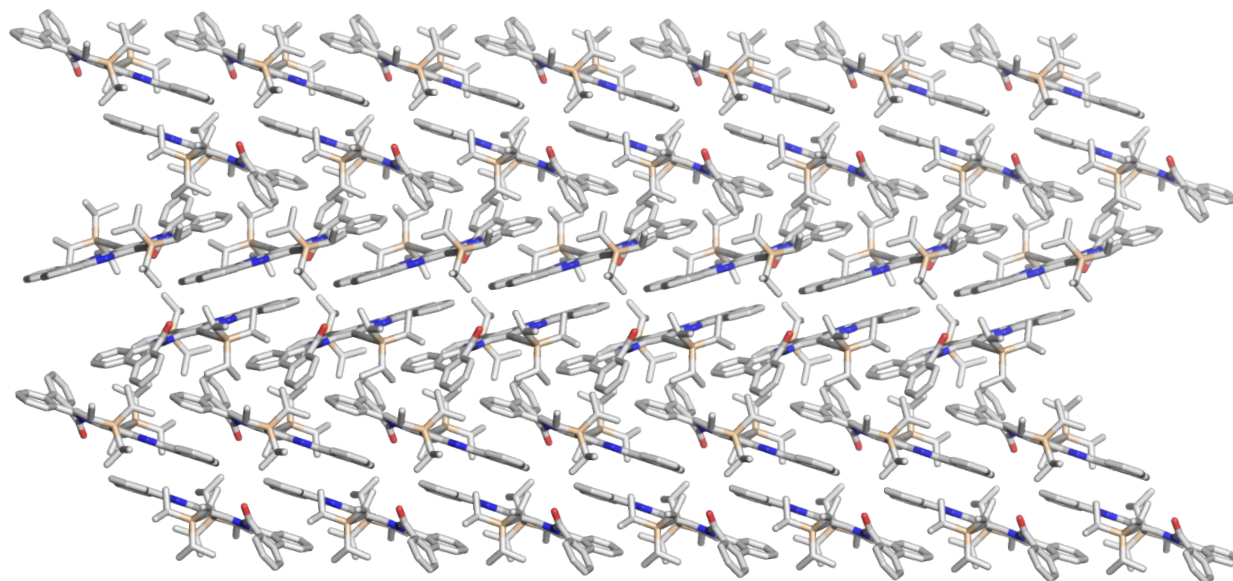


Figure 11 Herringbone arrangement of **15** from side-view (hydrogen atoms have been omitted for clarity).

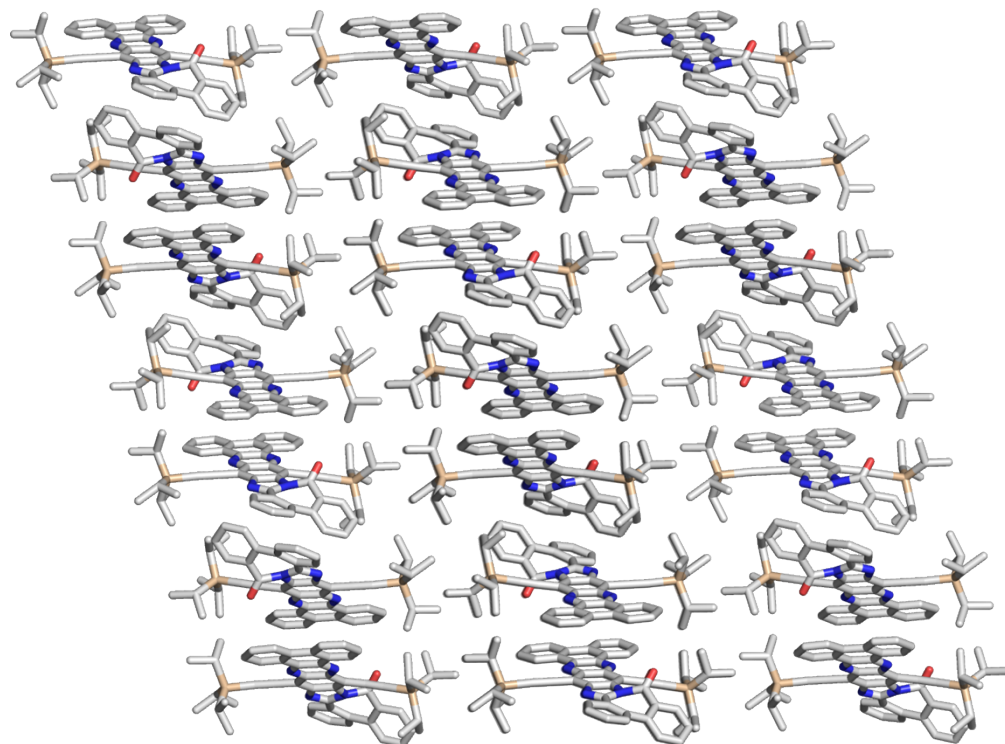


Figure 12 Arrangement of **15** from front-view (hydrogen atoms have been omitted for clarity).

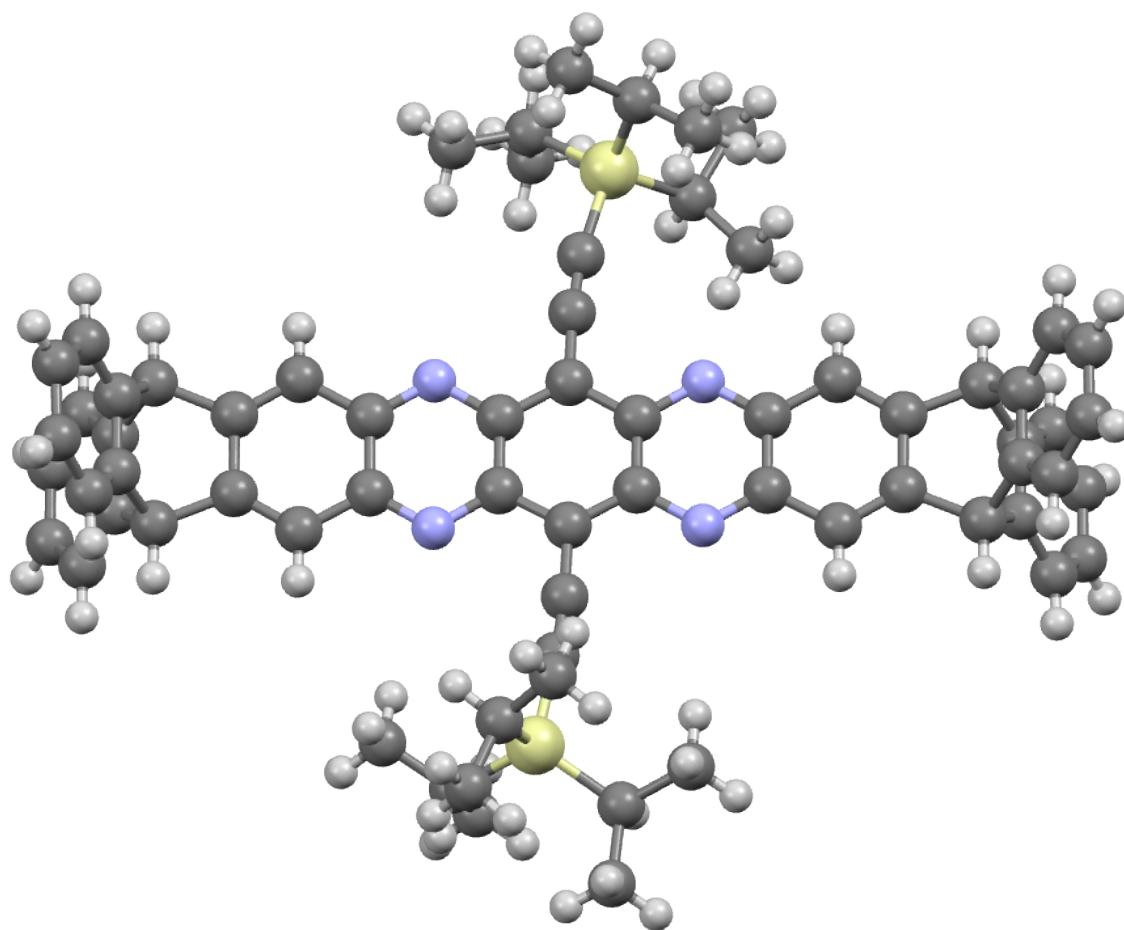


Figure 13 X-Ray crystal structure of **17**.

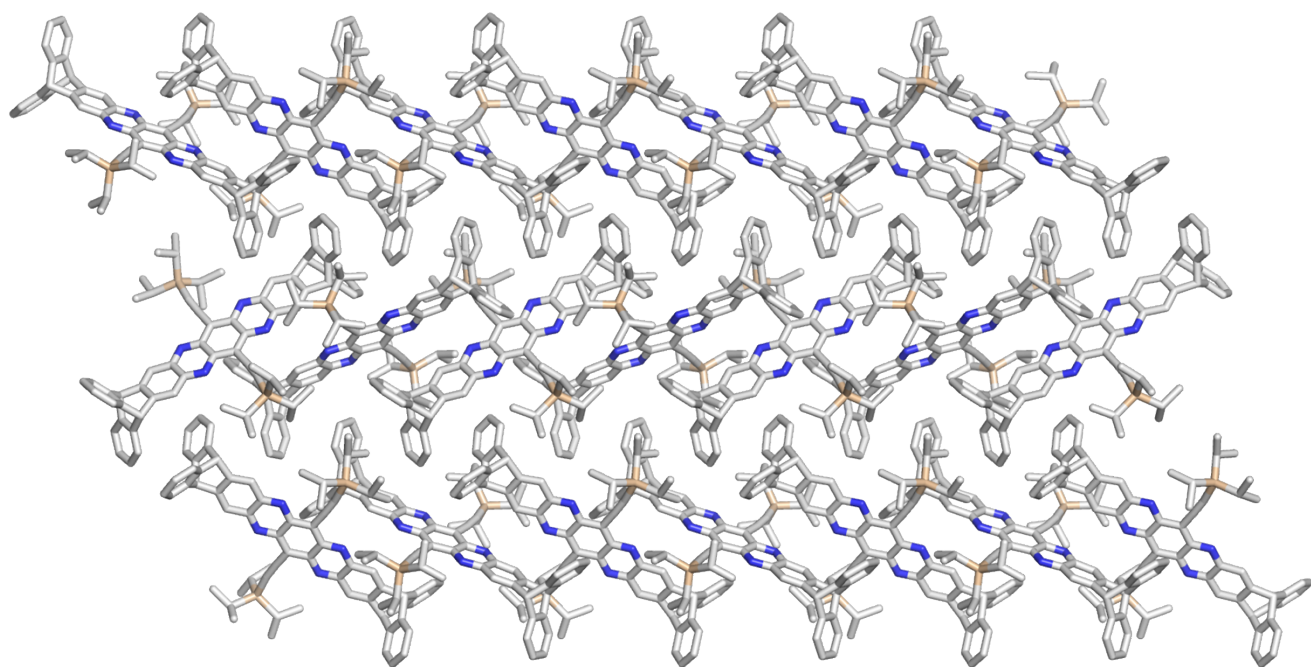


Figure 14 Arrangement of **17** from side-view (hydrogen atoms have been omitted for clarity).

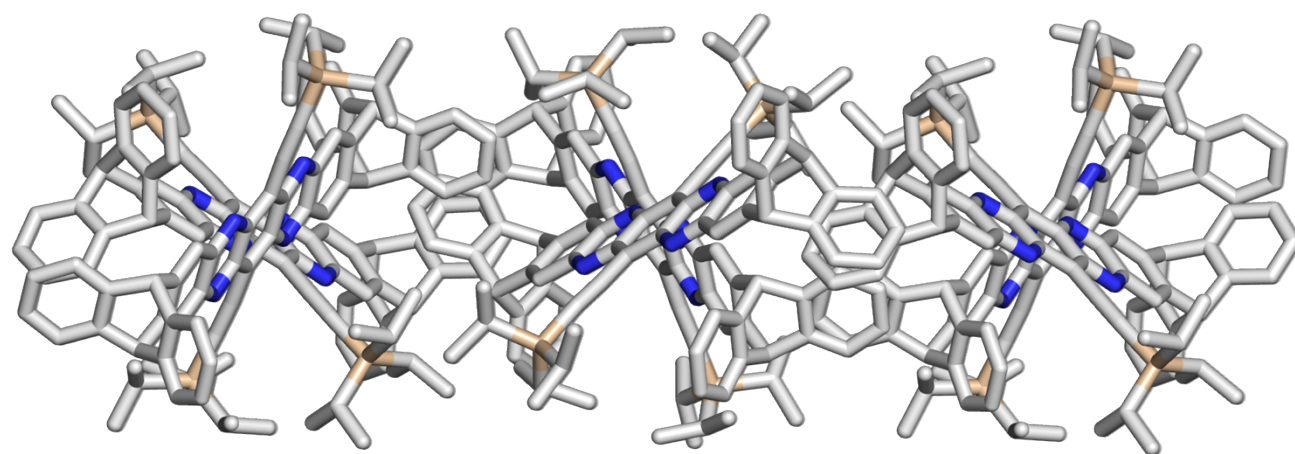


Figure 15 Twisted arrangement of **17** from front-view (hydrogen atoms have been omitted for clarity).

S8. ¹H- and ¹³C-NMR spectra

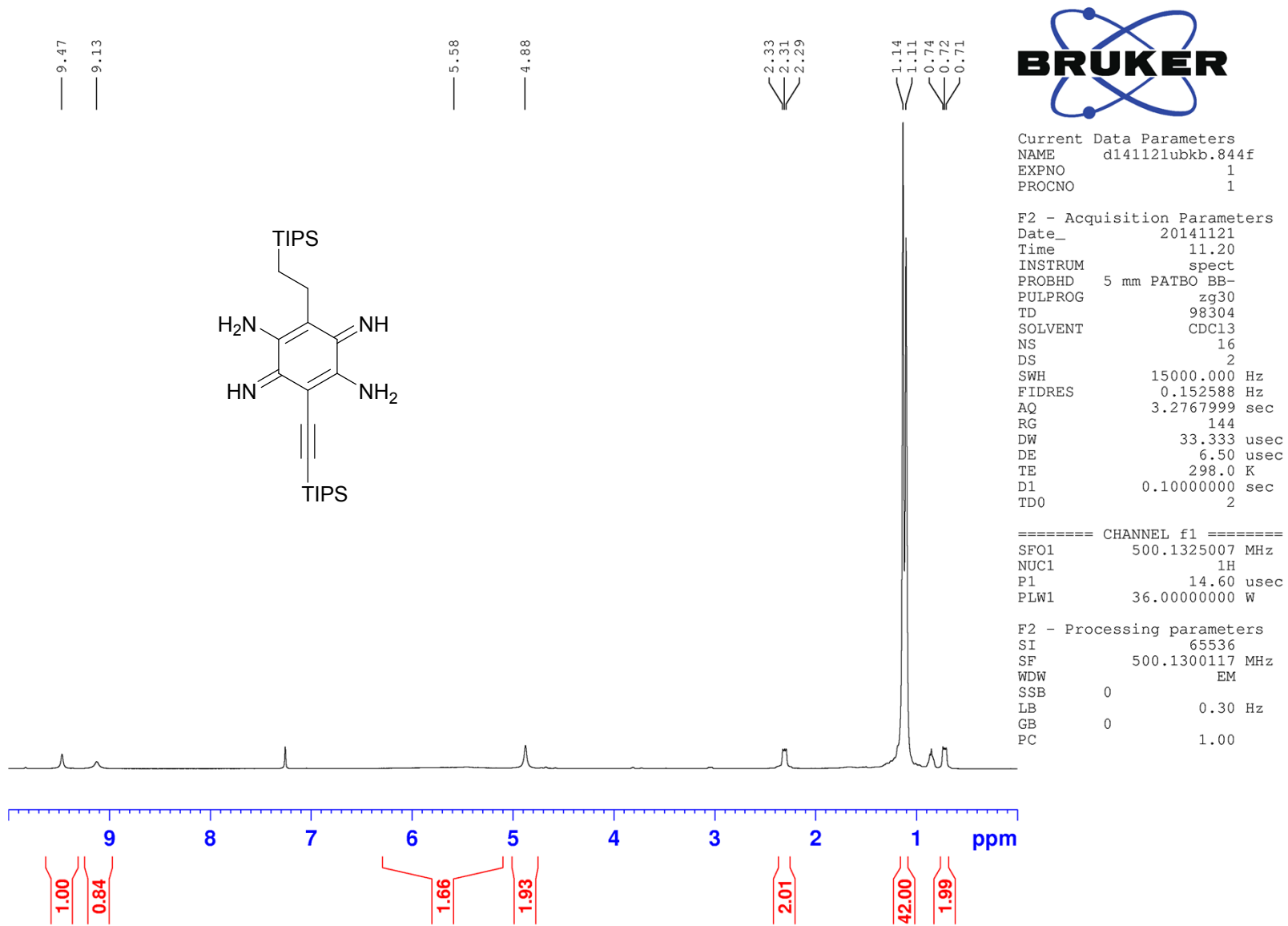
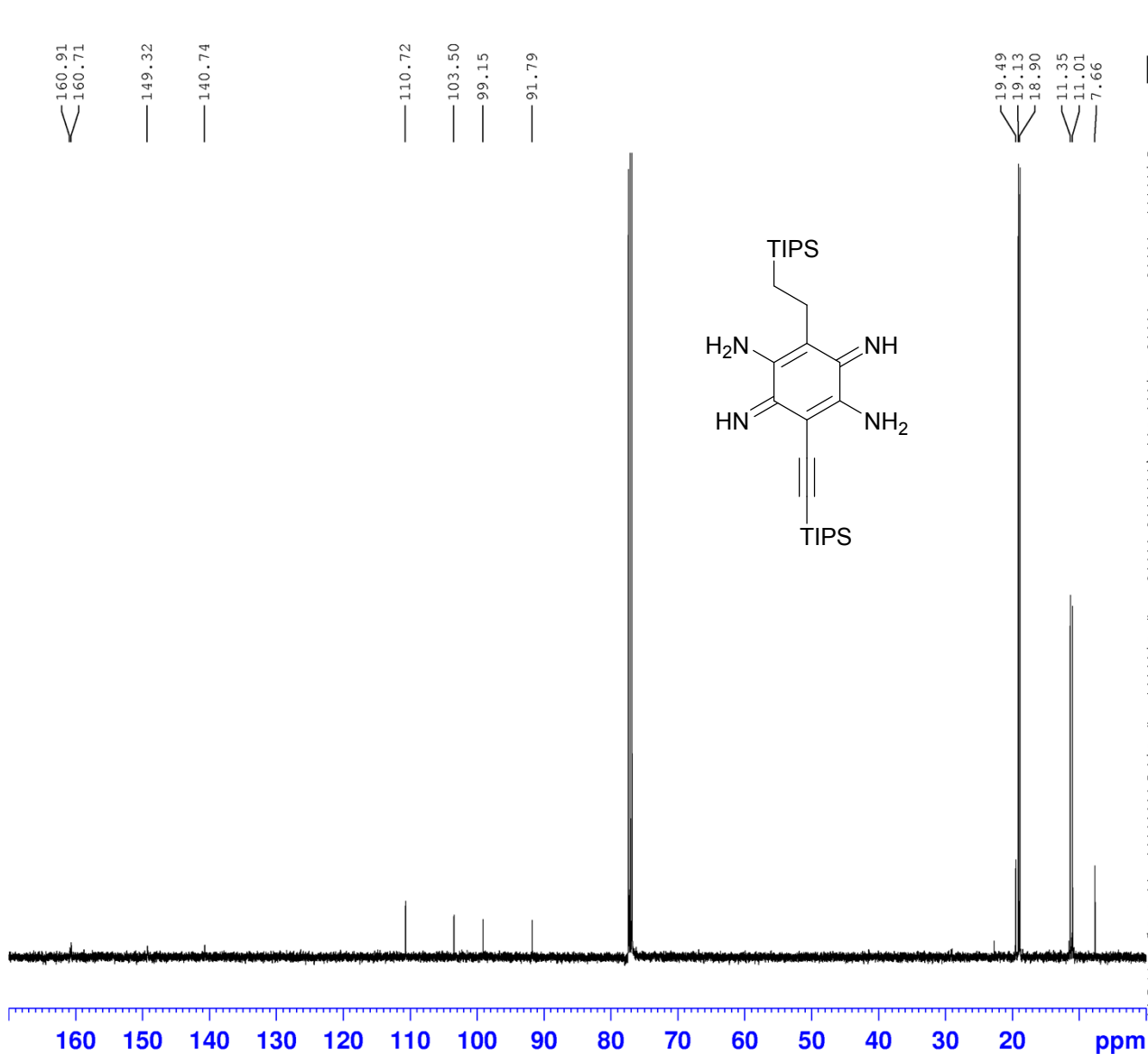


Figure 16 ¹H-NMR spectrum of 5 in CDCl₃.



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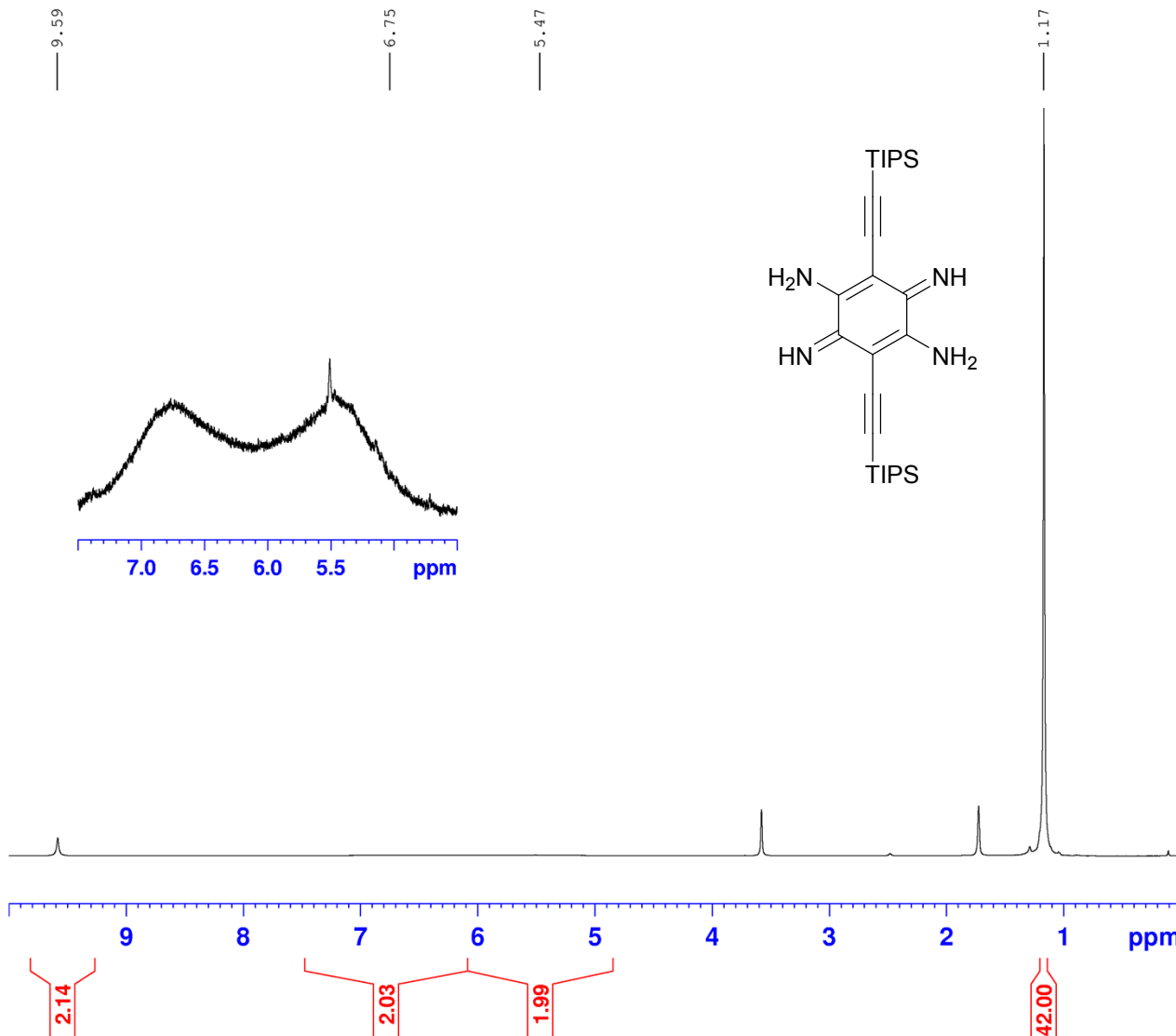
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 RG 1620
 DW 13.200 usec
 DE 6.50 usec
 TE 298.4 K
 D1 1.50000000 sec
 D11 0.03000000 sec
 TD0 64

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 P1 9.50 usec
 PLW1 63.00000000 W

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 PLW12 1.19900000 W
 PLW13 0.76738000 W

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 PC 1.40

Figure 17 ¹³C{¹H}-NMR spectrum of 5 in CDCl₃.



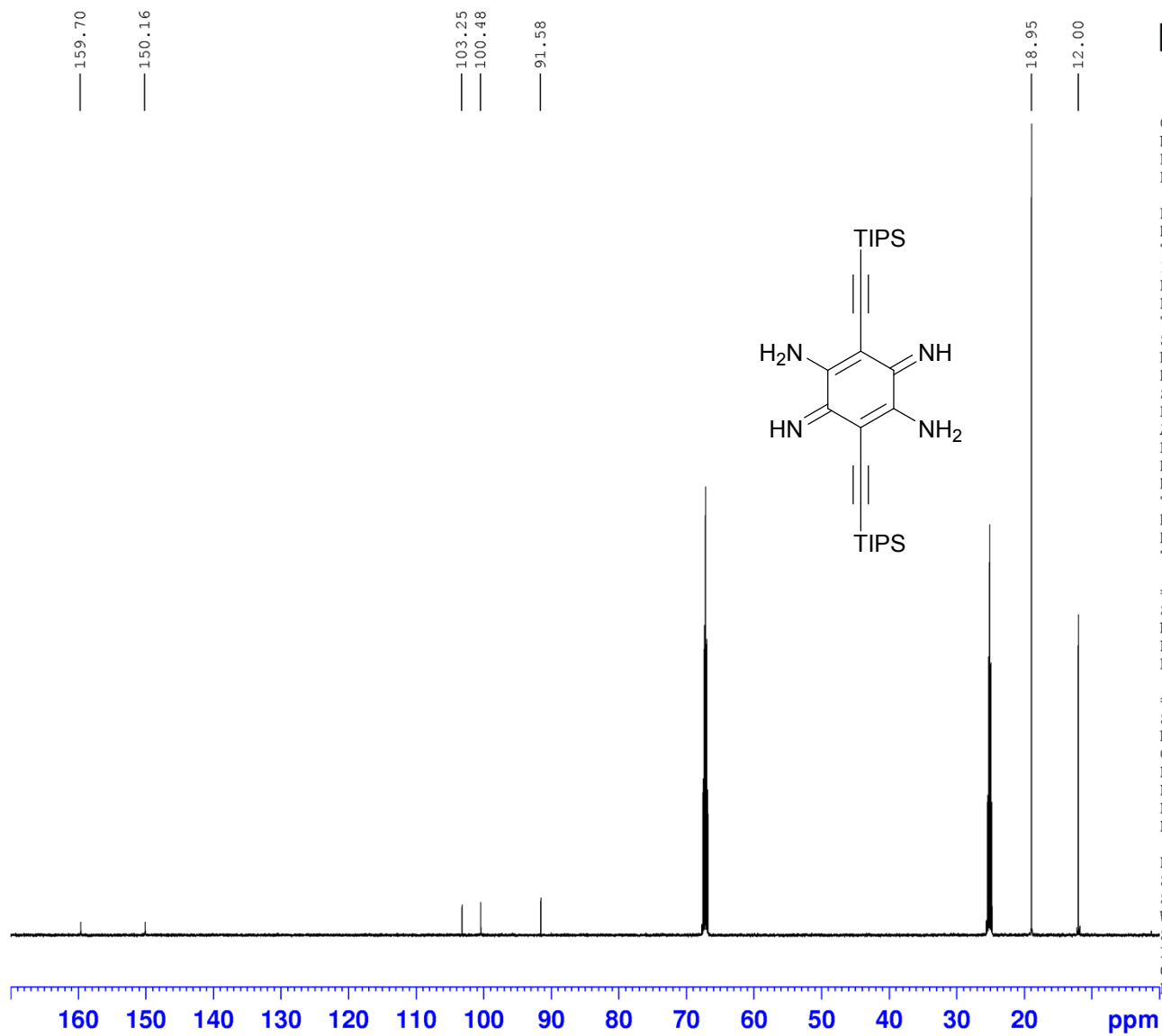
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Figure 18 ¹H-NMR spectrum of 6 in THF-d₈.



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 PROCNO 1

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 FIDRES 0.288992 Hz
 AQ 1.7301503 sec
 RG 1620
 DW 13.200 usec
 DE 6.50 usec
 TE 298.2 K
 D1 1.50000000 sec
 D11 0.03000000 sec
 TD0 64

==== CHANNEL f1 =====
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 NUC1 13C
 P1 9.50 usec
 PLW1 63.00000000 W

==== CHANNEL f2 =====
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 PLW13 0.76738000 W

F2 - Processing parameters
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 GB 0
 PC 1.40

Figure 19 ¹³C{¹H}-NMR spectrum of 6 in THF-d₈.

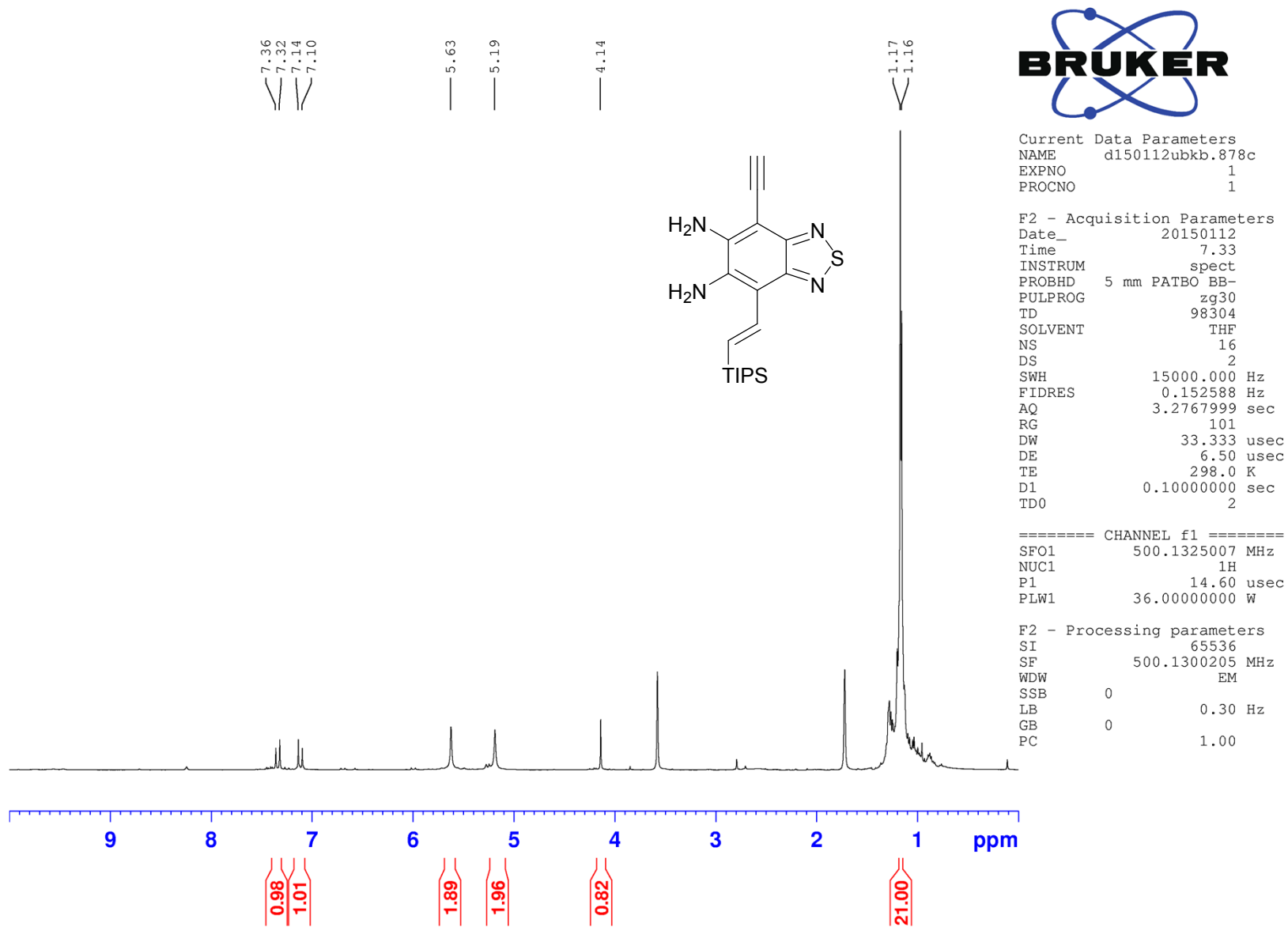
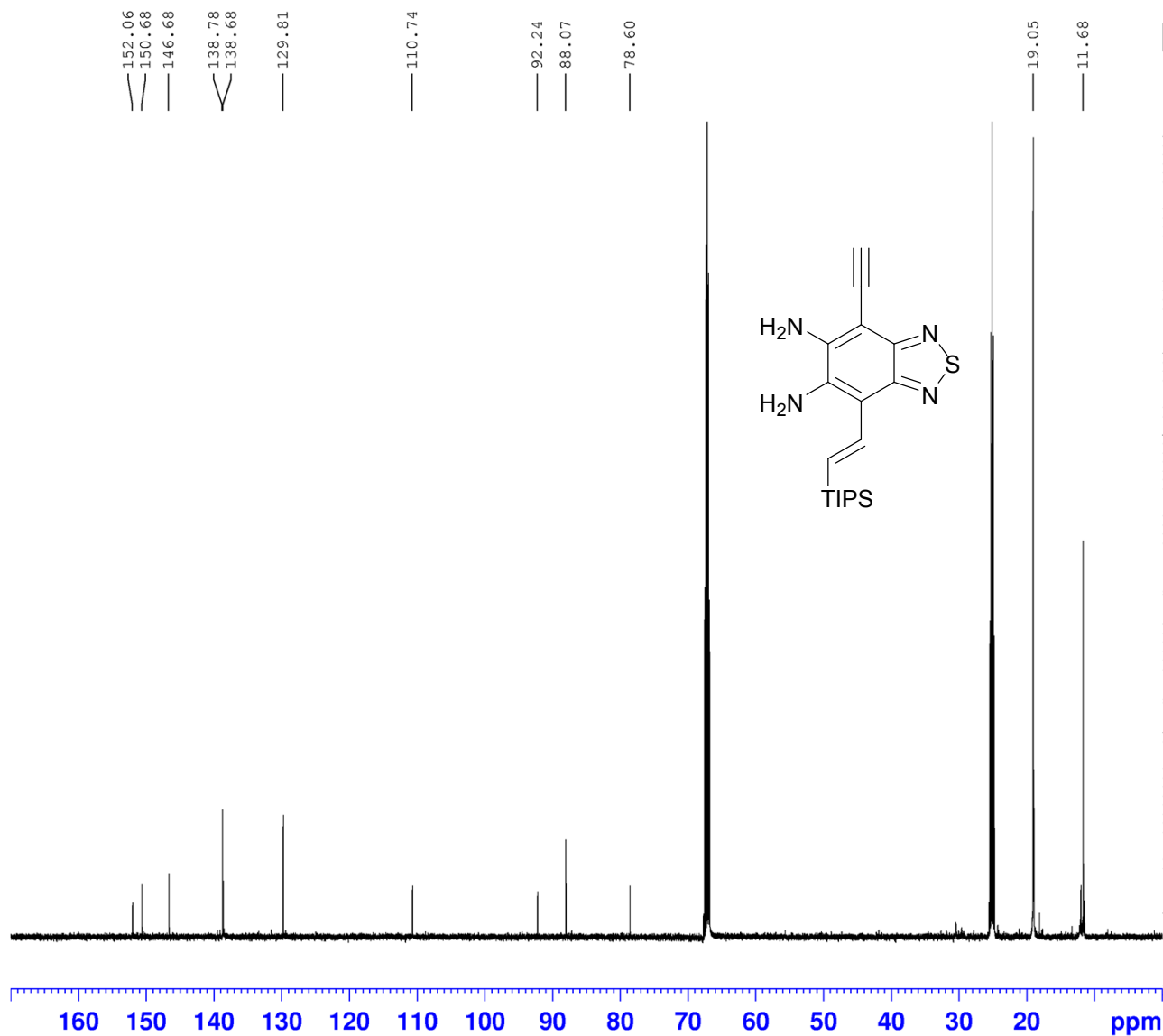


Figure 20 ¹H-NMR spectrum of 7 in THF-d₈.



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PROCNO 1

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FIDRES 0.288992 Hz
AQ 1.7301503 sec
RG 1620
DW 13.200 usec
DE 6.50 usec
TE 298.1 K
D1 1.50000000 sec
D11 0.03000000 sec
TD0 64

=====
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PLW1 63.00000000 W

=====
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PLW12 1.19900000 W
PLW13 0.76738000 W

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Figure 21 ¹³C{¹H}-NMR spectrum of 7 in THF-d₈.

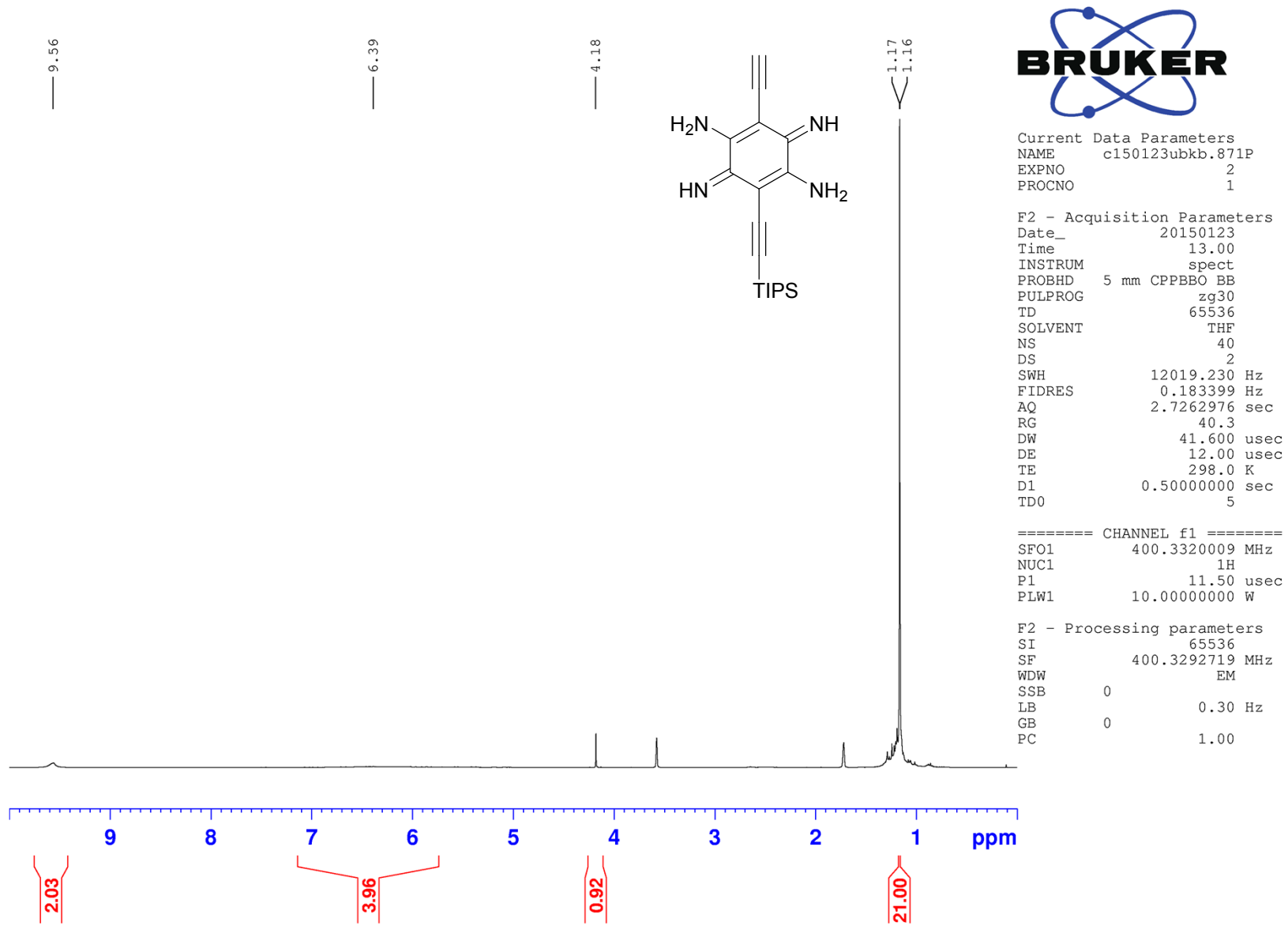


Figure 22 ¹H-NMR spectrum of 8 in THF-d₈.

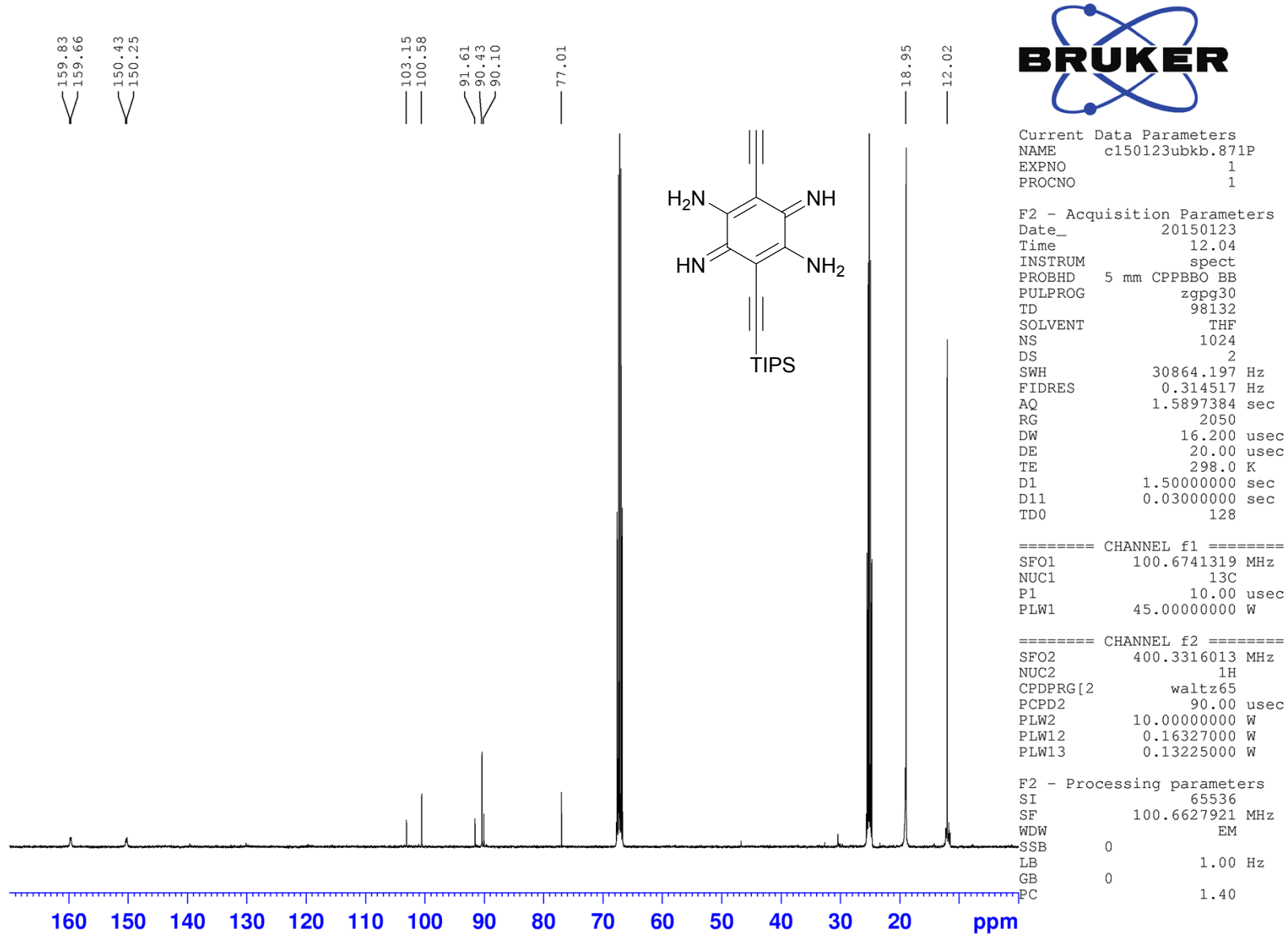
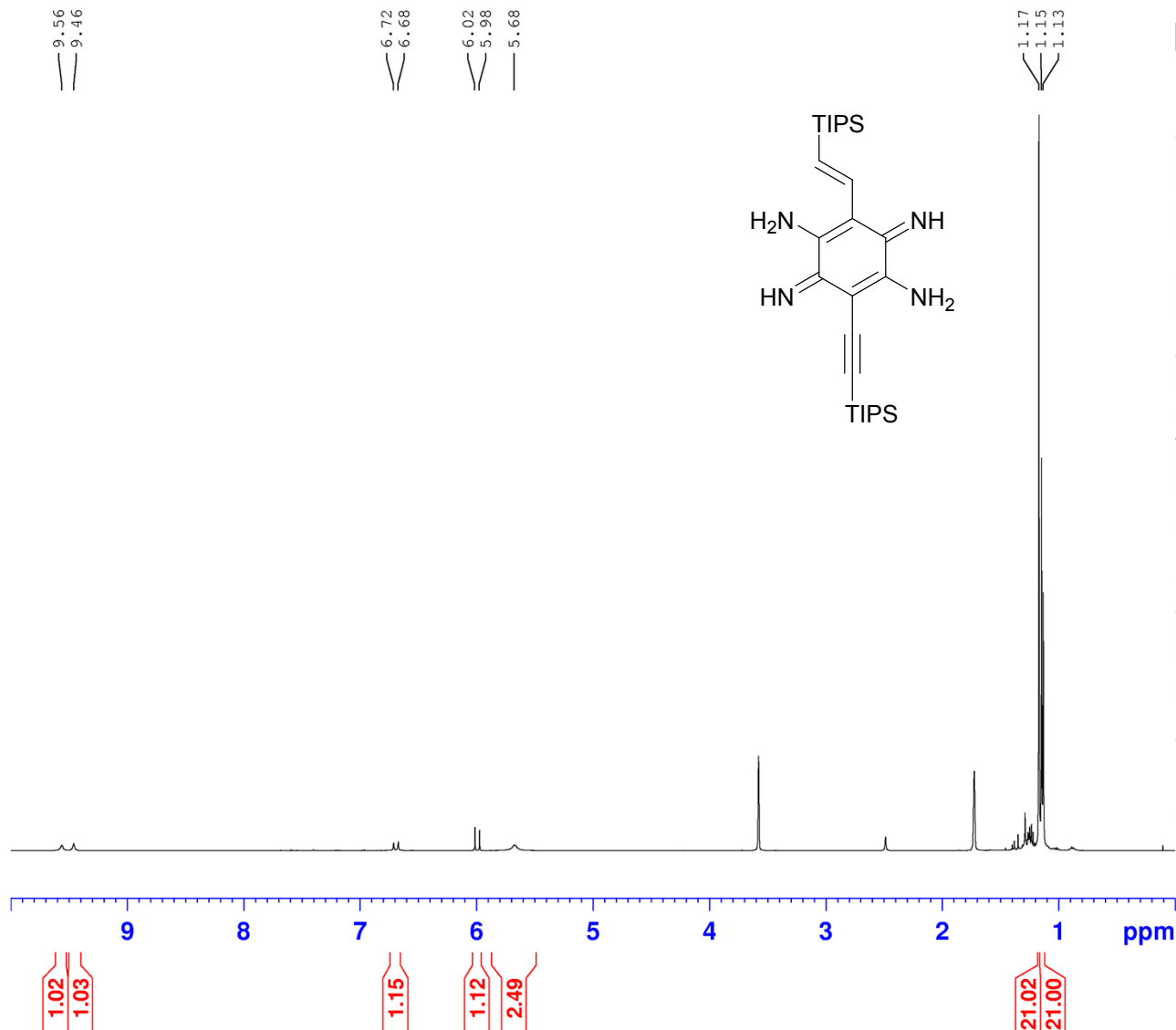


Figure 23 ¹³C{¹H}-NMR spectrum of 8 in THF-d₈.



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 PROCNO 1

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 RG 161
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 TE 295.0 K
 D1 0.10000000 sec
 TD0 4

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 SFO1 500.1325007 MHz
 NUC1 1H
 P1 14.60 usec
 PLW1 36.00000000 W

F2 - Processing parameters
 SI 65536
 SF 500.1300186 MHz
 WDW EM
 SSB 0
 LB 0.30 Hz
 GB 0
 PC 1.00

Figure 24 ¹H-NMR spectrum of 9 in THF-d₈.

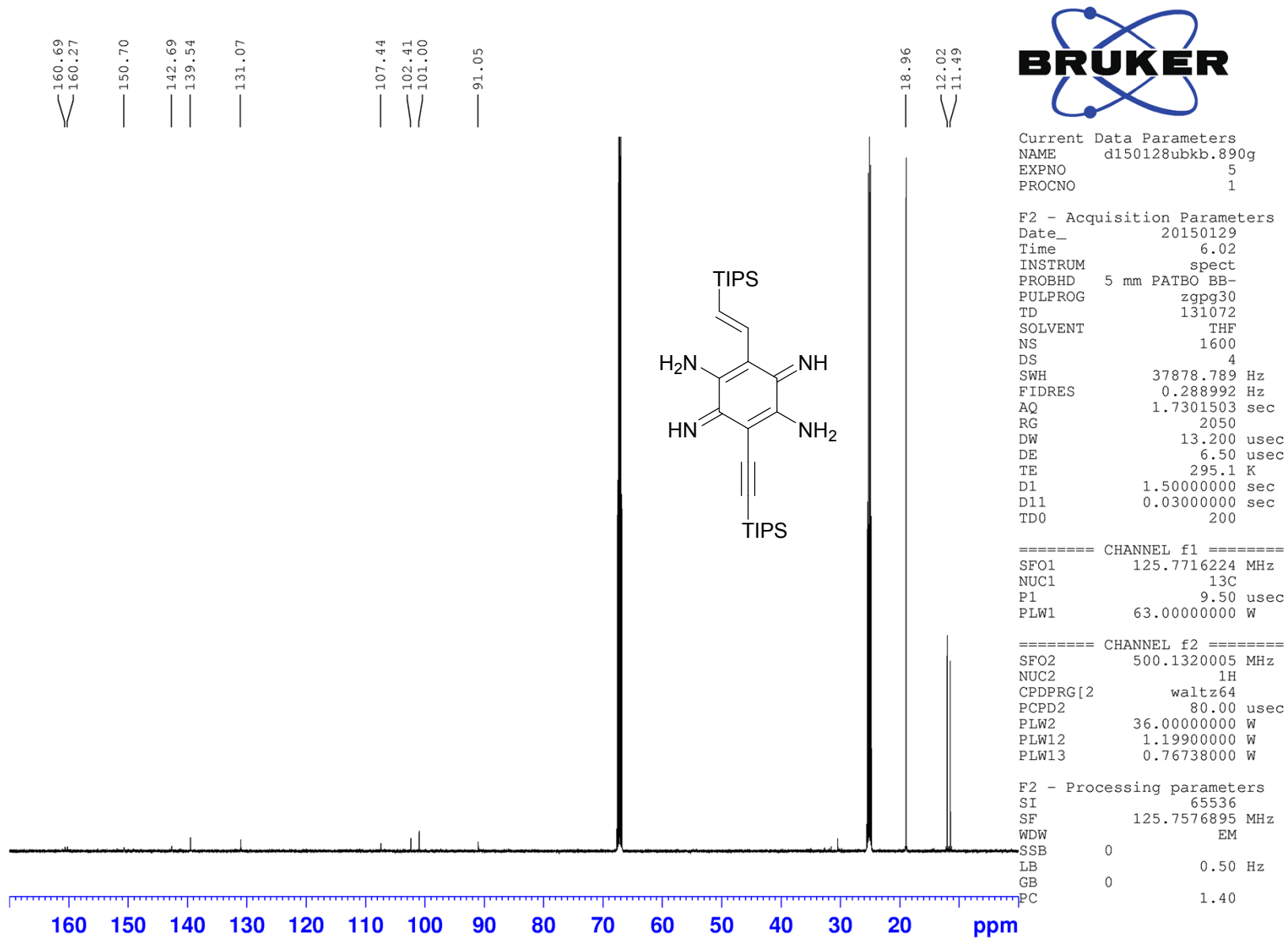


Figure 25 $^{13}\text{C}\{^1\text{H}\}$ -NMR spectrum of 9 in THF-d₈.

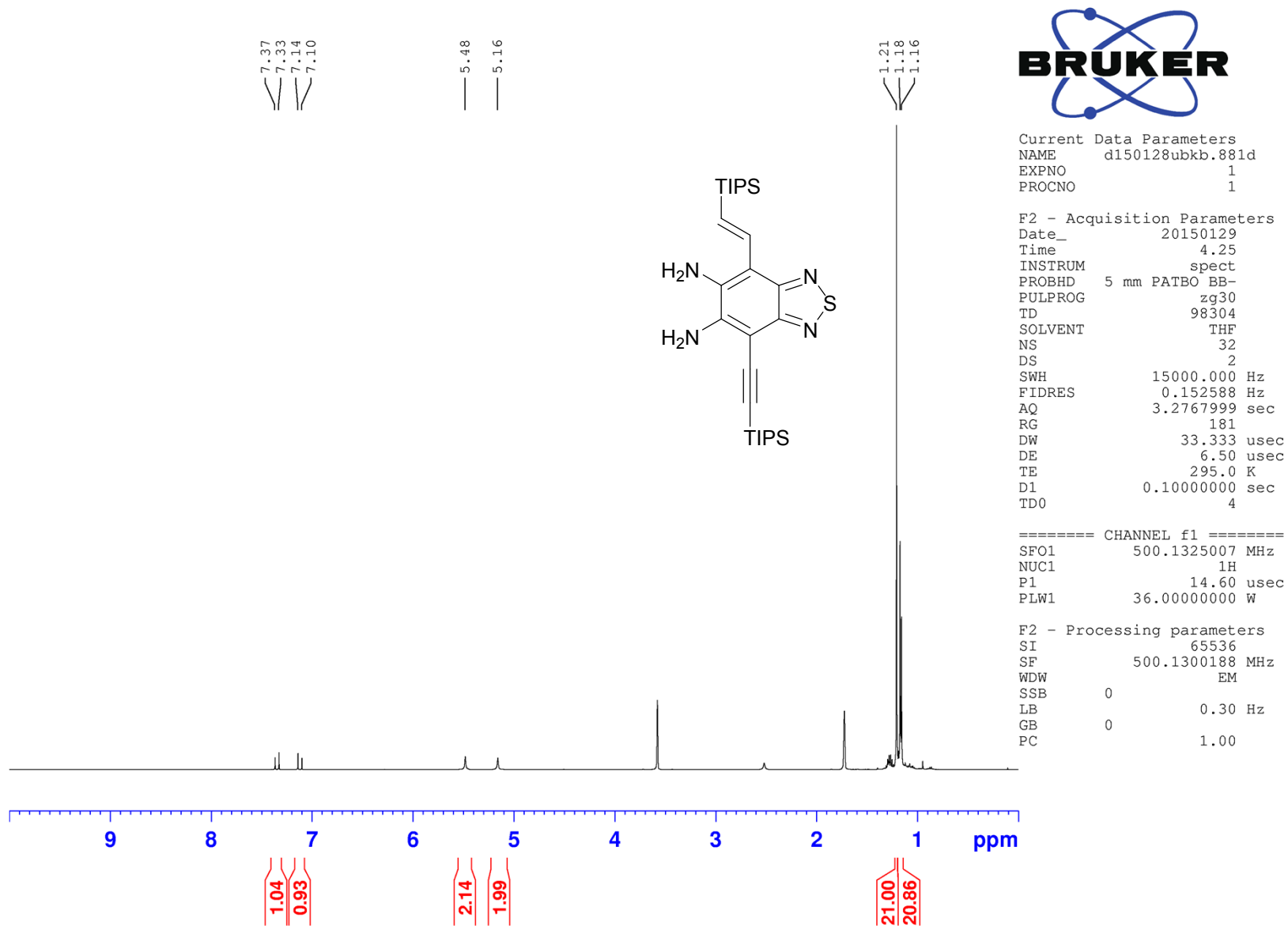
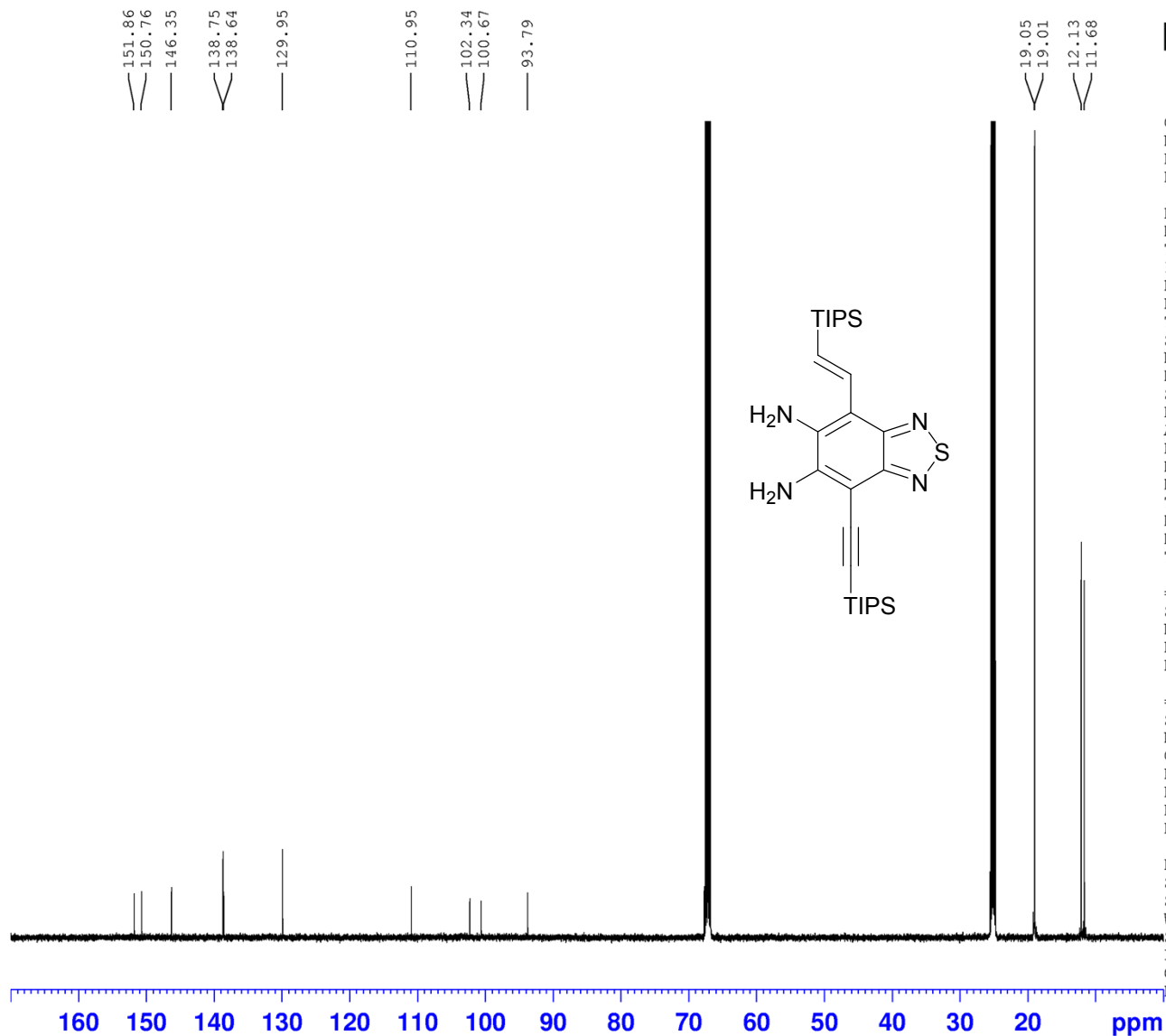


Figure 26 ¹H-NMR spectrum of 10 in THF-d₈.



Current Data Parameters
 NAME d150128ubkb.881d
 EXPNO 2
 PROCNO 1

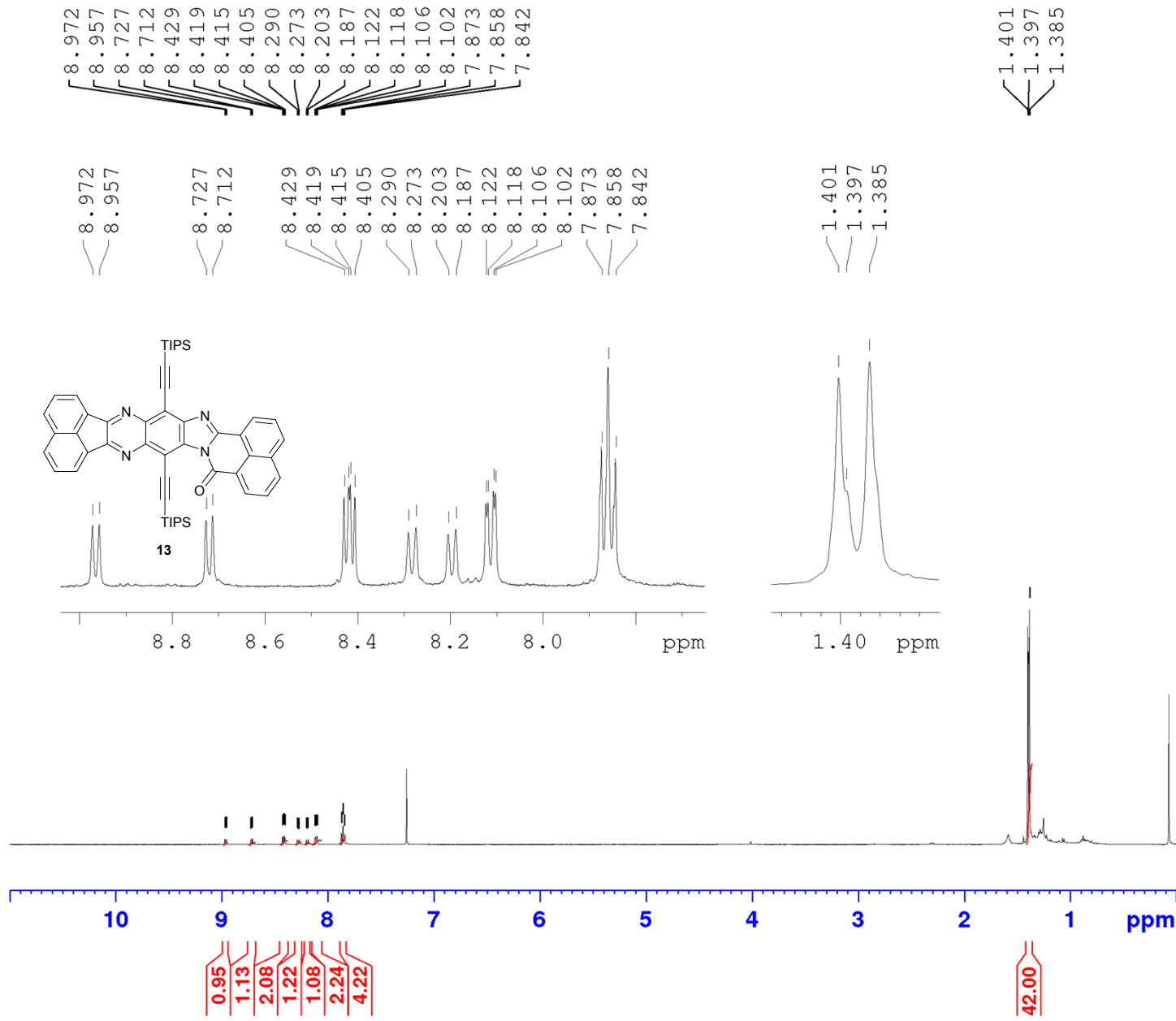
F2 - Acquisition Parameters
 Date_ 20150129
 Time 4.29
 INSTRUM spect
 PROBHD 5 mm PATBO BB-
 PULPROG zgpg30
 TD 131072
 SOLVENT THF
 NS 1600
 DS 4
 SWH 37878.789 Hz
 FIDRES 0.288992 Hz
 AQ 1.7301503 sec
 RG 1620
 DW 13.200 usec
 DE 6.50 usec
 TE 295.1 K
 D1 1.50000000 sec
 D11 0.03000000 sec
 TD0 200

==== CHANNEL f1 =====
 SFO1 125.7716224 MHz
 NUC1 13C
 P1 9.50 usec
 PLW1 63.00000000 W

==== CHANNEL f2 =====
 SFO2 500.1320005 MHz
 NUC2 1H
 CPDPRG[2] waltz64
 PCPD2 80.00 usec
 PLW2 36.00000000 W
 PLW12 1.19900000 W
 PLW13 0.76738000 W

F2 - Processing parameters
 SI 65536
 SF 125.7576896 MHz
 WDW EM
 SSB 0
 LB 0.50 Hz
 GB 0
 PC 1.40

Figure 27 $^{13}\text{C}\{^1\text{H}\}$ -NMR spectrum of 10 in THF-d8.



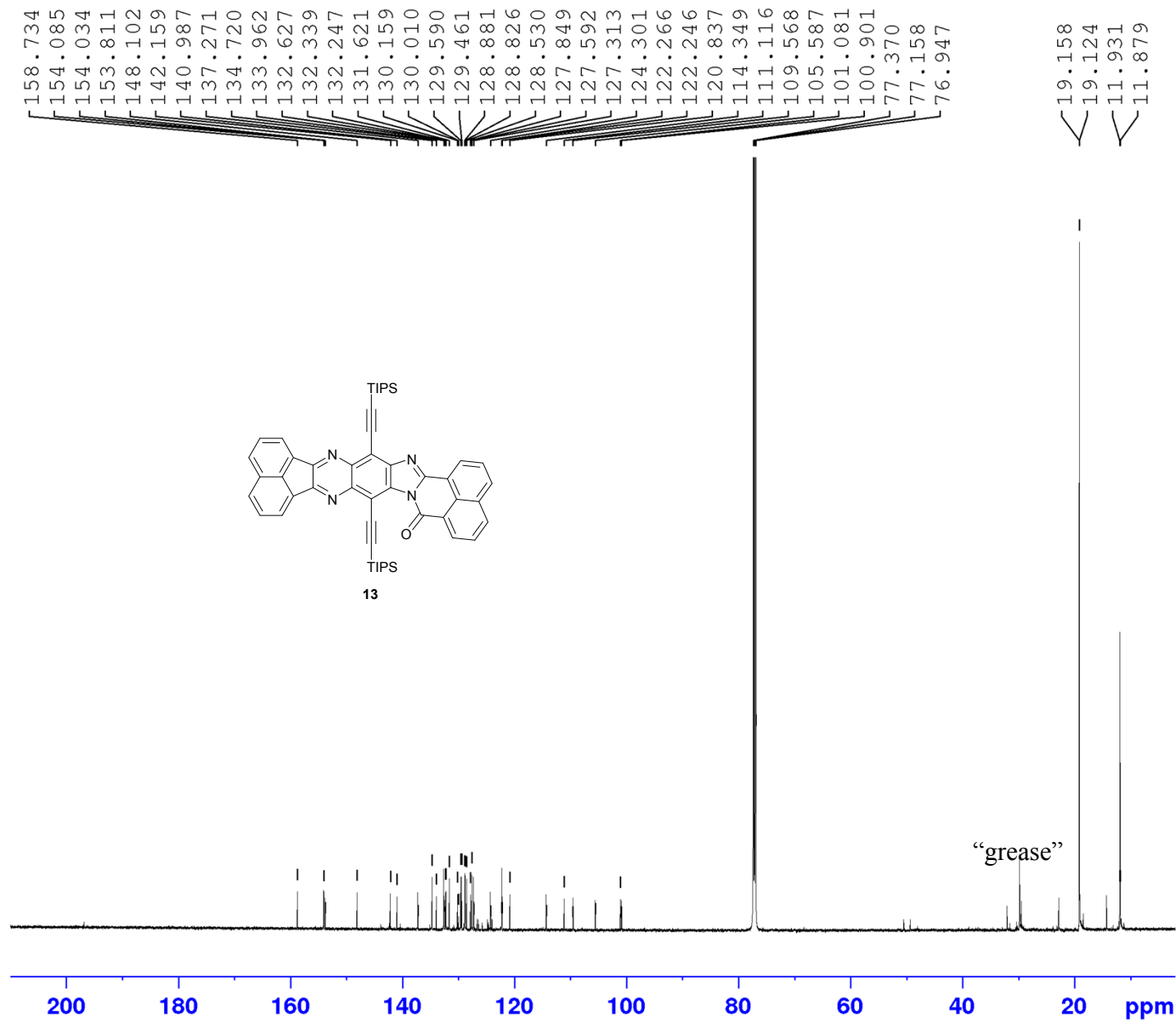
1.401
1.397
1.385

Current Data Parameters
NAME d150324ubpb.08903_24
EXPNO 1
PROCNO 1

F2 - Acquisition Parameters
Date_ 20150324
Time 9.15 h
INSTRUM spect
PROBHD Z119877_0003 (zg30)
PULPROG 98304
TD 287
SOLVENT CDC13
NS 16
DS 2
SWH 15000.000 Hz
FIDRES 0.152588 Hz
AQ 3.2767999 sec
RG 287
DW 33.333 usec
DE 6.50 usec
TE 295.0 K
D1 0.10000000 sec
TD0 2
SFO1 500.1325007 MHz
NUC1 1H
P1 14.60 usec
PLW1 36.00000000 W

F2 - Processing parameters
SI 65536
SF 500.1300127 MHz
WDW EM
SSB 0
LB 0.30 Hz
GB 0
PC 1.00

Figure 28 ¹H-NMR spectrum of 13 in CDCl₃.

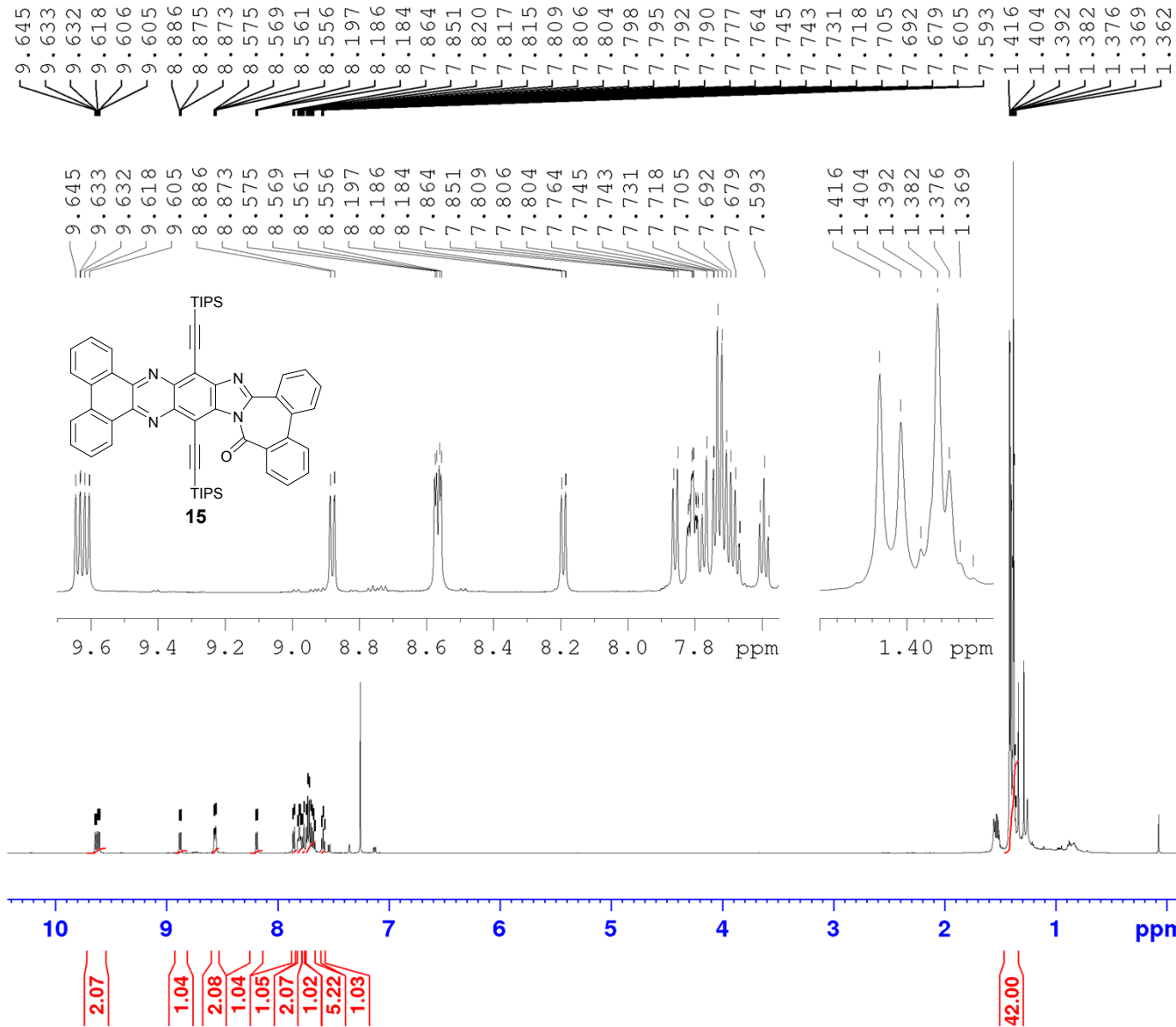


Current Data Parameters
 NAME e150413ubpb.08903
 EXPNO 1
 PROCNO 1

F2 - Acquisition Parameters
 Date_ 20150413
 Time 12.22 h
 INSTRUM spect
 PROBHD z132808_0001 (
 PULPROG zgpg30
 TD 98132
 SOLVENT CDCl3
 NS 4096
 DS 4
 SWH 45454.547 Hz
 FIDRES 0.463198 Hz
 AQ 1.0794520 sec
 RG 2050
 DW 11.000 usec
 DE 18.00 usec
 TE 295.0 K
 D1 2.00000000 sec
 D11 0.03000000 sec
 TD0 512
 SFO1 150.9480335 MHz
 NUC1 13C
 P1 12.50 usec
 PLW1 87.00000000 W
 SFO2 600.2462301 MHz
 NUC2 1H
 CPDPRG[2] waltz64
 PCPD2 70.00 usec
 PLW2 7.50000000 W
 PLW12 0.08910700 W
 PLW13 0.04482000 W

F2 - Processing parameters
 SI 65536
 SF 150.9314547 MHz
 WDW EM
 SSB 0
 LB 1.00 Hz
 GB 0
 PC 1.40

Figure 29 $^{13}\text{C}\{^1\text{H}\}$ -NMR spectrum of 13 in CDCl_3 .

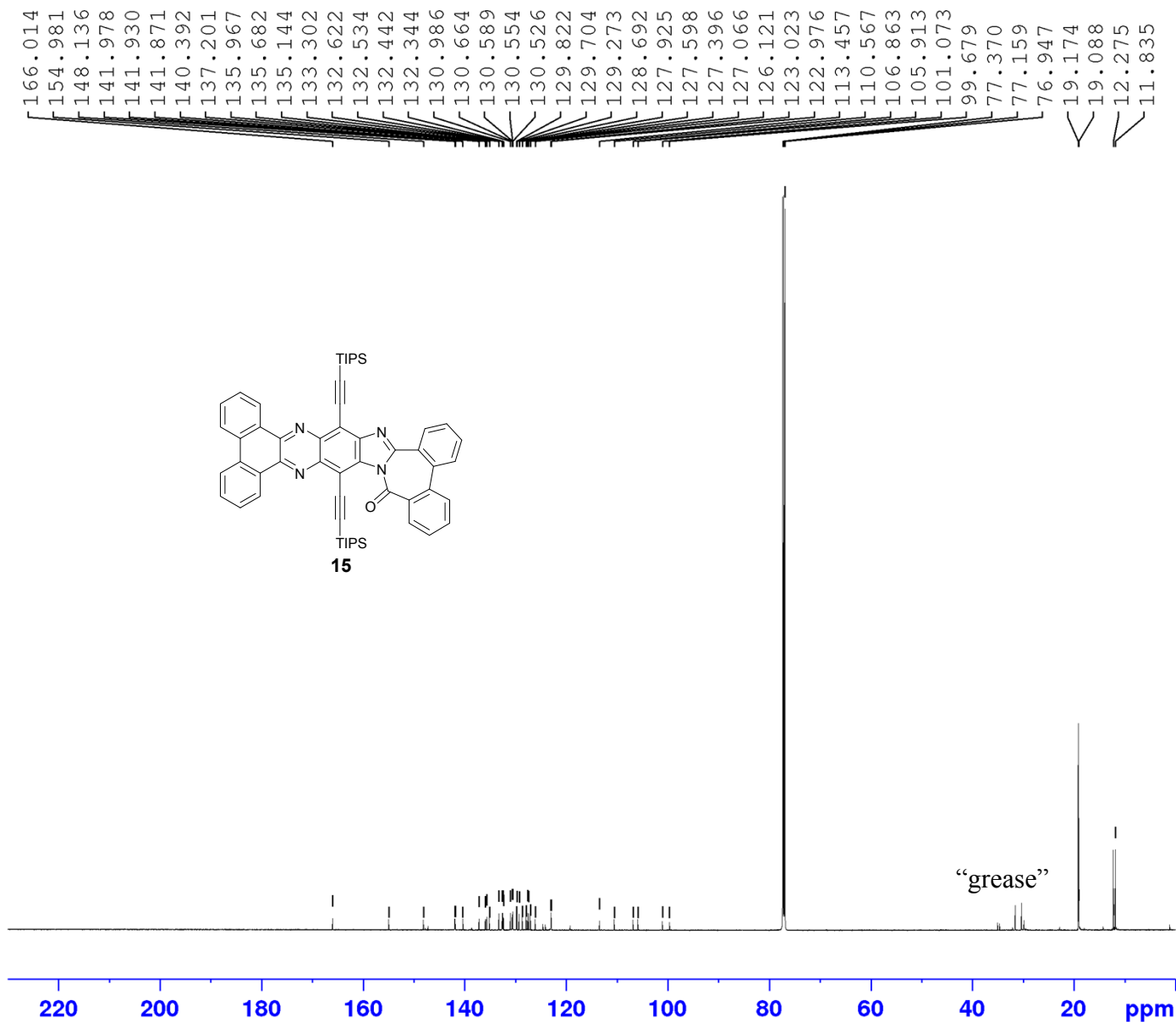


Current Data Parameters
 NAME e150424ubpb.08804
 EXPNO 2
 PROCNO 1

F2 - Acquisition Parameters
 Date_ 20150424
 Time 17.31 h
 INSTRUM spect
 PROBHD z132808_0001 (
 PULPROG zg30
 TD 131072
 SOLVENT CDCl3
 NS 128
 DS 2
 SWH 18028.846 Hz
 FIDRES 0.137549 Hz
 AQ 3.6350634 sec
 RG 13.85
 DW 27.733 usec
 DE 12.00 usec
 TE 295.0 K
 D1 0.10000000 sec
 TD0 16
 SFO1 600.2468302 MHz
 NUC1 1H
 P1 7.63 usec
 PLW1 7.50000000 W

F2 - Processing parameters
 SI 65536
 SF 600.2440159 MHz
 WDW EM
 SSB 0
 LB 0.30 Hz
 GB 0
 PC 1.00

Figure 30 ¹H-NMR spectrum of **15** in CDCl₃.



Current Data Parameters
 NAME e150424ubpb.08804
 EXPNO 1
 PROCNO 1

F2 - Acquisition Parameters
 Date_ 20150424
 Time 13.55 h
 INSTRUM spect
 PROBHD z132808_0001 (
 PULPROG zgpg30
 TD 98132
 SOLVENT CDCl3
 NS 4096
 DS 4
 SWH 45454.547 Hz
 FIDRES 0.463198 Hz
 AQ 1.0794520 sec
 RG 2050
 DW 11.000 usec
 DE 18.00 usec
 TE 295.0 K
 D1 2.00000000 sec
 D11 0.03000000 sec
 TD0 512
 SFO1 150.9480335 MHz
 NUC1 13C
 P1 12.50 usec
 PLW1 87.00000000 W
 SFO2 600.2462301 MHz
 NUC2 1H
 CPDPRG[2] waltz64
 PCPD2 70.00 usec
 PLW2 7.50000000 W
 PLW12 0.08910700 W
 PLW13 0.04482000 W

F2 - Processing parameters
 SI 65536
 SF 150.9314545 MHz
 WDW EM
 SSB 0
 LB 1.00 Hz
 GB 0
 PC 1.40

Figure 31 $^{13}\text{C}\{^1\text{H}\}$ -NMR spectrum of 15 in CDCl_3 .

S9. Cartesian coordinates of computational studied molecules

Cartesian coordinates of compound 13							
				C	-4.77120	1.39055	-0.24174
				C	-5.39940	2.61514	-0.19096
C	-1.10697	0.92078	-0.31171	C	-6.81900	2.64540	-0.17152
C	-1.10925	-0.51746	-0.38062	C	-7.58643	1.49401	-0.20228
C	0.11424	-1.27375	-0.43969	C	-6.96444	0.21598	-0.25536
C	1.28615	-0.52103	-0.51183	C	-7.59207	-1.05985	-0.29342
C	1.29564	0.89563	-0.36050	C	-6.82961	-2.21382	-0.34543
C	0.12130	1.64367	-0.26349	C	-5.40999	-2.18868	-0.36303
C	0.15779	3.05114	-0.12167	C	-4.77597	-0.96641	-0.32578
C	0.05494	-2.68264	-0.31432	Si	0.29972	6.10138	0.11580
C	-0.03976	-3.87977	-0.12871	C	-0.01870	6.58810	1.94720
C	0.22144	4.25934	-0.00086	C	-1.00745	6.70448	-1.15361
N	-2.27380	1.63381	-0.26089	C	2.07155	6.60407	-0.42915
C	-3.37378	0.93523	-0.27684	C	-0.42795	8.06081	2.14317
C	-3.37712	-0.51615	-0.32989	C	-1.00747	5.66040	2.67732
N	-2.28043	-1.22046	-0.37405	C	-2.43869	6.27744	-0.77878
N	2.64889	-0.86953	-0.59000	C	-0.95717	8.19818	-1.51939
C	3.36240	0.33693	-0.43153	C	2.45618	8.05856	-0.09726
N	2.58935	1.37614	-0.30389	C	3.14615	5.64502	0.11753
C	3.24519	-2.05036	-1.10297	Si	-0.12886	-5.70847	0.10923
C	4.72907	-2.07696	-1.00802	C	0.08276	-6.51277	-1.62359
C	5.46858	-0.90865	-0.68120	C	1.37152	-6.12393	1.23190
C	4.81154	0.32285	-0.41453	C	-1.85944	-6.10570	0.84214
C	6.89366	-0.96354	-0.64931	C	-0.60788	-5.70221	-2.73602
C	7.61489	0.22100	-0.35212	C	-0.36305	-7.98558	-1.69433
C	6.95445	1.40507	-0.10498	C	1.75922	-7.61144	1.30521
C	5.54878	1.46165	-0.13631	C	1.25798	-5.51670	2.64186
C	5.39040	-3.25821	-1.28986	C	-2.40694	-5.00634	1.77194
C	6.79607	-3.32048	-1.23949	C	-1.95233	-7.48113	1.53120
C	7.53206	-2.19849	-0.92876	H	8.69962	0.18317	-0.32615
O	2.58575	-2.93807	-1.59041	H	7.51771	2.30473	0.11818
C	-5.56072	0.21349	-0.27310	H	5.02781	2.39075	0.06273

H	4.80716	-4.13317	-1.55192	H	-0.43162	-6.16338	-3.71517
H	7.29717	-4.25824	-1.45357	H	-1.69349	-5.66120	-2.59280
H	8.61649	-2.24646	-0.90051	H	0.12651	-8.62159	-0.95249
H	-4.83281	3.53963	-0.16668	H	-0.13662	-8.40714	-2.68073
H	-7.31696	3.60884	-0.13147	H	-1.44353	-8.08395	-1.55011
H	-8.66984	1.56687	-0.18604	H	1.95368	-8.03748	0.31773
H	-8.67577	-1.12917	-0.28191	H	2.67273	-7.74129	1.89782
H	-7.33212	-3.17533	-0.37425	H	0.98349	-8.21927	1.77997
H	-4.84718	-3.11477	-0.40467	H	1.02777	-4.44847	2.61226
H	0.96164	6.45372	2.42857	H	2.19854	-5.63509	3.19315
H	-0.73378	6.14234	-2.05872	H	0.47883	-6.00694	3.23381
H	2.06620	6.50158	-1.52428	H	-2.41173	-4.02598	1.29137
H	0.26570	8.76434	1.67646	H	-3.43450	-5.23903	2.07752
H	-0.47128	8.30845	3.21031	H	-1.81472	-4.91819	2.68804
H	-1.42230	8.25796	1.73183	H	-1.61972	-8.30338	0.89311
H	-0.70958	4.61196	2.61619	H	-2.98703	-7.69593	1.82339
H	-1.07392	5.92950	3.73831	H	-1.35284	-7.51149	2.44564
H	-2.01702	5.73574	2.26230				
H	-2.50152	5.20814	-0.56017				
H	-3.13226	6.48843	-1.60159				
H	-2.80770	6.82184	0.09643				
H	0.02132	8.49803	-1.90263				
H	-1.69118	8.42517	-2.30172				
H	-1.19092	8.84283	-0.66710				
H	1.74743	8.79258	-0.48796				
H	3.43840	8.30178	-0.51933				
H	2.53016	8.21578	0.98341				
H	2.95586	4.60680	-0.16065				
H	4.13686	5.92346	-0.26254				
H	3.20200	5.68665	1.21127				
H	1.16422	-6.47926	-1.81936				
H	2.19402	-5.60129	0.72161				
H	-2.51729	-6.13677	-0.03957				
H	-0.24393	-4.67449	-2.77863				
				Cartesian coordinates of compound 15			
				C	-0.98963	0.93500	0.53420
				C	-0.90928	-0.50217	0.52305
				C	0.33284	-1.19289	0.26434
				C	1.44453	-0.37274	0.09272
				C	1.36129	1.05284	0.06539
				C	0.16681	1.73732	0.27639
				C	0.10390	3.14927	0.22120
				C	0.31731	-2.60448	0.15053
				C	0.21661	-3.81114	0.02913
				C	0.07729	4.36363	0.15763
				N	-2.16510	1.54785	0.77111
				C	-3.23993	0.81103	0.99285
				C	-3.16325	-0.62096	0.97627
				N	-2.01489	-1.23646	0.74307

C	-4.51840	1.47237	1.25458	C	-0.91466	8.40260	1.63762
C	-5.67987	0.69773	1.48937	C	2.33257	8.11177	-0.09636
C	-5.60268	-0.77251	1.46952	C	2.49267	6.08022	-1.60558
C	-4.36668	-1.41669	1.21830	C	-1.13989	5.63969	-2.56609
C	-4.58928	2.87659	1.27350	C	-0.88590	8.08631	-1.99315
C	-5.78711	3.52182	1.52003	Si	-0.10445	-5.63201	-0.07425
C	-6.94239	2.76640	1.75292	C	-0.92506	-5.94575	-1.78512
C	-6.88587	1.38284	1.73730	C	1.61171	-6.48305	0.00960
C	-6.73145	-1.58498	1.69533	C	-1.30669	-6.05967	1.36201
C	-6.64552	-2.96689	1.67554	C	-1.87667	-4.81170	-2.20983
C	-5.41833	-3.59275	1.42726	C	-1.65152	-7.29966	-1.90177
C	-4.29391	-2.82074	1.20122	C	1.63929	-7.97115	-0.38506
N	2.78963	-0.64440	-0.20165	C	2.33225	-6.25619	1.35141
C	3.42057	0.60183	-0.36380	C	-1.07640	-5.23372	2.64188
N	2.59310	1.60006	-0.20953	C	-1.36356	-7.56106	1.70775
C	3.44885	-1.90995	-0.05284	H	-3.68037	3.43635	1.09179
C	4.37357	-2.27399	-1.16409	H	-5.83009	4.60596	1.53357
C	5.35437	-1.42375	-1.71065	H	-7.88815	3.26211	1.94750
C	5.75871	-0.13389	-1.09750	H	-7.79830	0.83032	1.92217
C	4.86702	0.79438	-0.51416	H	-7.69603	-1.13411	1.89030
C	7.12700	0.19423	-1.11515	H	-7.53536	-3.56222	1.85387
C	7.60340	1.39863	-0.61867	H	-5.34950	-4.67560	1.41308
C	6.71081	2.32789	-0.08556	H	-3.33216	-3.27815	1.00776
C	5.36155	2.02115	-0.03240	H	7.83315	-0.52788	-1.50706
C	4.17145	-3.55536	-1.69279	H	8.66810	1.60668	-0.64044
C	4.88016	-3.98885	-2.80453	H	7.06647	3.27620	0.30266
C	5.81005	-3.13499	-3.39408	H	4.65259	2.72224	0.38836
C	6.04772	-1.88132	-2.84485	H	3.43265	-4.19432	-1.22544
O	3.13942	-2.65454	0.84007	H	4.69909	-4.97690	-3.21371
Si	0.09864	6.20615	0.00475	H	6.35430	-3.44461	-4.28023
C	-0.51887	6.91406	1.68044	H	6.77649	-1.23111	-3.31414
C	1.94780	6.63270	-0.27606	H	0.34086	6.81692	2.35970
C	-1.10089	6.67056	-1.42260	H	2.44345	6.07135	0.52990
C	-1.67228	6.08865	2.28056	H	-2.09301	6.65798	-0.94675

H	-1.40876	5.03671	2.40425	H	-2.29634	-5.77909	0.97069
H	-1.95803	6.48167	3.26342	H	-1.37959	-3.84052	-2.23332
H	-2.56702	6.13070	1.64913	H	-2.28447	-5.00593	-3.20884
H	-0.12316	9.04932	1.25125	H	-2.72989	-4.72242	-1.52888
H	-1.16439	8.76178	2.64275	H	-1.01868	-8.15338	-1.64880
H	-1.80155	8.56151	1.01618	H	-2.00960	-7.45450	-2.92619
H	2.06004	8.49464	0.89037	H	-2.53073	-7.33785	-1.25125
H	3.41643	8.24010	-0.20207	H	1.24340	-8.14340	-1.38916
H	1.86249	8.75766	-0.84363	H	2.66868	-8.34848	-0.37554
H	2.28900	5.01282	-1.72094	H	1.06717	-8.59479	0.30750
H	3.57943	6.21468	-1.66379	H	2.38757	-5.19849	1.61729
H	2.06171	6.59952	-2.46717	H	3.35983	-6.63633	1.30696
H	-1.36561	4.63440	-2.20569	H	1.83419	-6.78473	2.17011
H	-1.90646	5.91238	-3.30125	H	-1.10127	-4.15947	2.45151
H	-0.18617	5.58915	-3.09991	H	-1.84875	-5.46008	3.38646
H	-0.88844	8.86291	-1.22481	H	-0.11028	-5.46116	3.10178
H	-1.67894	8.33539	-2.70792	H	-1.56495	-8.19526	0.84144
H	0.06200	8.16214	-2.53441	H	-2.15472	-7.75358	2.44181
H	-0.08833	-5.95576	-2.49970	H	-0.42708	-7.90434	2.15693
H	2.18462	-5.94431	-0.76013				

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