# **Supporting Information**

### Au nanoflowers-Ag nanoparticles assembled SERS-active substrates

## for sensitive MC-LR detection

Yuan Zhao<sup>\*</sup>, Xuan Yang, Yaodong Luo, Ruipeng Yu, Lingling Zhang, Yaxin Yang and Qijun Song<sup>\*</sup> The Key Lab of Food Colloids and Biotechnology, Ministry of Education, School of Chemical and Material Engineering, Jiangnan University, Wuxi, Jiangsu, 214122, PRC.

\*Corresponding author. E-mail addresses: <u>zhaoyuan@jiangnan.edu.cn (Y. Zhao)</u>, <u>qsong@jiangnan.edu.cn (Q.</u>

Song).

# **Captions:**

#### **Experimental section**

 Table S1. Comparison of the LODs of different assay for MC-LR detection.

Table S2. The detail sequences of probes and aptamers of MC-LR.

**Fig. S1** Statistical analysis of diameters for Au NFs (A), Ag NPs (B) and the number of Ag NPs on the surface of Au NFs (C).

Fig. S2 Zeta potential (A) and DLS spectra (B) of Au NFs-Ag NPs core-satellite assemblies.

**Fig. S3** The enlarged Y-shaped DNA for the preparation of Au NFs-Ag NPs core-satellite assemblies. The distance between Au NFs and Ag NPs was calculated to 4 nm.

Fig. S4 SERS spectra of Au NFs-Ag NPs core-satellite assemblies.

**Fig. S5** DLS data of Au NFs-Ag NPs core-satellite assemblies in the present of different concentration of MC-LR.

**Fig. S6** SERS spectra of Au NFs-Ag NPs core-satellite assemblies after embedding in 5 nM MC-LR, MC-RR, AFB1, FB1, OTA and BSA solution.

Table S3. Detection of MC-LR spiked in Tai lake water.

**Fig. S7** (A) The representative SERS spectra of Au NFs-Ag NPs core-satellite assemblies for the detection of MC-LR samples from Table S3. (B) Statistical analysis of the SERS peak at 1330 cm<sup>-1</sup> under different concentration of MC-LR.

**Fig. S8** HPLC chromatogram for the detection of MC-LR in Tai lake water. (A) Tai lake water, (B)  $1\mu$ M MC-LR in ultrapure water, (C) 4.5  $\mu$ M MC-LR spiked in Tai lake water, (D) 9.0  $\mu$ M MC-LR spiked in Tai lake water.

Table S4. Detection of MC-LR in Tai lake water using HPLC and this method.

#### **Experimental section**

#### 1.1 Synthesis and characterization of Au NFs

An aliquot of 1.25 mL 5 mM HAuCl<sub>4</sub> solution was added into 23.75 mL Millipore-Q water. The mixture was heated at 100 °C for 5 min under vigorous stirring. An amount of 500  $\mu$ L 1% sodium citrate was quickly added and the reaction was kept for 10 min. And then, an amount of 500  $\mu$ L the above solution was added into 1 mL 0.1 M PBS containing 500  $\mu$ L 1% PVP under vigorous stirring. An amount of 300  $\mu$ L 10 mM HAuCl<sub>4</sub> solution and 300  $\mu$ L 10 mM NH<sub>2</sub>OH-HCl solution were synchronously added. After 2 h, the mixture was centrifuged at 2400 g for 5 min, and the precipitates were dissolved in 100  $\mu$ L Millipore-Q water. The plasmonic property of Au NFs was determined by UV-vis spectrometer, and the detailed structures were characterized through transmission electron microscope (TEM) and scanning electron microscope (SEM).

#### 1.2 Preparation of Ag NPs

 $14 \pm 2.5$  nm Ag NPs were synthesized according to the previous reported methods.<sup>1</sup> Briefly, an amount of 20 mL Millipore-Q water was mixed with 5 mL 1% PVP in an ice-water bath. An aliquot of 600 µL freshly prepared 0.1 M NaBH<sub>4</sub> solution was quickly added into the above solution. And then, 5 mL 1% PVP and 5 mL 10 mM AgNO<sub>3</sub> solution were added drop wise to the above solution under agitation. The suspension was kept at 80 °C for 2 h to remove the excess NaBH<sub>4</sub>. An aliquot of 2 mL Ag NPs was centrifuged at 9600 g for 10 min, and the precipitates were dissolved in 200 µL Millipore-Q water.

The size of Ag NPs was chosen according to the assembly steric hindrance. The small sized Ag NPs could be to the maximum assembled around the larger Au NFs, resulting in the formation of more "hot spot" and the enhancement of SERS signal. Ag NPs with the size below  $14 \pm 2.5$  nm

were difficult to centrifugal. Most of the small Ag NPs would be lost during the assembled process due to the hard manipulation. Ag NPs with the size above  $14 \pm 2.5$  nm were difficult to synthesize and showed irregular sphere structure. Bigger sized Ag NPs were not stable and may lead to aggregations when used for assembly.

#### 1.3 Surface functionalization of Au NFs and Ag NPs

An amount of 100  $\mu$ L Au NFs was used to react with 10  $\mu$ L 10  $\mu$ M probe 1 through Au-SH covalent bond.<sup>2, 3</sup> The molar ratio of Au NFs and probe 1 was about 1:100 (Table S1, ESI†). The mixture of Au NFs and probe 1 was incubated in 0.5× TBE buffer containing 50 mM NaCl. After 12 h, the solution was centrifuged to remove the unreacted probe 1. The precipitates were washed for three times and finally dissolved in 100  $\mu$ L 1× TBE buffer. Similarly, an amount of 200  $\mu$ L above-prepared Ag NPs was modified by probe 2 with the molar ratios of 1:10 according to the procedure described above. Ag NPs-probe 2 conjugates were further modified by 4-nitrothiophenolate (4-NTP) as Raman label. The final concentration of 4-NTP was 2  $\mu$ M.

#### 1.4 Fabrication of Au NFs-Ag NPs assembled SERS-active substrate

An amount of 50  $\mu$ L Au NFs-probe 1 solution, 100  $\mu$ L 4-NTP-Ag NPs-probe 2 solution and 100  $\mu$ L 1  $\mu$ M aptamers against MC-LR were mixed with 500  $\mu$ L 1× TBE buffer. After 12 h, the solution was centrifuged at 2400 g for 5 min, and the precipitates were dissolved in 100  $\mu$ L Millipore-Q water. Au NFs-Ag NPs core-satellite assemblies were fabricated depending on the formation of Y-shaped dsDNA hybrid configuration.

Au NFs-Ag NPs core-satellite assemblies employed as SERS-active substrates were mixed with 200  $\mu$ L different concentration of MC-LR solutions, *i.e.*, 0, 0.01, 0.05, 0.1, 0.5, 1, 5, 10 and 50 nM. The mixture was incubated in 20 mM Tris-HCl solution for 30 min at room temperature

and was centrifuged to remove the disassembled Ag NPs. The precipitates were finally dissolved and washed by 70% ethanol aqueous solution, in order to purify the residual Y-shaped dsDNA driven NP assemblies. SERS spectra of assemblies were determined through the LabRam-HR800 Micro-Raman spectrometer with the accumulation time of 10 s. SERS spectra were further managed through a Lab-spec 5.0 software. The surface charge density and hydrodynamic diameters of assemblies were measured through a Zetasizer Nano ZS system (ZetaPALS, Brookhaven Instruments Co. Ltd., UK).

#### 1.5 Evaluation of the selectivity and reproducibility of Raman aptasensors

The selectivity of Raman aptasensors was determined in the presence of 5 nM other mycotoxin and proteins, involving MC-RR, aflatoxin B1 (AFB1), fumonisin B1 (FB1), ochratoxin A (OTA) and bovine serum albumin (BSA). The reproducibility of detection is investigated by detecting the different concentration of MC-LR standard solutions spiked in Tai lake water (Wuxi), including 0.10, 0.20, 0.50, 1.00, 2.00 and 5.00 ng/mL. The results were validated through HPLC equipped by UV array detector set at 238 nm. Applied analytical column was ultimate AQ C18 ( $4.6\times250$  nm). Volume of injected samples was 10 µL. The mobile phase was 36% methyl cyanide aqueous solution containing 0.05% trifluoroacetic acid. The flow rate was kept at 1.0 mL min<sup>-1</sup>. Considering the low amount of MC-LR in Tai lake water and the LODs for HPLC method, different concentration of micromolar level MC-LR (0, 4.50, 9.00 µM) was spiked into the Tai lake water and was detected by HPLC. The samples were first diluted for 1000 times and then were detected by SERS-active substrates.

Assav	MC-LR	Refs	
	(pM)		
Voltammetric Aptasensor	1.9	4	
Surface-enhanced Fluorescence Immunosensor	7.0	5	
Raman aptasensor	8.6	This work	
Electrochemical Impedance Biosensor	18.0	6	
Quartz Crystal Microbalance Sensor	40.0	7	
Colorimetric Detection	50.0	8	
Surface Plasmon Resonance Based	100.0	9	
Immunoassay	190.0		
Immunosensor Based Nuclear		10	
Magnetic Resonance	000.0		
Electrochemical Biosensor 930.0		11	
Immunochromatographic assay	1000.0	12	

Table S1. Comparison of the LODs of different assay for MC-LR detection.

Table S2. The detail sequences of probes and aptamers of MC-LR.

Name	Sequences (5´-3´)		
Probe 1	-SH-(CH2)6-CTG TGA CGG TAA TT		
Probe 2	-SH-(CH2)6-TGG TAT GGT CAC AG		
Probe 3	GCG GAG ATG GGG CAT AAT GAG GTG GTA TGG GTA ATT GTC ATG GTG GTC GTC TTT GGC GCC		
Aptamers	GGC GCC AAA CAG GAC CAC CAT GAC AAT TAC CCA TAC CAC CTC ATT ATG CCC CAT CTC CGC		



**Fig. S1** Statistical analysis of diameters for Au NFs (A), Ag NPs (B) and the number of Ag NPs on the surface of Au NFs (C).



Fig. S2 Zeta potential (A) and DLS spectra (B) of Au NFs-Ag NPs core-satellite assemblies.



**Fig. S3** The enlarged Y-shaped DNA for the preparation of Au NFs-Ag NPs core-satellite assemblies. The distance between Au NFs and Ag NPs was calculated to 4 nm.



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Fig. S5 DLS data of Au NFs-Ag NPs core-satellite assemblies in the present of different concentration of MC-LR.



Fig. S6 SERS spectra of Au NFs-Ag NPs core-satellite assemblies after embedding in 5 nM MC-LR, MC-RR, AFB1, FB1, OTA and BSA solution.

Polluted water <sup>a</sup>	Original concentration (ng/mL) <sup>b</sup>	Spiked concentration (ng/mL)	Detected concentration (mean <sup>c</sup> $\pm$ SD <sup>d</sup> , ng/mL)	Recovery (mean $\overset{c}{\pm}$ SD $\overset{d}{}$ )
1	0.58	0.10	$0.65\pm0.07$	$95.59 \pm 2.85$
2	0.73	0.20	$0.89\pm0.05$	$95.70\pm2.63$
3	0.26	0.50	$0.74 \pm 0.04$	$97.37\pm3.07$
4	0.33	1.00	$1.28\pm0.21$	$96.24\pm3.18$
5	1.54	2.00	$3.38\pm0.43$	$95.48 \pm 3.42$
6	1.06	5.00	$5.92\pm0.76$	$97.70\pm3.56$

<b>Fable S3.</b> Detection of MC-LR spiked	in	Tai	lake	water.
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<sup>a</sup>Polluted water samples from six different areas of Wuxi city. <sup>b</sup>MC-LR was concentrated and then detected by HPLC. <sup>c</sup>The mean of three experiments. <sup>d</sup>SD = standard deviation.



Fig. S7 (A) The representative SERS spectra of Au NFs-Ag NPs core-satellite assemblies for the detection of MC-LR samples from Table S3. (B) Statistical analysis of the SERS peak at 1330 cm<sup>-1</sup> under different concentration of MC-LR.



Fig. S8 HPLC chromatogram for the detection of MC-LR in Tai lake water. (A) Tai lake water, (B) 1μM MC-LR in ultrapure water, (C) 4.5 μM MC-LR spiked in Tai lake water, (D) 9.0 μM MC-LR spiked in Tai lake water.

Samples	Spiked concentration (µM)	Detected concentration by HPLC (mean <sup>a</sup> $\pm$ SD <sup>b</sup> , $\mu$ M)	Detected concentration by this method $(mean^{a} \pm SD^{b}, \mu M)$
	0	0	$(0.58 \pm 0.15) \times 10^{-3}$
Tai lake	4.50	$4.56\pm0.62$	$4.52\pm0.24$
water	9.00	$9.08 \pm 1.27$	$9.03\pm0.57$

Table S4. Detection of MC-LR in Tai lake water using HPLC and this method.

<sup>a</sup>The mean of three experiments.  ${}^{b}SD = standard deviation.$ 

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