# Enantioselective Hydrogenation of Cyclic Imines Catalysed by Noyori-Ikariya Half-Sandwich Complexes and Their Analogues

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### 1. Experimental

### 1.1 General

Reactions with oxygen- and moisture-sensitive materials were carried out under argon atmosphere using standard Schlenk techniques. Solvents were dried using activated molecular sieves (4Å).

<sup>1</sup>H and <sup>13</sup>C NMR spectra were recorded on Bruker AVANCE III 400 MHz, 600 MHz and 700 MHz spectrometers with reference to the solvent residual signal as an internal standard (CD<sub>3</sub>CN:  $\delta_{\rm H}$  1.950 ppm,  $\delta_{\rm C}$  118.69 and 1.39 ppm, DMSO- $d_6$ :  $\delta_{\rm H}$  2.500 ppm,  $\delta_{\rm C}$ 39.600 ppm, CDCl<sub>3</sub>:  $\delta_{\rm H}$  7.265 ppm,  $\delta_{\rm C}$  77.00 ppm, CD<sub>2</sub>Cl<sub>2</sub>:  $\delta_{\rm H}$  5.320 ppm,  $\delta_{\rm C}$  54.00 ppm). <sup>1</sup>H NMR, <sup>13</sup>C NMR, COSY, gHSQCAD, gHMBCAD and 2D J-resolved spectra were measured using the manufacturers' software (Topspin 2.1 and 3.2, Bruker Biospin GmbH, Rheinstetten, Germany). The chemical shifts are given in  $\delta$  scale [ppm] and coupling constants in Hz. The digital resolution enabled reporting the  $\delta$  of <sup>1</sup>H to 3, coupling constants to 1, and <sup>13</sup>C to 2 decimal places. Signals assigned from 2D experiments are reported to 2 (<sup>1</sup>H) and 1 (<sup>13</sup>C) decimal places, respectively.

High resolution mass spectrometry measurements were carried out on LTQ Orbitrap Velos (Thermo Fisher Scientific, USA), UltrafleXtreme<sup>™</sup> MALDI-TOF/TOF (Bruker Daltonics, Germany) and GC-MS (Agilent 7890A GC + OA-TOF Waters GCT Premier MS) spectrometers. Electrospray ionization (ESI), matrix-assisted laser desorption/ionization (MALDI), or atmospheric pressure chemical ionization (APCI) were used as an ion sources.

For imines **1–6**, GC analyses were carried out on a Varian CP-3800 GC equipped with a Varian 1177 autosampler, FID and and a non-polar column (Varian VF-1, 60 m × 0.25 mm × 0.25 µm). Nitrogen (99.99%) was used as the carrier gas at a flow rate of 0.5 mL/min. The injector was heated to 300 °C, the injected volume was 1 µL and split ratio 1:25. The detector was heated to 270 °C for the measurement of conversion and 250 °C for the analysis of *ee* using a pre-column derivatization method previously described by us<sup>9e</sup>. In the case of **7–12**, the conversion and ee were acquired on a Agilent 7693/7890 GC equipped with the chiral J&W CycloSilTM-B (30 m × 0.25 mm × 0.25 µm) column, Agilent 5975C Triple-Axis MSD detector and helium as the carrier gas.

IR spectra were measured on Nicolet 6700 FTIR instrument.

Crystallographic measurements were done with four circle CCD diffractometer Gemini by Oxford Diffraction, Ltd., with graphite monochromated Cu K $\alpha$  radiation ( $\lambda$  = 1.54187 Å). The crystal structure was solved by charge flipping method using program Superflip<sup>1</sup> and refined with the Jana2006 program package<sup>2</sup> by full-matrix least squares technique on F. The molecular structure plot was prepared using ORTEP III.<sup>3</sup> Supramolecular interactions were viewed in Mercury.<sup>4</sup>

## 1.2 Chemicals

The following compounds were purchased from commercial sources and used as received:

(*S*,*S*)-*N*-*p*-toluenesulfonyl-1,2-diphenylethylenediamine (98 %, ABCR), Dichloro( $\eta^{5}$ -Cp\*)rhodium(I) dimer (99 %, Strem Chemicals), Dichloro( $\eta^{6}$ -*p*-cymene)ruthenium(II) dimer (Sigma-Aldrich), [RuCl( $\eta^{6}$ -*p*-cymene)(*S*,*S*)-TsDPEN] (**A**, Sigma-Aldrich), [RuCl( $\eta^{6}$ -mesitylene)(*S*,*S*)-TsDPEN] (**C**, Sigma-Aldrich), [RuCl(*S*,*S*)-Teth-TsDPEN] (**D**, Strem Chemicals), [RuCl(*S*,*S*)-Ts-DENEB] (**E**, Strem Chemicals), 6,7-diethoxy-1-methyl-3,4-dihydroisoquinoline (**7**, Acros Organics), 7-methoxy-1-methyl-3,4-dihydro-β-carboline (harmaline, **11**, TCI).

The following known compounds were synthesized according to previously reported procedures:

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6,7-dimethoxy-1-methyl-3,4-dihydroisoquinoline (1),<sup>9b</sup>

1-methyl-3,4-dihydroisoquinoline (2),<sup>9b</sup>

6-methoxy-1-methyl-3,4-dihydroisoquinoline (3),<sup>9b</sup>

7-methoxy-1-methyl-3,4-dihydroisoquinoline (4),<sup>9b</sup>

1-phenyl-3,4-dihydroisoquinoline (5),<sup>9f</sup>

1-(4-trifluoromethylphenyl)-3,4-dihydroisoquinoline (6),<sup>9f</sup>

6,7-dimethoxy-1-isopropyl-3,4-dihydroisoquinoline (8),<sup>9g</sup>

1-isopropyl-3,4-dihydroisoquinoline (9),<sup>9g</sup>

1-methyl-3,4-dihydro-\beta-carboline (harmalane, 10),<sup>9h</sup>

3-methyl-1,2-benzoisothiazole-1,1-dioxide (12).<sup>9i</sup>
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The syntheses of the following compounds are based on published procedures:

[Ru( $\eta^{6}$ -*p*-cymene)(HN-CHPhCHPhNSO<sub>2</sub>C<sub>6</sub>H<sub>4</sub>-*p*-CH<sub>3</sub>)] (**B**),<sup>5</sup> [RhCl( $\eta^{5}$ -Cp\*)(*S*,*S*)-TsDPEN] (**F**),<sup>6a</sup> 4-(4-methylcyclohexa-1,4-dienyl)butan-1-ol (**13a**),<sup>7</sup> [( $\eta^{6}$ -(4-(*p*-methylphenyl)butanol)RuCl<sub>2</sub>]<sub>2</sub> (**14a**),<sup>8</sup> 4-(4-methylcyclohexa-1,4-dienyl)butane (**15a**),<sup>7</sup> [RuCl( $\eta^{6}$ -(*p*-methylphenyl)butanol)(*S*,*S*)-TsDPEN] (**G**),<sup>6b</sup> [RuCl( $\eta^{6}$ -(*p*-methylphenyl)butane)(*S*,*S*)-TsDPEN] (**H**).<sup>6b</sup>

### 1.3 Synthetic procedures

### 1.3.1 Synthesis of [Ru( $\eta^6$ -p-cymene)(HN-CHPhCHPhNSO<sub>2</sub>C<sub>6</sub>H<sub>4</sub>-p-CH<sub>3</sub>)] (B)

 $[RuCl_2(\eta^6-p-cymene)]_2$  (125.4 mg, 0.197 mmol), (*S*,*S*)-TsDPEN (150.0 mg, 0.409 mmol) were dissolved in dichloromethane (2.9 mL), KOH (163.8 mg, 2.920 mmol) was added and the mixture was stirred at room temperature for 5 min. The reaction mixture was washed with water (2.9 mL), dried over CaH<sub>2</sub> and filtered. The solvent was evaporated under reduced pressure to afford a dark purple solid. Yield: 148 mg, (61 %).



<sup>1</sup>H NMR (600.23 MHz, CD<sub>3</sub>CN, 300.0 K):  $\delta$  1.201 (3H, d, J = 6.9, H-7"), 1.292 (3H, d, J = 6.9, H-7), 2.146 (3H, s, H-1), 2.296 (3H, s, CH<sub>3</sub>-*para*<sup>Ts</sup>), 2.642 (1H, dd, J = 6.9, 6.9, H-6), 3.938 (1H, d, J = 4.9, H-3'), 4.286 (1H, s, H-2'), 5.328 (1H, d, J = 6.0, H-4), 5.414 (1H, d, J = 6.0, H-3"), 5.627 (1H, d, J = 6.0, H-4"), 5.768 (1H, d, J = 6.0, H-3), 6.945 (2H, m, H-*meta*<sup>TS</sup>), 7.125 (2H, m, H-*ortho*), 7.15 (2H, m, H-*meta*), 7.17 (2H, m, H-*para*, H-*para*'), 7.211 (2H, m, H-*ortho*<sup>Ts</sup>), 7.248 (2H, m, H-*meta*'), 7.443 (1H, br. s, H-4'), 7.482 (2H, m, H-*ortho*').

<sup>13</sup>C NMR (150.93 MHz, CD<sub>3</sub>CN, 300.0 K): δ 20.07 (C-1), 21.31 (CH<sub>3</sub>-para<sup>Ts</sup>), 23.41 (C-7"), 23.57 (C-7), 32.71 (C-6), 72.33 (C-2'), 77.00 (C-4), 80.59 (C-4"), 81.54 (C-3"), 81.96 (C-3'), 84.74 (C-3), 89.06 (C-2), 100.04 (C-5), 127.09 (C-para'), 127.32 (C-ortho<sup>Ts</sup>), 127.40 (C-ortho), 127.65 (Cpara), 128.06 (C-ortho'), 128.38 (C-meta'), 128.83 (C-meta), 129.46 (C-meta<sup>Ts</sup>), 141.35 (Cpara<sup>Ts</sup>), 143.81 (C-ipso<sup>Ts</sup>), 145.61 (C-ipso), 148.45 (C-ipso').

HRMS (ESI<sup>+</sup>) calculated for C<sub>31</sub>H<sub>35</sub>N<sub>2</sub>O<sub>2</sub>RuS [M+H]<sup>+</sup>: 601.1457; found: 601.1459.

### 1.3.2 Synthesis of [RhCl( $\eta^{5}$ -Cp\*)(*S*,*S*)-TsDPEN] (F)

[RhCl<sub>2</sub>Cp\*]<sub>2</sub> (154.5 mg, 0.25 mmol) and (*S*,*S*)-TsDPEN (183 mg, 0.5 mmol) were dissolved in dichloromethane (5 mL), triethylamine (139.1  $\mu$ L, 101 mg, 1 mmol) was added and the solution was stirred for 20 min at room temperature. The reaction mixture was washed with water (2.5 mL), dried over MgSO<sub>4</sub> and filtered. The solvent was evaporated under reduced pressure to give an orange solid. Yield: 300 mg, 94 %. m.p. 219–220 °C (dec.).



<sup>1</sup>H NMR (400.00 MHz,  $CD_2Cl_2$ , 303.2 K):  $\delta$  1.817 (15H, s,  $CH_3^{Cp}$ ), 2.291 (3H, s,  $CH_3^{-}para^{Ts}$ ), 3.447 (1H, m, H-4'), 3.713 (1H, ddd, J = 10.8, 13.6, 2.3, H-3'), 3.903 (1H, m, H-4'), 3.972 (1H, d, J = 10.8, H-2'), 6.29 (2H, m, H-*ortho*), 6.653 (2H, m, H-*ortho*'), 6.780 (2H, m, H-*meta*'), 6.85 (2H, m, H-*meta*^{Ts}), 6.86 (1H, m, H-*para*'), 7.13 (2H, m, H-*meta*), 7.14 (1H, m, H-*para*), 7.311 (2H, m, H-*ortho*^{Ts}).

<sup>13</sup>C NMR (100.59 MHz,  $CD_2Cl_2$ , 303.2 K):  $\delta$  9.98 ( $CH_3^{Cp}$ ), 21.42 ( $CH_3$ -*para*<sup>Ts</sup>), 69.91 (C-2'), 72.14 (C-3'), 94.52 ( $C^{cp}$ ), 94.61 ( $C^{cp}$ ), 126.88 (C-*para*'), 127.58 (C-*meta*'), 127.62 (C-*ortho*), 128.32 (C-*meta*<sup>Ts</sup>), 128.55 (C-*ortho*<sup>Ts</sup>), 128.83 (C-*para*), 129.01 (C-*meta*), 129.35 (C-*ortho*'), 139.89 (C-*para*<sup>Ts</sup>), 140.06 (C-*ipso*), 140.32 (C-*ipso*'), 142.38 (C-*ipso*<sup>Ts</sup>).

HRMS (ESI<sup>+</sup>) calculated for C<sub>31</sub>H<sub>36</sub>N<sub>2</sub>O<sub>2</sub>RhS [M–Cl]<sup>+</sup>: 603.1553; found 603.1556.

#### 1.3.3 Synthesis of 4-(4-methylcyclohexa-1,4-dienyl)butan-1-ol (13a)

1,2-Bis(diphenylphosphino)ethane (DPPE) (102.0 mg, 0.256 mmol), CoBr<sub>2</sub> (54.3 mg, 0.248 mmol), ZnI<sub>2</sub> (157.6 mg, 0.494 mmol) and Zn powder (31.8 mg, 0.486 mmol) were mixed with tetrahydrofuran (33 mL) at 70 °C and the mixture was stirred for 15 min. Isoprene (1.00 g, 14.7 mmol, 1.47 mL) and 5-hexyn-1-ol (1.18 g, 12.1 mmol, 1.33 mL) were added and then further stirred for 2 hours at 70 °C. The solvent was evaporated under reduced pressure and the crude oily product was purified by flash column chromatography (hexane/EtOAc = 3:1) to afford colourless oil consisting of **13a** and side-products **13b** and **13c**. Yield: 1.86 g (93 %).



**13a**: <sup>1</sup>H NMR (600.23 MHz, CDCl<sub>3</sub>, 293.2 K): δ 1.479 (2H, m, H-3), 1.56 (2H, m, H-2), 1.663 (3H, s, H-11), 1.995 (2H, dd, *J* = 7.5, 7.5, H-4), 2.574 (4H, s, H-7, H-10), 3.635 (2H, dd, *J* = 6.4, 6.4, H-1), 5.407 (1H, s, H-9), 5.421 (1H, s, H-6).

<sup>13</sup>C NMR (150.93 MHz, CDCl<sub>3</sub>, 293.2 K): δ 22.98 (C-11), 23.50 (C-3), 29.81 (C-10), 31.57 (C-7), 32.35 (C-2), 36.73 (C-4), 62.83 (C-1), 118.45 (C-6), 118.57 (C-9), 131.22 (C-8), 134.60 (C-5).

**13b**: <sup>1</sup>H NMR (600.23 MHz, CDCl<sub>3</sub>, 293.2 K): δ 1.493 (2H, m, H-3), 1.545 (2H, m, H-2), 1.758 (3H, s, H-11), 2.076 (2H, dd, *J* = 7.4, 7.4, H-4), 2.093 (2H, m, H-7), 2.10 (2H, m, H-6), 3.635 (2H, dd, *J* = 6.4, 6.4, H-1), 5.582 (1H, d, *J* = 6.6, H-10), 5.590 (1H, d, *J* = 6.6, H-9).

<sup>13</sup>C NMR (150.93 MHz, CDCl<sub>3</sub>, 293.2 K): δ 22.92 (C-11), 23.73 (C-3), 26.92 (C-6), 28.66 (C-7), 32.34 (C-2), 36.61 (C-4), 62.81 (C-1), 119.02 (C-10), 119.25 (C-9), 133.01 (C-8), 135.97 (C-5).

**13c**: <sup>1</sup>H NMR (600.23 MHz, CDCl<sub>3</sub>, 293.2 K): δ 1.543 (2H, m, H-3), 1.565 (2H, m, H-2), 1.676 (3H, s, H-11), 2.008 (2H, m, H-4), 2.472 (2H, t, *J* = 8.0, H-6), 2.667 (2H, br. s., H-9), 3.635 (2H, m, H-1), 5.421 (2H, m, H-8, H-10).

<sup>13</sup>C NMR (150.93 MHz, CDCl<sub>3</sub>, 293.2 K): δ 23.19 (C-11), 23.39 (C-3), 27.64 (C-9), 32.36 (C-2), 33.79 (C-6), 36.86 (C-4), 62.83 (C-1), 118.29, 118.48 (C-8, C-10), 131.20, 131.22 (C-5, C-7).

HRMS (ESI<sup>+</sup>) calculated for C<sub>11</sub>H<sub>17</sub>O [M–H]<sup>-</sup> : 165.1279; found: 165.1277. IR: v<sub>max</sub> = 3622, 3442, 3082, 3011, 2935, 2879, 2861, 2820, 1662, 1610, 1475, 1448, 1439, 1384, 1053, 1022 cm<sup>-1</sup>.

#### **1.3.4** Synthesis of $[(\eta^{6}-(4-(p-tolyl)butan-1-ol)RuCl_{2}]_{2}$ (14a)

Diene **13a** (1.50 g, 9.02 mmol) was dissolved in ethanol (70 mL) and RuCl<sub>3</sub>·xH<sub>2</sub>O (387 mg, 1.48 mmol, trihydrate assumed) was added. The reaction mixture was stirred for 18 h under reflux. After the reaction, the mixture was hot-filtered to remove Ru black. The volatiles were removed on a rotary evaporator giving red-brown oil. The oil was washed with cold and degassed EtOH (3 × 3 mL) and hexane (3 × 3 mL). The resulting red-brown solid was dried in vacuum (1 Torr). Yield: 372 mg (75 %). M.p. 168–171 °C (dec.).



**14a**: <sup>1</sup>H NMR (400.00 MHz, DMSO-*d*<sub>6</sub>, 303.2 K): δ 1.461 (2H, m, H-8), 1.584 (2H, m, H-7), 2.082 (3H, s, H-1), 2.383 (2H, t, *J* = 7.7, H-6), 3.403 (2H, t, *J* = 6.3, H-9), 5.751 (2H, m, H-3), 5.786 (2H, m, H-4).

<sup>13</sup>C NMR (100.58 MHz, DMSO-*d*<sub>6</sub>, 303.2 K): *δ* 17.86 (C-1), 25.89 (C-7), 31.69 (C-6), 32.00 (C-8), 60.38 (C-9), 86.93 (C-3), 87.01 (C-4), 99.62 (C-2), 101.71 (C-5).

**14b**: <sup>1</sup>H NMR (400.00 MHz, DMSO- $d_6$ , 303.2 K):  $\delta$  1.565 (2H, m, H-9), 1.576 (2H, m, H-8), 2.129 (3H, s, H-1), 2.423 (2H, br. t., *J* = 7.6, H-7), 3.412 (2H, t, *J* = 6.4, H-10), 5.449 (1H, d, *J* = 5.7, H-3), 5.484 (1H, d, *J* = 5.7, H-5), 5.640 (1H, br. s., H-11), 5.964 (1H, t, *J* = 5.7, H-4).

<sup>13</sup>C NMR (100.58 MHz, DMSO-*d*<sub>6</sub>, 303.2 K): δ 18.52 (C-1), 25.64 (C-8), 32.10 (C-9), 32.47 (C-7), 60.39 (C-10), 80.19 (C-5), 81.09 (C-3), 84.18 (C-11), 88.33 (C-4), 106.58 (C-2), 109.62 (C-6).

HRMS (ESI<sup>+</sup>) calculated for  $C_{22}H_{32}O_2Ru_2Cl_3$  [M–Cl]<sup>+</sup>: 636.9555; found 636.9556. IR:  $v_{max}$ =3482, 3067, 3047, 3008, 2944, 2869, 1530, 1495, 1450, 1377, 1065, 1033, 809 cm<sup>-1</sup>.

#### 1.3.5 Synthesis of [RuCl( $\eta^{6}$ -(4-(p-tolyl)butan-1-ol)(S,S)-TsDPEN] (G)

Dimer **14a** (250.0 mg, 0.372 mmol) and ligand (*S*,*S*)-TsDPEN (272.5 mg, 0.744 mmol) were dissolved in dichloromethane (8.3 mL). Triethylamine (207.4  $\mu$ L, 150.5 mg, 1.487 mmol) was added thereto and the solution was stirred for 1 hour at room temperature. The orange precipitate was filtered off, washed with water (3 × 2 mL) and dried in vacuum (1 Torr). Yield: 415 mg, (84 %). m.p. 205–207 °C (dec.).



<sup>1</sup>H NMR (700.13 MHz, DMSO- $d_6$ , 303.2 K):  $\delta$  1.542 (2H, m, H-8), 1.693 (2H, m, H-7), 2.214 (3H, s, CH<sub>3</sub>-*para*<sup>Ts</sup>), 2.247 (3H, s, H-1), 2.568 (1H, m, H-6u), 2.599 (1H, m, H-6d), 3.179 (1H, dd, *J* = 10.3, 12.5, H-4'), 3.474 (2H, m, H-9), 3.557 (1H, m, H-3'), 3.708 (1H, d, *J* = 11.2, H-2'), 4.455 (1H, t, *J* = 5.1, 9-OH), 5.496 (1H, d, *J* = 5.6, H-3), 5.595 (2H, s, H-3", H-4"), 5.641 (1H, d, *J* = 5.6, H-4), 6.530 (2H, d, *J* = 7.4, H-*ortho*'), 6.673 (2H, dd, *J* = 7.4, 7.4, H-*meta*'), 6.778 (1H, t, *J* = 7.4, H-*para*'), 6.805 (2H, m, H-*ortho*), 6.819 (2H, d, *J* = 7.9, H-*meta*<sup>Ts</sup>), 7.043 (2H, d, *J* = 7.9, H-*ortho*<sup>Ts</sup>), 7.10 (3H, m, H-*meta*, H-*para*), 7.354 (1H, d, *J* = 7.8, H-4').

<sup>13</sup>C NMR (176.06 MHz, DMSO-*d*<sub>6</sub>, 303.2 K): δ 18.35 (C-1), 20.82 (CH<sub>3</sub>-para<sup>Ts</sup>), 26.49 (C-7), 32.11 (C-6), 32.17 (C-8), 60.48 (C-9), 68.58 (C-2'), 71.10 (C-3'), 81.95 (C-3"), 81.97 (C-4), 83.17 (C-3), 83.95 (C-4"), 95.09 (C-2), 96.65 (C-5), 125.96 (C-para'), 126.62 (C-meta', C-ortho<sup>Ts</sup>), 127.15 (C-ortho), 127.56 (C-meta<sup>Ts</sup>), 127.64 (C-para), 128.19 (C-meta), 128.93 (C-ortho'), 138.11 (C-para<sup>Ts</sup>), 139.24 (C-ipso'), 139.90 (C-ipso), 143.78 (C-ipso<sup>Ts</sup>).

HRMS (ESI<sup>+</sup>) calculated for C<sub>32</sub>H<sub>37</sub>N<sub>2</sub>O<sub>3</sub>RuS [M–Cl]<sup>+</sup>: 631.1568; found 631.1569. IR:  $v_{max}$  = 3455, 3279, 3226, 3084, 3064, 3030, 2933, 2861, 1599, 1494, 1453, 1378, 1270, 1128, 1105, 1084, 1039, 821, 813, 689 cm<sup>-1</sup>. [ $\alpha$ ]<sup>20</sup><sub>589</sub> = +66.7° (*c* 0.397, DMSO).

#### 1.3.6 Synthesis of 4-(4-methylcyclohexa-1,4-dienyl)butane (15a)

DPPE (102.0 mg, 0.256 mmol),  $CoBr_2$  (54.3 mg, 0.248 mmol),  $ZnI_2$  (157.6 mg, 0.494 mmol) and Zn powder (31.8 mg, 0.486 mmol) were mixed with tetrahydrofuran (33 mL) at 70 °C and the mixture was stirred for 15 min. Isoprene (1.00 g, 14.7 mmol, 1.47 mL) and 1-hexyne (1.00 g, 12.1 mmol, 1.39 mL) were added and then further stirred for 2 hours at 70 °C. The solvent was evaporated under reduced pressure and the crude oily product was purified by flash column chromatography (hexane/EtOAc = 3:1) to afford colourless oil consisting of **15a–c.**<sup>1</sup> Yield: 1.54 g (84 %).



**15a**: <sup>1</sup>H NMR (600.23 MHz, CDCl<sub>3</sub>, 303.2 K): δ 0.899 (3H, t, *J* = 7.2, H-1), 1.306 (2H, m, H-2), 1.389 (2H, m, H-3), 1.674 (3H, s, H-11), 1.968 (2H, t, *J* = 7.5, H-4), 2.584 (4H, m, H-7, H-10), 5.416 (1H, m, H-6), 5.421 (1H, m, H-9).

<sup>13</sup>C NMR (150.93 MHz, CDCl<sub>3</sub>, 303.2K):  $\delta$  14.01 (C-1), 22.44 (C-2), 23.05 (C-11), 29.70 (C-3), 29.92 (C-10), 31.61 (C-7), 36.81 (C-4), 117.97 (C-6), 118.72 (C-9), 131.28 (C-8), 135.15 (C-5).

**15b**: <sup>1</sup>H NMR (600.23 MHz, CDCl<sub>3</sub>, 303.2 K): δ 0.899 (3H, t, *J* = 7.2, H-1), 1.306 (2H, m, H-2), 1.398 (2H, m, H-3), 1.767 (3H, s, H-11), 2.049 (2H, t, *J* = 7.6, H-4), 2.102 (4H, m, H-6, H-7), 5.578 (1H, m, H-10), 5.602 (1H, m, H-9).

<sup>13</sup>C NMR (150.93 MHz, CDCl<sub>3</sub>, 303.2K): δ 14.00 (C-1), 22.41 (C-2), 22.98 (C-11), 27.05 (C-6), 28.72 (C-7), 29.88 (C-3), 36.69 (C-4), 118.60 (C-10), 119.34 (C-9), 132.83 (C-8), 136.73 (C-5).

**15c**: <sup>1</sup>H NMR (600.23 MHz, CDCl<sub>3</sub>, 303.2 K): δ 0.899 (3H, m, H-1), 1.315 (2H, m, H-2), 1.375 (2H, m, H-3) 1.687 (3H, s, H-11), 1.979 (2H, m, H-4), 2.482 (2H, t, *J* = 8.0, H-6), 2.678 (2H, br. s., H-9), 5.412 (2H, m, H-8, H-10).

<sup>13</sup>C NMR (150.93 MHz, CDCl<sub>3</sub>, 303.2K): *δ* 14.01 (C-1), 22.46 (C-2), 23.26 (C-11), 27.69 (C-9), 29.58 (C-3), 33.89 (C-6) 36.95 (C-4), 117.80 (C-10), 118.48 (C-8), 131.37 (C-7, C-5).

<sup>&</sup>lt;sup>1</sup> Small amount (9d) of aromatic impurity was formed as a decomposition product, which has been assigned for the sake of completeness.

**15d**: <sup>1</sup>H NMR (600.23 MHz, CDCl<sub>3</sub>, 303.2 K): δ 0.899 (3H, m, H-1), 1.291 (2H, m, H-2), 1.577 (2H, m, H-3), 2.318 (3H, s, H-*para*-CH<sub>3</sub>), 2.570 (2H, t, *J* = 7.9, H-4), 7.074 (2H, m, H-*ortho*), 7.090 (2H, m, H-*meta*).

<sup>13</sup>C NMR (150.93 MHz, CDCl<sub>3</sub>, 303.2K): δ 13.96 (C-1), 20.98 (C-*para*-CH<sub>3</sub>), 22.36 (C-2), 33.80 (C-3), 35.19 (C-4), 128.26 (C-*ortho*), 128.88 (C-*meta*), 134.91 (C-*para*), 139.81 (C-*ipso*).

HRMS (EI) calculated for  $C_{11}H_{18}$  [M<sup>+•</sup>]: 150.1409; found 150.1410. IR:  $v_{max}$  = 3080, 3036, 2960, 2930, 2873, 2859, 2820, 1662, 1610, 1467, 1457, 1379 cm<sup>-1</sup>.

### 1.3.7 Synthesis of $[(\eta^6-(1-buty)-4-methy)Benzene)RuCl_2]_2$ (16a)

Diene **15a** (1.547 g, 10.3 mmol) was dissolved in ethanol (80 mL) and RuCl<sub>3</sub>·xH<sub>2</sub>O (538 mg, 2.1 mmol, trihydrate assumed) was added. The reaction mixture was stirred for 18 h under reflux. After the reaction, the mixture was hot-filtered to remove Ru black. The volatiles were removed on a rotary evaporator giving red-brown oil. The oil was washed with cold and degassed EtOH (3 × 3 mL) and hexane (3 × 3 mL). The resulting red-brown solid was dried in vacuum (1 Torr). Yield: 988 mg (75 %). M.p. 192 °C (dec.).



**16a**: <sup>1</sup>H NMR (700.13 MHz, DMSO-*d*<sub>6</sub>, 303.2 K): δ 0.894 (3H, t, *J* = 7.3, H-9), 1.333 (2H, m, H-8), 1.539 (2H, m, H-7), 2.080 (3H, s, H-1), 2.379 (2H, t, *J* = 7.4, H-6), 5.745 (2H, d, *J* = 5.5, H-3), 5.789 (2H, d, *J* = 5.5, H-4).

<sup>13</sup>C NMR (176.06 MHz, DMSO- $d_6$ , 303.2K):  $\delta$  13.76 (C-9), 17.85 (C-1), 21.85 (C-8), 31.27 (C-7), 31.55 (C-6), 86.92 (C-3), 86.98 (C-4), 99.59 (C-2), 101.70 (C-5).

**16b**: <sup>1</sup>H NMR (700.13 MHz, DMSO-*d*<sub>6</sub>, 303.2 K): δ 0.893 (3H, m, H-10), 1.347 (2H, m, H-9), 1.538 (2H, m, H-8), 2.126 (3H, s, H-1), 2.418 (2H, t, *J* = 7.6, H-7), 5.445 (1H, d, J = 5.2, H-3), 5.483 (1H, d, *J* = 5.2, H-5), 5.643 (1H, s, H-11), 5.958 (1H, t, *J* = 5.2, H-4).

<sup>13</sup>C NMR (176.06 MHz, DMSO-*d*<sub>6</sub>, 303.2K): *δ* 13.78 (C-10), 18.52 (C-1), 21.96 (C-9), 31.03 (C-8), 32.35 (C-7), 80.15 (C-5), 81.06 (C-3), 84.16 (C-11), 88.32 (C-4), 106.53 (C-2), 109.55 (C-6).

HRMS (MALDI) calculated for  $C_{22}H_{32}Ru_2Cl_3$  [M–Cl]<sup>+</sup>: 604.9651; found 604.9659. IR:  $v_{max}$  =3067, 3045, 3009, 2951, 2927, 2868, 2859, 1530, 1497, 1466, 1377, 1034, 811 cm<sup>-1</sup>.

#### 1.3.8 Synthesis of [RuCl( $\eta^{6}$ -(1-butyl-4-methylbenzene)(*S*,*S*)-TsDPEN] (H)

Dimer **16a** (200.0 mg, 0.312 mmol) and ligand (*S,S*)-TsDPEN (228.9 mg, 0.625 mmol) were dissolved in dichloromethane (7 mL). Triethylamine (174.2  $\mu$ L, 126.4 mg, 1.249 mmol) was added thereto and the solution was stirred for 1 hour at room temperature and washed with water (3 × 2 mL). Removal of solvents gave the crude product. The product was purified by flash chromatography (gradient elution from hexane to EtOAc to MeOH) and dried in vacuum (1 Torr) to give a dark orange solid. Yield: 225 mg, (56 %). m.p. 149 °C (dec.).



<sup>1</sup>H NMR (600.23 MHz, CDCl<sub>3</sub>, 303.2 K):  $\delta$  0.964 (3H, t, *J* = 7.3, H-9), 1.421 (2H, m, H-8), 1.654 (2H, m, H-7), 2.215 (3H, s, CH<sub>3</sub>-para<sup>Ts</sup>), 2.249 (3H, s, H-1), 2.582 (2H, m, H-6), 3.187 (1H, m, H-4'u), 3.556 (1H, m, H-3'), 3.712 (1H, d, *J* = 11.2, H-2'), 5.498 (1H, d, *J* = 5.4, H-3a), 5.592 (2H, s, H-3b, H-4b), 5.639 (1H, d, *J* = 5.4, H-4a), 6.534 (2H, d, *J* = 7.5, H-*ortho*'), 6.676 (2H, t, *J* = 7.5, H-*meta*'), 6.782 (1H, t, *J* = 7.5, H-*para*'), 6.802 (2H, m, H-*ortho*), 6.818 (2H, m, H-*meta*<sup>Ts</sup>), 7.048 (2H, m, H-*ortho*<sup>Ts</sup>), 7.097 (1H, m, H-*para*), 7.102 (2H, m, H-*meta*), 7.343 (1H, m, H-4'd).

<sup>13</sup>C NMR (150.93 MHz, CDCl<sub>3</sub>, 303.2K): δ 13.80 (C-9), 18.30 (C-1), 20.77 (CH<sub>3</sub>-para<sup>Ts</sup>), 21.94 (C-8), 31.92 (C-7), 31.94 (C-6), 68.56 (C-2'), 71.11 (C-3'), 81.91 (C-3b), 81.94 (C-4a), 83.17 (C-3a), 83.83 (C-4b), 95.01 (C-2), 96.68 (C-5), 125.94 (C-para'), 126.58 (C-meta'), 126.62 (C-ortho<sup>Ts</sup>), 127.11 (C-ortho), 127.51 (C-meta<sup>Ts</sup>), 127.61 (C-para), 128.16 (C-meta), 128.89 (C-ortho'), 138.09 (C-para<sup>Ts</sup>), 139.27 (C-ipso'), 139.87 (C-ipso), 143.74 (C-ipso<sup>Ts</sup>).

HRMS (ESI<sup>+</sup>) calculated for C<sub>32</sub>H<sub>37</sub>N<sub>2</sub>O<sub>2</sub>RuS [M–Cl]<sup>+</sup>: 615.1619; found 615.1608. IR:  $v_{max}$  = 3275, 3219, 3085, 3061, 3029, 2956, 2927, 2871, 1599, 1534, 1494, 1466, 1378, 1269, 1132, 1129, 1085, 811, 618 cm<sup>-1</sup>. [ $\alpha$ ]<sup>20</sup><sub>589</sub> = +28.5° (*c* 0.134, DMSO).

#### 1.3.9 AH in batch pressure reactors – optimized setup

The substrate (44  $\mu$ mol) was weighed into a GC vial and dissolved in methanol (0.5 mL). Trifluoroacetic acid (3.4  $\mu$ L, 44  $\mu$ mol) was added thereto, and the mixture was stirred for 5 min. The catalyst solution (0.44  $\mu$ mol of catalyst; concentration 5.4 mg/1 mL in methanol) was added to the mixture and the vial was closed by a pierced square of Parafilm. The vial was placed in an autoclave, which was closed, purged (3 × 5 bar) and filled with hydrogen (15 bar).

After 6 h, the autoclave was opened and a sample of the reaction mixture (200  $\mu$ L) was mixed with a saturated solution of Na<sub>2</sub>CO<sub>3</sub> (1 mL). The solution was extracted with diethyl ether (3 × 1.5 mL) and the organic phase was dried over Na<sub>2</sub>SO<sub>4</sub> for 1 h. The ether solution was stripped to dryness in a stream of air. The stripped residue was dissolved in acetonitrile (GC analysis) or acetonitrile-*d*<sub>3</sub> (NMR analysis).

#### 1.3.10 ATH in an NMR spectrometer

The reactions were monitored *in situ* in an NMR spectrometer at 30 °C by following the same protocol as described in our previous publications.<sup>9a</sup>

Acetonitrile- $d_3$  (volume calculated so as to reach 730 µL of total volume of the reaction mixture), formic acid (13.1 µL, 0.347 mmol) and triethylamine (19.3 µL, 0.139 mmol) were mixed in an NMR tube. The catalyst (0.275 µmol) dissolved in acetonitrile- $d_3$  (conc. 5.4 mg/mL) was added to the mixture. Finally, after 5 minutes, the substrate (55 µmol) was added to the reaction mixture. The reaction was followed by <sup>1</sup>H NMR.

In the end of the kinetic experiment, a sample of the reaction mixture (200  $\mu$ L) was mixed with a saturated solution of Na<sub>2</sub>CO<sub>3</sub> (1 mL). The solution was extracted with diethyl ether (3 × 1.5 mL) and the organic phase was dried over Na<sub>2</sub>SO<sub>4</sub> for 1 h. The ether solution was stripped to dryness in a stream of air. The stripped residue was dissolved in acetonitrile (0.8 mL) and enantioselectivity was measured by GC.<sup>9</sup>e

#### 1.3.11 AH of 1 on a synthetic scale

Substrate **1** (90 mg, 0.44 mmol) was weighed into a vial and dissolved in methanol (5 mL). Trifluoroacetic acid (34  $\mu$ L, 0.44 mmol) was added thereto, and the mixture was stirred for 5 min. The catalyst solution (0.0044 mmol of catalyst; concentration 5.4 mg/1 mL in methanol) was added to the mixture, and the vial was closed by a pierced square of Parafilm. The vial was placed in an autoclave, which was closed, purged (3 × 5 bar) and filled with hydrogen (15 bar).

After 6 h, the autoclave was opened and the reaction mixture was alkalized with a solution of NaOH. The solution was extracted with diethyl ether ( $3 \times 5$  mL) and filtered through a plug of silica gel to remove the catalyst. The organic phase was dried over Na<sub>2</sub>SO<sub>4</sub>, filtered and evaporated. Pale yellow oil was obtained. Yield: 84 mg (92%). 97% *ee* and >95% purity.

# 2. Supplementary results

## 2.1 Asymmetric hydrogenation experiments

	$H_{3}CO \xrightarrow{H_{2}} N$	(A (1 mol%)) solvent, RT $H_3C$		
Entry	Concentration (mM)	Volume (mL)	Solvent	Conv (%)
1	44	1	acetonitrile	0
2	88	0.5	acetonitrile	0
3	176	0.25	acetonitrile	0
4	44	1	DMSO	2
5	88	0.5	DMSO	4
6	176	0.25	DMSO	3
7	44	1	MeOH	6
8	88	0.5	МеОН	10
9	176	0.25	MeOH	6

Table S1. Screening of reaction conditions in the AH of 1 with catalyst A.<sup>a</sup>

<sup>a</sup> Amount of substrate  $n = 44 \mu mol$ , catalyst loading 1 mol%,  $p(H_2) = 15$  bar, RT, 6 h.

Table S2.	Screening	of acid	additives	in the	AH of 1	with catal	yst <b>A</b> .ª

H₃CO H₃CO		$\begin{array}{c} \underline{H_2, \mathbf{A} (1 \text{ mol}\%)} \\ \text{MeOH, acid} \\ \text{RT} \end{array} \xrightarrow{H_3CO} \\ H_3CO \end{array}$	NH
Entry	Acid	Acid/Substrate <sup>b</sup>	Conv (%)
1	HBF4 (48%)	1	19
2	TfOH	1	9
3	CF <sub>3</sub> COOH	1	57
4	CF <sub>3</sub> COOH	0.5	18
5	CF <sub>3</sub> COOH	0.75	42
6	CF <sub>3</sub> COOH	1.25	52
7	CF₃COOH	1.5	54

<sup>a</sup> Amount of substrate  $n = 44 \mu mol$ , concentration of substrate c = 88 mM, catalyst loading 1 mol%,  $p(H_2) = 15$  bar, RT, 6 h. <sup>b</sup> Molar ratio.



**Fig. S1** ATH of imines 1–4, 8–10 and 12 catalysed by complexes A, G and H using HCOOH/Et<sub>3</sub>N (5:2) in CD<sub>3</sub>CN at 0.5 mol% catalyst loading and a temperature of 30 °C. In parentheses: *ee*, TON after 50 min, TOF at 20% conversion in  $h^{-1}$ .



### 2.2 Synthesis of G and H, structural elucidation of 13a-c

Scheme S1 Synthesis of complexes G and H.



**Fig. S2** Parts of a <sup>1</sup>H-<sup>13</sup>C HMBC NMR spectrum of isolated mixture of **13a**–c. a) **13b**: Correlation of methyl 11 with methylene 7, correlation between methylenes 4 and 6;

b) 13c: Correlation of methyl 11 and methylene 4 to common methylene 6





Fig. S3 Solid-state molecular structure of complex G



**Fig. S4** The molecule of dichloromethane crystallized with **G** in an equimolar ratio. The crystal structure was held together by O-H···Cl hydrogen bonds of 2.39(5) Å between the hydroxyl group (donor) and chloride (acceptor) as evident from the packing view down the *c* axis. Such a hydrogen bond has been observed for similar compounds, although in an intramolecular manner.<sup>10</sup>

1.21 onoclinic, <i>P</i> 2 <sub>1</sub> 0
onoclinic, <i>P</i> 2 <sub>1</sub> 0
0
77870(10)
.4690(4)
7911(2)
4.5931(14)
56.08(5)
937
$38 \times 0.24 \times 0.18$
487
936
09
54
0
034
54
0291
0405
44
.91

Table S3 Crystallographic and structure refinement for complex G

The title compound crystallizes in the monoclinic space group  $P2_1$ . The molecular structure is given in Figure 1. The single-crystal structure of  $C_{33}H_{39}Cl_3N_2O_3RuS$  is built up of discrete moieties of  $C_{32}H_{37}Cl_2N_2O_3RuS$  and dichloromethane with the two formula units in the asymmetric unit.

The absolute structure was tested by introducing twinning and refining volume fractions. The inversion was used as a merohedral twinning operation. In this case, the volume fraction of the inversion twin is the Flack parameter.<sup>11</sup> This parameter refined to a final value of -0.008(7), which confirms that the above configuration is the correct absolute structure.

Crystallographic data (including structure factors) for the structure reported in this article has been deposited with the Cambridge Crystallographic Center, CCDC No. 1021406 Copies of the data can be obtained free of charge on application to The Director, CCDC, 12 Union Road, Cambridge CB2 1EZ, UK, fax: +44 1223 336 033, e-mail: deposit@ccdc.cam.ac.uk or http://www.ccdc.cam.ac.uk.

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# 4. Copies of NMR spectra

## 4.1 4-(4-methylcyclohexa-1,4-dienyl)butan-1-ol (13a-c)





## 4.2 [( $\eta^{6}$ -(4-(p-methylphenyl)butanol)RuCl<sub>2</sub>]<sub>2</sub> (14a,b)

BV011-3 solvent: DMSO-d6 temp: 303,2 K date: 29 MAR 2013



BV011-3 solvent: DMSO-d6 temp: 303,2 K date: 29 MAR 2013 109.62 106.58 τ, . 7 0 0 0 0 N H 0 0  $0 \infty$ 0 100004 00 H01900 mm 686014 in co Current Data Parameters 9. 6 201.00 201.00 201.00 201.00 20.000 20.000 20.000 20.000 20.000 20.000 20.000 20.0000 . . . . BV011-3 NAME 00 200 EXPNO 2 T O  $\infty$   $\infty$   $\infty$   $\infty$   $\infty$   $\infty$ 00  $\dashv$   $\dashv$ 12 V 512 V/  $\backslash$ 1 1/ PROCNO 3 F2 - Acquisition Parameters Date\_ 20130329 Time 10.56 INSTRUM spect PROBHD 5 mm BBI 1H/D-PULPROG zqpq TD 65536 SOLVENT DMSO NS 2304 DS 4 25252.525 Hz SWH FIDRES 0.385323 Hz AQ 1.2976128 sec RG 2050 19.800 usec DW DE 6.50 usec ΤE 303.2 K 1.00000000 sec D1 0.03000000 sec D11 TDO 1 ====== CHANNEL fl ======= SF01 100.5921496 MHz NUC1 13C P1 13.90 usec PLW1 89.36699677 W ====== CHANNEL f2 ======= SFO2 400.0016000 MHz NUC2 1H CPDPRG[2 bi\_waltz65\_256 PCPD2 90.00 usec 15.10200024 W PLW2 PLW12 0.09135900 W PLW13 0.07400100 W F2 - Processing parameters SI 131072 SF 100.5801248 MHz WDW EM SSB 0 LB 0 Hz GB 0 1.40 PC Т 110 100 90 80 70 60 50 40 30 20 10 ppm

## 4.3 [RuCl( $\eta^{\circ}$ -(p-methylphenyl)butanol)(S,S)-TsDPEN] (G)

BV023-1 solvent: DMSO-d6 temp: 303,2 K date: 9 Apr 2013



S24

solvent: DMSO-d6 temp: 303,2 K date: 9 Apr 2013 804486466777 710 001010100100 ഗര 5770 0 00 ω 60 25 . . . . . . . . . . . 00 0400 <u>й</u>, 4 യന Current Data Parameters • • 000000000000000 . . . . ٠. -NAME BV023-1 4 m m m n n n n n n n n 90 ммнн  $\sim \infty$ 0 0 00 0 0 0 ດັດ  $\infty \infty \infty \infty$ r 0 õ m H N N M M EXPNO 2 2 ζ. 1 J  $\backslash /$ PROCNO F2 - Acquisition Parameters Date\_ 20130409 Time 14.19 INSTRUM spect PROBHD 5 mm CPTCI 1H/ PULPROG zgpg30 TD 130892 SOLVENT DMSO 4096 NS DS 32 41666.668 Hz SWH FIDRES 0.318329 Hz 1.5707040 sec AQ 2050 RG DW 12.000 usec DE 18.00 usec ΤE 303.3 K 1.00000000 sec D1 D11 0.03000000 sec TDO 1 ====== CHANNEL f1 ======= 176.0654325 MHz SF01 NUC1 13C P1 12.50 usec -1.00000000 W PLW1 ====== CHANNEL f2 ======= 700.1328005 MHz SFO2 NUC2 1H CPDPRG[2 mlev PCPD2 65.00 usec PLW2 -1.00000000 W PLW12 -1.00000000 W PLW13 -1.00000000 W F2 - Processing parameters SI 262144 SF 176.0479018 MHz EM WDW SSB 0 LB 1.00 Hz GB 0 PC 1.40 ..... 150 140 90 80 70 60 50 40 30 20 130 120 110 100 10 ppm

BV023-1

S25

## 4.4 4-(4-methylcyclohexa-1,4-dienyl)butane (15a-c)





# 4.5 [( $\eta^{6}$ -(4-(p-methylphenyl)butane)RuCl<sub>2</sub>]<sub>2</sub> (16a,b)

BV-037-1 solvent: DMSO temp: 303.2 K date: 24 Jul 2014





S29

## 4.6 [RuCl( $\eta^{6}$ -(p-methylphenyl)butane)(S,S)-TsDPEN] (H)

BV038-1a solvent: DMSO temp: 303.2 K date : 20 Aug 2014



