

Structural transition and superconductivity in hydrothermally synthesized FeX (X=S,Se)

Supplementary

Table S1: Rietveld refinement of FeSe_{hydro} in the triclinic low-temperature phase (10 K)

Space group	$P\bar{1}$ (Nr. 2)				
a (pm)	376.59(2)				
b (pm)	376.66(2)				
c (pm)	547.93(1)				
α (°)	90.024(4)				
β (°)	89.943(4)				
γ (°)	90.168(2)				
Volume (Å ³)	77.72(1)				
Atom	Wyck.	x	y	z	b_{eq}
Fe	$2i$	0.2328(7)	0.25*	0.00*	0.44(2)
Se	$2i$	0.7458(5)	0.25*	0.2663(1)	0.20(1)
R_{wp}	3.25				
R_{exp}	1.84				
χ^2	1.76				
Fe - Fe (pm)	256.9(2), 257.7(2), 275.2(2), 276.0(2)				
Fe - Se (pm)	234.3(2), 238.35(3), 238.41(3), 242.2(2)				
Se-Fe-Se (°)	113.93(1), 114.14(1), 104.43(6), 104.38(1), 109.89(1), 110.14(1)				

*Coordinates fixed after refinements converged to $y = 0.25$ respectively $z = 0$ within the *esd*'s.

Figure S1: Rietveld fit of FeSe_{hydro} at 10 K. Insert: Splitting of the (220)_{tetra} Bragg reflection into the doublet [(220)_{triclinic}, ($\bar{2}\bar{2}0$)_{triclinic}].

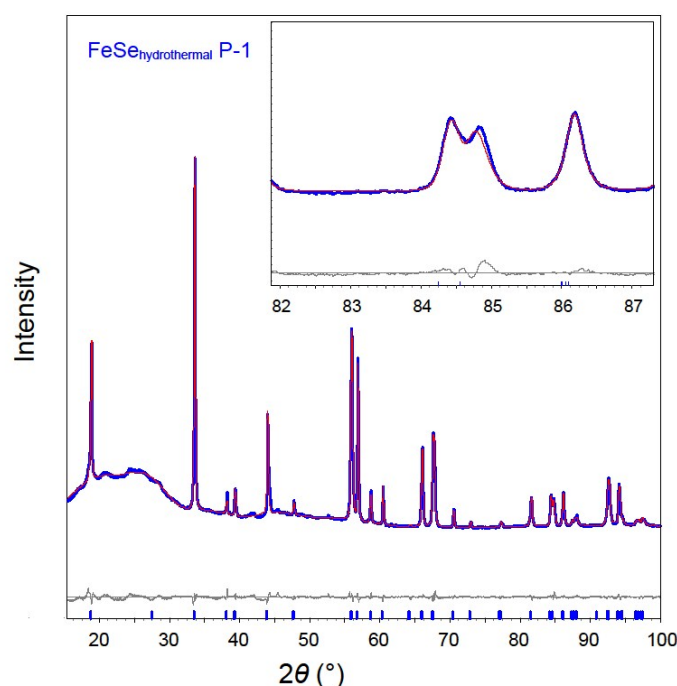
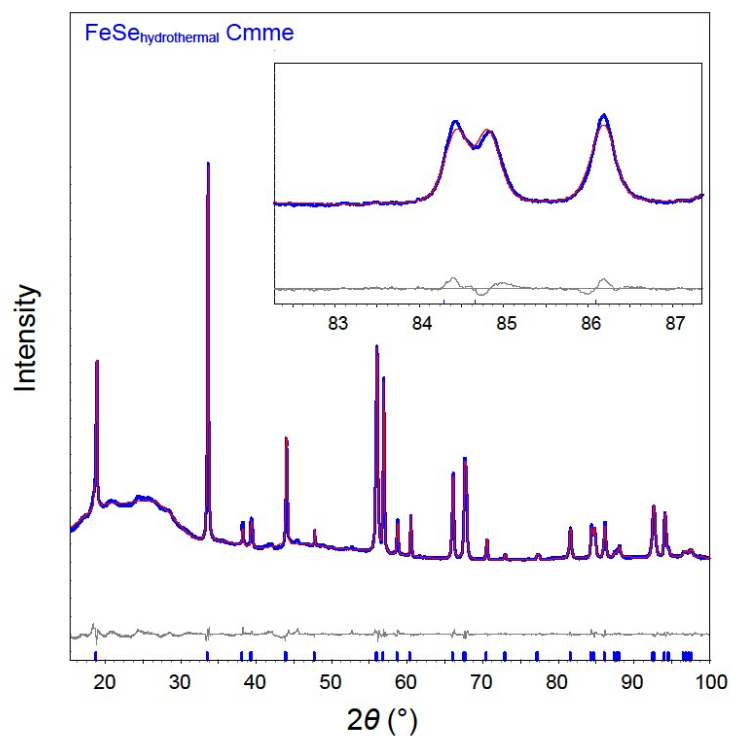


Figure S2: Rietveld-fits of $\text{FeSe}_{\text{hydro}}$ at 10 K: orthorhombic ($Cmme$) versus mixture of tetragonal ($P4/nmm$) and orthorhombic ($Cmme$) symmetry illustrated at the $(220)_{\text{tetra}}$ Bragg reflection splitted into the doublet $[(040)_{\text{ortho}}, (400)_{\text{ortho}}]$.

a) Orthorhombic ($Cmme$)



b) Orthorhombic ($Cmme$) and tetragonal ($P4/nmm$) mixed

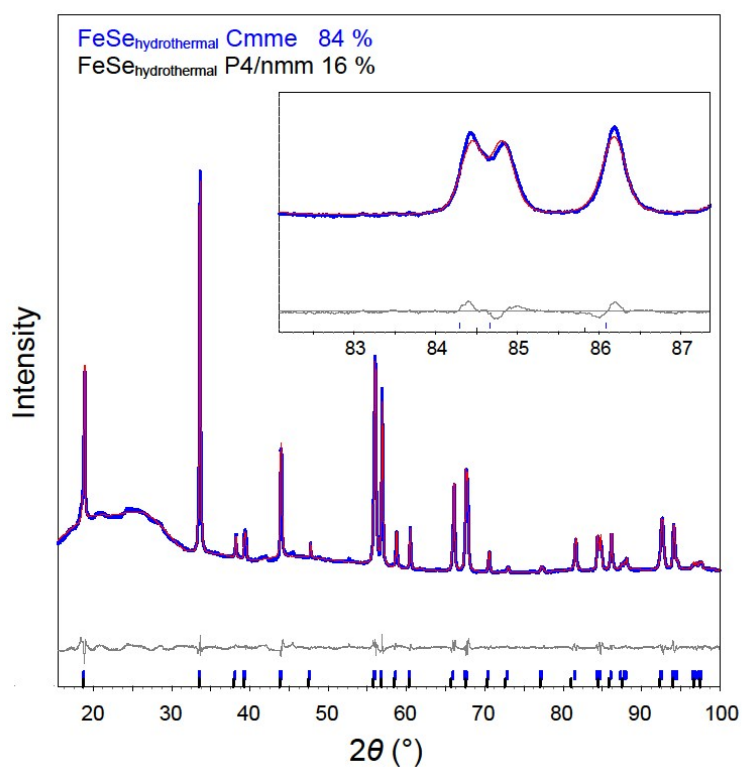


Figure S3: Temperature evolution of the unit cell volume in FeSe synthesized via hydrothermal (black) and conventional (blue) reaction method, respectively.

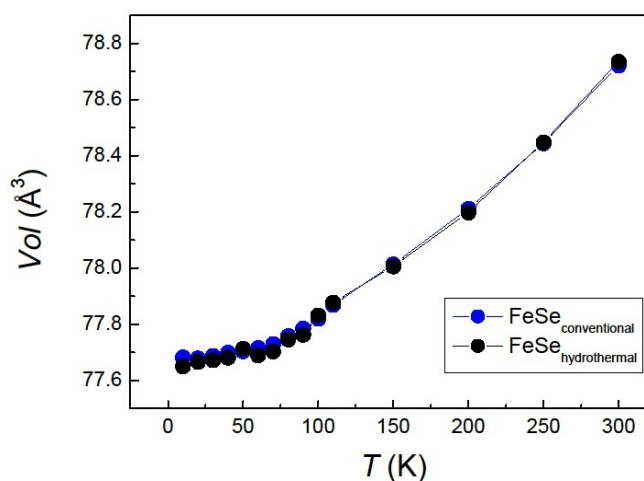


Table S2: Crystallographic data of FeS (single crystal)

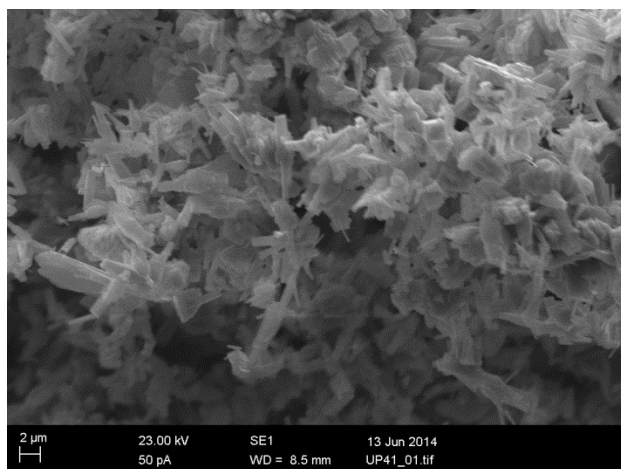
Formula	FeS
Formula weight (g·mol ⁻¹)	87.9
Crystal System	Tetragonal
Space group	<i>P4/nmm</i> O2 (No. 129)
<i>a</i> , <i>c</i> (pm)	368.06(3), 502.83(7)
<i>V</i> (nm ³)	0.06812(1)
<i>Z</i>	2
<i>d</i> _{calc} (g·cm ³)	4.2859
μ (Mo-K α) (mm ⁻¹)	11.809
Crystal Size (μ m ³)	20 × 20 × 2
Temperature (K)	293
Wavelength (pm)	λ = 71.073
θ range (deg.)	4.1 – 30.3
<i>hkl</i> range	–4 → +5; –5 → +4; –7 → +7
Tot., Uniq. Data, <i>R</i> _{int}	1248, 81, 0.108
<i>N</i> _{Refl} , <i>N</i> _{Par}	81, 6
<i>R</i> 1, <i>wR</i> 2, <i>S</i>	0.0805, 0.0903, 1.99
$\Delta\rho_{\min}$, $\Delta\rho_{\max}$, /eÅ ⁻³	–3.79, 3.92

Atomic positions and equivalent displacement parameters (Å ²)					
Atom	<i>Wyck</i>	<i>x</i>	<i>y</i>	<i>z</i>	<i>U</i> _{eq}
Fe	2 <i>a</i>	0	0	0	0.0088(6)
S	2 <i>c</i>	0	½	0.255(1)	0.0096(9)

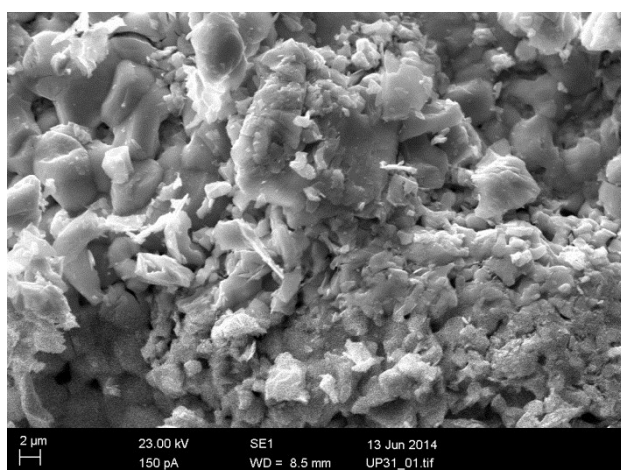
Selected bond lengths (pm) and angles (deg)					
Fe-Fe	260.3(1)		S-Fe-S	110.3(1) ×2	109.1(1) ×4
Fe-S	224.3(1)				

Figure S4. SEM images of the samples showing morphology

a) hydrothermally prepared FeSe



b) conventionally prepared FeSe



c) hydrothermally prepared FeS

