

Electronic Supplementary Information (ESI) for

Enantiopure anion templated synthesis of zeolitic metal–organic framework

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Experimental Section

General information: In this paper, all the chemical reagents were purchased from Energy-Chemical and used without further purification. The chiral compound (*R*)-H₃CIA ((*S*)-H₃CIA) was synthesized from L-(*-*)-lactic acid methyl ester (D-(*-*)-lactic acid methyl ester) and dimethyl 5-hydroxyisophthalate by our group. ¹H NMR (400 MHz) and ¹³C NMR (100 MHz) spectra were recorded in deuterated trifluoroacetic acid (TFA-D, δ 11.50) solutions using a Bruker AVANCE 400 spectrometer. Chemical shifts are reported as δ values in parts per million (ppm) relative to tetramethylsilane (TMS) for all recorded NMR spectra. Elemental analysis was carried out by the analysis center of our institute. FT-IR spectrum was measured as KBr pellets on a Nicolet Magna 750 FT-IR spectrometer in the range of 400~4000cm⁻¹. The powder X-ray diffraction (PXRD) analysis was recorded on a Rigaku Dmax2500 diffractometer with Cu K α radiation (λ =1.54056Å) from 5.00–50.00°. Thermal stability study was performed on a NETSCHZ STA-449C thermoanalyzer with a heating rate of 10°C/min in a range of 30–800°C under nitrogen and atmosphere. Luminescence spectra were recorded on a Shimadzu RF-5301 spectrophotometer.

Measurements of solid CD spectra: The mixture of sample and 100 mg dry KCl powder was well grounded and then pressed into a disk for the CD measurement with a MOS-450 spectropolarimeter.

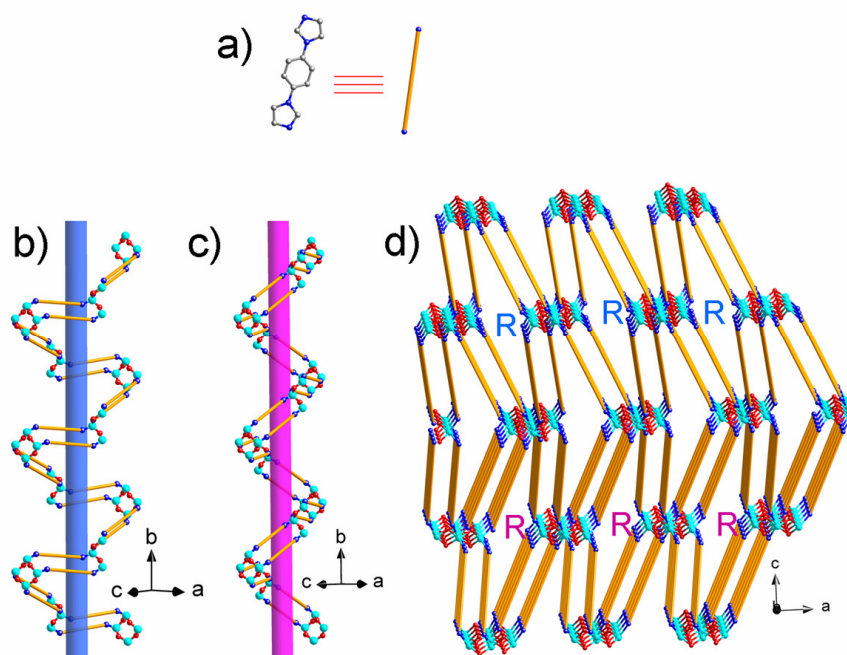


Figure S1. Schematic illustrations of complex **1-D**: a) the 1,4-DIB ligand is simplified as a line; b) the right-handed helical A-chain in complex **1-D**; c) the right-handed helical B-chain in complex **1-D**; d) the 3D framework of **1-D** consisting of two types of right-handed helical chains along b-axis.

Absolute helicity often goes with chirality in some compounds, such as proteins, nucleic acids, carbohelicenes and so on, which is a special form of one-dimensional chirality. Such a phenomenon also exists in complex **1-D** and one of prominent structural features is the presence of two types of helical chains: right-handed helical A-chain and right-handed helical B-chain (Figure S1), dictated by the chirality of induction agent. As depicted in Figure 2b-c, two types of helical chains were all built from $\text{Zn}_4(\mu_2\text{-OH})_4$ units and parallel 1,4-DIB ligands running along the b-axis. These helixes are generated around the crystallographic 4_3 screw axis and contain four $\text{Zn}_4(\mu_2\text{-OH})_4$ units and four pairs of 1,4-DIB ligands per turn, respectively. As a result, these helical chains formed a 3D cationic homochiral framework at a ratio of 1:1.

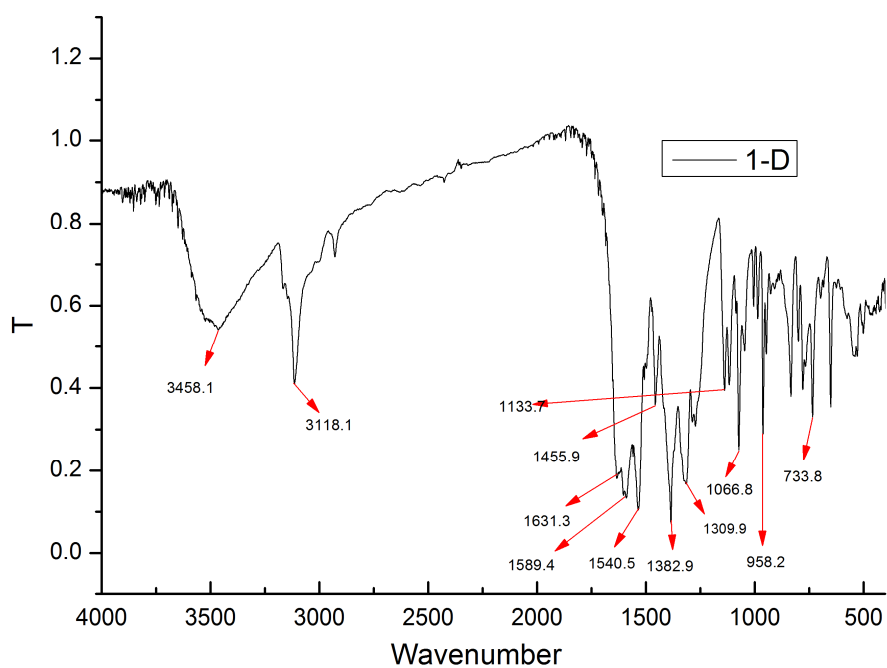


Figure S2. The IR spectra of **1-D**

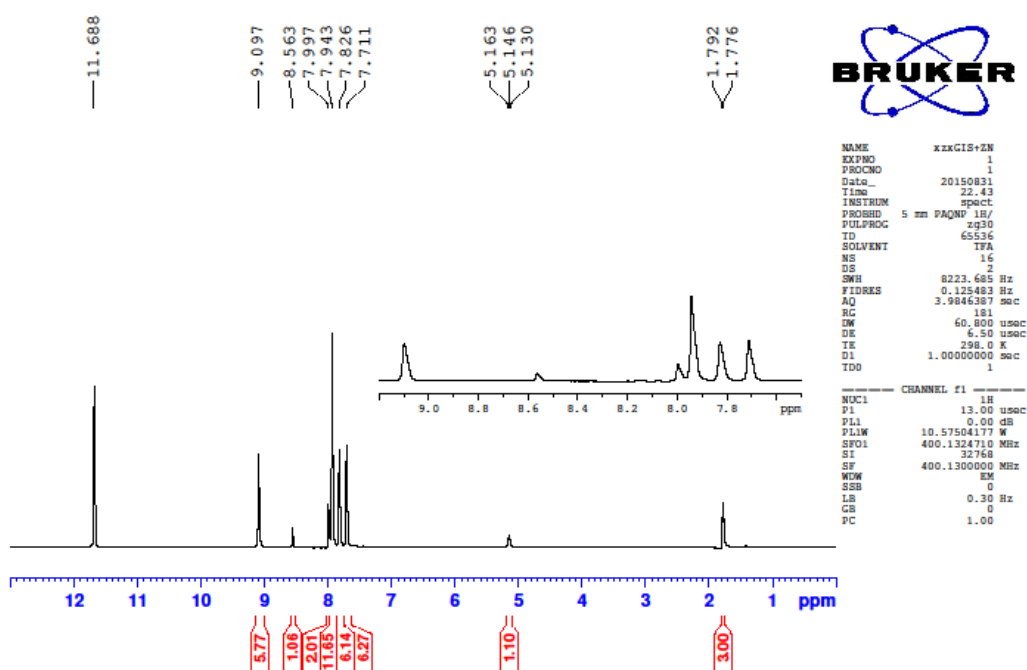


Figure S3. ¹H NMR spectra of **1-D** in TFA-D

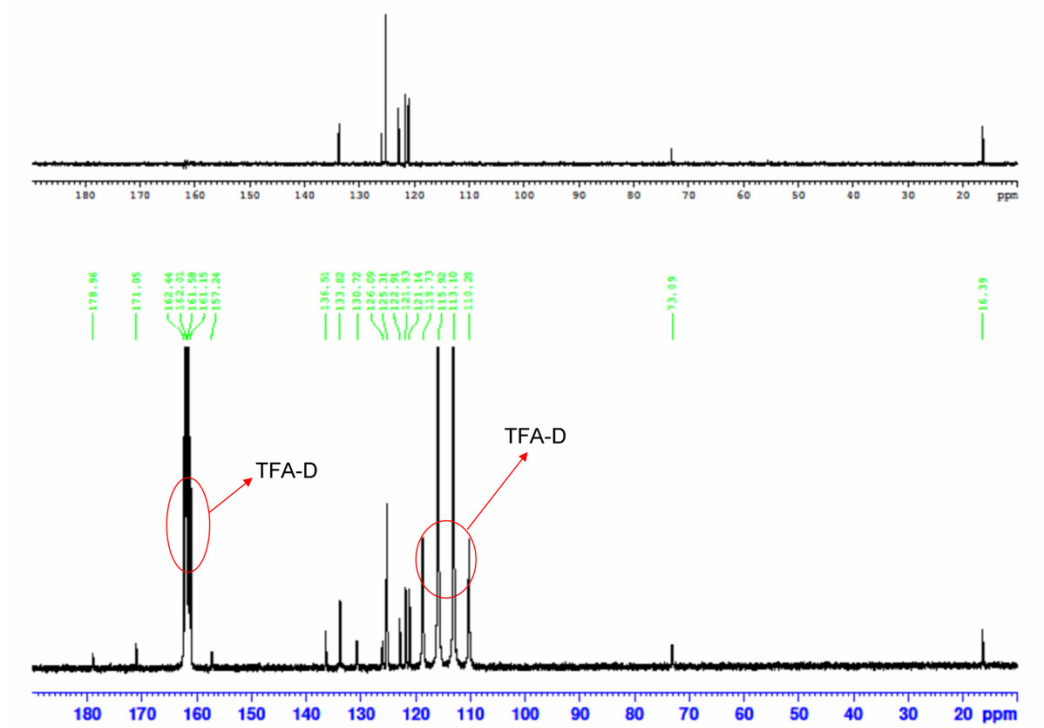


Figure. S4 ^{13}C NMR spectra of **1-D** in TFA-D

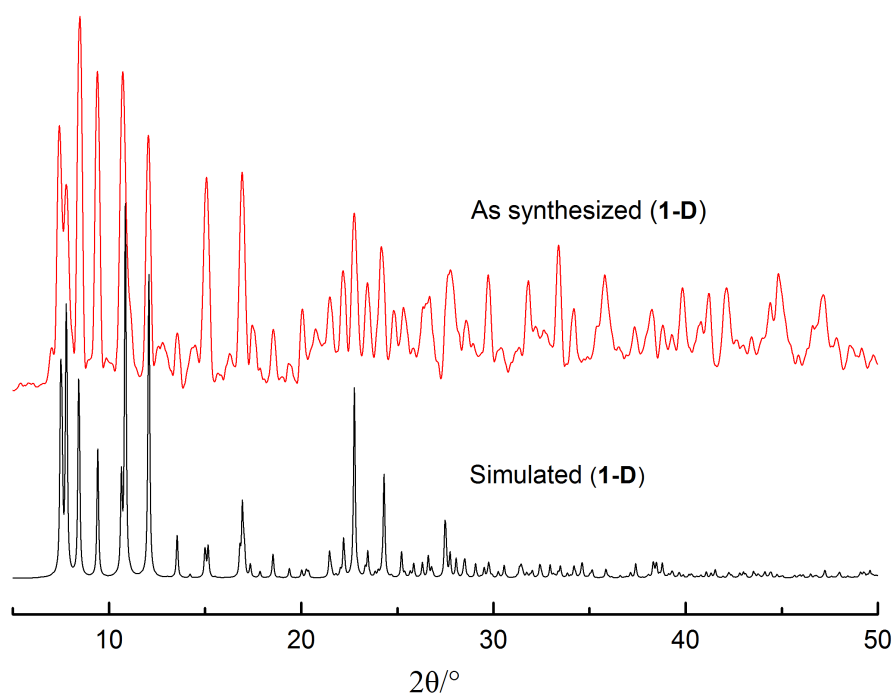


Figure S5. PXRD patterns of **1-D**

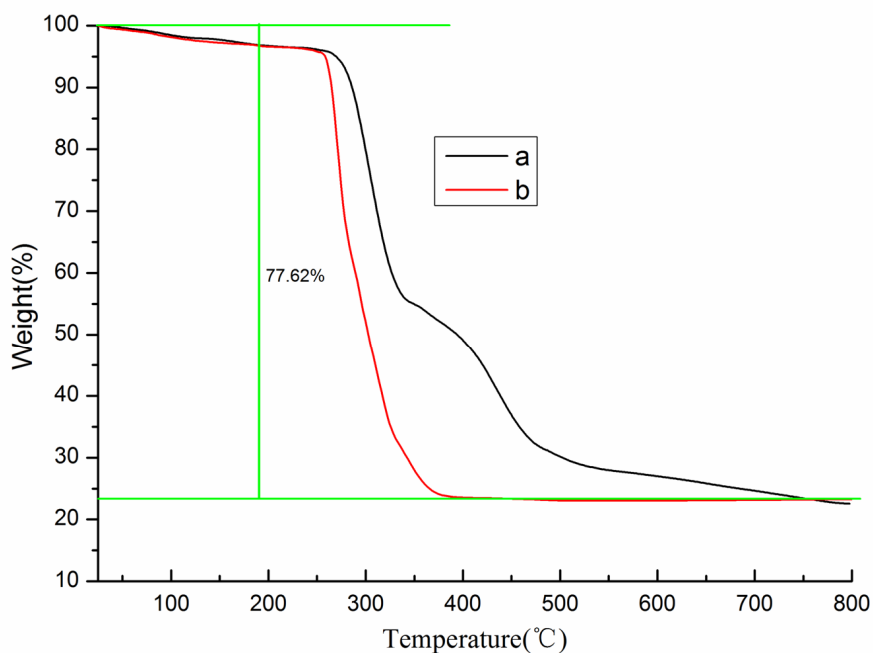


Figure S6. Thermogravimetric diagram of **1-D** under nitrogen (a) and atmosphere (b)

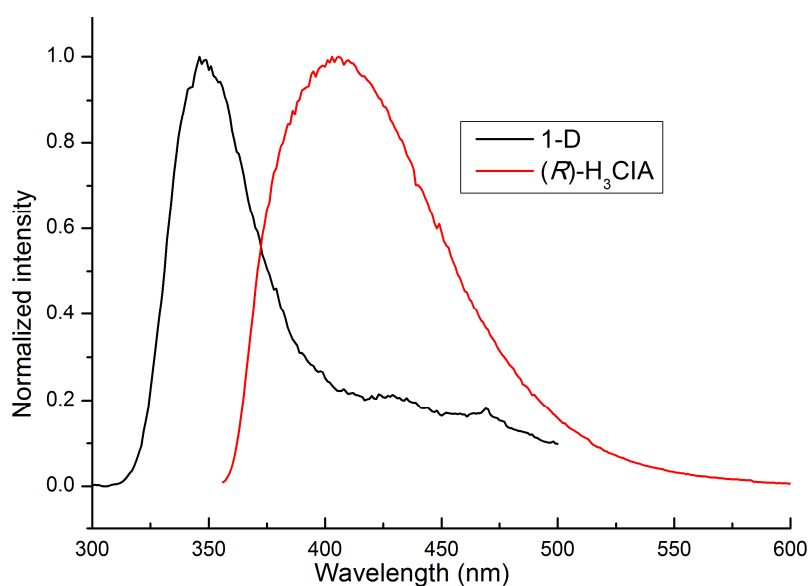


Figure S7. The fluorescent emission spectra of **1-D** and *(R)*-H₃CIA

At room temperature the photoluminescence properties of **1-D** were studied in the solid state. Excitation of the microcrystalline sample of **1-D** at 270 nm produces the intense luminescence with the peak maxima at 349 nm. In order to understand the nature of above emission band better, the photoluminescent property of guest molecule *(S)*-H₃CIA is also tested. As shown in Figure S6, upon excitation at 336 nm, *(S)*-H₃CIA has similar emissions at 404 nm. In comparison to *(R)*-H₃CIA ligand, the

emission maxima in **1-D** has changed and indicates blue shift, which may be attributed to a charge transfer transition between cationic frameworks and anionic guest molecules.¹

References:

1. (a) Z. C. Hu, B. J. Deibert, J. Li, *Chem. Soc. Rev.*, 2014, **43**, 5815; (b) Y. J. Cui, Y. F. Yue, G. D. Qian, B. L. Cheng, *Chem. Rev.*, 2012, **112**, 1126.