

Electronic Supplementary Information
For

A Sensitive Fluorescent Sensor for the Detection of Endogenous Hydroxyl Radical in Living Cells, Bacteria and Direct Imaging of its Ecotoxicity in Living Zebra Fish

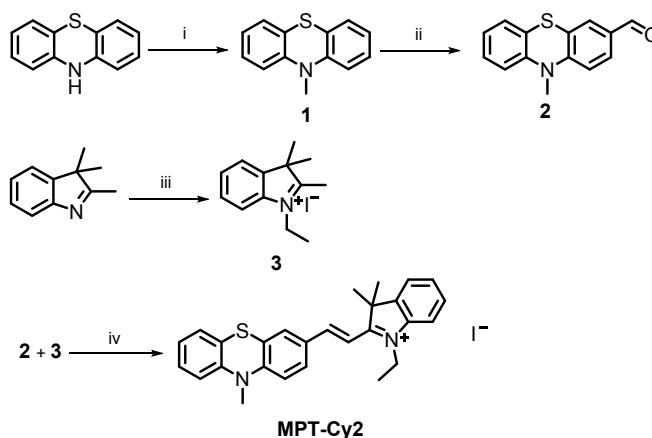
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Dye synthesis

The synthetic route for **MPT-Cy2** (3-[2-(1',3',3'-thimethyl-indolium-2'-yl)vinyl]-10- methyl-phenothiazine) is outlined in Scheme S1.



Scheme 1. Synthetic route of dye **MPT-Cy2**. (i) NaH, DMF, CH₃I, room temperature, 2 h, 78%. (ii) POCl₃, DMF, 95 °C, 12 h, 65 %; (iii) Toluene, CH₃CH₂I, reflux, 10 h, 92%; (iv) Piperidine, Ethanol, reflux, 12 h 72%.

Intermediate products **1**, **2** and **3** were synthesized according to literature procedures.^{1,2}

Synthesis of 3-[2-(1',3',3'-thimethyl-indolium-2'-yl) vinyl]-10- methyl-phenothiazine (MPT-Cy2)

1-Ethyl-2, 3, 3- trimethylindolenium quaternized salt **3** (1.58 g, 5.0 mmol), and aldehyde **2** (2.5 g, 4.5 mmol) were added to a 100 mL flask with 50 mL ethanol, followed by catalytic piperidine (1.0 mL). The resulting mixture was stirred for 12 h under reflux. The residue was recrystallized from ethanol to give the desired product in 70% yield. ¹H-NMR (400 MHz, d₆-DMSO) δ 8.36 (d, *J* = 16.0 Hz, 1H), 8.11 (m, 2H), 7.88 (t, *J* = 7.2 Hz, 2H), 7.59 (m, 3H), 7.26 (m, 2H), 7.09 (m, 3H), 4.69 (d, *J* = 8.0 Hz, 2H), 3.45 (s, 3H), 1.78 (s, 6H), 1.45 (t, 3H). ¹³C-NMR (101 MHz, DMSO) δ181.16, 153.34, 147.52, 144.17, 141.77, 139.93, 134.41, 130.15, 129.89, 129.10, 128.31, 127.43,

124.69, 124.02, 123.43, 118.21, 117.53, 116.35, 114.53, 108.64, 106.11, 73.81, 52.15, 31.17, 26.84, 14.18, 11.42. Mp: 210-212 °C. **HRMS-ESI:** m/z calcd. M^+ for $C_{27}H_{27}N_2S^+$, 411.1889; found, 411.1861.

3-[2-(1',3',3'-thimethyl-indolium-2'-yl)vinyl]-5-oxo-10-methyl-phenothiazine (OMPT-Cy2)

¹H NMR (400 MHz, d₆-DMSO) δ 9.02 (s, 1H), 8.60 (d, $J = 1.2$ Hz, 2H), 8.09 (s, 1H), 7.84 (t, 2H), 7.79 (m, 4H), 7.64 (m, 2H), 7.44 (d, 1H), 4.72 (m, 2H), 3.92 (s, 3H), 1.84 (s, 6H), 1.49 (t, 3H). **¹³C-NMR** (101 MHz, DMSO) δ 206.94, 181.16, 153.34, 150.12, 144.17, 144.06, 140.93, 133.52, 129.65, 128.99, 128.65, 128.32, 127.33, 124.29, 123.26, 123.11, 121.43, 116.18, 115.26, 110.13, 55.48, 52.35, 37.26, 31.15, 26.25, 19.02, 14.12. **HRMS-ESI:** m/z calcd. M^+ for $C_{27}H_{27}N_2OS^+$, 427.1839; found, 427.1856.

Preparation of stock solutions for generation of ROS³⁻⁶

(a) H₂O₂

H₂O₂ was diluted appropriately in water. The concentration of H₂O₂ was determined based on the molar extinction coefficient at 240 nm (43.6 M⁻¹ cm⁻¹). Then, a H₂O₂ stock solution in water was prepared.

(b) •OH

To a solution of H₂O₂ in 100 μ M sodium phosphate buffer at pH 7.4 as a cosolvent, the FeSO₄ solution (10 μ M) was added at room temperature. Then, •OH was generated from Fe²⁺ and H₂O₂ (Fenton reaction).

(c) OCl⁻

NaOCl solution was diluted appropriately in 0.1 M NaOH aq. The concentration of OCl⁻ was determined based on the molar extinction coefficient at 292 nm (350 M⁻¹ cm⁻¹). Then, a OCl⁻ stock solution in 0.1 M NaOH aq. was prepared.

(d) Generation of •O₂⁻

Superoxide (•O₂⁻) was added as solid KO₂.

(e) ¹O₂

A solution of NaMoO₄ was added to a solution of H₂O₂ in 0.1 M sodium phosphate buffer at pH 7.4 as a cosolvent at room temperature.

Live cell incubation

HeLa cells were cultured in DEME (Invitrogen) supplemented with 10% FCS (Invitrogen). One day before imaging cells were seeded into 24-well flatbottomed plates. The next day, the cells were incubated with 8.0 μ M dye for 40 min at 37 °C under 5% CO₂ and washed with phosphate-buffered saline (PBS) three times.

HeLa cells pre-treated with PMA (2 ng mL⁻¹) for 40 min and then incubated with **MPT-Cy2** (8 μ M) plus MitoTracker Deep Red FM (1 μ M) for 30 min at 37 °C. The cells were washed with PBS buffer and the fluorescence images were acquired

Strains and growth conditions

Four bacteria were grown aerobically in standard luria-bertani medium (LB) or anaerobically in

lactate medium (LM) containing 2.0 g/l lactate, 2.0 g/l yeast extract, 12.8 g/l Na₂HPO₄·7H₂O, 3 g/l KH₂PO₄, 0.5 g/l NaCl, and 1.0 g/l NH₄Cl at 33°C (as described previously).⁷ The decabromodiphenyl ether (5 M) was performed to bacteria at 37 °C for 2 h for each assay in triplicate.

TiO₂NP Exposure

Commercially available TiO₂ particles were obtained from Aladdin, China. Stock solutions (0.5 mg mL⁻¹) of the NPs were prepared by sonication in water for 30 min. Zebra fish were continuously exposed to TiO₂NP suspensions and illumination (14 h per day) for 5 days spanning initial embryogenesis through larval development. The photocatalytic experiments were performed using a solar simulator apparatus (Model No. 92190) with a 1600-W xenon lamp. All the data are expressed as mean ± standard deviation (SD) of three independent experiments.

Zebra fish embryos and zebra fish husbandry and exposure

Zebra fish embryos were collected immediately after fertilization and placed into 48-well plates and treated with concentrations of 0.5 mg mL⁻¹ TiO₂NP (5 nm and 40nm) suspensions and illumination. Zebra fish were maintained at 25 °C under a constant temperature of 20 ± 1 °C with a 16:8 h light: dark photoperiod.⁸⁻⁷ The zebrafish embryos were exposed to 0.5 mg/mL concentrations of the 5 and 40 nm diameter nanoparticles in beakers with 20 fish per beaker over a period of 5 days and their survival rate was monitored. The exposure, collection, and analysis experiment was replicated 3 times.

Zebrafish and zebrafish embryos were incubated with the **MPT-Cy2** probe (5 μM) after culture in aerated tap water containing TiO₂NPs of 5 or 40 nm diameter for 30 min. Zebra fish embryos and zebra fish were washed with PBS buffer and the fluorescence images were acquired

Fluorescence imaging

Fluorescence imaging in cells, tumor slices and zebrafish were obtained with spectral confocal multiphoton microscopes (Zeiss LSM 700 confocal laser scanning microscope).

Photostability

MPT-Cy2, and **OMPT-Cy2** were dissolved in DMSO-water (5:5 v) at a concentration of 10.0 μM, respectively. The solutions were irradiated under a 500 W iodine tungsten lamp for 2 h at a distance of 250 mm away. An aqueous solution of sodium nitrite (50.0 g/L) was placed between the samples and the lamp as a heat filter. The photostabilities were expressed in the terms of remaining absorption (%) calculated from the changes of absorbance at the absorption maximum before and after irradiation by iodine tungsten lamp.

Table S1 Spectral properties of **MPT-Cy2** and **OMPT-Cy2** in different solvents

Dyes	solvents	λ_{abs} (nm)	λ_{em} (nm)	$\epsilon \times 10^4$ (nm)	Φ (%)
MPT-Cy2	DMSO	392/ 552	487	2.56	0.0006
	H ₂ O	385/ 545	476	2.55	0.0005
	acetone	383/ 545	475	2.46	0.0007
	methanol	391/ 551	477	3.21	0.0005
	DCM	406/ 580	491	3.10	0.0004
OMPT-Cy2	DMSO	346/ 492	483/ 628	5.67	0.051
	H ₂ O	347/ 455	477/ 593	5.12	0.013
	acetone	347/ 479	476/ 604	5.98	0.037
	methanol	342/ 475	475/ 620	5.10	0.039
	DCM	351/ 479	488/ 623	5.24	0.035

Table S2 Physicochemical Characterization of TiO₂ NPs

Parameters	TiO ₂ NPs-rutile	TiO ₂ NPs- anatase
particle size (nm)	40 nm	5 nm
purity	99.8%	99.8%
surface area (m ² /g)	30	147
zeta potential (mV at pH 7.5)	-32.4	-41.6

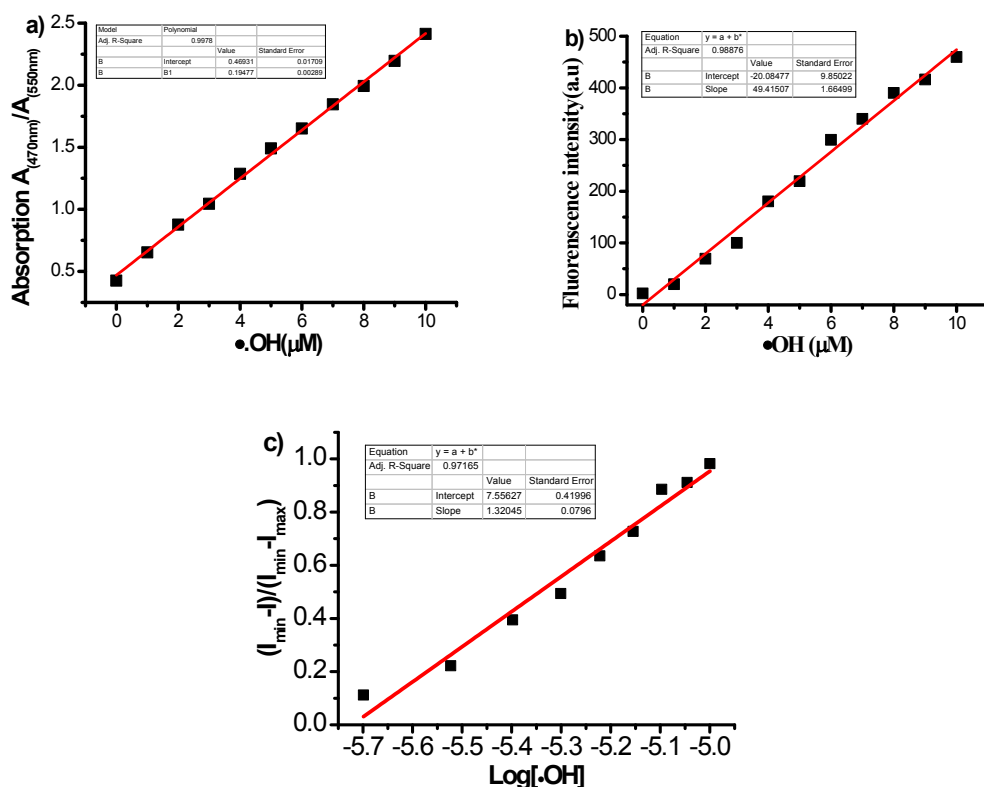


Fig. S1. a) Plot of the absorption intensity ratios at 470 nm and 550nm of **MPT-Cy2** (10 μM) upon addition of $\bullet\text{OH}$ (0–10 μM); b) the fluorescence spectra of **MPT-Cy2** (10 μM) upon the addition of $\bullet\text{OH}$ (0-10 μM); c) the response of the fluorescence signal to changing $\bullet\text{OH}$ concentrations. A linear regression curve was then fitted to these fluorescence intensity data, and the point at which this line crossed the ordinate axis was considered as the

detection limit (1.16×10^{-6} M). $Y = 7.55627 + 1.32045 * X$, $R = 0.97165$. The detection limit was calculated based on the method reported in the previous literature.¹⁰ Conditions: each spectrum was recorded after 3 min in water.

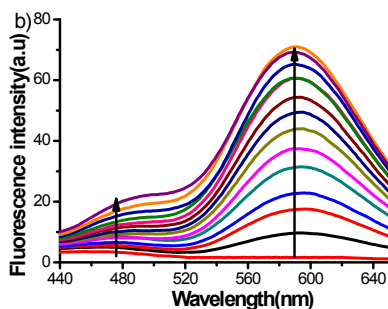


Fig. S2. Changes in the fluorescence emission spectrum of MPT-Cy2 (10 μM) with increases of the •OH concentration (0–30 μM) at 595 nm in water, $\lambda_{\text{ex}} = 350$ nm.

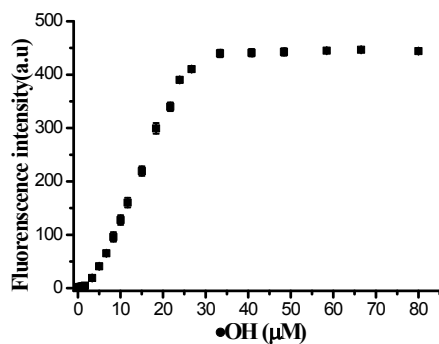


Fig. S3. Changes in the fluorescence spectrum of MPT-Cy2 (30 μM) upon addition of •OH (0-80 μM). Each spectrum was recorded after 3 min in water.

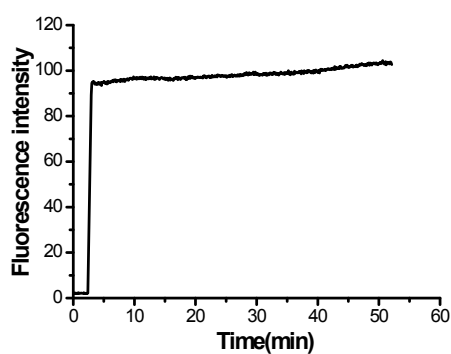


Fig. S4. Time dependent fluorescence intensity changes of MPT-Cy2 (10 μM) at 590 nm in the presence of 50 equiv •OH in water (60min).

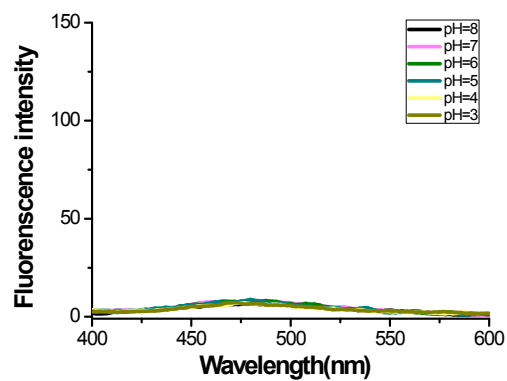


Fig. S5. Influence of pH on fluorescence for MPT-Cy2 (20 μ M) from pH=3 to pH= 8.

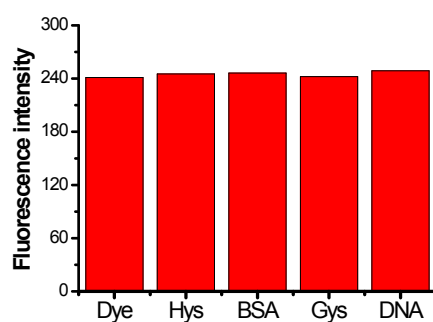


Fig. S6. The fluorescence spectra titration of OMPT-Cy2 (2.0 μ M) in PBS buffer (20.0 mM pH = 7.4) by Hys(25.0 mM); Cys (200 μ M); BSA (200 μ M); DNA (200 μ M) during 120 min.

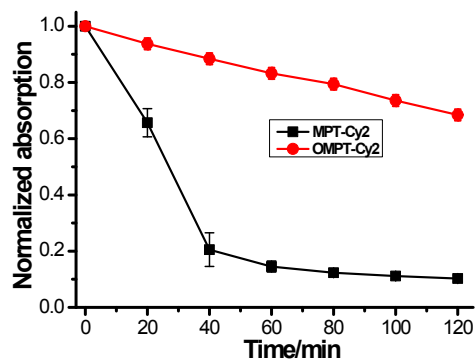


Fig. S7. Photo-fading of dyes (MPT-Cy2 and OMPT-Cy2) in solvent mixture with the ratio of DMSO-water 5: 5 v/v with radiation by a 500 W iodine-tungsten lamp. MPT-Cy2: $\lambda_{\text{abs}} = 520$ nm, OMPT-Cy2: $\lambda_{\text{abs}} = 460$ nm.

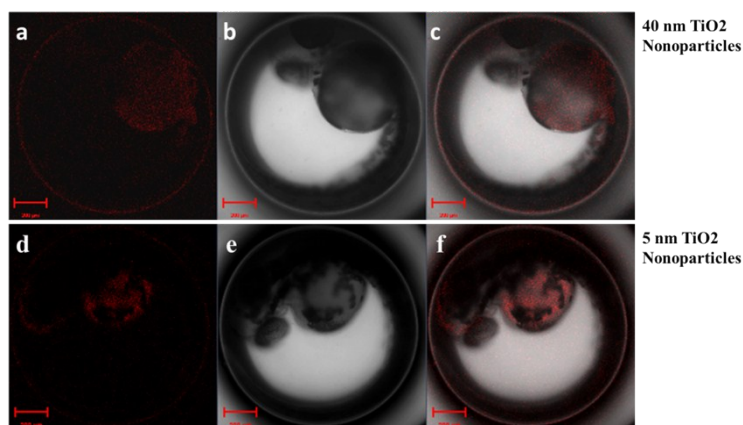


Fig. S8. Confocal fluorescence microscopic images of zebra fish embryos were incubated with the **MPT-Cy2** probe ($5\mu\text{M}$) after culture in aerated tap water containing two different sized nanoparticles (5nm and 40nm) for 40 min. (a-b) Fluorescence image of the embryo with $5\mu\text{M}$ **MPT-Cy2** after culture in aerated tap water containing 40nm sized nanoparticles; (c-d) Fluorescence image of the embryo with $5\mu\text{M}$ **MPT-Cy2** after culture in aerated tap water containing 5nm sized nanoparticles.

In vitro testing: the fluorescent substance, **OMPT-Cy2**, was generated when $\bullet\text{OH}$ was added slowly to the **MPT-Cy2** solution. To confirm the formation of this substance, the partial MS and HPLC spectra of the reaction of **PTE-Cy2** with $\bullet\text{OH}$ are shown in Fig S8 and S9.

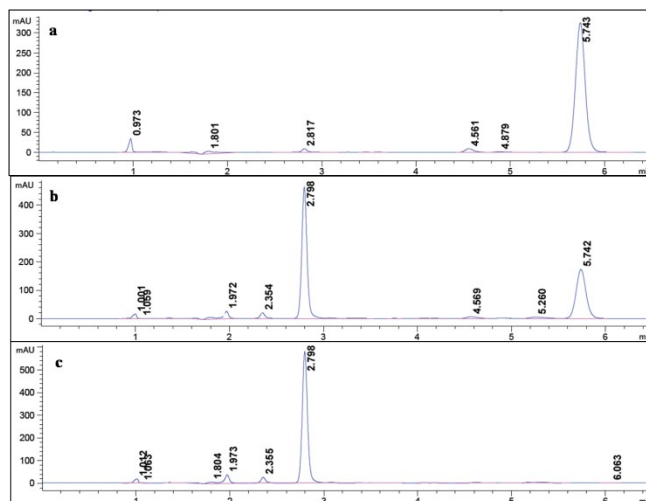


Fig. S9. HPLC chromatograms of probe **MPT-Cy2** after reaction with $\bullet\text{OH}$, a) just **MPT-Cy2** ($10\mu\text{M}$) exist in the solution, b) **MPT-Cy2** ($10\mu\text{M}$) upon addition of $\bullet\text{OH}$ ($6\mu\text{M}$) for 10 min, c) **MPT-Cy2** ($10\mu\text{M}$) upon addition of $\bullet\text{OH}$ ($15\mu\text{M}$) for 10 min.

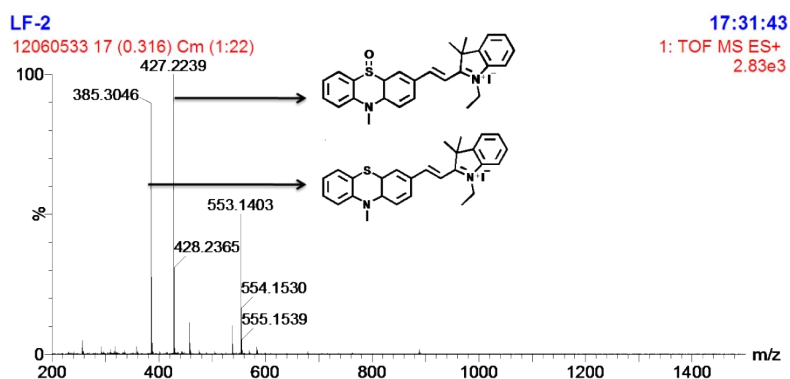


Fig. S10. MS monitoring oxidation of the \bullet OH with MPT-Cy2 process.

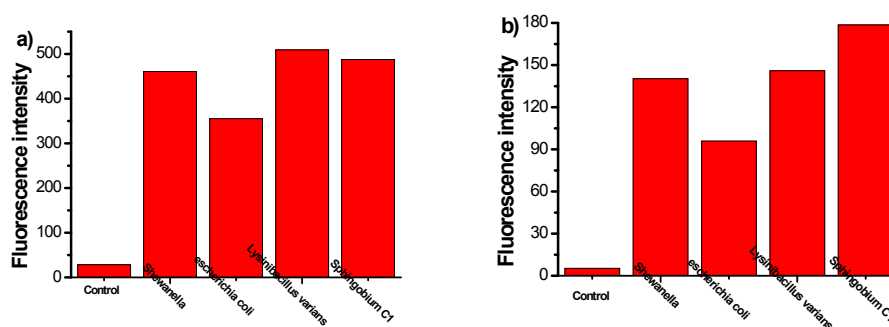


Fig. S11. Changes in the fluorescence intensity of DCFH-DA (a) and MPT-Cy2 (b) ($10\mu\text{M}$) with four species of bacteria in recovered supernatants.

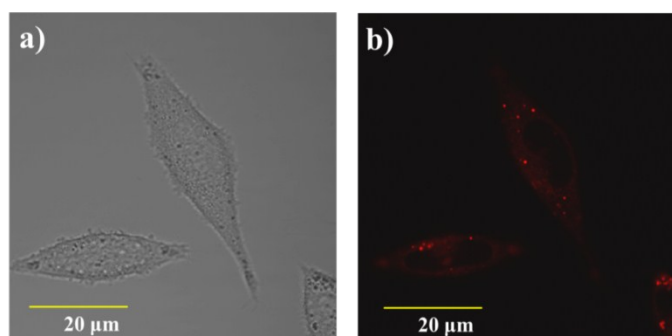


Fig. S12. The sensor MPT-Cy2 (20mM) was loaded into HeLa cells for 2.5 hours without PMA pre-treated.

We monitored the changes of the fluorescence in the cells for 4 hours without PMA pre-treated. There was almost non-fluorescent when the sensor MPT-Cy2 ($10\mu\text{M}$) was loaded into HeLa cells for 40 min. As shown in Fig.S12, there was only a little fluorescence signals came out gradually 2.5 hours later.

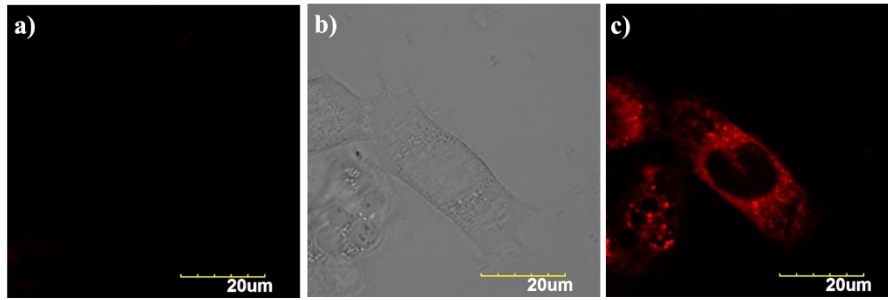


Fig. S13. HeLa cells pre-treated with TEMPOL (5 mM) for 1 h and then incubated with PMA (2 ng mL⁻¹) for another 1 h and finally with MPT-Cy2 (8 μM) for 40 min.

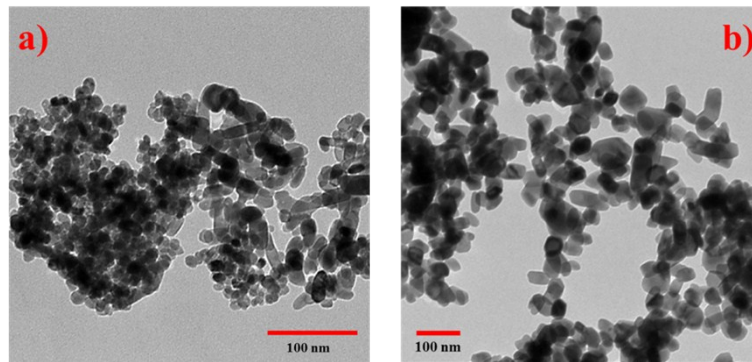
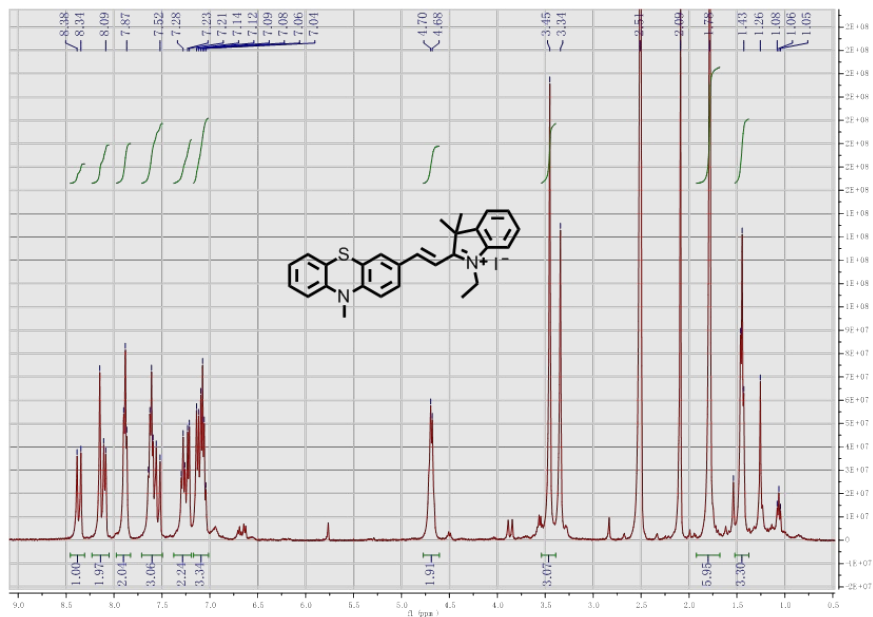


Fig. S14. TEM images of 5 nm TiO₂NPs (a) and 40 nmTiO₂NPs (b).



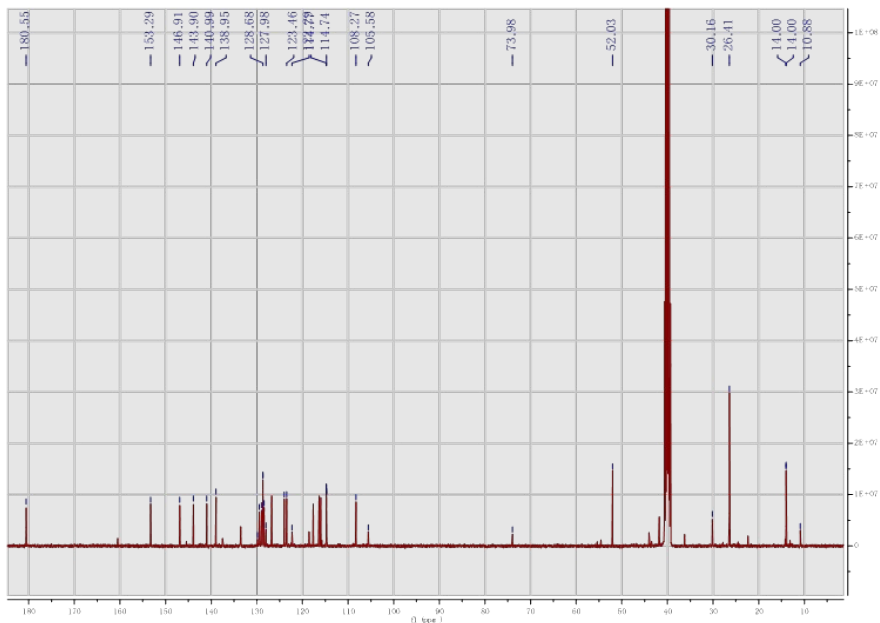
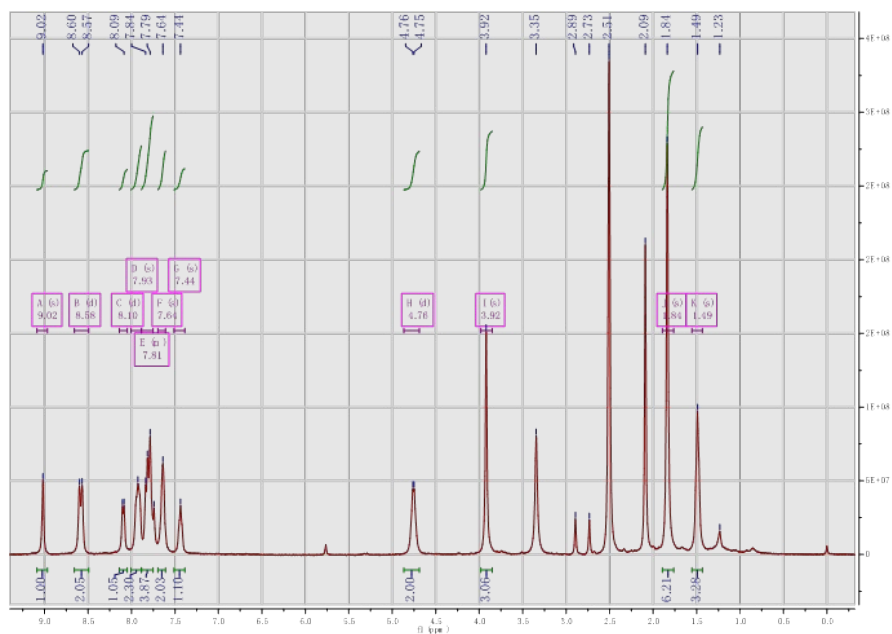


Fig. S15. ^1H and ^{13}C NMR spectra of MPT-Cy2 in DMSO.



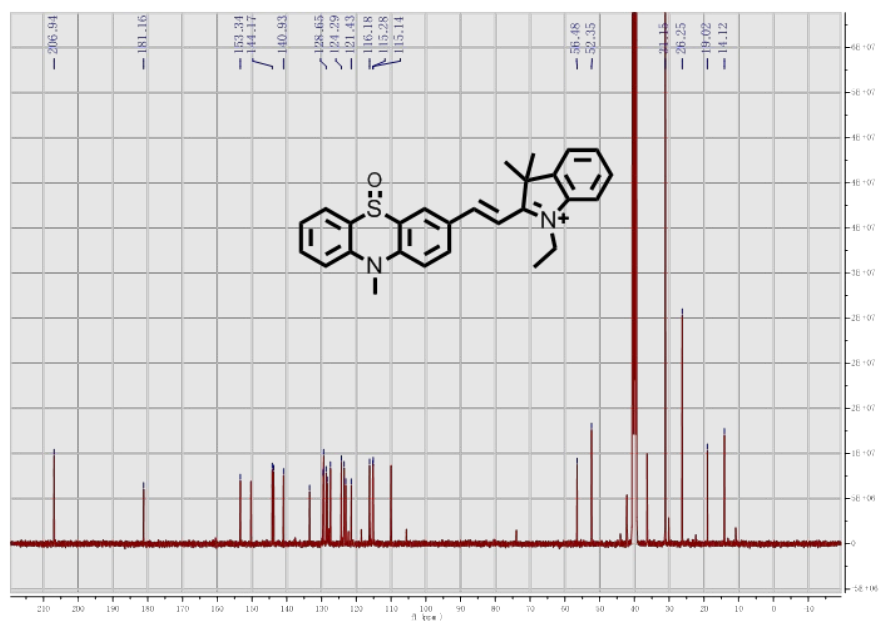


Fig.S16. ^1H and ^{13}C NMR spectra of OMPT-Cy2 in DMSO.

References

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