# Supporting Information

# Concomitant Polymorphs of the *p-tert*-Butylcalix[4]arene Analogue *p-iso*-Propylcalix[4]arene

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#### Single Crystal X-Ray Diffraction (SCXRD)

Single crystal X-ray diffraction data were collected on a Bruker SMART APEX-II CCD area-detector diffractometer equipped with an Oxford Cryosystems Cryostream 700Plus cryostat. A multilayer monochromator with Mo<sub>Ka</sub> radiation ( $\lambda = 0.71073$  Å) from a sealed tube was used.

Data reduction was carried out by means of the standard procedure using the Bruker software package SAINT and the absorption corrections and the correction of other systematic errors were performed using SADABS. The structures were solved by direct methods using SHELXS (2008) and refined using SHELXL (2008). X-Seed was used as the graphical interface for the SHELX program suite. Hydrogen atoms were placed in calculated positions using riding models.

CCDC deposit numbers contain the supplementary crystallographic data for this paper. These data can be obtained free of charge from the Cambridge Crystallographic Data 75 Centre via www.ccdc.cam.ac.uk/data request/cif. CCDC numbers are: 1048933, 1048934 and 1048935.

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Form	Ip	IIp	III <sub>p</sub>
Formula	C40 H48 O4	C40 H48 O4	C40 H48 O4
Formula Weight	592.78	592.78	592.78
Crystal System	Monoclinic	Monoclinic	Monoclinic
Space group	$P2_1/c$ (No. 14)	<i>C</i> 2/ <i>c</i> (No. 15)	$P2_1/n$ (No. 14)
a [Å]	9.4324(11)	24.851(3)	17.6812(15)
<i>b</i> [Å]	15.3488(18)	9.4124(13)	9.5252(8)
<i>c</i> [Å]	23.578(3)	29.215(4)	19.9837(17)
β [°]	98.751(2)	103.211(2)	98.363(2)
V [Å <sup>3</sup> ]	3373.8(7)	6652.8(15)	3329.8(5)
Z	4	8	4
$D_{calc} [g/cm^3]$	1.167	1.184	1.183
Mu(MoKa) [ /mm ]	0.074	0.075	0.074
F(000)	1280	2560	1280
Crystal Size [mm]	0.10 x 0.12 x 0.20	0.06 x 0.10 x 0.23	0.06 x 0.10 x 0.24
Temperature (K)	100(2)	100(2)	100(2)
Radiation [Å] MoKa	0.71073	0.71073	0.71073
Theta Min-Max [°]	1.6, 28.3	1.4, 28.3	1.7, 26.0
Dataset	-11: 6; -20: 16; -30: 28	-33: 32 ; -12: 12 ; -37: 37	-12: 21 ; -11: 11 ; -24: 21
Tot., Uniq. Data, R(int)	14481, 7453, 0.065	36604, 7870, 0.028	17118, 6530, 0.048
Observed data $[I > 2.0 \text{ sigma}(I)]$	3730	6706	4449
N <sub>ref</sub> , N <sub>par</sub>	7453, 409	7870, 445	6530, 419
R, wR2, S	0.0658, 0.1775, 0.95	0.0601, 0.1600, 1.02	0.0529, 0.1363, 1.03
Max. and Av. Shift/Error	0.00, 0.00	0.00, 0.00	0.00, 0.00

Table S1: Tabulated cry	ystallographic data f	for Forms I <sub>p</sub> , II	p and IIIp
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Min. and Max. Resd. Dens. [e/Å <sup>3</sup> ]	-0.28, 0.25	-0.29, 0.50	-0.28, 0.29	

### **Differential Scanning Calorimetry (DSC)**

DSC thermograms were obtained using a TA Instruments Q20 differential scanning calorimeter. Heating rates of 10, 15 and 20 K min<sup>-1</sup> was used in the range 25 to 315 °C. The sample was purged with  $N_2$  at flow rate 50 mL min<sup>-1</sup>. Sample sizes ranged between 4 and 8 mg.



Figure S1. Thermogram of the toluene-*p-iso*-propylcalix[4]arene solvate.





**Figure S2**. Thermograms of the *i*Pc-toluene solvate carried out at various heating rates. (a) 10 K min<sup>-1</sup>, (b) 15 K min<sup>-1</sup>, and (c) 20 K min<sup>-1</sup>.

#### **Powder X-Ray Diffraction (PXRD)**

The diffractograms were recorded with a Bruker D2 PHASER with Lynxeye 1D detector and Nifiltered copper K $\alpha$  radiation (30kV, 10mA generator parameters; restricted by a 1.0 mm divergence slit and a 2.5° Soller collimator) with a 0.02° step width.



**Figure S3**. PXRD profiles of the recrystallized phases from the melt. (a) Form  $III_p$  obtained from the melt of the *i*Pc-toluene solvate at heating rate of 10 K min<sup>-1</sup>, (b) A mixture of forms  $II_p$  and  $III_p$  obtained from the melt of the *i*Pc-toluene solvate at heating rate of 20 K min<sup>-1</sup>. The PXRD for the

recrystallized phases obtained from melting the *i*Pc-toluene solvate at heating rate of 15 K min<sup>-1</sup> matched form  $II_p$  as the quantity of form  $II_p$  was very small.

## **Calculated PXRDs**





**Figure S4**. Calculated PXRD profiles for the three polymorphs of *i*Pc, with form  $III_p$  at the top, form  $II_p$  in the middle, and form  $I_p$  at the bottom.