## **Supporting Information**

## Aluminum zinc oxide nanostructures with customized size and shape by non-aqueous synthesis<sup>†</sup>

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Figure S1: Influence of the solvent on the morphology of AZO nanocrystals – XRD patterns of the particles prepared in the solvent mixtures (BnOH and BnNH<sub>2</sub>)



Figure S2: Influence of the solvent on the morphology of AZO nanocrystals – TEM images of the particles prepared in the solvent mixtures, from left to right 25 vol% BnOH/75 vol% BnNH<sub>2</sub>, 50 vol% BnOH/25 vol% BnOH/25 vol% BnOH<sub>2</sub>, 75 vol% BnOH/25 vol% BnOH<sub>2</sub> (BnOH and BnNH<sub>2</sub>)



Figure S3: TGA thermograms of washed and dried products demonstrating the influence of the volume ratio of benzylamine/benzyl alcohol on the mass of bound organics at the particle surface (note: the larger weight loss of the sample from 25 vol% BnNH<sub>2</sub>/75 vol % BnOH is essentially due to incomplete drying prior to the analysis, visible as larger weight loss at low temperatures)



Figure S4: Influence of the AI precursor on the morphology of AZO nanocrystals synthesized in benzyl alcohol – XRD patterns



Figure S5: Influence of the AI precursor on the morphology of AZO nanocrystals synthesized in benzylamine – XRD patterns



Figure S6: Influence of the precursor concentration of the crystallite size in benzyl alcohol – XRD patterns



Figure S7: Influence of the precursor concnetration of the crystallite size in benzylamine – XRD patterns



Figure S8: Influence of the Al content on the aspect ratio for AZO nanoparticles synthesized in benzylamine (spherical particles) and benzyl alcohol (rod-shaped particles)



Figure S9: Bandgap energies of AZO nanospheres obtained in benzylamine and rod-shaped AZO nanoparticles synthesized in benzyl alcohol for different AI doping levels



Figure S10: Influence of the scale on the aspect ratio of synthesized spherical and rod-shaped AZO nanoparticles (50 g/L, 2.5 mol% AI, 200 °C, 24 h)