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Structural diversity in coordination polymers with semirigid Lewis acidity ligand: structures and properties

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Table S1 Distances(Å) and angles (°) of hydrogen bondings for **2** and **4**[a]

D-H···A	Distance (D···A)	D-H-A	Angle (D-H-A)
		2	
O(1W)-H(1WA)···O(7)#1	3.183(7)	O(1W)-H(1WA)-O(7)#1	156(8)
O(1W)-H(1WB)···O(1W)#2	3.076(6)	O(1W)-H(1WB)-O(1W)#2	142(8)
		4	
O(1W)-H(1WA)···O(7)#3	2.23(12)	O(1W)-H(1WA)-O(7)#3	139(16)
O(1W)-H(1WB)···O(2)#4	1.98(7)	O(1W)-H(1WB)-O(2)#4	160(18)

[a] Symmetry transformations used to generate equivalent atoms: #1 x-1/2, -y+5/2, z+1/2; #2 -x+3/2, y+1/2, -z+3/2; #3 -x, -y+2, -z+1; #4 -x+1/2, -y+3/2, -z+1.

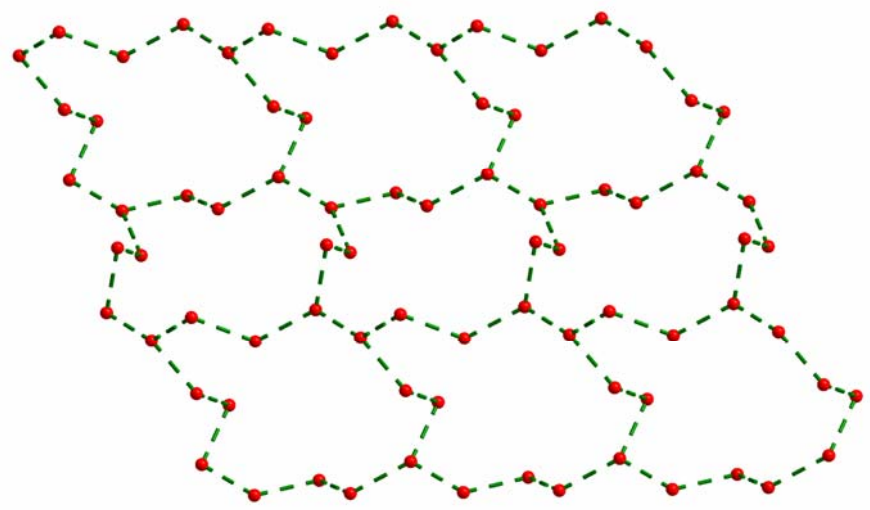

Fig. S1 View of 2D layered water cluster in complex **1**.

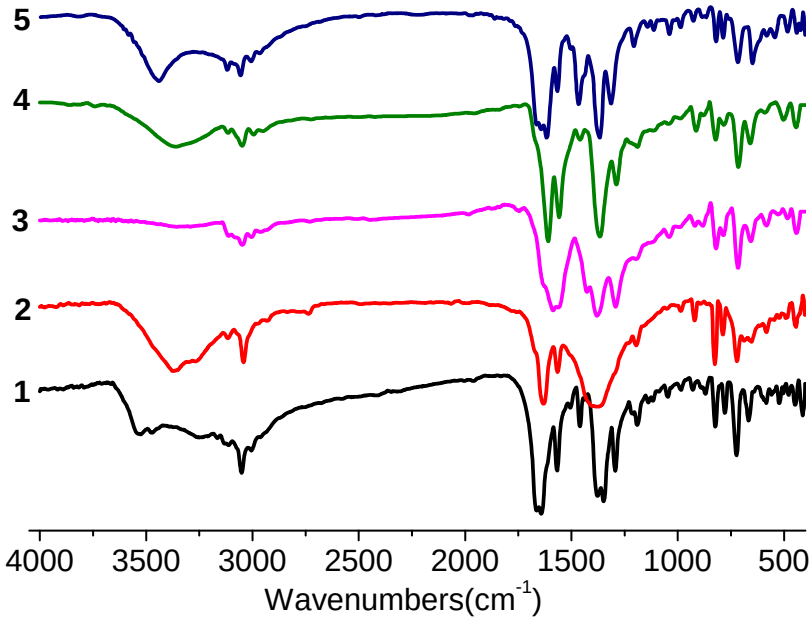

Fig. S2 The IR spectra of complexes **1**–**5**.

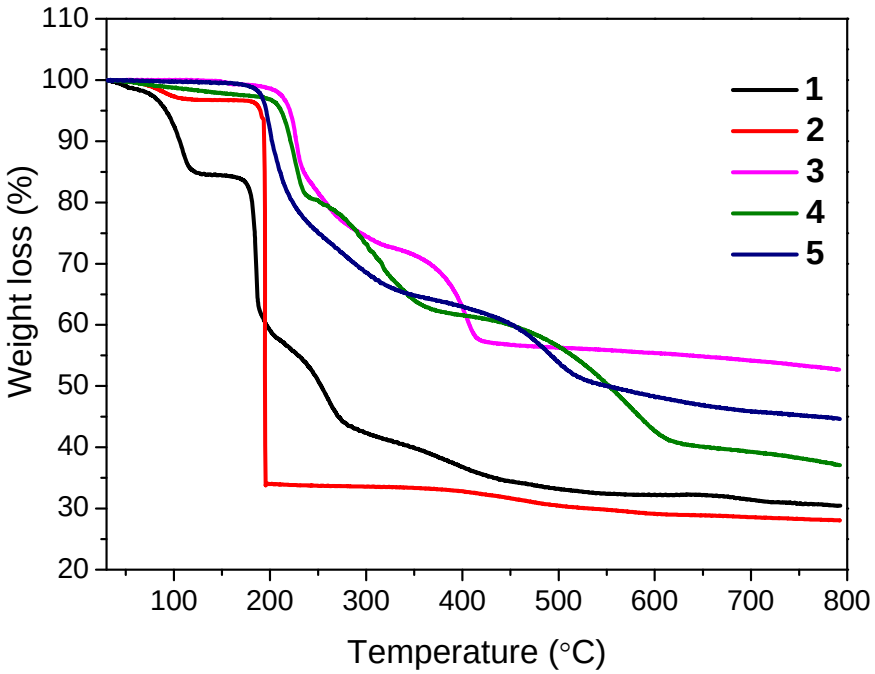

Fig. S3 TGA plots of complexes **1**–**5**.

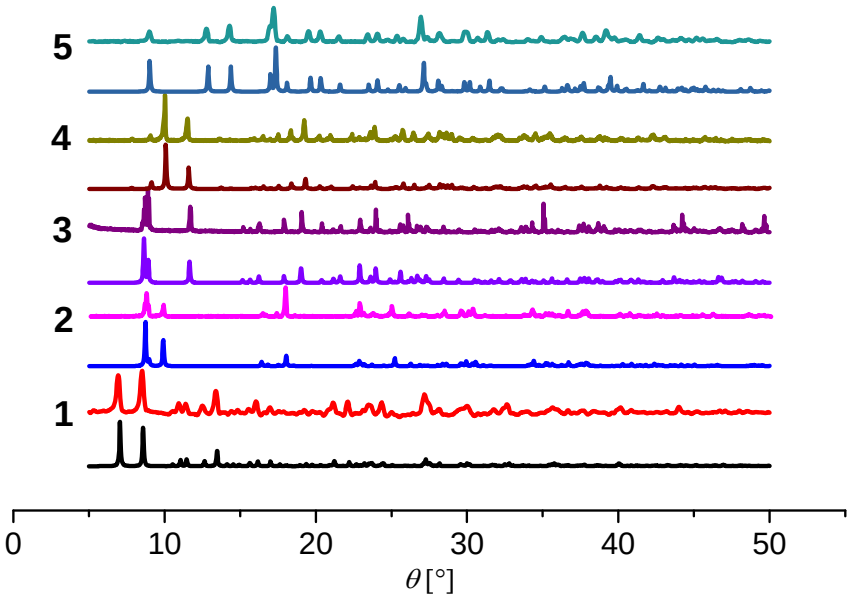

Fig. S4 Powder X-ray diffraction (PXRD) pattern of complexes **1**–**5**. Down: calculated from single crystal data; Top: experimental.