

Solid State Studies of Phthalazinylhydrazones and Triazolophthalazines The Role of Nitro Group

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Synthesis and spectral analysis

2-(2-nitrobenzylidene)-1-(phthalazin-1(2*H*)-ylidene)hydrazinium chloride hemihydrate (**1**): orange rectangular prism crystals; mp 213°C; IR (KBr, cm⁻¹): 3395 (νNH, νOH), 3130-2700 (νCH), 1612 (νC=N, δNH, νCC_{ring}, δOH), 1590 (νC=N_{ring}), 1562 (νCC_{ring}), 1520 (νNO₂, δNH, νC=N, νCC_{ring}), 1505 (νCC_{ring}), 1470 (δNH, δOH), 1440 (νCC_{ring}), 1360 (νNO₂, δNH, δCH), 1280 (νCN_{ring}), 1100 (νCN), 1050 (νNN), 875 (δCH), 745 (δNO₂, δCH), 590 (δNO₂, δCH).

2-(3-nitrobenzylidene)-1-(phthalazin-1(2*H*)-ylidene)hydrazinium chloride (**2**): yellow hexagonal plate crystals; mp 280°C; IR (KBr, cm⁻¹): 3435 (νNH), 3100-2700 (νCH), 1610 (νC=N, δNH, νCC_{ring}), 1590 (νC=N_{ring}, νCC_{ring}), 1530 (νNO₂, δNH, νC=N), 1504 (νCC_{ring}), 1475 (δNH), 1350 (νNO₂, δNH, δCH), 1275 (νCN_{ring}), 1090 (νCN), 1050 (νNN), 890 (δCH), 800 (δCH), 760 (δNO₂, δCH), 730 (δCH), 590 (δNO₂, δCH).

2-(3-nitrobenzylidene)-1-(phthalazin-1(2*H*)-ylidene)hydrazine (**2A**): orange needle crystals; mp 236°C; IR (KBr, cm⁻¹): 3338 (νNH), 3100-2830 (νCH), 1617 (νC=N, δNH, νCC_{ring}), 1600 (νCC_{ring}), 1580 (νC=N_{ring}), 1565 (νCC_{ring}), 1520 (νNO₂, δNH, νC=N), 1480 (δNH), 1435 (νCC_{ring}), 1378 (δCH, δNH), 1355 (νNO₂, δNH, δCH), 1260 (νCN_{ring}), 1082 (νCN), 1007 (νNN), 890 (δCH), 778 (δCH), 735 (δNO₂, δCH), 673 (δCH, δCC), 590 (δNO₂, δCH).

2-(4-nitrobenzylidene)-1-(phthalazin-1(2*H*)-ylidene)hydrazinium chloride methanol solvate (**3**): yellow needle crystals; mp 288°C; IR (KBr, cm⁻¹): 3420 (νNH), 3110-2740 (νCH), 1613 (νC=N, δNH, νCC_{ring}, δOH), 1590 (νC=N_{ring}, νCC_{ring}), 1515 (νNO₂, δNH, νCC_{ring}, νC=N), 1475 (δNH, δOH), 1340 (νNO₂, δNH, δCH), 1275 (νCN_{ring}), 1090 (νCN), 1045 (νNN), 1025 (νCO), 840 (δCH), 760 (δNO₂, δCH), 590 (δNO₂, δCH).

2-(4-nitrobenzylidene)-1-(phthalazin-1(2*H*)-ylidene)hydrazine (**3A**): orange prism crystals; mp 260°C; IR (KBr, cm⁻¹): 3290 (νNH), 3085-2880 (νCH), 1613 (νC=N, δNH, νCC_{ring}), 1590 (νC=N_{ring}, νCC_{ring}), 1567 (νCC_{ring}), 1515 (νNO₂, δNH, νCC_{ring}, νC=N), 1475 (δNH), 1385 (δCH, δNH), 1345 (νNO₂), 1260 (νCN_{ring}), 1105 (νCN, νNN_{ring}), 1014 (νNN), 850 (δCH), 750 (δNO₂, δCH), 590 (δNO₂, δCH).

2-[3-(4-nitrophenyl)prop-2-en-1-ylidene]-1-(phthalazin-1(2*H*)-ylidene)hydrazinium chloride methanol solvate (**4**): yellow prism crystals; mp 150°C; IR (KBr, cm⁻¹): 3470 (νNH, νOH), 3100-2800 (νCH), 1630 (νCC), 1610 (νC=N, δNH, νCC_{ring}), 1590 (νC=N_{ring}), 1560 (νCC_{ring}), 1510 (νNO₂, δNH, νCC_{ring}, νC=N), 1475 (δNH, δOH), 1380 (δNH, δCH), 1340 (νNO₂), 1280 (νCN_{ring}), 1100 (νCN), 1040 (νNN), 1000 (νCO, δCH), 830 (δCH), 750 (δNO₂, δCH), 600 (δNO₂, δCH).

2-(3-phenylprop-2-en-1-ylidene)-1-(phthalazin-1(2*H*)-ylidene)hydrazinium chloride methanol solvate (**5**): yellow prism crystals; mp 190°C; IR (KBr, cm⁻¹): 3235 (νNH, νOH), 3120-2700 (νCH), 1625 (νCC), 1612 (νC=N, δNH, νCC_{ring}), 1595 (νC=N_{ring}, νCC_{ring}), 1500 (νCC_{ring}, νC=N), 1470 (δNH, δOH), 1445 (νCC_{ring}), 1375 (δNH, δCH), 1275 (νCN_{ring}), 1090 (νCN), 1045 (νNN), 1030 (νCO), 980 (δCH), 750 (δCH), 690 (δCH), 590 (δCC, δCH).

1-[(3-phenylprop-2-en-1-ylidene)hydrazinylidene]-1,2-dihydrophthalazine (**5A1**): yellow plate crystals; mp 138°C; IR (KBr, cm⁻¹): 3235 (νNH), 3080-2880 (νCH), 1624 (νCC), 1615 (νC=N, δNH), 1602 (νCC_{ring}), 1592 (νC=N_{ring}), 1575 (νCC_{ring}), 1526 (νCC_{ring}, δNH, νC=N), 1475 (δNH, δCH), 1448 (νCC_{ring}), 1380 (δCH, δNH), 1354 (δCH, δNH), 1245 (νCN_{ring}), 1145 (νCN, νNN_{ring}), 1000 (νNN), 970 (δCH), 745 (δCH), 690 (δCH), 660 (δCC), 590 (δCC, δCH).

1-[(3-phenylprop-2-en-1-ylidene)hydrazinylidene]-1,2-dihydrophthalazine (**5A2**): yellow needle crystals; mp 128°C; IR (KBr, cm⁻¹): 3235 (νNH), 3080-2835 (νCH), 1623 (νCC), 1615 (νC=N, δNH), 1602 (νCC_{ring}), 1590 (νC=N_{ring}), 1574 (νCC_{ring}), 1521 (νCC_{ring}, δNH, νC=N), 1498 (νCC_{ring}), 1473 (δNH, δCH), 1446 (νCC_{ring}), 1377 (δCH, δNH), 1355 (δCH, δNH), 1243 (νCN_{ring}), 1143 (νCN, νNN_{ring}), 1012 (νNN), 968 (δCH), 742 (δCH), 687 (δCH), 662 (δCC), 585 (δCC, δCH).

3-(2-nitrophenyl)[1,2,4]triazolo[3,4-*a*]phthalazine (**1B**): yellow rectangular prism crystals (recrystallized by slow evaporation from MeOH/MeCN); mp 279°C; IR (KBr, cm⁻¹): 3090-2870 (νCH), 1625 (νCC, νC=N), 1600 (νCC), 1574 (νCC), 1525 (νCC, νC=N, νNO₂), 1460 (νCC, νCN), 1382 (δCH), 1355 (νNO₂, νCN), 1220 (νNN, δCH), 1100 (νNN), 960 (δCC, δCH, δCN), 850 (δNCN), 750 (δCH), 710 (δNO₂), 695 (δCNN), 590 (δCC, δCH, δNO₂).

3-(3-nitrophenyl)[1,2,4]triazolo[3,4-*a*]phthalazine (**2B**): yellow needle crystals (recrystallized by slow evaporation from MeOH/DMF); mp 287°C; IR (KBr, cm⁻¹): 3130-2830 (νCH), 1628

(ν_{CC} , $\nu_{\text{C=N}}$), 1600 (ν_{CC}), 1578 (ν_{CC}), 1560 (ν_{CC}), 1540 (ν_{NO_2}), 1526 (ν_{CC}), 1510 ($\nu_{\text{C=N}}$), 1460 (ν_{CC} , ν_{CN}), 1370 (δ_{CH}), 1355 (ν_{NO_2} , ν_{CN}), 1267 (δ_{CH}), 1220 (ν_{NN} , δ_{CH}), 1087 (ν_{NN}), 974 (δ_{CC} , δ_{CH} , δ_{CN}), 873 (δ_{NCN}), 830 (δ_{NO_2} , δ_{CC}), 780 (δ_{CH}), 765 (δ_{CH}), 745 (δ_{CH}), 726 (δ_{CH}), 700 (δ_{NO_2} , δ_{CNN}), 595 (δ_{CC} , δ_{CH} , δ_{NO_2}).

3-(4-nitrophenyl)[1,2,4]triazolo[3,4-*a*]phthalazine (**3B**): orange rectangular plate crystals (recrystallized by slow evaporation from MeOH/MeCN); mp 308°C; IR (KBr, cm^{-1}): 3120-2820 (ν_{CH}), 1623 (ν_{CC} , $\nu_{\text{C=N}}$), 1600 (ν_{CC}), 1560 (ν_{CC}), 1512 (ν_{CC} , $\nu_{\text{C=N}}$, ν_{NO_2}), 1470 (ν_{CC} , ν_{CN}), 1450 (ν_{CC}), 1382 (δ_{CH}), 1345 (ν_{NO_2} , ν_{CN}), 1220 (ν_{NN} , δ_{CH}), 1100 (ν_{NN}), 964 (δ_{CC} , δ_{CH} , δ_{CN}), 855 (δ_{NCN} , δ_{CH}), 756 (δ_{CH}), 704 (δ_{NO_2}), 695 (δ_{CNN}), 590 (δ_{CC} , δ_{CH} , δ_{NO_2}).

3-[2-(4-nitrophenyl)ethenyl][1,2,4]triazolo[3,4-*a*]phthalazine (**4B**): orange plate crystals (recrystallized by slow evaporation from MeOH/MeCN and MeOH/DMF); mp 324°C; IR (KBr, cm^{-1}): 3140-2800 (ν_{CH}), 1622 (ν_{CC} , $\nu_{\text{C=N}}$), 1593 (ν_{CC} , ν_{NO_2}), 1510 (ν_{CC} , $\nu_{\text{C=N}}$, ν_{NO_2}), 1470 (ν_{CC} , ν_{CN}), 1450 (ν_{CC}), 1390 (δ_{CH}), 1340 (ν_{NO_2} , ν_{CN}), 1225 (ν_{NN} , δ_{CH}), 1103 (ν_{NN}), 974 (δ_{CC} , δ_{CH} , δ_{CN}), 840 (δ_{NCN} , δ_{CH}), 746 (δ_{CH}), 705 (δ_{NO_2}), 694 (δ_{CNN}), 590 (δ_{CC} , δ_{CH} , δ_{NO_2}).

3-[2-phenylethenyl][1,2,4]triazolo[3,4-*a*]phthalazine (**5B**): pale yellow prism crystals (recrystallized by slow evaporation from MeOH/DMF); mp 197°C; IR (KBr, cm^{-1}): 3100-2820 (ν_{CH}), 1638 (ν_{CC}), 1626 (ν_{CC} , $\nu_{\text{C=N}}$), 1596 (ν_{CC}), 1576 (ν_{CC}), 1558 (ν_{CC}), 1523 (ν_{CC} , $\nu_{\text{C=N}}$), 1505 (ν_{CC}), 1470 (ν_{CC} , ν_{CN}), 1450 (ν_{CC}), 1385 (δ_{CH}), 1350 (ν_{CN}), 1205 (ν_{NN} , δ_{CH}), 1115 (ν_{NN}), 963 (δ_{CC} , δ_{CH} , δ_{CN}), 845 (δ_{NCN} , δ_{CH}), 760 (δ_{CH}), 754 (δ_{CH}), 696 (δ_{CNN}), 590 (δ_{CC} , δ_{CH}).

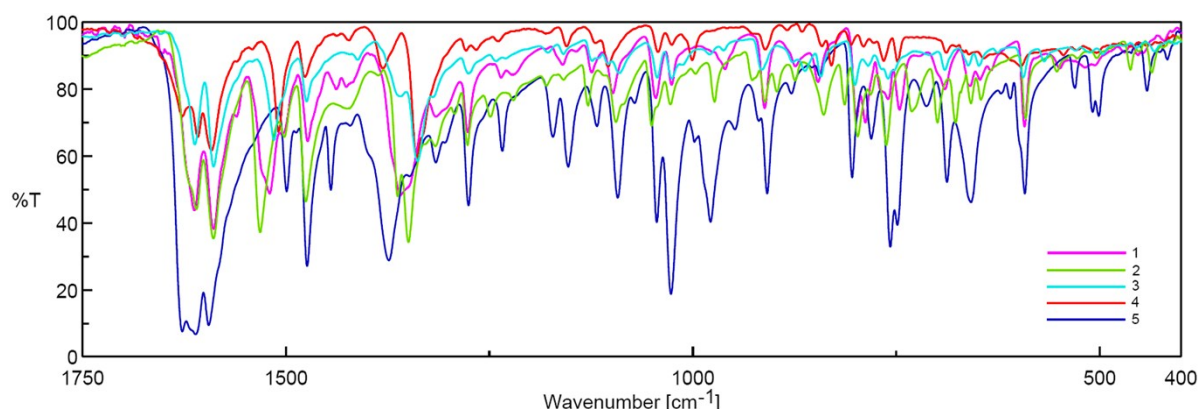


Fig. S1. The partial IR spectra of hydrazonium chlorides (**1-5**).

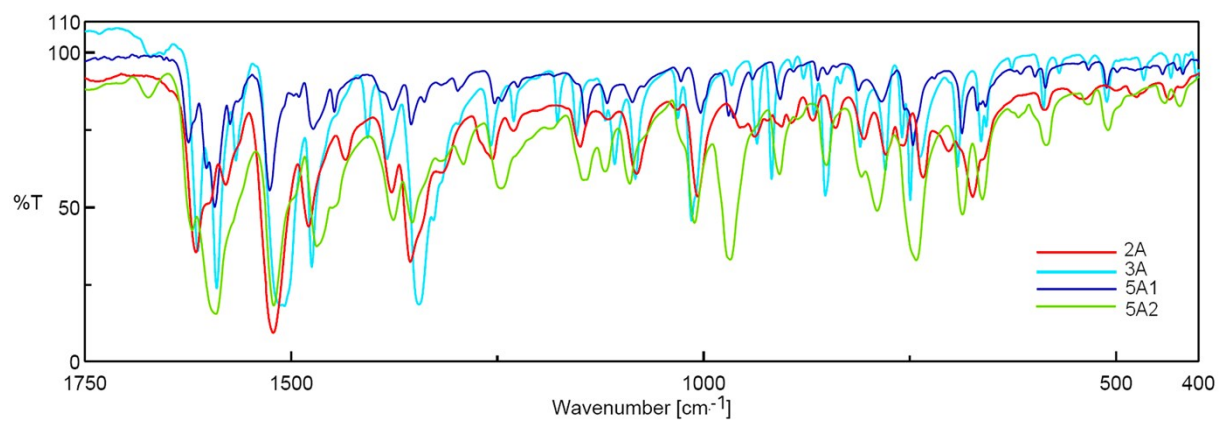


Fig. S2. The partial IR spectra of phthalazinylhydrazones (**2A**, **3A**, **5A1**, **5A2**).

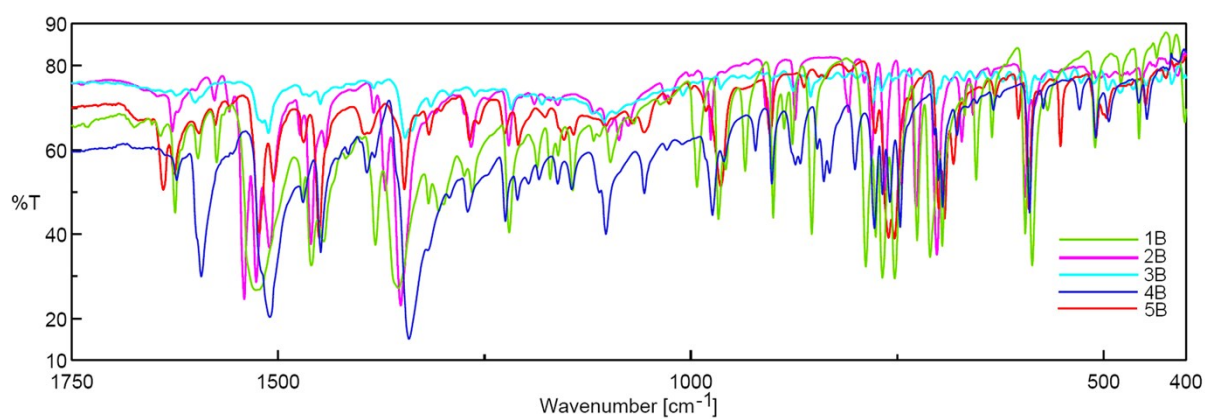


Fig. S3. The partial IR spectra of triazolophthalazines (**1B-5B**).

Table S1. Some characteristic vibration frequencies of phthalazinylhydrazones.

Assignment	1	2	2A	3	3A	4	5	5A1	5A2
vNH, vOH	3395	3435	3338	3420	3290	3470	3235	3235	3235
vCH	3090, 3020, 2765	3086, 3020, 2968, 2920, 2872, 2723	3082, 2954, 2854	3084, 3020, 2922, 2856, 2767	3074, 3024, 2925, 2850	3100, 3043, 2924, 2831	3106, 3039, 2928, 2822, 2757	3077, 3055, 3028, 2950	3078, 3025, 2993, 2950, 2855
vCC						1630	1625	1624	1623
vC=N, δ NH	1612	1610	1617	1613	1613	1610	1612	1615	1615
vCC _{ring}			1600					1602	1602
vC=N _{ring}	1590	1590	1580	1590	1590	1590	1595	1592	1590
vCC _{ring}	1562		1565		1567	1560		1575	1574
vC=N _{ring} , vCC _{ring}	1520	1530	1520	1515	1515	1510	1500	1526	1521
vCC _{ring}	1505	1504							1498
δ NH	1470	1475	1480	1475	1475	1475	1470	1475	1473
vCC _{ring}	1440		1435				1445	1448	1446
δ NH, δ CH			1378		1385	1380	1375	1380	1377
vNO ₂ , δ NH, δ CH	1360	1350	1355	1340	1345	1340		1354	1355
vCN _{ring}	1280	1275	1260	1275	1260	1280	1275	1245	1243
vCN	1100	1090	1082	1090	1105	1100	1090	1145	1143
vNN	1050	1050	1007	1045	1014	1040	1045	1000	1012
δ NO ₂	745	760	735	760	750	750			

Table S2. Some characteristic vibration frequencies of triazolophthalazines.

Assignment	1B	2B	3B	4B	5B
vCH	3085, 3048, 3030, 2874	3104, 3057, 3018, 2960, 2886	3098, 3076, 2925, 2850	3107, 3062, 2929, 2836	3057, 2036, 3000, 2920, 2848
vCC					1638
vCC, vC=N	1625	1628	1623	1622	1626
vCC	1600, 1574	1600, 1578, 1560	1600, 1560	1593	1596, 1576, 1558
vNO ₂	1525	1540	1512	1510	
vC=N	1525	1510	1512	1510	1523
vCC, vCN	1460	1460	1470	1470	1470
vNO ₂ , vCN	1355	1355	1345	1340	1350
vNN, δ CH	1220	1220	1220	1225	1205
vNN	1100	1087	1100	1103	1115
δ NCN, δ CH	850	873	855	840	845

Structural studies

The description of X-ray measurement of **5B**

The crystals of the compound **5B** were well shaped and gave discrete diffraction. All reflections were in the places of the frames, in which they should appear, i.e. the placement of the reflections was in agreement with those calculated from the orientation matrix. There were no observable reflections beyond the places calculated from the orientation matrix. The reflections were well shaped, i.e. shapes of the reflections fit well the Gaussian curve with a 2-dimensional domain. Solution of the structure in the statistically best space group ($P2_1/c$, $a = 9.6808(12)$ Å, $b = 5.1248(7)$ Å, $c = 13.8869(18)$ Å, $\alpha = 90^\circ$, $\beta = 100.306(11)^\circ$, $\gamma = 90^\circ$, 100.0(1) K) leads to the model (Fig. S4 (a)) with overlapped 2-phenylethenyl and triazolophthalazine moieties by symmetry (the symmetry center located at the gravity center of the molecule). The domains can be separated in the initial model (Fig. S4 (b)) but the refinement leads to the abnormal interatomic distances, impossible isotropic displacement parameters, unacceptable goodness of fit and $R1 = 0.7288$. Because the wrongly interpreted statistic may lead to introduction of the non-existing symmetry centre (the internal molecular symmetry may cause the false statistics, because the molecularly equivalent groups of atoms may not be connected by crystallographic symmetry), as well as, because the symmetry centre forces the occupation factors of both domains to be equal 0.5, the structure was solved and refined in the maximal non-isomorphic subgroup, which does not possess this ambiguous symmetry element (space group Pc). Solution of the structure in the Pc space group leads to the model (Fig. S4 (c)) still possessing the overlapped 2-phenylethenyl and triazolophthalazine moieties. The domains can be separated in the initial model (Fig. S4 (d)) but the refinement leads to the abnormal interatomic distances, impossible isotropic displacement parameters, unacceptable goodness of fit and $R1 = 0.6528$. The slight increase of the structure quality originates mainly from the unequal occupation of both domain (0.63843:0.36157), what proves that the $P2_1/c$ space group was incorrect (the internal molecular symmetry leads to false statistics, because the molecularly equivalent groups of atoms were not connected by crystallographic symmetry). The introduction of the anisotropy to the displacement parameters of the Pc space group refinement leads to the enormous prolation of the anisotropic displacement ellipsoids. The two-domain disorder model (with overlapped 2-phenylethenyl and triazolophthalazine moieties of the opposite domains) was modified and two further domains were introduced to the refinement (each domain was

separated into the two domains slightly relocated along the major axis of spheroid). The refinement of four-domain model was stable only with the isotropic displacement parameters. The close inspection of the atoms positions of this model shows some deviation from the Pc space group symmetry. Thus the measurement data were reduced in the triclinic crystal system. The unit cell parameters ($a = 5.1248(7) \text{ \AA}$, $b = 9.6808(12) \text{ \AA}$, $c = 13.8869(18) \text{ \AA}$, $\alpha = 100.306(11)^\circ$, $\beta = 90.030(9)^\circ$, $\gamma = 90.033(8)^\circ$, $100.0(1) \text{ K}$) show some deviations from the values required for monoclinic crystal system symmetry. The refinement of the structural model of two molecules, of which each was disordered unequally over two positions (with the anisotropy introduced to the displacement parameters of all non-hydrogen atoms), in the $P1$ space group, leads to the stable convergence with the $R1 = 0.2579$. The quality of the model is unsatisfactory for the full structural description and analysis of the structural parameters, but it allows to confirm that the $P1$ space group is the correct one and that studied compound is the solvent-free 3-[2-phenylethenyl][1,2,4]triazolo[3,4-a]phthalazine.

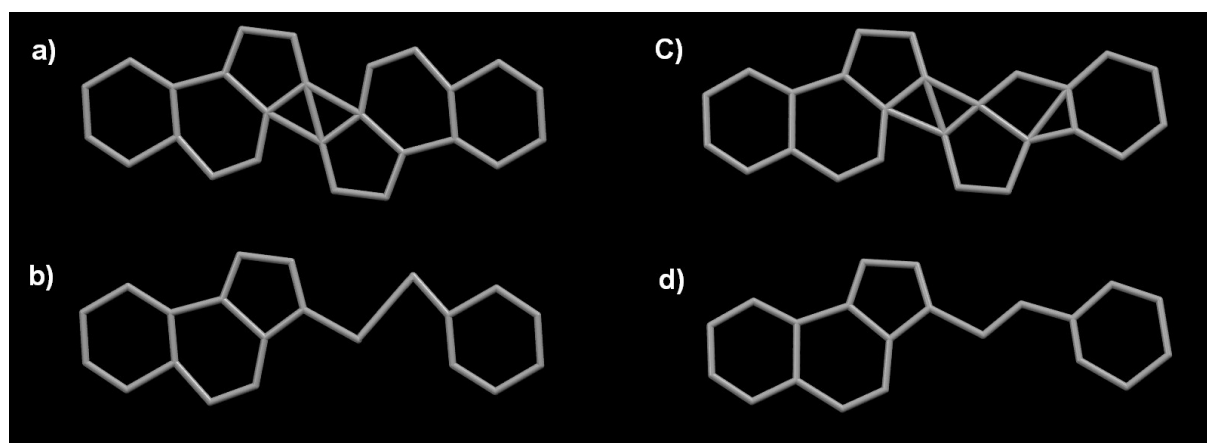


Fig. S4. The initial model of the **5B** after the solution in a) the $P2_1/c$ space group, c) the Pc space group; the separated domain of the **5B** in b) the $P2_1/c$ space group, d) the Pc space group.

Table S3. Selected structural data for compounds **1-5** and **2A, 3A, 5A1, 5A2** [Å].

	N2-C2	N1-N2	C1-N1	C1-C4	C1-N3	N3-N4	N4-C9	N5-O1	N5-O2
1	1.287(3)	1.370(2)	1.333(2)	1.446(2)	1.335(2)	1.380(2)	1.274(3)	1.190(3)/ 1.300(9)	1.243(5)/ 1.145(9)
	1.289(3)	1.370(2)	1.337(2)	1.445(3)	1.337(2)	1.383(2)	1.283(2)	1.224(2)	1.213(2)
2	1.2946(15)	1.3742(13)	1.3371(15)	1.4471(15)	1.3360(15)	1.3778(13)	1.2815(15)	1.2289(14)	1.2273(14)
2A	1.2927(14)	1.3670(13)	1.3685(14)	1.4601(15)	1.3080(14)	1.3925(13)	1.2839(15)	1.2264(13)	1.2257(12)
3	1.2962(16)	1.3735(14)	1.3337(15)	1.4468(16)	1.3375(15)	1.3827(13)	1.2821(16)	1.2239(16)	1.2293(15)
3A	1.2928(13)	1.3685(11)	1.3660(12)	1.4582(13)	1.3125(13)	1.3889(11)	1.2845(13)	1.2206(12)	1.2272(12)
4	1.2972(16)	1.3745(15)	1.3362(15)	1.4447(16)	1.3406(16)	1.3801(13)	1.2874(16)	1.2269(14)	1.2286(14)
5	1.2944(16)	1.3734(13)	1.3354(14)	1.4396(17)	1.3386(14)	1.3813(14)	1.2889(15)	-	-
5A1	1.2914(14)	1.3639(12)	1.3744(13)	1.4597(14)	1.3060(13)	1.3985(12)	1.2853(14)	-	-
5A2	1.2936(13)	1.3578(11)	1.3724(12)	1.4610(13)	1.3092(13)	1.3960(11)	1.2855(13)	-	-

Table S4. Hydrogen bonds in studied compounds [Å, °].

	D-H	H...A	D...A	D-H...A
Compound 1				
N1—H1N...O99	0.94	1.83	2.684(2)	149.6
N3—H3N...Cl1 ⁱ	0.87	2.25	3.1000(16)	163.9
N21—H21N...Cl1 ⁱⁱ	0.94	2.21	3.0415(16)	146.1
N23—H23N...Cl21	0.98	2.15	3.0988(15)	160.6
O99—H99O...Cl1	0.91	2.21	3.1149(17)	172.2
O99—H99P...Cl21	0.94	2.16	3.0897(16)	171.3
C8—H8...Cl1 ⁱ	0.95	2.71	3.633(2)	163.9
C9—H9...O1	0.95	2.25	2.828(3)	118.6
C9—H9...Cl1 ⁱ	0.95	2.67	3.4814(19)	143.4
C15—H15...O99	0.95	2.53	3.461(3)	167.0
C22—H22...O99 ⁱⁱⁱ	0.95	2.51	3.448(2)	171.6
C29—H29...O21	0.95	2.43	2.883(2)	109.2
C29—H29...Cl21	0.95	2.64	3.4260(18)	140.7
C35—H35...Cl1 ⁱⁱ	0.95	2.75	3.6937(19)	175.0

Compound **1B**

C6—H6•••O1 ^{iv}	0.95	2.54	3.4677(14)	164.9
C8—H8•••O1 ^v	0.95	2.60	3.5221(14)	164.1

Compound **2**

N1—H1N•••C11 ^{vi}	0.88	2.39	3.1825(10)	150.9
N3—H3N•••C11	0.89	2.26	3.1322(10)	164.1
C2—H2•••O1 ^{vi}	0.95	2.44	3.3101(14)	152.1
C6—H6•••O1 ^{vii}	0.95	2.58	3.4432(14)	151.0
C7—H7•••O2 ^{vii}	0.95	2.54	3.2801(15)	134.8
C8—H8•••C11	0.95	2.63	3.5567(12)	164.4
C9—H9•••C11	0.95	2.81	3.6026(12)	142.0
C15—H15•••C11 ^{vi}	0.95	2.63	3.5599(12)	166.8

Compound **2A**

N1—H1N•••N2 ^{viii}	0.89	2.25	2.9500(13)	135.1
C2—H2•••O1 ^{viii}	0.95	2.59	3.4019(14)	143.6
C7—H7•••O2 ^{ix}	0.95	2.57	3.4869(14)	162.4

Compound **2B**

C2—H2•••N4 ^x	0.93	2.57	3.481(4)	165.7
C5—H5•••N3 ^x	0.93	2.55	3.459(4)	166.8
C6—H6•••O22 ⁱⁱ	0.93	2.52	3.223(4)	132.7
C15—H15•••O1 ^x	0.93	2.55	3.178(4)	125.1
C15—H15•••N2	0.93	2.36	2.998(4)	125.8
C22—H22•••N24 ^{xi}	0.93	2.48	3.406(3)	171.1
C25—H25•••N23 ^{xi}	0.93	2.57	3.466(4)	162.3
C35—H35•••N22	0.93	2.36	2.996(4)	125.8

Compound **3**

N1—H1N•••C11 ^{xii}	0.89	2.25	3.0893(10)	158.5
N3—H3N•••C11	0.90	2.20	3.0831(11)	166.8
O91—H91O•••C11	0.91	2.20	3.114(2)	175.0
C2—H2•••O91 ^{xiii}	0.95	2.41	3.336(2)	166.5
C8—H8•••C11	0.95	2.67	3.5818(12)	161.8
C9—H9•••C11	0.95	2.66	3.4885(12)	146.4
C11—H11•••C11 ^{xii}	0.95	2.79	3.5859(12)	141.8
C14—H14•••O91 ^{xiv}	0.95	2.41	3.299(3)	156.3
C6—H6•••O1 ^{xv}	0.95	2.93	3.8593(16)	166.7
C6—H6•••O2 ^{xv}	0.95	2.62	3.2684(15)	125.7
C5—H5•••O2 ^{xv}	0.95	2.69	3.3016(15)	122.4

Compound 3A				
N1—H1N•••N2 ^{xvi}	0.92	2.20	2.9391(12)	137.7
C6—H6•••O2 ^{xvii}	0.95	2.48	3.1139(13)	124.2
Compound 3B				
C7—H7•••O2 ^{xviii}	0.95	2.48	3.3266(18)	148.2
C11—H11•••N2	0.95	2.48	3.0735(19)	120.8
C12—H12•••N4 ^{xix}	0.95	2.39	3.2515(18)	151.2
C15—H15•••O1 ^x	0.95	2.50	3.3808(17)	154.7
Compound 4				
N1—H1N•••O91	0.90	1.87	2.7311(13)	159.5
N3—H3N•••C11	0.92	2.27	3.1638(11)	164.4
O91—H91O•••C11 ^{xx}	0.88	2.18	3.0540(9)	174.4
C5—H5•••O1 ^{xxi}	0.95	2.53	3.0812(16)	117.4
C8—H8•••C11	0.95	2.70	3.6136(13)	160.5
C9—H9•••C11	0.95	2.83	3.6079(12)	139.9
C13—H13•••O91 ^x	0.95	2.49	3.3820(15)	157.2
C14—H14•••O1 ^{xxii}	0.95	2.52	3.3199(14)	142.0
C16—H16•••O2 ^{xxiii}	0.95	2.51	3.1019(16)	120.2
Compound 4B				
C13—H13•••N3 ^x	0.95	2.56	3.4247(16)	151.0
C17—H17•••O1 ^{xxiv}	0.95	2.34	3.2151(16)	153.1
C7—H7•••O2 ^{xxv}	0.95	2.61	3.3737(17)	137.7
Compound 5				
N1—H1N•••O91 ^{xxvi}	0.99	1.80	2.7396(13)	155.9
N3—H3N•••C11	0.89	2.22	3.1072(10)	169.1
O91—H91O•••C11	0.93	2.12	3.0431(8)	169.6
C2—H2•••O91 ^{xxvii}	0.95	2.52	3.3548(14)	146.8
C5—H5•••O91 ^{xxvii}	0.95	2.57	3.3912(14)	144.3
C8—H8•••C11	0.95	2.82	3.6470(12)	146.4
C9—H9•••C11	0.95	2.81	3.5471(12)	134.6
Compound 5A1				
N1—H1N•••N2 ^{xxviii}	0.93	2.14	2.9258(13)	140.3

Compound **5A2**

N1—H1N•••N2 ^{xxix}	0.93	2.15	2.9561(12)	143.8
C13—H13•••N3 ^{xxx}	0.95	2.62	3.5655(13)	172.8

Symmetry transformations used to generate equivalent atoms: (i) $-x+1, -y+2, -z$; (ii) $-x+2, -y+2, -z$; (iii) $-x+2, -y+1, -z$; (iv) $-x, -y+1, -z+1$; (v) $x+1/2, -y+3/2, z+1/2$; (vi) $-x+1/2, y-1/2, -z+3/2$; (vii) $x-3/2, -y+1/2, z+1/2$; (viii) $-x+2, -y, -z+1$; (ix) $x+1, y+1, z$; (x) $x, y+1, z$; (xi) $x-1, y, z$; (xii) $-x+1, y+1/2, -z+1/2$; (xiii) $-x, y+1/2, -z+1/2$; (xiv) $x+1, y, z$; (xv) $x-2, -y+3/2, z+1/2$; (xvi) $-x, -y+3, -z+1$; (xvii) $x+1, -y+3/2, z+1/2$; (xviii) $x-1, y+1, z+1$; (xix) $x, y-1, z$; (xx) $-x, -y, -z+1$; (xxi) $x, y-1, z+1$; (xxii) $-x+1, -y+1, -z$; (xxiii) $-x+1, -y, -z$; (xxiv) $x, y-1, z$; (xxv) $x, y-2, z+1$; (xxvi) $x+1/2, -y+1/2, -z+1$; (xxvii) $x, -y+1/2, z+1/2$; (xxviii) $-x+2, -y+2, -z+1$; (xxix) $-x+1, -y+2, -z+1$; (xxx) $-x+1, y-1/2, -z+3/2$.

Table S5. Stacking interactions [$\text{\AA}$, $^\circ$]. Cg(1)-C(8) indicate the centroids of the aromatic rings containing N2, C8, C12, N22, C28, C32, N4, N24 atoms respectively, α is a dihedral angle between planes I and J, β is an angle between Cg(I)-Cg(J) vector and normal to plane I and d_p is a perpendicular distance of Cg(I) on ring J plane.

Cg(I)•••Cg(J)	Cg•••Cg	α	β	d_p
Compound 1				
Cg(1)•••Cg(2) ⁱ	3.6941(11)	1.84(9)	24.16	-3.3643(7)
Cg(1)•••Cg(3) ⁱⁱ	3.6531(11)	6.54(9)	16.34	3.3663(7)
Cg(2)•••Cg(1) ⁱ	3.6942(11)	1.84(9)	24.40	-3.3706(8)
Cg(2)•••Cg(3) ⁱⁱ	3.8295(11)	8.34(9)	23.96	3.6532(8)
Cg(3)•••Cg(1) ⁱⁱ	3.6530(11)	6.53(9)	22.86	3.5055(8)
Cg(3)•••Cg(2) ⁱⁱ	3.8296(11)	8.34(9)	17.44	3.4997(8)
Cg(4)•••Cg(5) ⁱⁱⁱ	3.5033(11)	0.63(9)	16.44	-3.3521(7)
Cg(4)•••Cg(6) ^{iv}	3.6701(10)	8.09(8)	25.27	-3.3167(7)
Cg(5)•••Cg(4) ⁱⁱⁱ	3.5033(11)	0.63(9)	16.90	-3.3601(8)
Cg(5)•••Cg(5) ⁱⁱⁱ	3.9875(11)	0.0	32.47	-3.3643(8)
Cg(5)•••Cg(6) ^{iv}	3.8212(11)	8.64(9)	29.90	-3.5595(8)
Cg(6)•••Cg(4) ^v	3.6701(10)	8.09(8)	25.35	3.3188(7)
Cg(6)•••Cg(5) ^v	3.8212(11)	8.64(9)	21.33	3.3125(7)
Compound 1B				
Cg(7)•••Cg(2) ^{vi}	3.5194(7)	2.54(6)	17.72	-3.3779(4)
Cg(1)•••Cg(2) ^{vi}	3.7267(7)	1.57(5)	25.23	-3.3823(4)
Cg(2)•••Cg(7) ^{vi}	3.5194(7)	2.54(6)	16.30	-3.3525(4)
Cg(2)•••Cg(1) ^{vi}	3.7267(7)	1.57(5)	24.82	-3.3712(4)
Cg(2)•••Cg(2) ^{vii}	3.8849(7)	0.0	31.50	3.3124(4)
Cg(3)•••Cg(3) ⁱ	3.7635(7)	0.0	18.72	-3.5644(4)
Compound 2				
Cg(1)•••Cg(2) ^{viii}	3.7368(8)	1.38(5)	23.26	3.4126(5)
Cg(1)•••Cg(3) ^{ix}	3.6548(7)	9.77(5)	22.94	3.0970(5)
Cg(2)•••Cg(1) ^{viii}	3.7367(8)	1.38(5)	24.04	3.4330(5)
Cg(2)•••Cg(3) ^{ix}	3.7006(8)	9.09(6)	24.97	3.3303(5)
Cg(3)•••Cg(1) ^x	3.6549(7)	9.77(5)	32.07	-3.3658(5)
Cg(3)•••Cg(2) ^x	3.7006(8)	9.09(6)	25.85	-3.3548(5)
Cg(3)•••Cg(3) ^x	3.4806(7)	0.0	22.21	-3.2223(5)

Compound 2A

Cg(1)•••Cg(1) ^{xi}	3.7858(7)	0.0	27.67	3.3529(5)
Cg(1)•••Cg(2) ^{xi}	3.6101(7)	0.87(5)	20.85	3.3605(5)
Cg(2)•••Cg(2) ^{xi}	3.7858(7)	0.0	26.88	3.3766(5)
Cg(3)•••Cg(3) ^{xi}	3.7859(7)	0.0	27.00	3.3732(5)

Compound 2B

Cg(7)•••Cg(7) ^{xii}	3.2961(18)	0.0	9.04	3.2552(13)
Cg(2)•••Cg(3) ⁱⁱ	3.9032(19)	4.75(16)	29.21	-3.5252(14)
Cg(8)•••Cg(8) ^{xiii}	3.2788(14)	0.0	4.40	-3.2691(10)
Cg(8)•••Cg(4) ^{xiii}	3.9873(15)	2.07(12)	35.75	-3.3142(10)
Cg(4)•••Cg(6) ⁱ	3.9295(16)	9.43(12)	31.78	3.5646(9)
Cg(5)•••Cg(6) ⁱ	3.8906(16)	7.99(13)	29.69	3.5664(11)

Compound 3

Cg(1)•••Cg(3) ^{ix}	3.9534(8)	16.36(5)	27.51	-3.8176(4)
Cg(3)•••Cg(1) ^{xiv}	3.9533(8)	16.36(5)	15.05	3.5063(5)

Compound 3A

Cg(1)•••Cg(2) ^v	3.6624(6)	1.46(5)	21.63	-3.3875(4)
Cg(2)•••Cg(1) ^{iv}	3.6624(6)	1.46(5)	22.34	3.4045(5)

Compound 3B

Cg(7)•••Cg(1) ^{xv}	3.6686(9)	1.24(7)	22.29	-3.4148(5)
Cg(7)•••Cg(2) ^{xv}	3.6641(9)	2.63(7)	18.69	-3.4330(5)
Cg(7)•••Cg(2) ^{xvi}	3.5938(9)	2.63(7)	19.32	3.3513(5)
Cg(1)•••Cg(7) ^{xv}	3.6687(9)	1.24(7)	21.44	-3.3945(5)
Cg(1)•••Cg(1) ^{xv}	3.6798(9)	0.0	22.14	-3.4086(5)
Cg(1)•••Cg(1) ^{xvi}	3.8772(9)	0.0	27.63	3.4348(5)
Cg(1)•••Cg(2) ^{xvi}	3.5554(9)	2.27(6)	16.74	3.4406(5)
Cg(2)•••Cg(7) ^{xv}	3.6641(9)	2.63(7)	20.46	-3.4708(6)
Cg(2)•••Cg(7) ^{xvi}	3.5938(9)	2.63(7)	21.17	3.3915(6)
Cg(2)•••Cg(1) ^{xvi}	3.5554(9)	2.27(6)	14.60	3.4046(6)

Compound 4

Cg(1)•••Cg(3) ^x	3.7167(8)	1.64(6)	11.60	-3.6352(5)
Cg(2)•••Cg(3) ^{xvii}	3.6951(8)	1.93(6)	20.13	3.4516(5)
Cg(3)•••Cg(1) ^x	3.7167(8)	1.64(6)	12.02	-3.6409(5)
Cg(3)•••Cg(2) ^{xvii}	3.6951(8)	1.93(6)	20.91	3.4694(5)

Compound 4B

Cg(7)•••Cg(1) ^{xviii}	3.5522(10)	1.13(6)	23.65	3.2782(5)
Cg(7)•••Cg(2) ^{xix}	3.6142(11)	1.52(7)	23.40	-3.2787(5)
Cg(7)•••Cg(2) ^{xviii}	3.5361(11)	1.52(7)	21.55	3.2843(5)
Cg(1)•••Cg(7) ^{xviii}	3.5522(10)	1.13(6)	22.65	3.2538(5)
Cg(1)•••Cg(1) ^{xviii}	3.5064(10)	0.0	21.68	3.2583(5)
Cg(1)•••Cg(2) ^{xix}	3.3839(10)	0.045(6)	12.15	-3.3137(5)
Cg(2)•••Cg(7) ^{xix}	3.6141(11)	1.52(7)	24.88	-3.3170(5)
Cg(2)•••Cg(7) ^{xviii}	3.5362(11)	1.52(7)	21.76	3.2890(5)
Cg(2)•••Cg(1) ^{xix}	3.3839(10)	0.45(6)	11.70	-3.3081(5)
Cg(3)•••Cg(3) ^{vi}	3.6014(11)	0.0	19.41	-3.3967(5)

Compound **5**

Cg(1)•••Cg(3) ^{xix}	3.5386(7)	3.45(6)	16.20	3.4251(4)
Cg(2)•••Cg(3) ^x	3.9811(7)	5.85(6)	26.49	-3.5037(5)
Cg(3)•••Cg(1) ^{xx}	3.5386(7)	3.45(6)	14.55	3.3981(5)
Cg(3)•••Cg(2) ^x	3.9811(7)	5.85(6)	28.35	-3.5633(5)

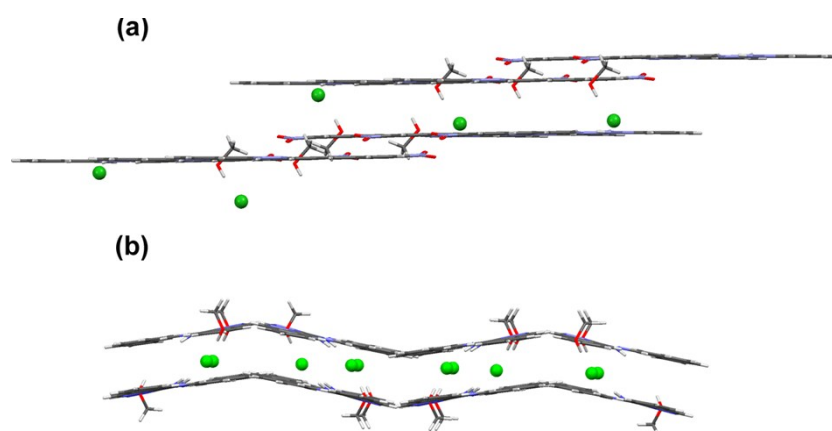
Compound **5A2**

Cg(1)•••Cg(1)xi	3.9130(6)	0	30.49	3.3719(4)
Cg(1)•••Cg(2)v	3.8076(7)	2.18(5)	26.05	-3.3909(4)
Cg(2)•••Cg(2)xi	3.9130(7)	0	28.67	3.4331(4)
Cg(3)•••Cg(3)xi	3.9130(7)	0	26.19	3.5112(5)

Symmetry transformations used to generate rings: (i) $-x+1, -y+1, -z$; (ii) $-x+1, -y+2, -z$; (iii) $-x+2, -y+1, -z$; (iv) $x, y-1, z$; (v) $x, y+1, z$; (vi) $-x+1, -y+1, -z+1$; (vii) $-x, -y+1, -z+1$; (viii) $-x, -y, -z+2$; (ix) $x-1, y, z$; (x) $-x+1, -y, -z+1$; (xi) $x, -y-1/2, z+1/2$; (xii) $-x+2, -y+1, -z+1$; (xiii) $-x+2, -y, -z$; (xiv) $x+1, y, z$; (xv) $-x, -y+2, -z+2$; (xvi) $-x-1, -y+2, -z+2$; (xvii) $-x, -y, -z+1$; (xviii) $-x+2, -y-1, -z+2$; (xix) $-x+1, -y-1, -z+2$; (xx) $x+1/2, -y+1/2, -z+1$.

Table S6. Selected structural data for **1B-4B** [Å]

	N2-C2	N1-N2	C1-N1	C1-C4	C1-N3	N3-N4	N4-C9	N5-O1	N5-O2
1B	1.2951(14)	1.3786(12)	1.3739(13)	1.4433(15)	1.3171(14)	1.3954(12)	1.3135(14)	1.2363(12)	1.2201(12)
2B	1.280(4)	1.374(3)	1.370(3)	1.422(4)	1.306(4)	1.375(4)	1.317(4)	1.205(4)	1.209(4)
	1.287(4)	1.385(3)	1.359(3)	1.421(4)	1.319(3)	1.372(3)	1.318(3)	1.203(3)	1.212(4)
3B	1.2991(18)	1.3847(15)	1.3732(18)	1.4374(19)	1.3187(18)	1.3781(16)	1.3217(18)	1.2340(16)	1.2313(15)
4B	1.3311(16)	1.3811(14)	1.3814(15)	1.4755(16)	1.3318(15)	1.4126(14)	1.3195(15)	1.2199(14)	1.2171(14)

Fig. S5. The comparison of crystal packing of **4** (a) and **5** (b).

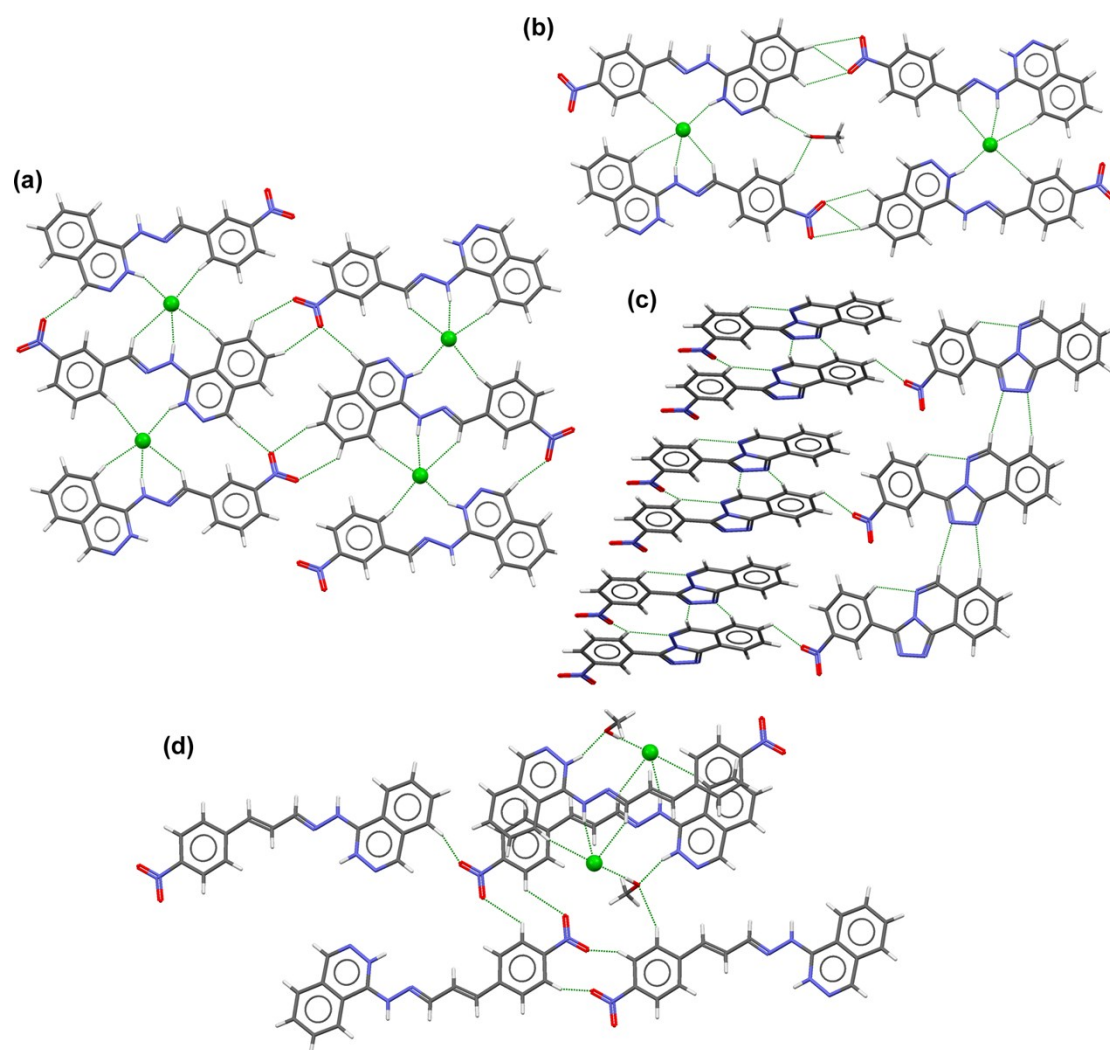


Fig. S6. The part of crystal packing including the C-H $\cdots$ O hydrogen bonds formed by the nitro group for **2** (a), **3** (b), **2B** (c) and **4** (d). For **3** the disordered methanol atoms were omitted.

Computational studies on O $\cdots$ O contacts

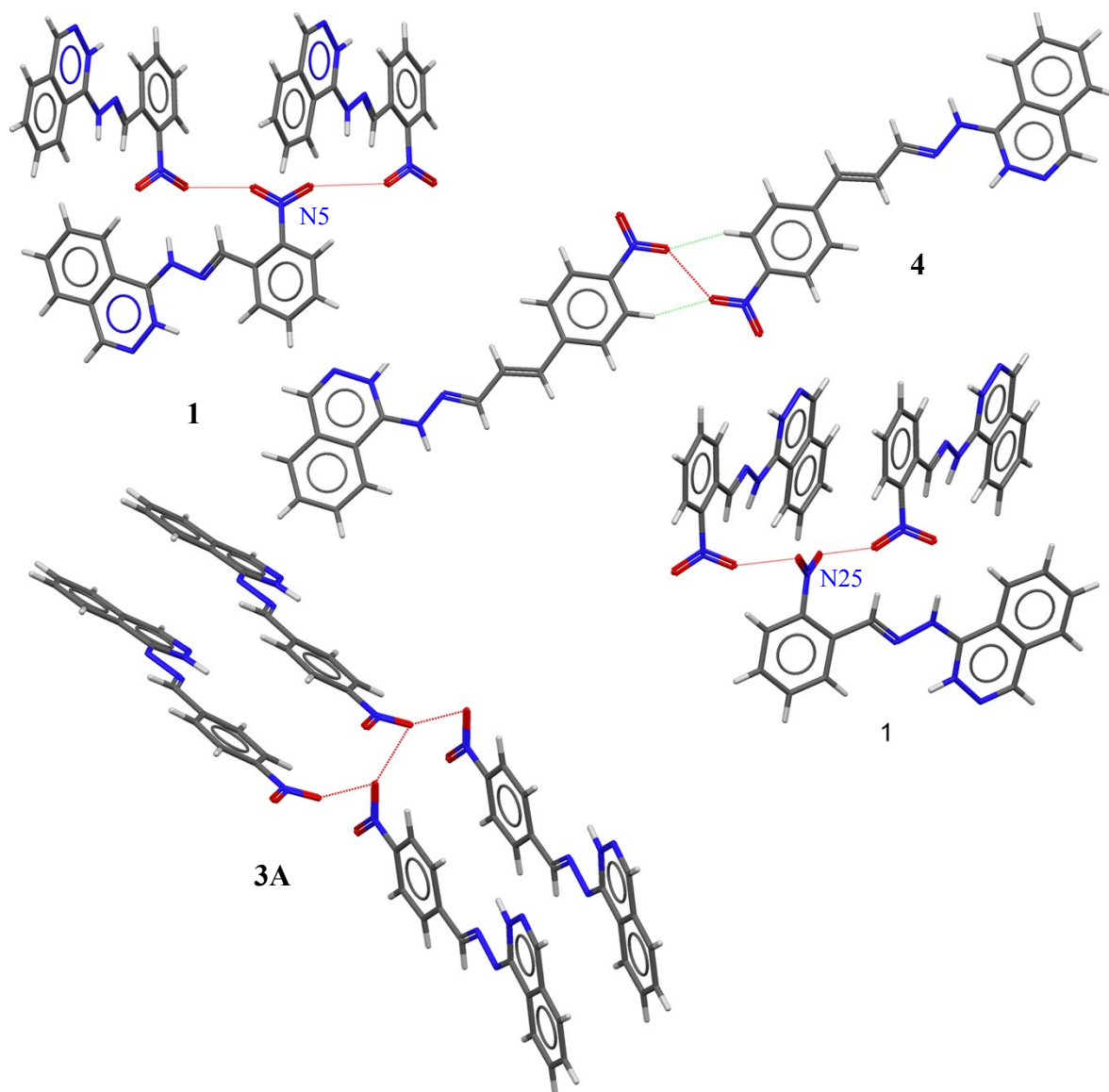


Fig. S7. The part of crystal packing showing the O $\cdots$ O contacts.

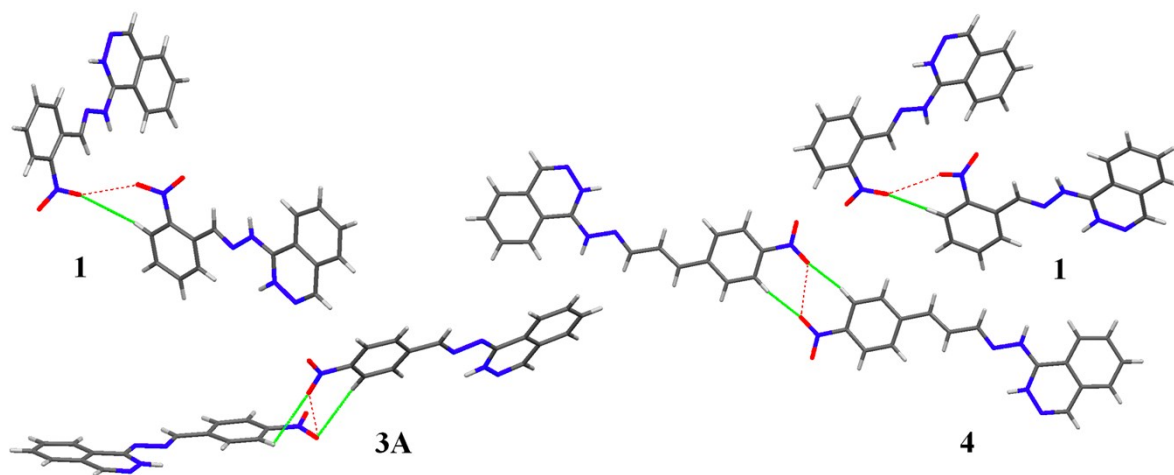


Fig. S8. The part of crystal packing showing the O $\cdots$ O contacts (marked in red) and the weak C-H $\cdots$ O hydrogen bonds (possessing the H $\cdots$ O distance longer than the sum of van der Waals radii, except for **4** (marked in green)).

Table S7. Hydrogen bonds that were included in the calculations of energy of the O $\cdots$ O interactions [$\text{\AA}$, $^\circ$].

D-H $\cdots$ A	d(D-H)	d(H $\cdots$ A)	d(D $\cdots$ A)	<(DHA)
Compound 1				
C12—H12 $\cdots$ O1 ⁱ	0.95	3.13	3.871	136.0
C32—H32 $\cdots$ O21 ⁱⁱ	0.95	2.71	3.514	142.3
Compound 3A				
C12—H12 $\cdots$ O1 ⁱⁱⁱ	0.95	2.82	3.684	152.1
C12—H12 $\cdots$ O1 ⁱⁱⁱ	0.95	2.89	3.303	107.3
Compound 4				
C14—H14—O1 ^{iv}	0.95	2.52	3.320	142.0

Symmetry transformations used to generate equivalent atoms: (i) $-x+1, y+1/2, -z+1/2$; (ii) $-x+2, y+1/2, -z+1/2$; (iii) $-x-1, y-1/2, -z+1/2$; (iv) $-x+1, -y+1, -z$.

Table S8. The hydrogen bonds with the shortest H $\cdots$ A distances that were considered in the calculations of the O $\cdots$ O interaction energies [$\text{\AA}$, $^\circ$].

D-H $\cdots$ A	d(D-H)	d(H $\cdots$ A)	d(D $\cdots$ A)	<(DHA)
Compound 1				
C10-H $\cdots$ O2 ⁱ	0.95	3.72	4.132	109.0
C12-H12 $\cdots$ O1 ⁱ	0.95	3.13	3.871	136.0
Compound 1				
C32-H32 $\cdots$ O21 ⁱⁱ	0.95	2.71	3.514	142.3
H32 $\cdots$ O21 ⁱⁱ	0.95	3.45	4.092	127.2
C30-H $\cdots$ O22 ⁱⁱ				
Compound 3A				
C12-H12 $\cdots$ O1 ⁱⁱⁱ	0.95	2.82	3.684	152.1
C12-H12 $\cdots$ O1 ⁱⁱⁱ	0.95	2.89	3.303	107.3
Compound 4				
C14-H14 $\cdots$ O1 ^{iv}	0.95	2.52	3.320	142.0

Symmetry transformations used to generate equivalent atoms: (i) $-x+1, y+1/2, -z+1/2$; (ii) $-x+2, y+1/2, -z+1/2$; (iii) $-x-1, y-1/2, -z+1/2$; (iv) $-x+1, -y+1, -z$.

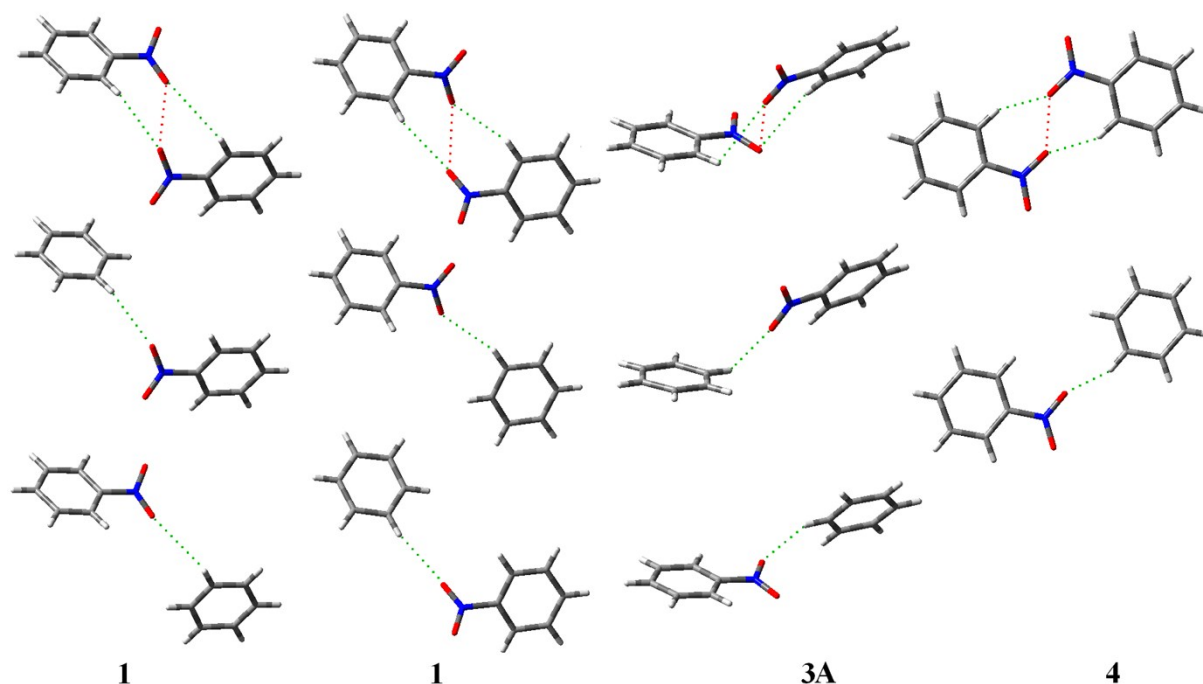


Fig. S9. The molecular moieties used for the theoretical calculations of the $O\cdots O$ interaction energies (All computed moieties were visualized by GaussView. GaussView, Version 5, Roy Dennington, Todd Keith, and John Millam, Semichem Inc., Shawnee Mission, KS, 2009.)

Table S9. The geometrical parameters of the $O\cdots O$ contacts [$\text{\AA}$, $^\circ$].

N-O $\cdots$ O	D(O $\cdots$ O)	$\angle$ (N-O $\cdots$ O)
Compound 1		
N5-O1 $\cdots$ O2 ⁱ	2.729(8)	137.0(3)
N5-O2 $\cdots$ O1 ⁱ	2.729(8)	151.1(3)
N25-O21 $\cdots$ O22 ⁱⁱ	2.742(2)	148.74(14)
N25-O22 $\cdots$ O21 ⁱⁱ	2.742(2)	134.38(14)
Compound 3A		
N5-O1 $\cdots$ O1 ⁱⁱⁱ	2.8629(14)	92.90(8)
N5-O1 $\cdots$ O1 ⁱⁱⁱ	2.8629(14)	145.28(8)
Compound 4		
N5-O1 $\cdots$ O1 ^{iv}	2.8619(13)	122.58(8)

Symmetry transformations used to generate equivalent atoms: (i) $-x+1, y+1/2, -z+1/2$; (ii) $-x+2, y+1/2, -z+1/2$; (iii) $-x-1, y-1/2, -z+1/2$; (iv) $-x+1, -y+1, -z$.