

Electronic supporting Information:

**A self-catenated *rob*-type porous coordination polymer  
constructed from triazolate and carboxylate ligands:  
fluorescent response to the reversible phase transformation**

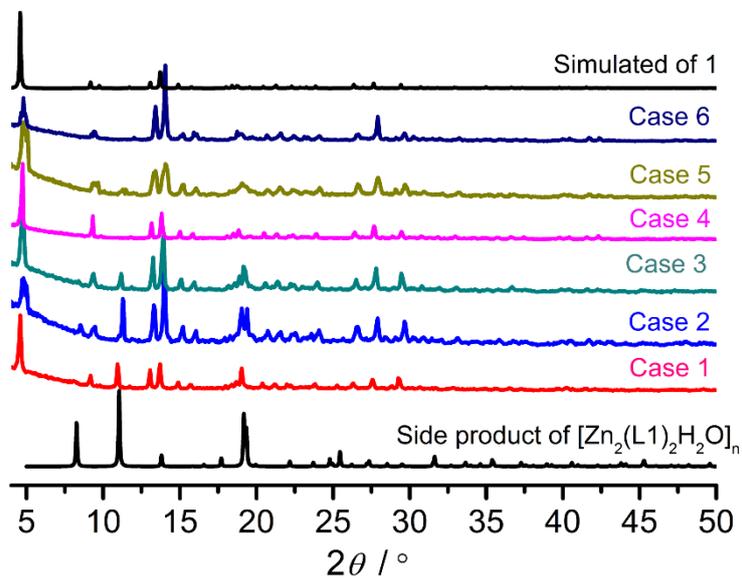
Mingli Deng,<sup>a</sup> Shijun Tai,<sup>b</sup> Weiquan Zhang,<sup>a</sup> Yongchen Wang,<sup>a</sup> Jiaying Zhu,<sup>a</sup> Jinsheng Zhang,<sup>b</sup> Yun Ling\*,<sup>a</sup> and Yaming Zhou <sup>a</sup>

<sup>a</sup> Shanghai Key Laboratory of Molecular Catalysis and Innovative Materials, Department of Chemistry, Fudan University, Shanghai, 200433, China.

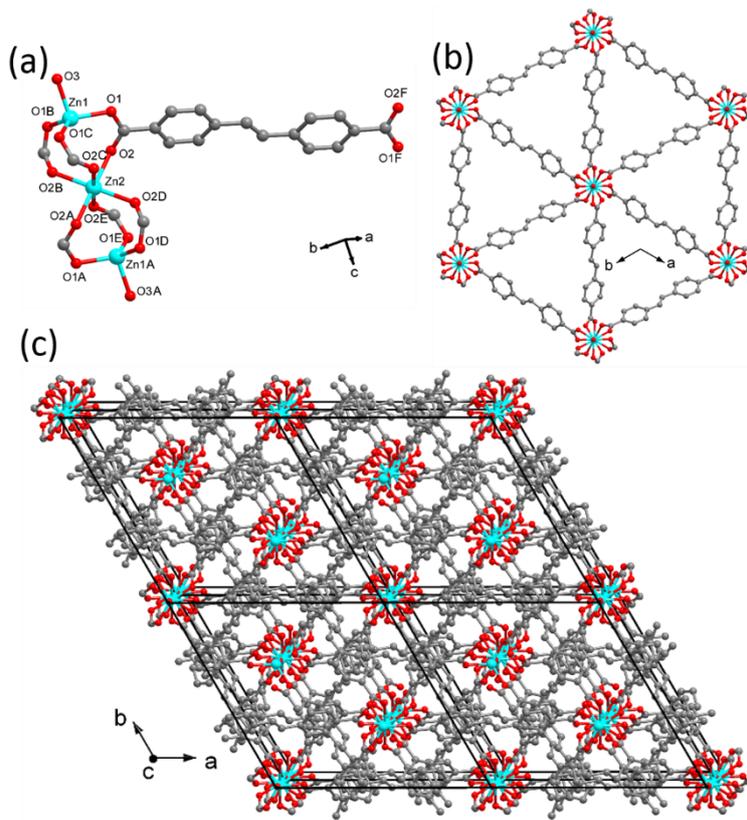
<sup>b</sup> College of Chemistry, Chemical Engineering and Environmental Engineering, Liaoning Shihua University, Liaoning, 113001, China.

Email: yunling@fudan.edu.cn (Ling, Y.)

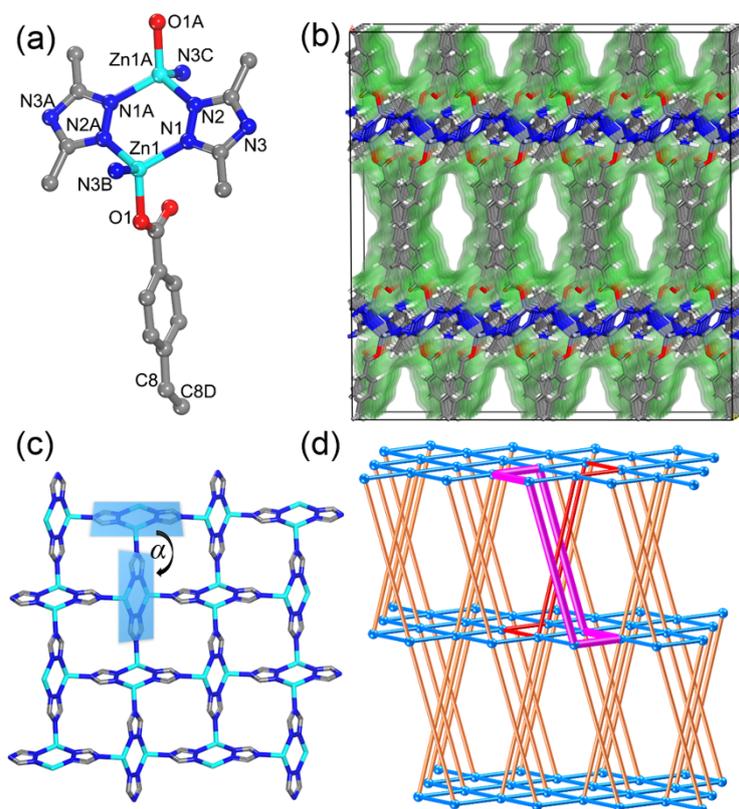
**Fig. S1.** the PXRD patterns obtained for synthesis pure phase of MA-11. (*Case 1*: 0.02 mL HAC, *Case 2*: 0.06 HNO<sub>3</sub>, *Case 3*: 0.06 mL HBF<sub>4</sub>, *Case 4*: HNO<sub>3</sub> (0.2 mL) and HBF<sub>4</sub> (0.06 mL), *Case 5*: 0.12 mL HNO<sub>3</sub>, *Case 6*: 0.06 mL HNO<sub>3</sub> and 0.06 mLHBF<sub>4</sub>)



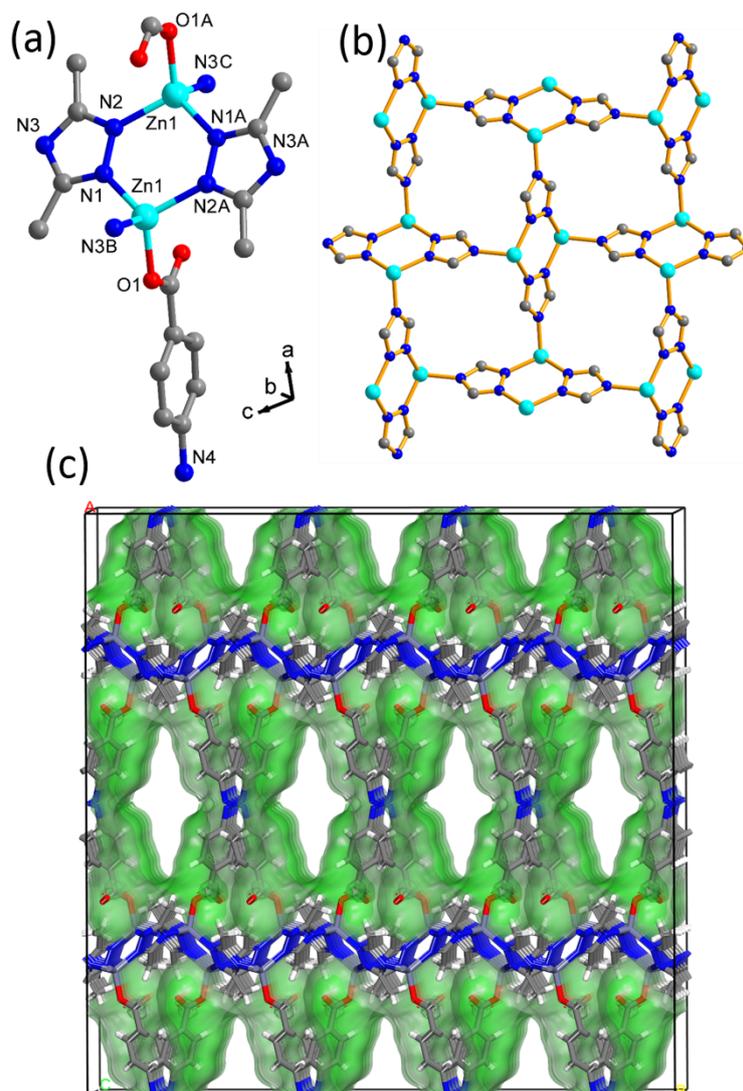
**Fig. S2.** the crystal structure of the side product obtained without additional acid. (a) the structure motif (A:  $2/3-x, 4/3-y, 1/3-z$ ; B:  $-x+y, 1-x, z$ ; C:  $1-y, 1+x-y, z$ ; D:  $2/3+x-y, 1/3+x, 1/3-z$ ; E:  $-1/3+y, 1/3-x+y, 1/3-z$ ; F:  $2/3-x, 1/3-y, 1/3-z$ ); (b) the two-dimensional layer structure; (c) the packing structure



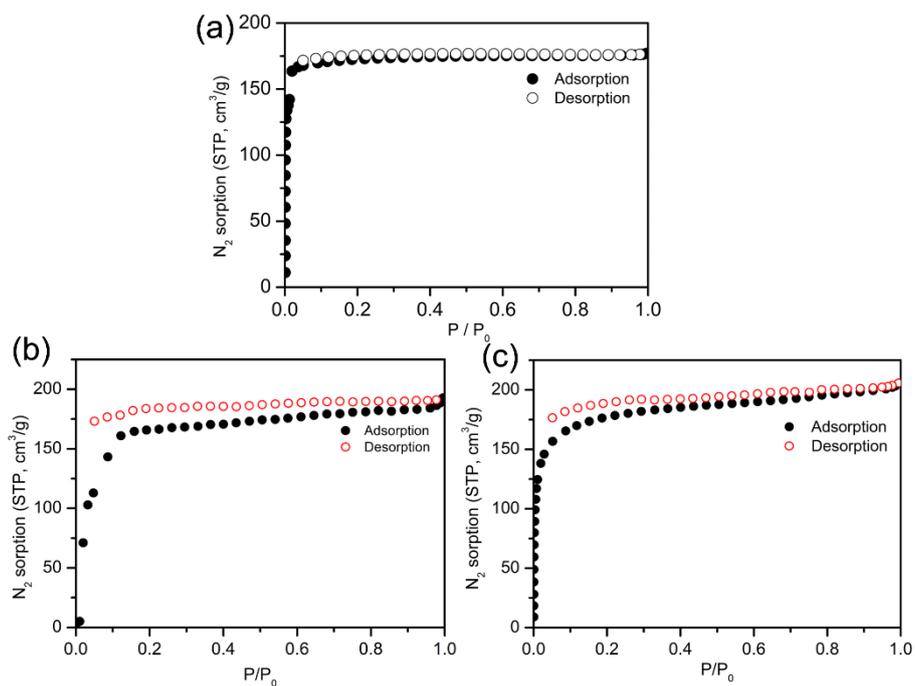
**Fig. S3.** (a) the structural motif of **MAC-11** with atomic label (symmetry codes: A:  $0.5-x, 0.5-y, -z$ ; B:  $0.5-x, 0.5+y, 0.5-z$ ; C:  $x, -y, -0.5+z$ ; D:  $-x, -y, -z$ ); (b) the three-dimensional structure of **MAC-11** showing 1D channel along *c* axis (the green: Connolly with probe atom radii of  $1.4 \text{ \AA}$ , cell:  $1 \times 4 \times 4$ ); (c) the two-dimensional layer formed exclusively by  $\text{Zn}_2(\text{dmtrz})_2$  dimers ( $\alpha$ : the dihedral angle); (d) the rob (48.66.8) net of 1 after considering of  $\text{Zn}_2(\text{dmtrz})_2$  as 6-connected node (the pink and red ring show the self-catenated hopf links).



**Fig. S4.** (a) the structure motif of **MAC-13** (symmetry codes: A:  $0.5-x, 0.5-y, -z$ ; B:  $0.5-x, -0.5+y, 0.5-z$ ; C:  $x, 1-y, -0.5+z$ ); (b) the two-dimensional (4,4) layer with the dihedral angle of  $82.84^\circ$ , which is slightly smaller than that in **1**; (c) the three-dimensional structure of **2** showing the accessible porous structure along  $c$  axis (the green surface: calculated by Connolly method with the probe atomic radii settled to be  $1.4 \text{ \AA}$ ).

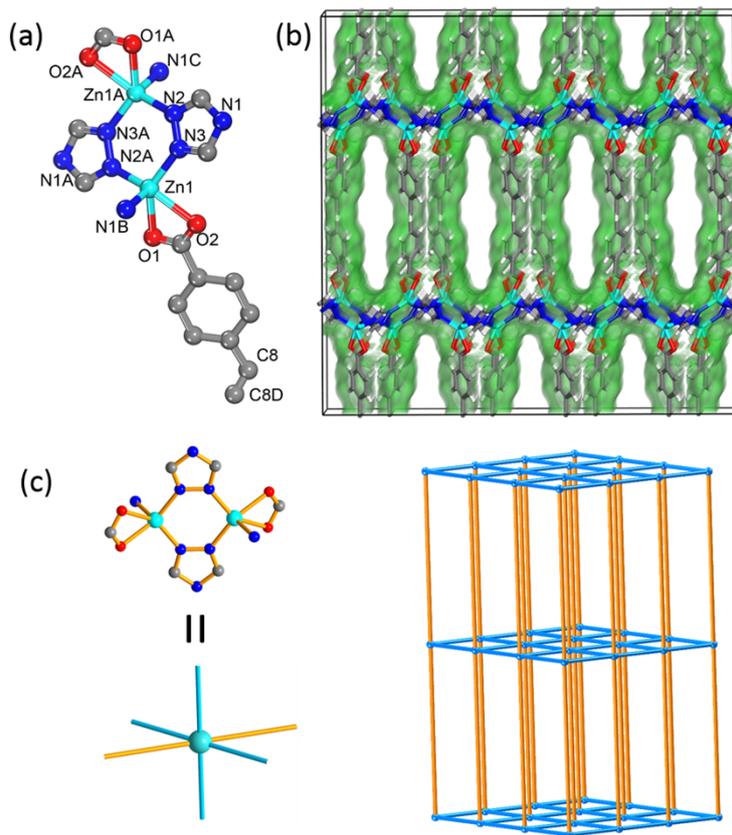


**Fig. S5.** N<sub>2</sub> sorption isothermal of **MAC-11**, **MAC-13** and **MAC-14** at 77 K

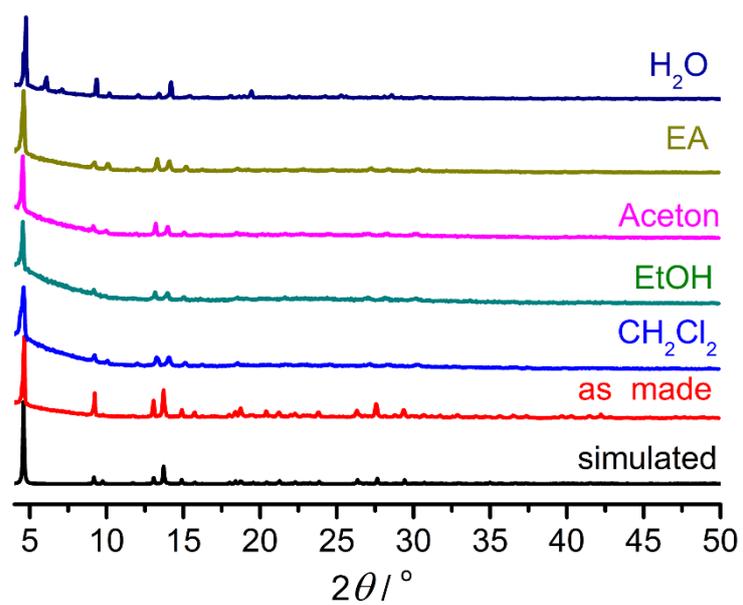


Compound	Surface area (m <sup>2</sup> /g)		Pore size (H-K, Å)	Pore Volume (micropore , Å)
	BET	Langmuir		
MAC-11	722	763	5.9	0.26
MAC-13	719	755	7.8	0.21
MAC-14	677	831	6.9	0.24

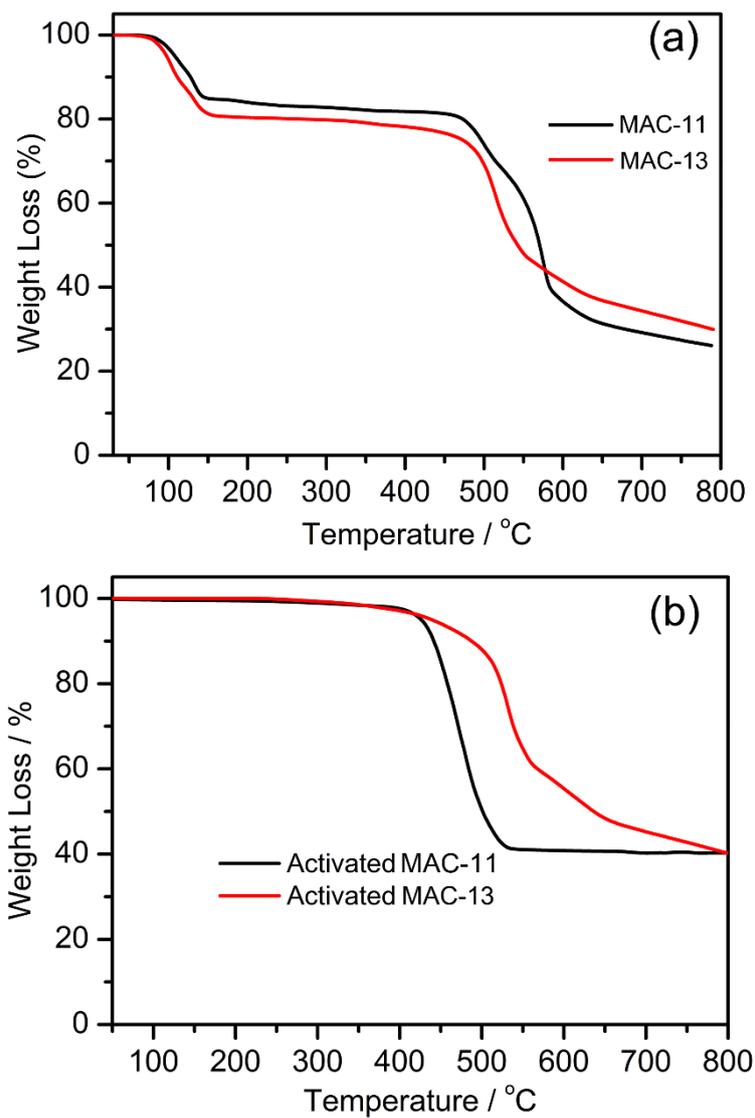
**Fig. S6.** (a) the structure motif of **MAC-14** (symmetry codes: A:  $0.5-x, 0.5-y, 1-z$ ; B:  $x, -y, -0.5+z$ ; C:  $0.5-x, 0.5+y, 1.5-z$ ; D:  $1-x, y, 0.5-z$ ); (b) the three-dimensional structure of **MAC-14** showing the accessible porous structure along  $c$  axis (the green surface: calculated by Connolly method with the probe atomic radii settled to be  $1.4 \text{ \AA}$ ); (c) considering the Zn-dmtrz dimer as a 6-connected node, **MAC-14** can be regarded as 6-connected  $pcu$ -type framework.



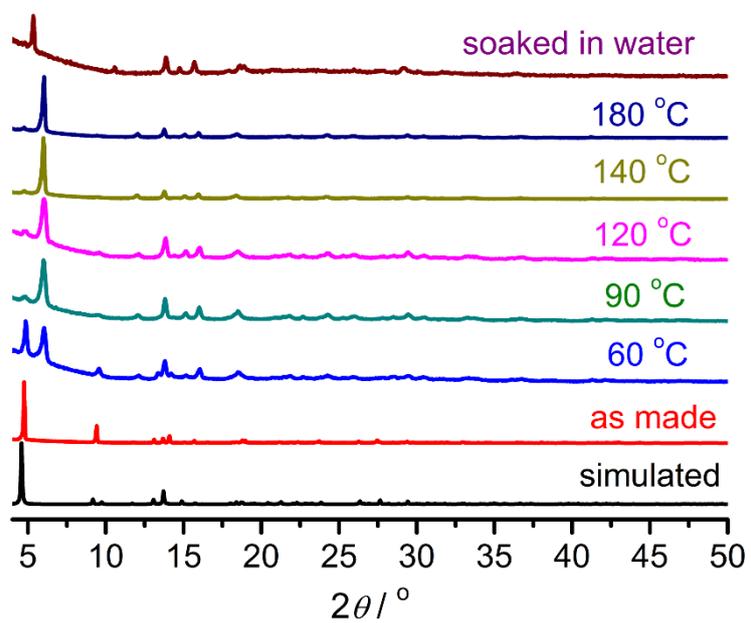
**Fig. S7.** The PXRD patterns of MAC-11 in common solvent.



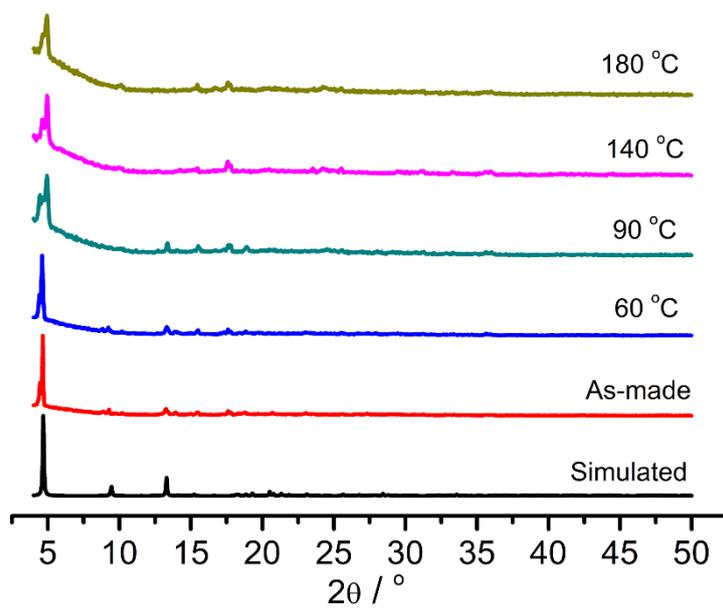
**Fig. S8.** (a) the TGA data of as-made **MAC-11** and **MAC-13** (the first step weigh loss of **MAC-11** agrees well with the weight loss of lattice water theoretically calculated from molecular weight, while **MAC-13** is not. It could be ascribed to the severe disorder of the lattice water in **MAC-13** that causes the difference between the TGA data and the molecular weight); (b) the TGA data of activated **MAC-11** and **MAC-13**.



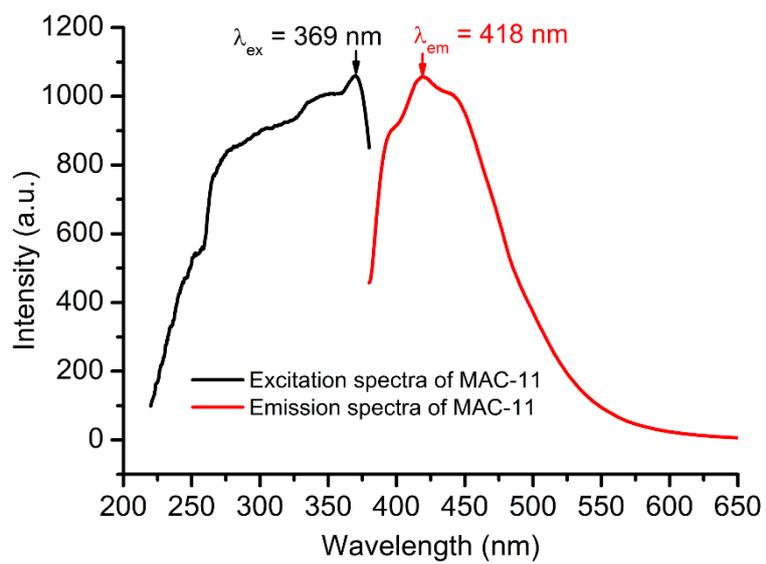
**Fig. S9** the temperature-dependent PXRD patterns of **MAC-13** (phase transformation:  $2\theta = \sim 4.6^\circ$  shifts to  $\sim 5.9^\circ$  and then shifts back to  $5.3^\circ$  in water solution after 1 h)



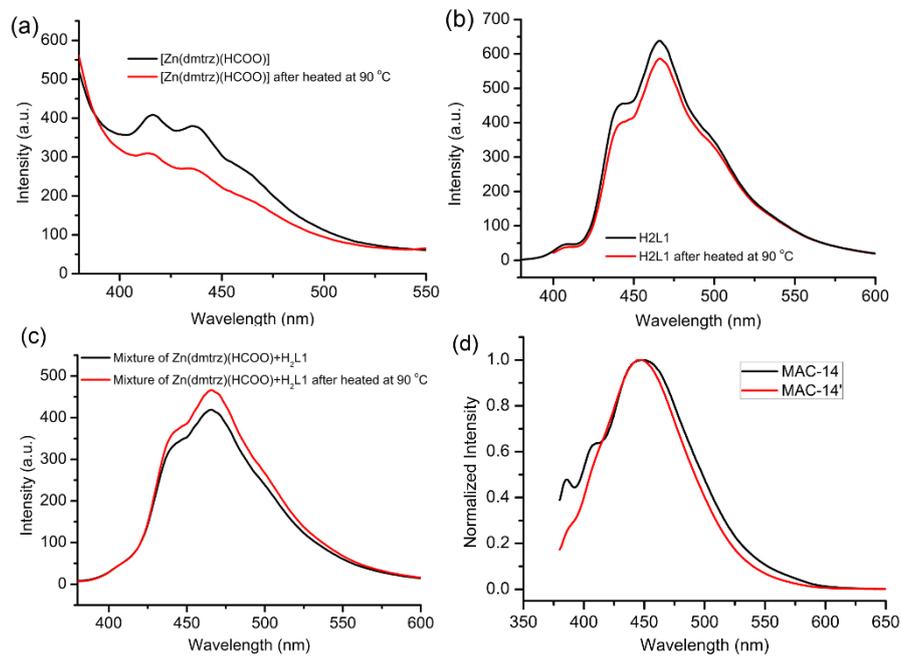
**Fig. S10.** the temperature-dependent PXRD patterns of **MAC-14** (the diffraction peak at  $2\theta = \sim 4.6^\circ$  slightly shifts to  $2\theta = \sim 4.9^\circ$ . The shift is much smaller than that of **MAC-11**)



**Fig. S11.** The excitation and emission spectra of **MAC-11**



**Fig. S12.** The emission spectra of 2D layer [Zn(dmtrz)(HCOO)] (a), H<sub>2</sub>L1 (b), their mixture (c) and MAC-14 before and after heated at 90 °C ( $\lambda_{\text{ex}} = 369 \text{ nm}$ )



**Fig. S13.** (a) The FT-IR spectra of MAC-11 and MAC-11' showing the disappearance of the vibration band at  $1675\text{ cm}^{-1}$ ; (b) the FT-IR spectra of MAC-13 and MAC-13' showing the similar change of the carboxylate groups as that of MAC-11 on the vibration band; (c) the FT-IR spectra of MAC-14 and MAC-14', indicating that there is no obvious change on the vibration band of carboxylate group.

