

A Metal-Organic Framework Based on Cyclotriphosphazene Functionalized Hexacarboxylate for Selective Adsorption of CO₂ and C₂H₆ from CH₄ at Room Temperature

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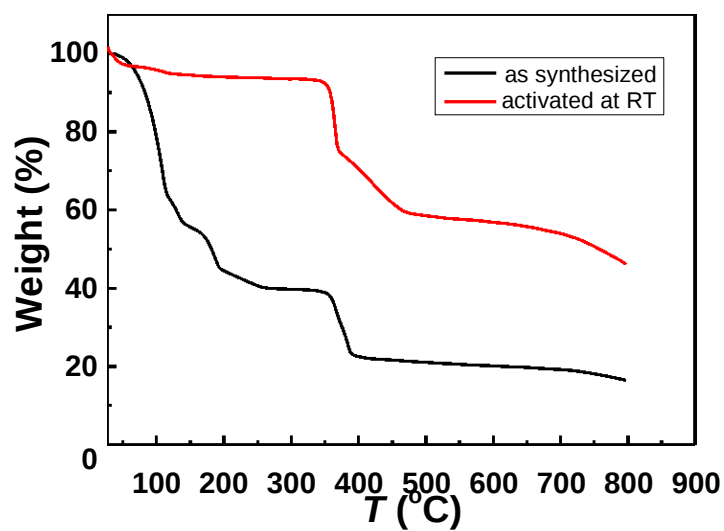


Fig. S1 TGA curve of as-synthesized **ZJNU-60** and activated **ZJNU-60a** under a nitrogen atmosphere at a heating rate of $5\text{ }^{\circ}\text{C min}^{-1}$.

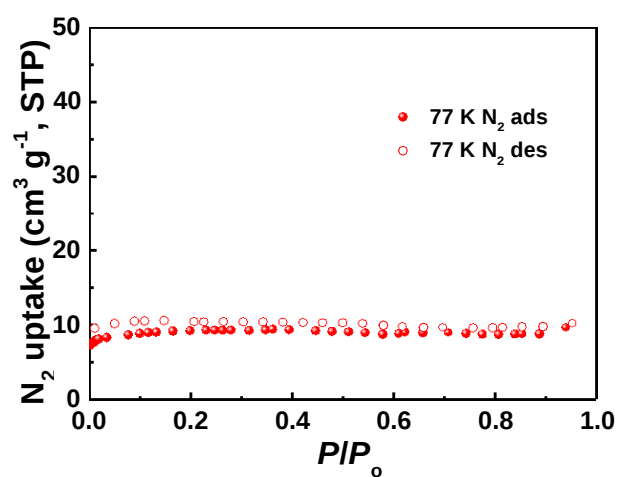


Fig. S2 N_2 adsorption-desorption isotherm of **ZJNU-60a** at 77 K. Solid and open symbols represent adsorption and desorption, respectively.

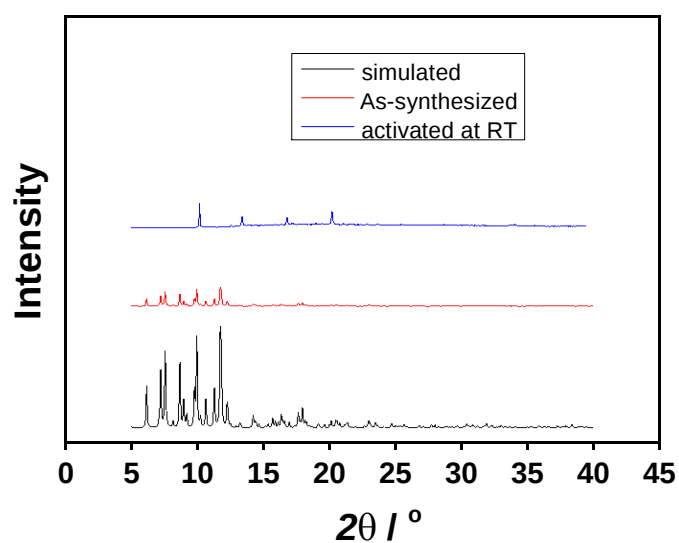


Fig. S3 PXRD patterns of as-synthesized ZJNU-60 and activated ZJNU-60a together with the simulated one.

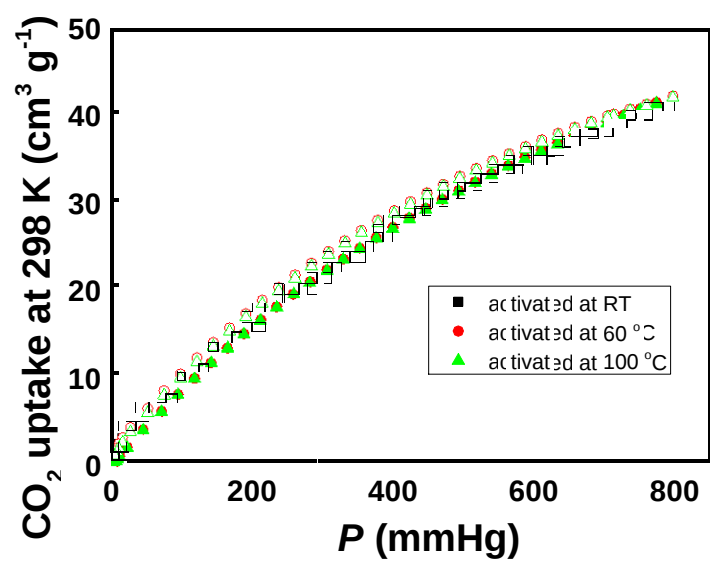


Fig. S4 CO₂ adsorption isotherms at 298 K of the samples activated at RT (black), 60 °C (red) and 100 °C (green), respectively. Solid and open symbols represent adsorption and desorption, respectively.

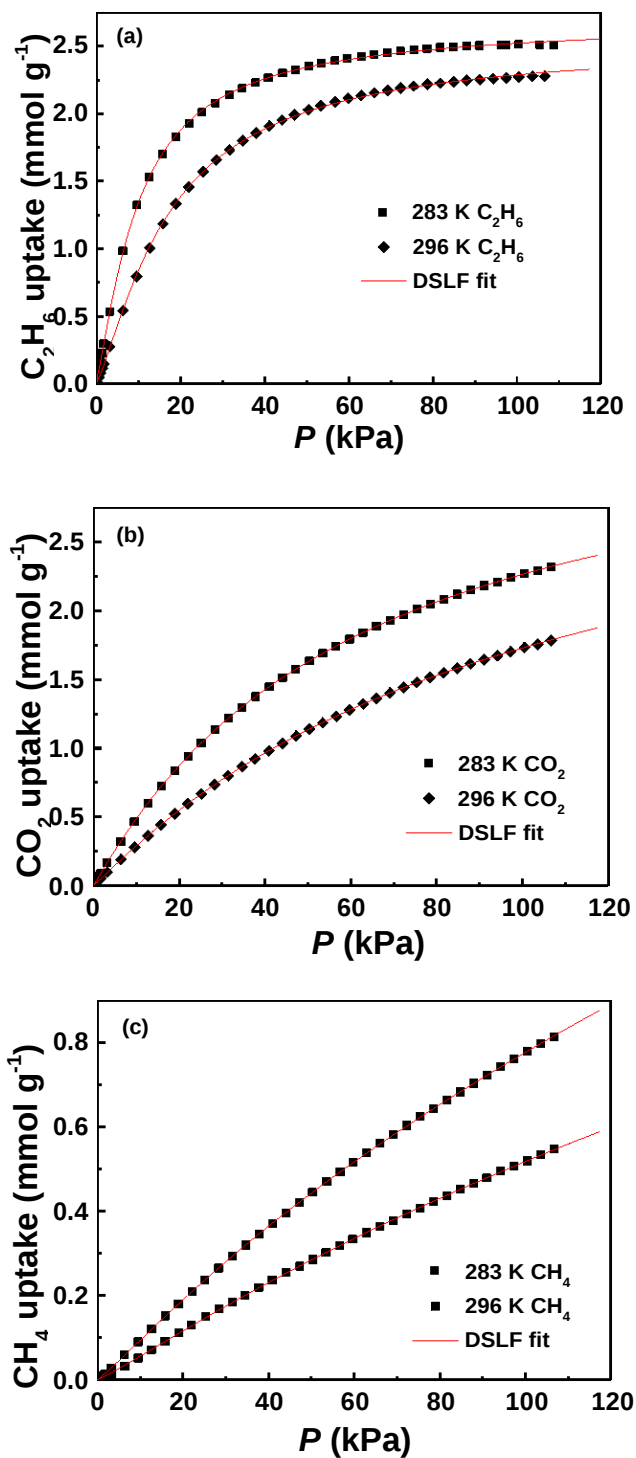


Fig. S5 Comparison of the pure-component isotherm data for (a) C₂H₆, (b) CO₂, and (c) CH₄ in ZJNU-60a with the fitted isotherms shown by continuous solid lines at 283 K, and 296 K.

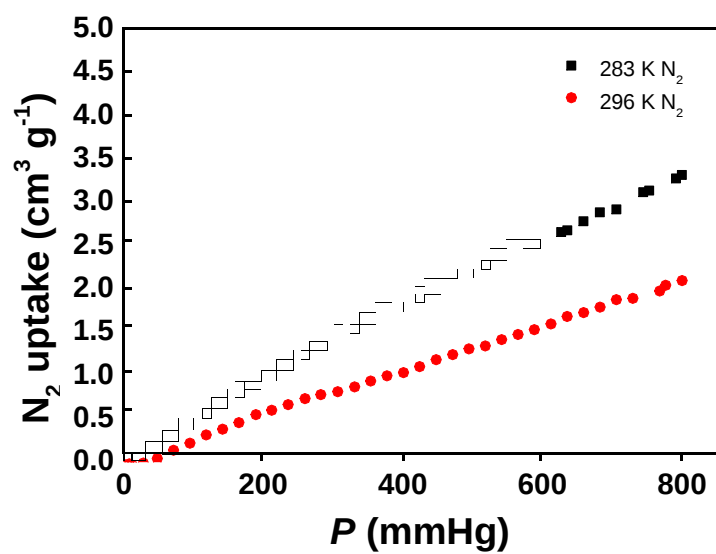


Fig. S6 N₂ adsorption isotherms of **ZJNU-60a** at 283 K and 296 K, respectively.

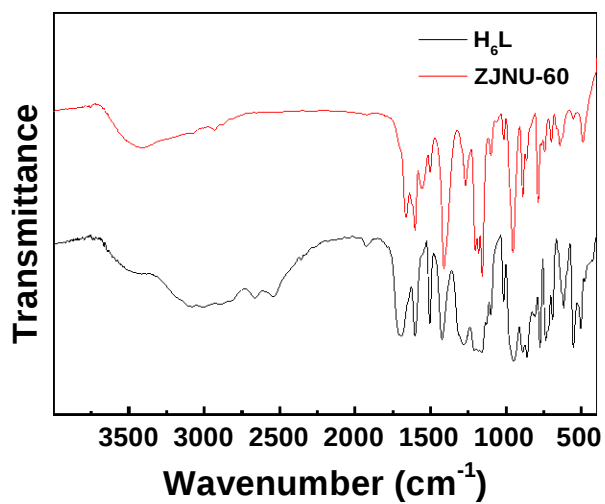
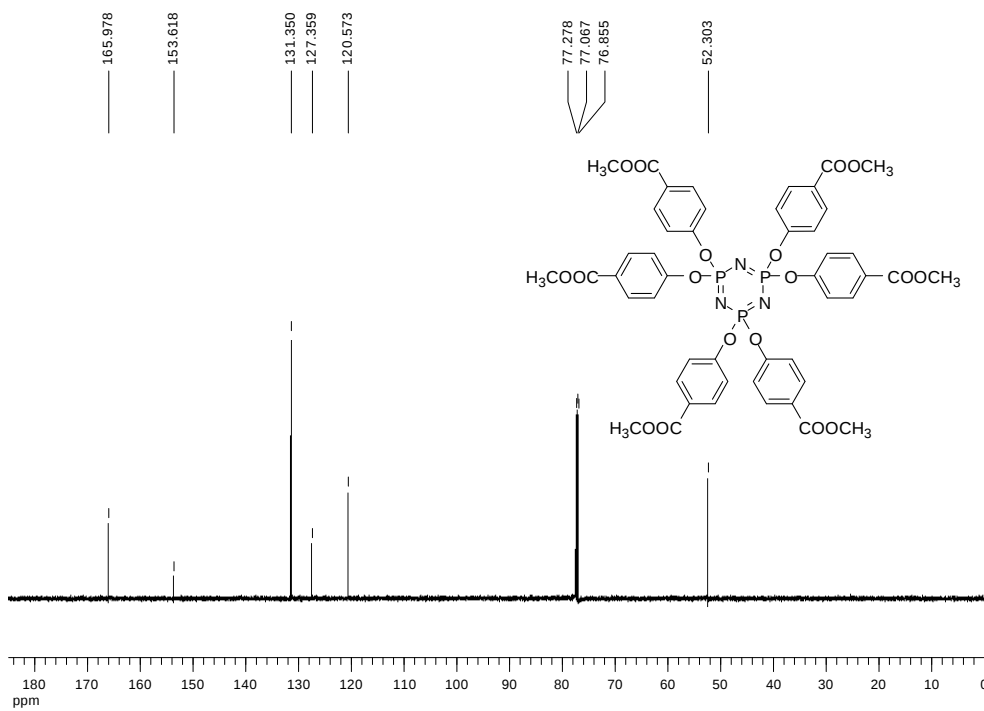
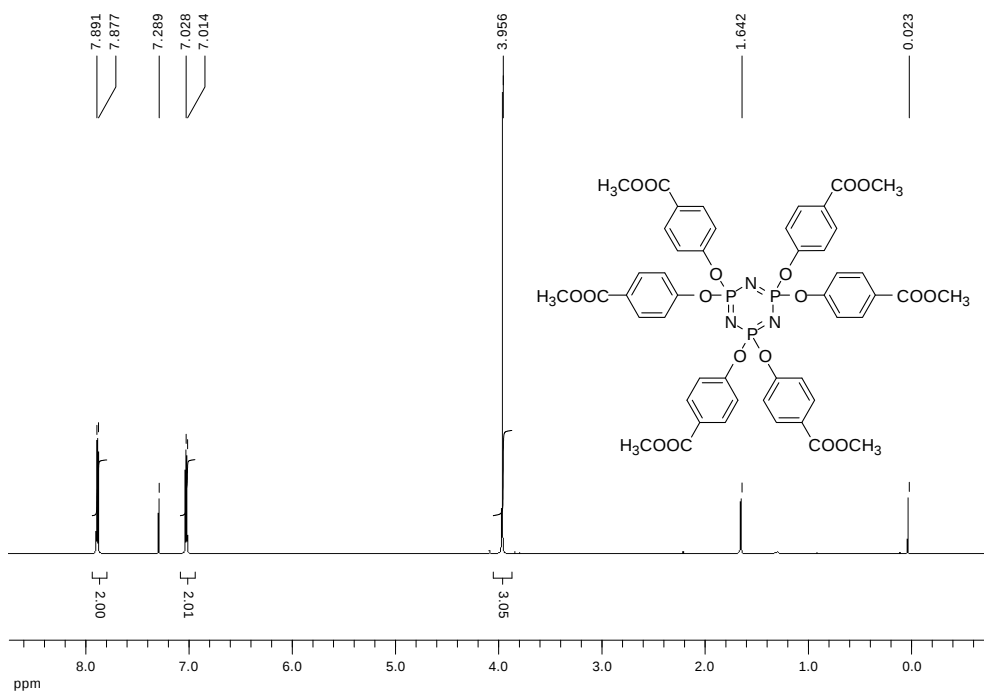


Fig. S7 FTIR spectra of organic ligand (black) and the as-synthesized **ZJNU-60** (red).



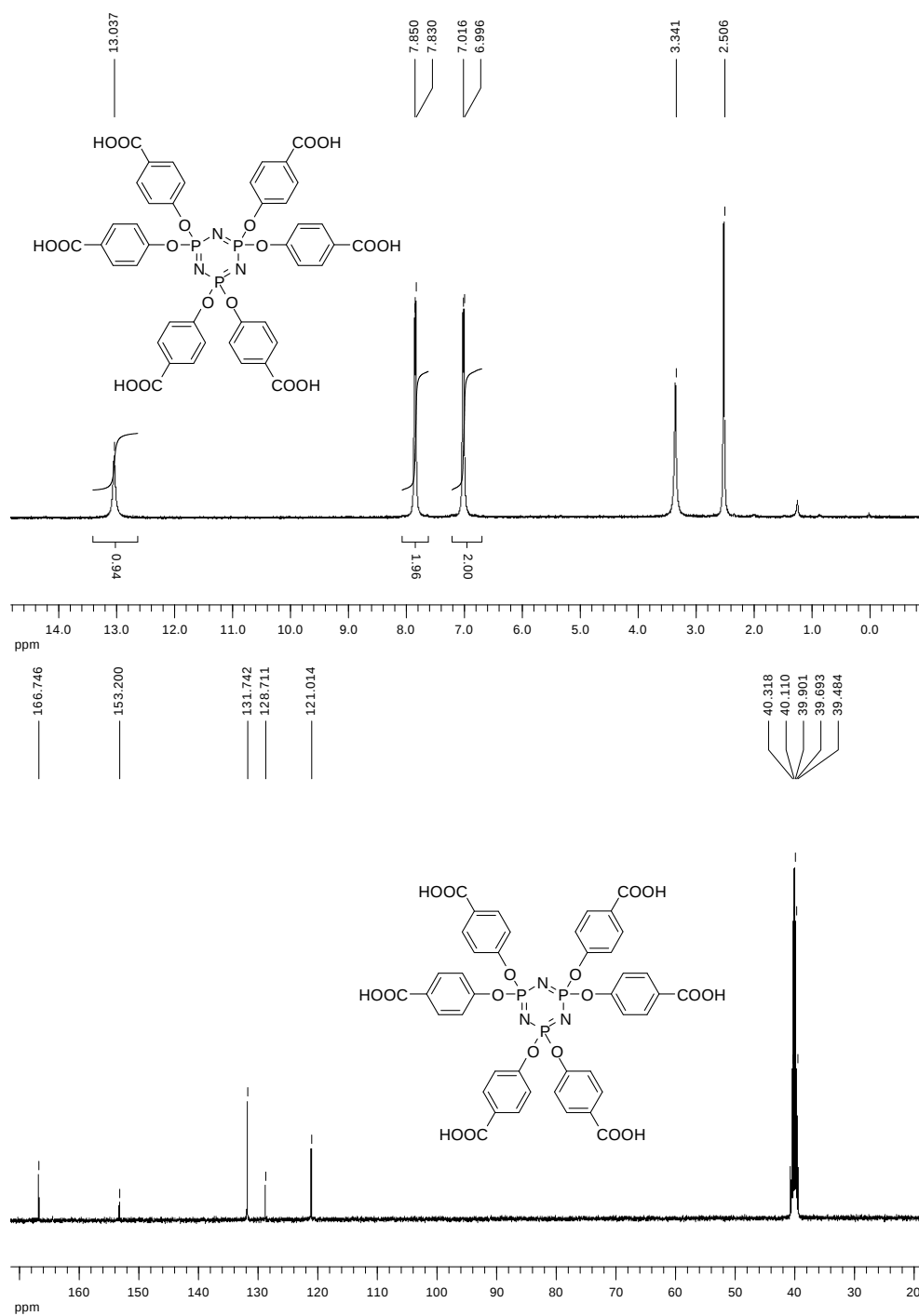


Fig. S8 ^1H NMR and ^{13}C NMR spectra.

Table S1 Dual-site Langmuir-Freundlich fit parameters for **ZJNU-60a**.

	Site A				Site B			
	$q_{A,sat}$ (mmol g ⁻¹)	b_{A0} (kPa ^{-ν_A})	E_A (kJ mol ⁻¹)	ν_A	$q_{B,sat}$ (mmol g ⁻¹)	b_{B0} (kPa ^{-ν_B})	E_B (kJ mol ⁻¹)	ν_B
C ₂ H ₆	4.79	2.56×10 ⁻¹³	57.815	0.27	2.44	2.41×10 ⁻⁹	39.66	1.31
CO ₂	3.13	6.59×10 ⁻⁹	34.511	0.98	0.57	9.01×10 ⁻⁶	16.46	1.25
CH ₄	5.67	4.68×10 ⁻⁹	29.031	1	0.35	5.54×10 ⁻⁸	27.36	1.24

Table S2 Crystal and structural refinement data for **ZJNU-60**

Empirical formula	C ₄₂ H ₃₀ Cu ₃ N ₃ O ₂₁ P ₃
Formula weight	1196.25
Temperature (K)	100 K
Wavelength (Å)	1.54184
Crystal system, space group	Triclinic, <i>P</i> -1
Unit cell dimensions	<i>a</i> = 12.6305(12) Å <i>b</i> = 14.5753(7) Å <i>c</i> = 22.114(3) Å <i>α</i> = 93.228(6) [°] <i>β</i> = 100.746(9) [°] <i>γ</i> = 98.776(6) [°]
Volume (Å ³)	3937.7(7)
Z, Calculated density (g cm ⁻³)	2, 1.009
Absorption coefficient (mm ⁻¹)	1.967
<i>F</i> (000)	1146
Crystal size (mm)	0.06 × 0.04 × 0.04
<i>θ</i> range for data collection (°)	3.08 to 74.73
Limiting indices	-15 ≤ <i>h</i> ≤ 14, -17 ≤ <i>k</i> ≤ 18, -26 ≤ <i>l</i> ≤ 27
Reflections collected / unique	41838 / 15694 [<i>R</i> _{int} = 0.1860]
Completeness to <i>θ</i> = 27.56	97.1 %
Absorption correction	Semi-empirical from equivalents
Max. and min. transmission	21.966 and 2.1075
Refinement method	Full-matrix least-squares on <i>F</i> ²
Data / restraints / parameters	15694 / 108 / 547
Goodness-of-fit on <i>F</i> ²	1.106
Final R indices [<i>I</i> > 2σ(<i>I</i>)]	<i>R</i> ₁ = 0.1921, <i>wR</i> ₂ = 0.4431
R indices (all data)	<i>R</i> ₁ = 0.2527, <i>wR</i> ₂ = 0.4910
Largest diff. peak and hole (e.Å ⁻³)	5.647 and -1.090
CCDC	1060803