

Supporting Information

‘Honeycomb’ nanotube assembly based on thiocalix[4]arene derivatives by weak interactions

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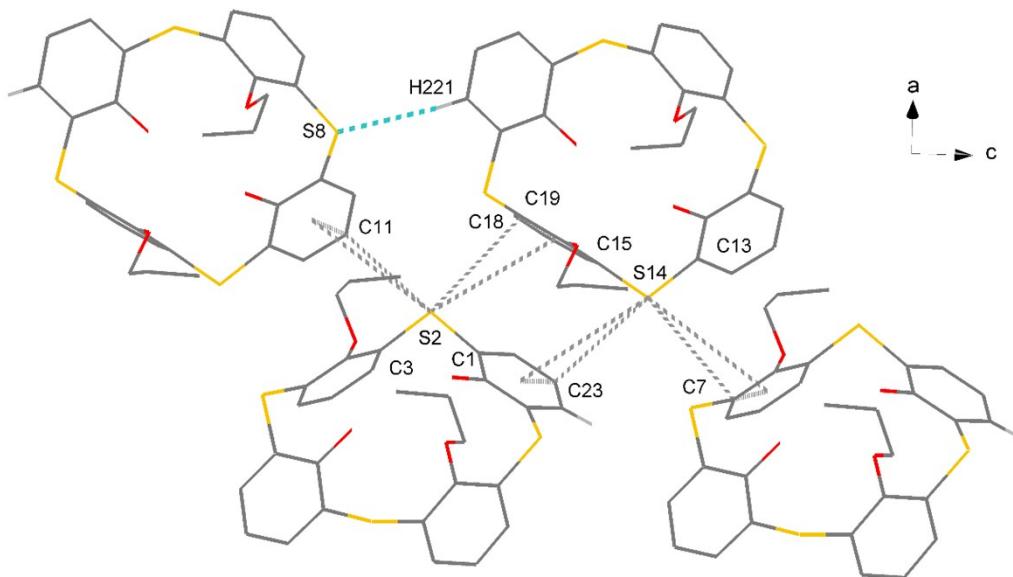


Figure S1 The intermolecular C–H \cdots S and S \cdots π interactions in crystals of **2**. C–H \cdots S and S \cdots π interactions are aqua and gray dot lines, respectively. Hydrogen atoms (except for those involved in hydrogen bonding) are omitted for clarity.

Atoms involved D–H \cdots A	Symmetry	H \cdots A	D \cdots A	\angle (DHA)
C(22) - H(221) \cdots S(8)	x, y, z+1	3.02(4)	3.795(3)	139(3)

Compound	Typ	S \cdots Cg	Symmetry	r/ \AA	d/ \AA	$\alpha/\alpha'(\text{ }^\circ)$	$\phi(\text{ }^\circ)$	
2	III	S(2) \cdots Cg(2)	1/3+x-y, 1/3+x, 2/3-z	-	4.194	C(11)	3.463	
		S(2) \cdots Cg(3)	1/3+x-y, 1/3+x, 5/3-z	-	4.104	C(18)-C(19)	3.650	
	III	S(14) \cdots Cg(1)	1/3+y, 2/3-x+y, 2/3-z	4.250	C(7)	72.62/157.94	59.95	
		S(14) \cdots Cg(4)	1/3+y, 2/3-x+y, 5/3-z	4.241	C(23)	75.15/156.22	58.92	
The parameter values of r , d , α , α' , and ϕ lie in the allowable range of S \cdots π interactions. ^{S1}								

Reference

S1 C. –Q. Wan, J. Han and C. W. M. Thomas, *New J. Chem.*, 2009, **33**, 707–712

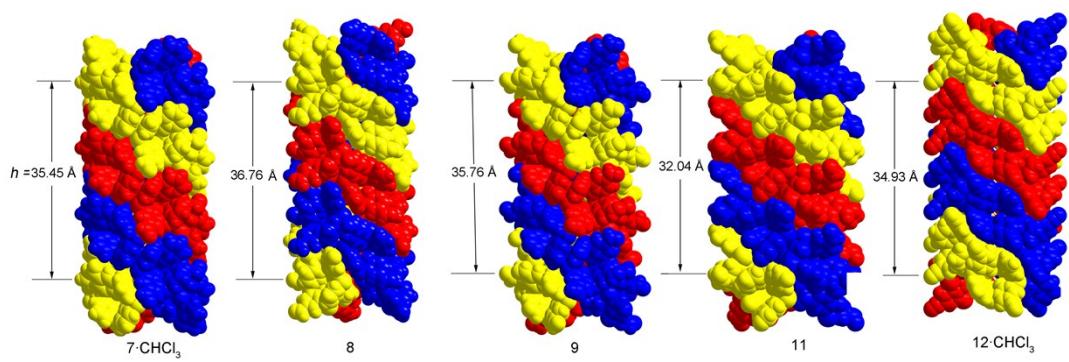


Figure S2

Space filling view of the triply helical nanotube motif are found in the crystal structure of five compounds. Side view showing helical nature of strands coloured yellow, red and blue. Other molecular components omitted for clarity.

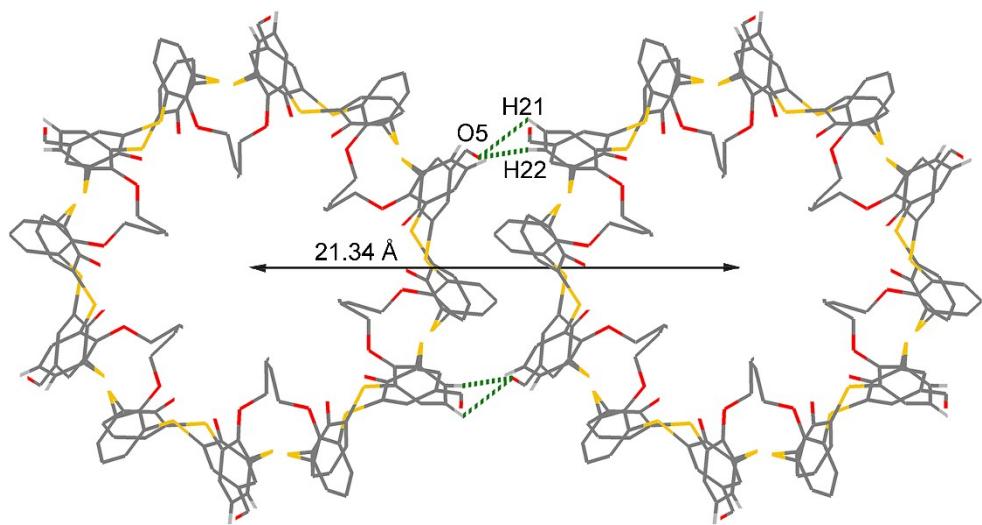


Figure S3 The C-H···O interactions are found in the packing between neighbouring hexameric discs of **7·CHCl₃**. The green dotted lines represent C-H···O interactions. Hydrogen atoms (except for those involved in hydrogen bonding) and CHCl₃ molecules not involved are omitted for clarity.

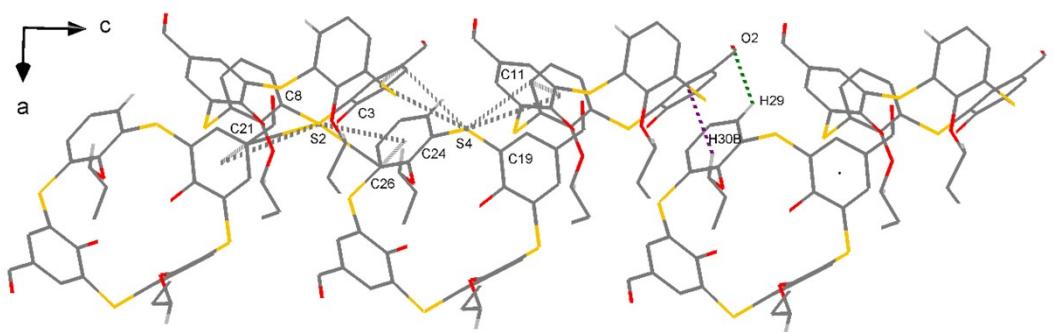


Figure S4 View along the crystallographic *b* axis, the intermolecular C-H···O, C-H···π and S···π interactions are green, violet and gray dot lines, respectively, observed in the triply helical nanotubes of **8**. Hydrogen atoms (except for those involved in hydrogen bonding) are omitted for clarity.

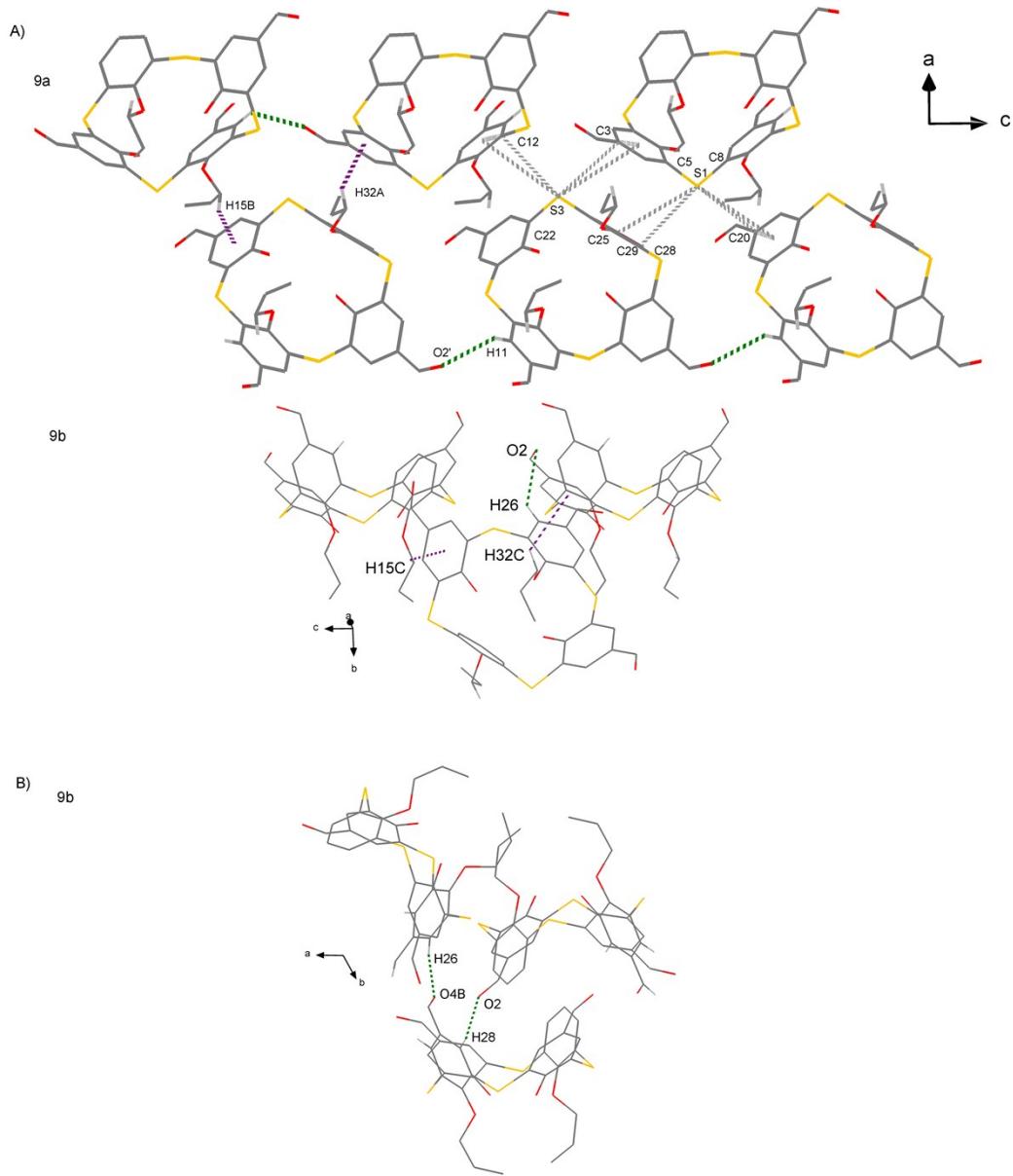


Figure S5

(A) View along the crystallographic *b* axis, the intermolecular C-H···O, C-H··· π and S··· π interactions are green, violet and gray dot lines, respectively, observed in the triply helical nanotubes of **9a** and **9b**.

(B) The C-H···O interactions observed in neighbouring nanotubes of **9b**. C-H···O interactions shown as dashed green lines.

Hydrogen atoms (except for those involved in hydrogen bonding) are omitted for clarity.

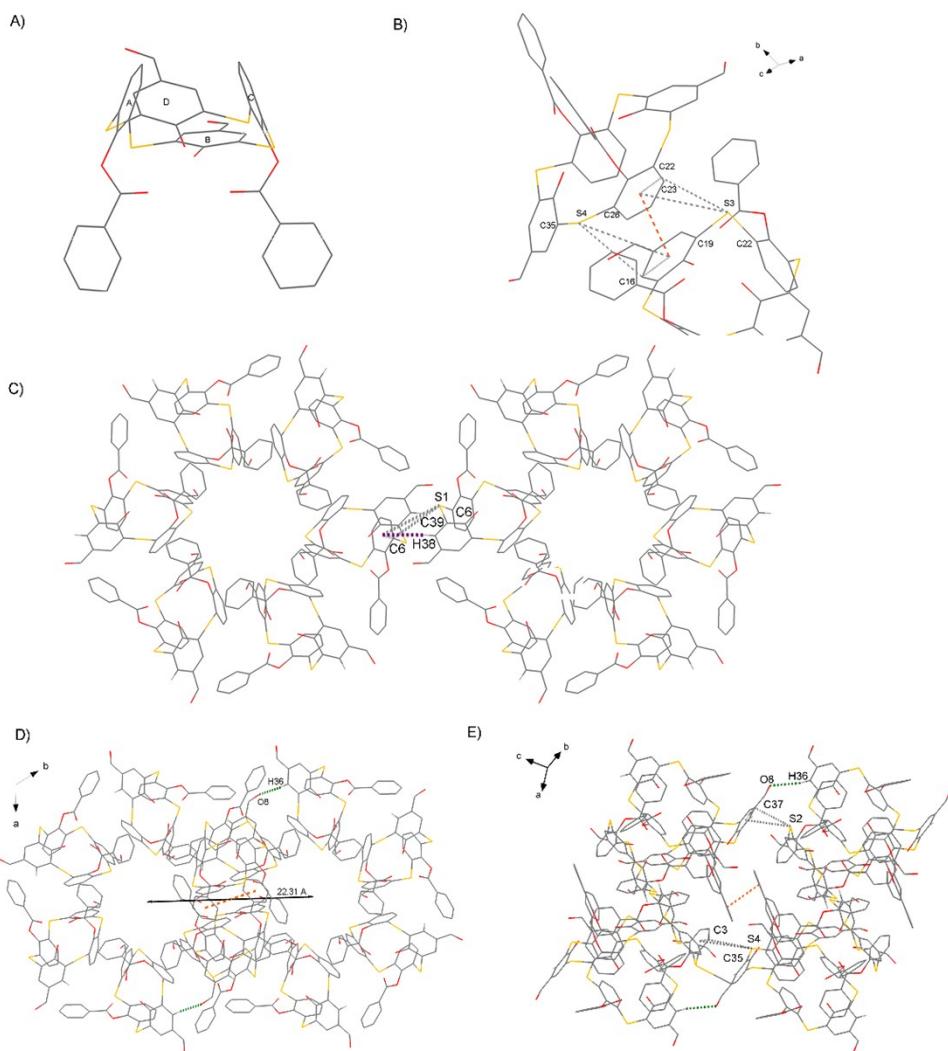


Figure S6

- (A) Schematic of the structure of **10**.
 - (B) The S···π interactions and π···π interactions observed in hexameric disc.
 - (C) Non-covalent interactions the same layer neighbouring hexameric of **10·H₂O** observed in 2D network.
 - (D) The different layer neighbouring hexameric of **10·H₂O** observed in 3D network.
 - (E) Non-covalent interactions the adjacent layer neighbouring hexameric of **10·H₂O** observed in 3D network.
- C-H···O interactions, C-H···π interactions, and S···π interactions are green, violet and gray dot lines, respectively. Hydrogen atoms (except for those involved in hydrogen bonding) and H₂O are omitted for clarity.

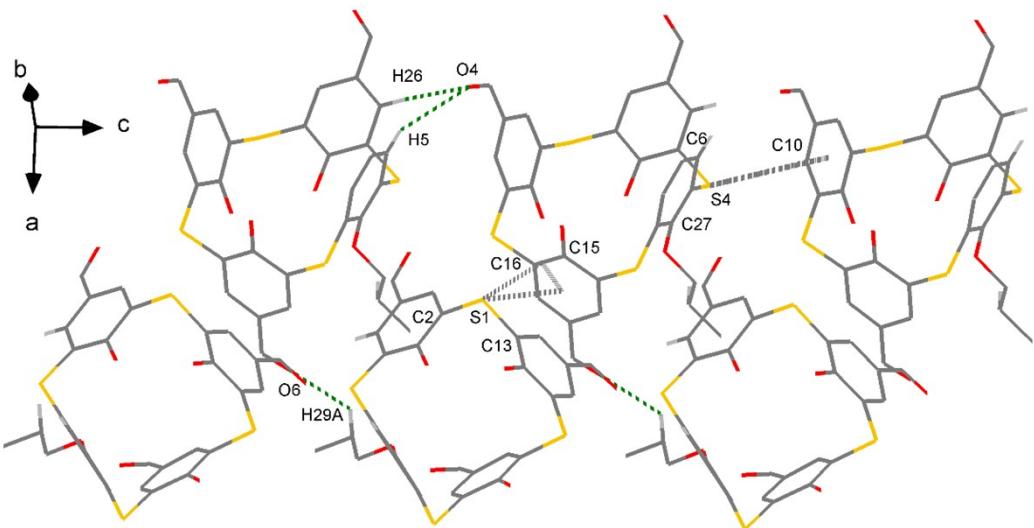


Figure S7 View along the crystallographic *b* axis, the intermolecular C-H···O and S··· π interactions are green and gray dot lines, respectively, observed in the triply helical nanotubes of **11**. Hydrogen atoms (except for those involved in hydrogen bonding) are omitted for clarity

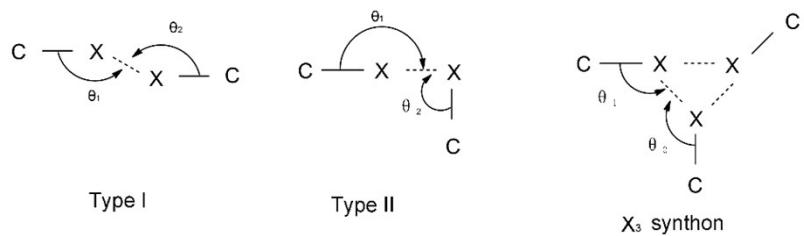


Figure S8 Schematic representations of Type I, Type II, and X_3 synthon halogen···halogen contacts. The angles: Type I ($\theta_1=\theta_2=140\text{--}180^\circ$) and Type II ($\theta_1=150\text{--}180^\circ$, $\theta_2=90\text{--}120^\circ$). The X_3 synthon is a trigonal array of halogen atoms with attractive electrophile-nucleophile Type II contacts.

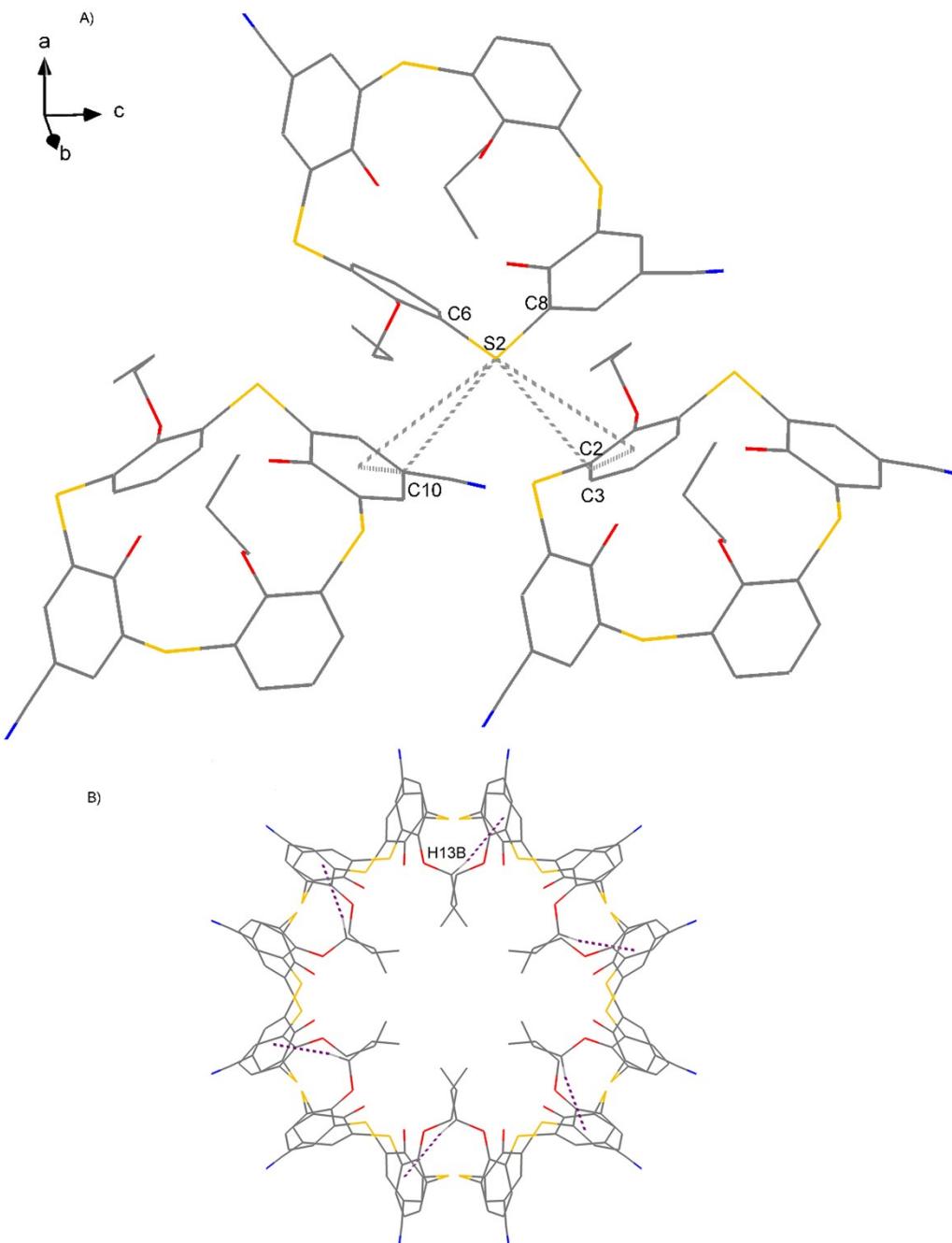


Figure S9

(A) The intermolecular $S\cdots\pi$ interactions are gray dot lines observed in the triply helical nanotubes of **12·CHCl₃**.

(B) Structure of **12·CHCl₃** showing hexameric discs through the $C-H\cdots\pi$ interactions. $C-H\cdots\pi$ interactions shown as dashed orange lines.

Figure S10 ^1H NMR and ESI – MS of compound 7

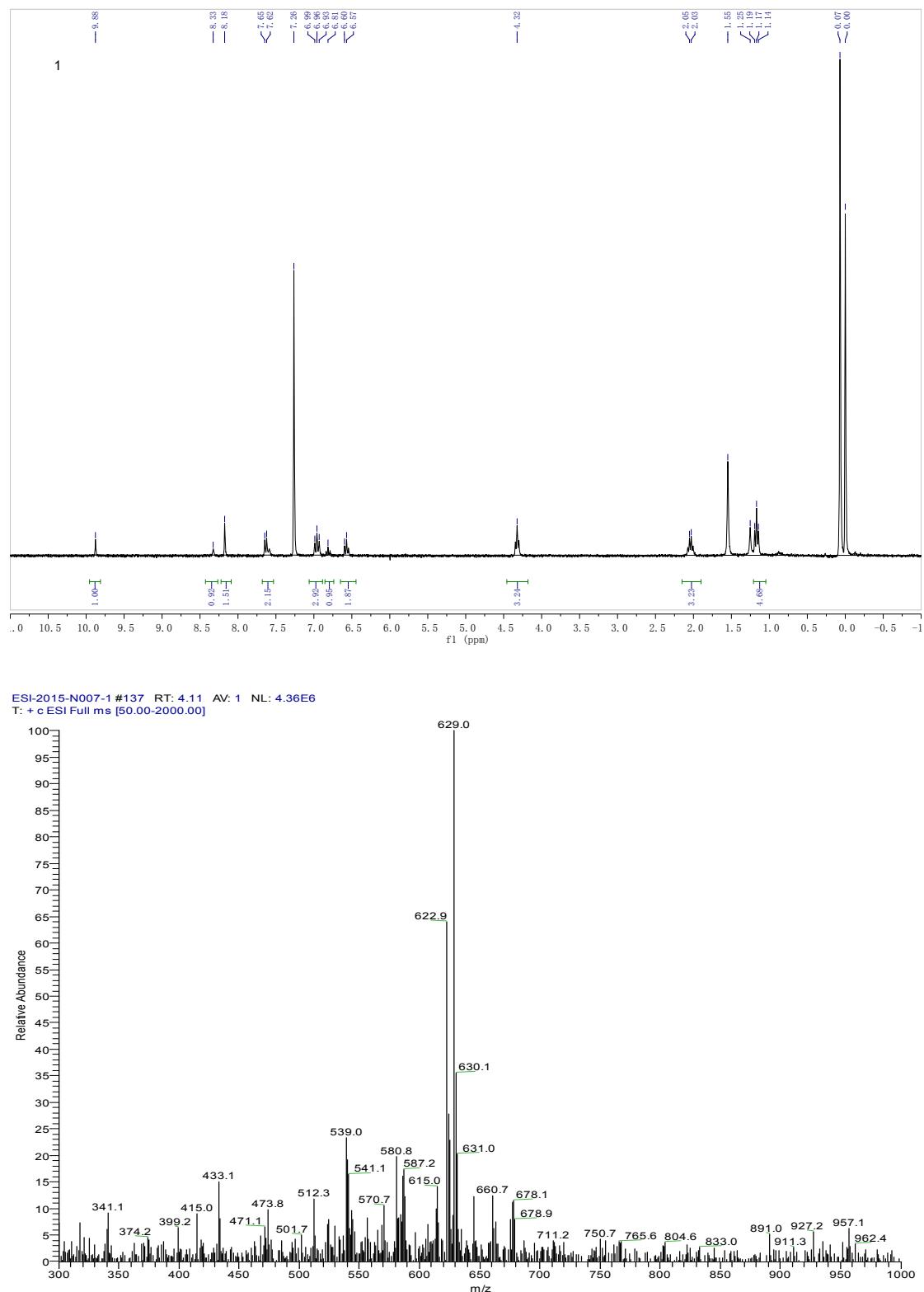


Figure S11 ^1H NMR and ESI – MS of compound **12**

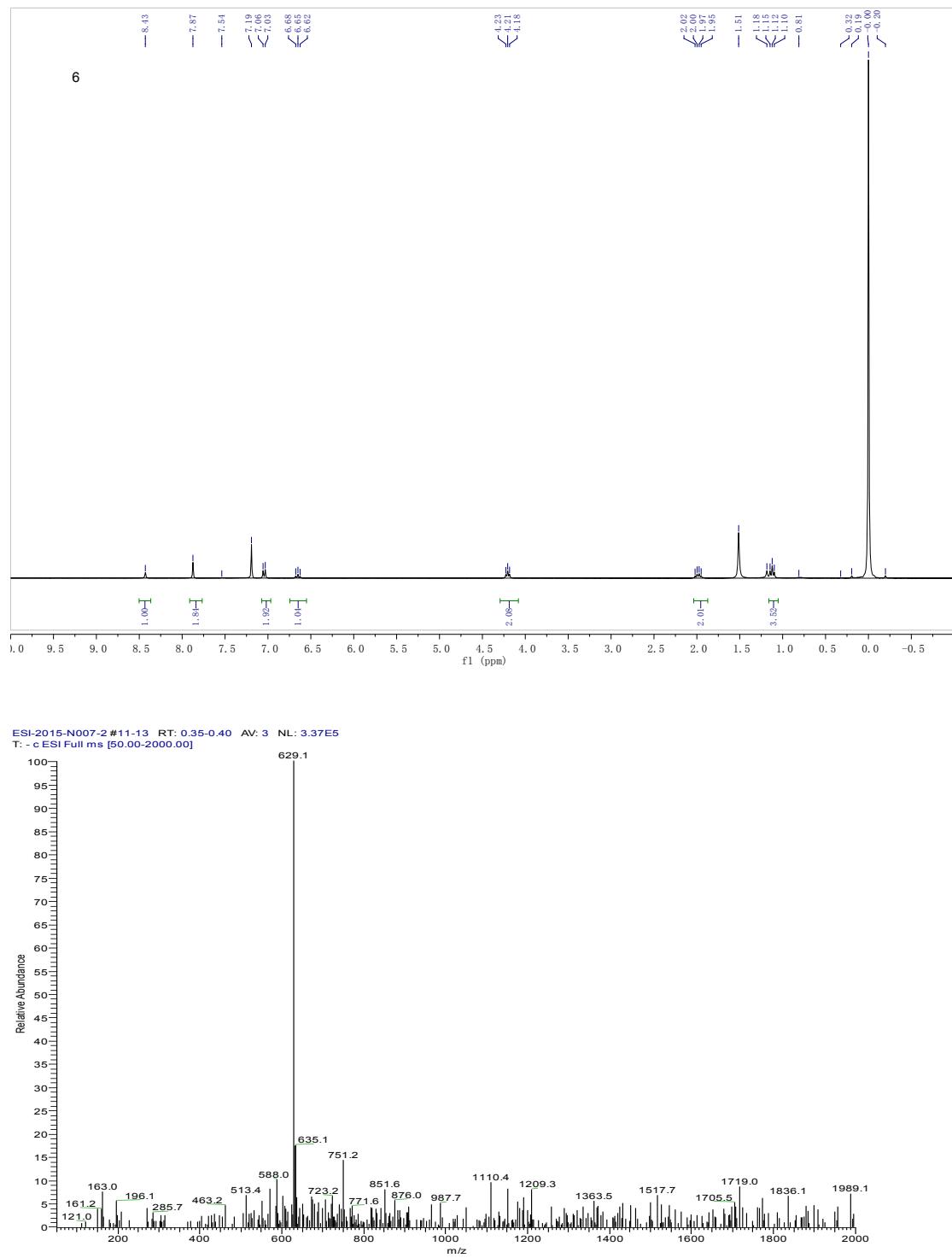


Table S1 Dihedral angles of aromatic rings and reference molecular plane R of the seven compounds

Compound	Plane AR (°)	Plane BR (°)	Plane CR (°)	Plane DR (°)
2	70.34	35.71	71.19	30.57
7·CHCl₃	71.09(68.14)	33.52(30.60)	73.51(76.51)	37.89(38.77)
8	77.15	32.69	65.91	44.37
9	75.11	43.26	69.95	38.63
10·H₂O	110.16	20.73	109.25	28.42
11	78.60(80.25)	89.92(89.43)	73.46(70.94)	43.25(35.64)
12·CHCl₃	68.73	43.97	68.73	43.97

() represents the disordered molecular

Table S2 Dihedral angles of the opposite aromatic rings of the seven compounds

Compound	Plane AC (°)	Plane BD (°)
2	38.48	113.72
7·CHCl₃	35.40	109.6(110.63)
8	36.96	102.97
9	34.97	98.10
10·H₂O	39.42	130.59
11	27.98(29.39)	46.84(52.80)
12·CHCl₃	42.54	92.06

() represents the disordered molecular

Table S3 The three types of $S \cdots \pi$ interactions in compounds **7-12** and their metric parameters

Compound	Type	$S \cdots Cg$	Symmetry			$\alpha/\alpha'(^{\circ})$	$\phi (^{\circ})$
				$r / \text{\AA}$	$d / \text{\AA}$		
7·CHCl₃	III	$S(2) \cdots Cg(1)$	$1/3+x-y, -1/3+x,$ $2/3-z$	4.357	C(6)	3.820	74.72/154.99
		$S(2) \cdots Cg(4)$	$1/3+x-y, -1/3+x,$ $5/3-z$	4.150	C(30)	3.599	168.48/71.26
	III	$S(4) \cdots Cg(2)$	$1/3+y, 2/3-x+y,$ $2/3-z$	4.193	C(14)	3.536	76.30/177.34
		$S(4) \cdots Cg(3)$	$1/3+y, 2/3-x+y,$ $5/3-z$	4.127	C(19) - C(20)	3.742	159.21/72.13
	8	$S(2) \cdots Cg(3)$	$1+y, 1-x+y, 2-z$	4.495	C(21)	3.701	165.26/61.41
		$S(2) \cdots Cg(4)$	$1+y, 1-x+y, 1-z$	4.216	C(26)	3.675	77.09/163.53
9	III	$S(4) \cdots Cg(1)$	$x-y, -1+x, 1-z$	4.258	C(5)	3.658	75.23/170.75
		$S(4) \cdots Cg(2)$	$x-y, -1+x, 2-z$	4.174	C(11)	3.572	163.49/67.04
	III	$S(1) \cdots Cg(3)$	$1/3+y, 2/3-x+y,$ $-1/3-z$	4.152	C(20)	3.739	175.33/74.07
		$S(1) \cdots Cg(4)$	$1/3+y, 2/3-x+y,$ $2/3-z$	4.149	C(28) - C(29)	3.684	69.63/159.33
	III	$S(3) \cdots Cg(1)$	$1/3+x-y, -1/3+x,$ $2/3-z$	4.252	C(3)	3.645	167.27/67.43
		$S(3) \cdots Cg(2)$	$1/3+x-y, -1/3+x,$ $-1/3-z$	4.167	C(12)	3.696	74.92/160.11
10·H₂O	II	$S(1) \cdots Cg(1)$	$1-x, -y, -z$	4.753	C(5)	3.775	118.0/83.91
		$S(2) \cdots Cg(6)$	$4/3-x+y, -2/3-x,$ $-1/3+z$	4.254	C(36) - C(37)	3.588	102.66/87.08
	II	$S(3) \cdots Cg(4)$	$1+y, 1-x+y, -z$	4.082	C(22) - C(23)	3.661	62.1/134.74
		$S(4) \cdots Cg(1)$	$2/3-y, -2/3+x-y,$ $-1/3+z$	4.891	C(3)	3.884	167.86/92.13
	III	$S(4) \cdots Cg(3)$	$x-y, -1+x-y, -z$	4.106	C(16)	3.664	58.79/135.44
		$S(1) \cdots Cg(3)$	$y, 1-x+y, 1-z$	3.979	C(15) - C(16)	3.551	164.08/75.55
11	II	$S(3) \cdots Cg(1)$	$1+x-y, x, 2-z$	3.620	C(5)	3.394	129.52/104.0
		$S(4) \cdots Cg(2)$	$x, y, 1+z$	4.260	C(10)	3.376	111.79/124.14
	III	$S(2) \cdots Cg(1)$	$-x+y, y, 1/2+z$	3.946	C(2)-C(3)	3.566	161.62/76.81
		$S(2) \cdots Cg(2)$	$y, -x+y, 2-z$	4.215	C(10)	3.778	71.82/173.38
12·CHCl₃	III	$S(2) \cdots Cg(1)$	$-x+y, y, 1/2+z$	3.946	C(2)-C(3)	3.566	161.62/76.81
		$S(2) \cdots Cg(2)$	$y, -x+y, 2-z$	4.215	C(10)	3.778	62.38

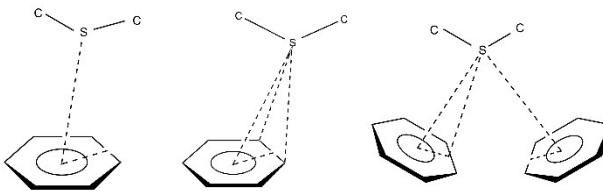


Table S4 $\pi \cdots \pi$ interactions for five compounds

Compound	Cg(I)…Cg(J)	Symmetry	α	Cg…Cg	Cg(I)_Perp	Cg(J)_Perp
7·CHCl₃	Cg(1)…Cg(1)	1-x,-y,1-z	0	4.050(3)	3.446(2)	3.447(2)
8	Cg(4)…Cg(4)	-x+5/3,-y+1/3,-z+4/3	0	4.047(2)	3.468(1)	3.468(1)
9	Cg(2)…Cg(2)	-x+1,-y,-z	0	4.708(3)	3.431(2)	3.431(2)
10·H₂O	Cg(3)…Cg(4)	1+y,1-x-y,-z	0	3.725(3)	28.216(5)	3.513(2)
	Cg(5)…Cg(5)	5/3-x,1/3-y,1/3-z	0	4.801(5)	-3.323(3)	-3.323(3)
11	Cg(4)…Cg(4)	-x+4/3,-y+5/3,-z+5/3	0	4.300(4)	-3.509(2)	-3.509(2)

Table S5 C-H… π interactions for six compounds

Compound	Atoms involved C-H…Cg	Symmetry	H…Cg	C…Cg	<C-H…Cg
7·CHCl₃	C(3) - H(3)…Cg(2)	1-x, -y, 1-z	3.183	4.064(4)	154.8
	C(4) - H(4A)…Cg(3)	1-x, -y, 1-z	3.179	3.818(4)	126.1
	C(5) - H(5)…Cg(4)	1-x, -y, 1-z	3.123	4.007(1)	155.8
	C(7) - H(7A)…Cg(2)	1/3+y, 2/3-x+y, 2/3-z	2.77	3.520(7)	132.6
	C(23) - H(23A)…Cg(4)	1/3+x-y, -1/3+x, 5/3-z	2.93	3.790(10)	136.1
	C(23) - H(23A)…Cg(5)	1/3+x-y, -1/3+x, 5/3-z	2.93	3.43(3)	134.0
	C(13) - H(13)…Cg(1)	1/3+y, 2/3-x+y, 2/3-z	3.138	3.603(4)	112.0
8	C(30) - H(30)…Cg(1)	x-y, x-1, -z+1	2.66	3.438(4)	136
9	C(15') - H(15B)…Cg(3)	1/3+y, 2/3-x+y, -1/3-z	2.96	3.89(3)	161
	C(32') - H(32A)…Cg(1)	1/3+x-y, -1/3+x, 2/3-z	2.92	3.77(5)	147
	C(15) - H(15C)…Cg(3)	1/3+y, 2/3-x+y, -1/3-z	2.90	3.64(2)	134
	C(32) - H(32C)…Cg(1)	1/3+x-y, -1/3+x, 2/3-z	2.96	3.61(2)	126
10·H₂O	C(38) - H(38)…Cg(1)	1-x, -y, 1-z	3.113	4.041(5)	175.9
11	C(24) - H(24)…Cg(1)	-x+4/3, -y+5/3, -z+5/3	2.65	3.476(9)	148
	C(25) - H(25)…Cg(2)	-x+4/3, -y+5/3, -z+5/3	2.90	3.442(6)	118
	C(25') - H(25')…Cg(2)	-x+4/3, -y+5/3, -z+5/3	2.80	3.526(12)	136
12·CHCl₃	C(13) - H(13B)…Cg(2)	-x+ y, y, -1/2+z	2.84	3.682(6)	146.6

Table S7 Hydrogen bonds of compound **7·CHCl₃**

Atoms involved D-H···A	Symmetry	H···A	D···A	∠(DHA)
C(24) - H(24A)···Cl(1)		2.86(5)	3.76(3)	134(3)
C(25) - H(25B)···Cl(2)		2.99(9)	4.053(9)	167(6)
C(7) - H(7B)···S(1)		3.02(5)	3.520(5)	112(3)
C(31) - H(31)···S(1)	x, y, 1+z	2.97(5)	3.757(5)	141(4)
C(31') - H(31)···S(1)	x, y, 1+z	2.97(5)	3.757(5)	119(3)
C(23) - H(23A)···S(2)		3.03(5)	3.572(6)	116(3)
C(7) - H(7B)···S(2)	y+1/3, -x+y+2/3, -z+2/3	3.02(6)	3.890(4)	146(4)
C(23) - H(23B)···S(3)		2.83(10)	3.569(7)	140(7)
C(23) - H(23B)···S(3')		2.18(11)	2.92(4)	138(7)
C(20) - H(20)···O(5)	x, y, 1+z	2.64(7)	3.263(6)	127(5)
C(21) - H(21)···O(5)	4/3-x+y, 2/3-x, 2/3+z	2.47(5)	3.128(5)	121(4)
C(22) - H(22)···O(5)	4/3-x+y, 2/3-x, 2/3+z	2.60(4)	3.046(5)	106(3)
O(2) - H(2)···S(1)		2.68(8)	3.077(3)	114(6)
O(4) - H(4)···S(3)		2.43(13)	3.071(6)	114(8)
O(4) - H(4)···S(3')		1.96(14)	2.794(10)	127(9)
O(4') - H(4)···S(3)		2.43(13)	3.38(13)	140(13)
O(4') - H(4)···S(3')		1.96(14)	3.01(11)	151(13)
O(2) - H(2)···O(1)		2.01(10)	2.758(4)	167(8)
O(4) - H(4)···O(3)		1.81(14)	2.834(10)	147(10)
O(4') - H(4)···O(3)		1.81(14)	2.67(4)	128(11)

Table S8 Hydrogen bonds of compound **8**

Atoms involved D-H···A	Symmetry	H···A	D···A	∠(DHA)
C(30) - H(30B)···S(2)	x-y, x-1, -z+1	3.05(3)	3.845(4)	138.8(6)
C(30) - H(30A)···S(4)		2.95(1)	3.453(4)	112.5(3)
C(12) - H(12)···O(2)	4/3-y, x-y-1/3, z+2/3	2.59(3)	3.388(4)	141.4(1)
C(13) - H(13)···O(2)	-x+y+5/3, -x+4/3, z+1/3	2.46(6)	3.266(4)	142.5(2)
C(29) - H(29)···O(2)	x-y, x-1, -z+1	2.87(1)	3.269(4)	106.4(2)
C(6) - H(6)···O(5)	-x+5/3, -y+1/3, -z+4/3	2.52(2)	3.399(4)	154.3(2)
C(7) - H(7)···O(5)	-x+5/3, -y+1/3, -z+4/3	2.58(3)	3.444(5)	150.8(1)
C(11) - H(11)···O(5)	-x+5/3, -y+1/3, -z+7/3	2.39(3)	3.176(4)	139.4(2)
O(1) - H(1A)···S(2)		2.64(3)	3.058(2)	114(3)
O(1B) - H(1B)···S(4)		2.61(3)	3.098(2)	120(3)
O(1) - H(1A)···O(3)		2.01(2)	2.781(3)	162(4)
O(4) - H(4A)···S4		1.93(2)	2.707(3)	157(3)

Table S9 Hydrogen bonds of compound **9**

Atoms involved D-H···A	Symmetry	H···A	D···A	∠(DHA)
C(15) - H(15C)···S(1)		2.71(13)	3.25(6)	119(8)
C(15') - H(15C)···S(1)		2.71(13)	3.58(4)	134(8)
C(2) - H(2)···S(2)	x, y, z+1	3.09(7)	3.909(5)	139(5)
C(15') - H(15D)···S(2)		3.03(7)	3.52(2)	117(5)
C(15) - H(15D)···S(3)	y+1/3, -x+y+2/3, -z-1/3	3.00(10)	4.05(5)	152(3)
C(15') - H(15D)···S(3)	y+1/3, -x+y+2/3, -z-1/3	3.00(10)	3.90(2)	162(5)
C(32) - H(32C)···S(3)		2.90(15)	3.51(2)	117(11)
C(32) - H(32D)···S(4)		3.01(10)	3.57(3)	112(8)
C(19) - H(19)···O(2)	-x+1, -y, -z	2.54(7)	3.421(9)	150 (4)
C(24) - H(24)···O(2)	-x+1, -y, -z	2.79(15)	3.464(13)	133(12)
C(26) - H(26)···O(2)	x-y+1/3, x-1/3, -z+2/3	2.99(4)	3.462(8)	108(3)
C(28) - H(28)···O(2)	-x+1, -y, -z+1	2.49(8)	3.302(9)	161(5)
C(11) - H(11)···O(2')	x, y, z-1	2.61(7)	3.30(2)	130(5)
C(28) - H(28)···O(2')	-x+1, -y, -z+1	2.95(7)	3.53(4)	128(4)
C(21) - H(21)···O(4B)	-x+y+4/3, -x+2/3, z-1/3	2.92(6)	3.65(3)	134(3)
C(26) - H(26)···O(4B)	-x+y+4/3, -x+2/3, z-1/3	2.09(6)	3.11(2)	164(4)
C(26) - H(26)···O(6)	-y+2/3, x-y-2/3, z+1/3	2.74(5)	3.452(9)	126(3)
O(1) - H(1')···S(1)		2.50(17)	3.014(7)	119(7)
O(5) - H(5)···S(3)		2.6(2)	2.982(5)	125(12)
O(1) - H(1)···S(4)		2.42(11)	2.993(10)	117(7)
O(1) - H(1')···O(3)		2.24(16)	2.893(9)	151(7)
O(1) - H(1)···O(7)		2.04(13)	2.922(7)	148(8)
O(5) - H(5)···O(7)		2.3(2)	2.942(7)	149(12)

Table S10 Hydrogen bonds of compound **10·H₂O**

Atoms involved D-H···A	Symmetry	H···A	D···A	∠(DHA)
C(11) - H(11)···O(1)	1-y, x-y, z	2.852(3)	3.693(8)	151.2(3)
C(36) - H(36)···O(8)	y+2/3, -x+y+1/3, -z+1/3	2.735(1)	3.657(6)	170.0(3)
O(7) - H(7)···S(1)		2.51(6)	3.034(3)	122.8(4)
O(3) - H(3)···S(3)		2.644(4)	3.080(3)	114.9(3)
O(7) - H(7)···O(2)		2.04(5)	2.725(4)	140.9(2)
O(3) - H(3)···O(6)		1.92(1)	2.663(4)	150.7(5)

Table S11 Hydrogen bonds of compound **11**

Atoms involved D-H···A	Symmetry	H···A	D···A	∠(DHA)
C(28) - H(28B)···S(2)	x, y, z+1	3.06(5)	4.006(4)	165.4(6)
C(28') - H(28D)···S(2')	x, y, z+1	3.01(5)	3.885(6)	150.4(3)
C(10) - H(10)···S(4')	x, y, -1+z	3.04(5)	3.500(5)	112.0(3)
C(28') - H(28D)···S(4')		2.59(1)	3.203(6)	121.2(4)
C(3) - H(3A)···O(2)	-x+y+1/3, -x+5/3, z-1/3	2.53(1)	3.233(4)	143.8(2)
C(7) - H(7)···O(2)	-y+5/3, x-y+4/3, z+1/3	2.53(1)	3.133(4)	122.5(4)
C(12) - H(12)···O(2)	-x+y+1/3, -x+5/3, z-1/3	2.75(2)	3.644(4)	168.2(1)
C(14) - H(14)···O(2)	-y+5/3, x-y+4/3, z-2/3	2.90(1)	3.490(4)	122.5(3)
C(5) - H(5)···O(4)	x, y, z+1	2.82(1)	3.702(5)	159.2(3)
C(10) - H(10)···O(4)	-x+4/3, -y+5/3, -z+2/3	2.63(1)	3.310(4)	130.9(1)
C(26) - H(26)···O(4)	x, y, z+1	2.91(1)	3.778(6)	155.1(5)
C(26) - H(26)···O(5)	-x+4/3, -y+5/3, -z+5/3	2.73(1)	3.539(7)	145.8(3)
C(29) - H(29A)···O(6)	y, -x+y+1, -z+2	2.54(1)	3.174(13)	122.9(2)
C(17') - H(17')···O(6')	y, -x+y+1, -z+1	2.43	3.34(\$)	165.5(6)
O(3) - H(3)···S(1)		2.54(2)	3.030(3)	119.8(2)
O(5) - H(5A)···S(2)		2.57(1)	3.072(5)	120.7(4)
O(1) - H(1)···S(4)		2.52(1)	3.000(2)	118.7 (1)
O(1) - H(1)···S(4')		2.48(1)	3.041(4)	126.7 (2)
O(3) - H(3)···O(1)		2.36	2.919(3)	126.5(1)
O(5) - H(5A)···O(4)	-x+4/3, -y+5/3, -z+2/3	2.02	2.690(5)	138.2(5)
O(5') - H(5')···O(4)	-x+4/3, -y+5/3, -z+2/3	2.58	2.99(1)	112.9(5)
O(1) - H(1)···O(7)		2.19	2.985(6)	163.6(3)
O(1) - H(1)···O(7')		2.28	3.040(16)	155.0(1)

Table S12 Hydrogen bonds of compound **12·CHCl₃**

Atoms involved D - H···A	Symmetry	H···A	D···A	$\angle(\text{DHA})$
O(2) - H(2)···S(2)		2.61	3.059(7)	114.8
O(2) - H(2)···O(1)		2.31	2.810(8)	119.6

Table S13 The Cl···Cl interactions of compound **12·CHCl₃**

	Symmetry	$d(\text{Cl}\cdots\text{Cl})/\text{\AA}$	$d(\text{C}\cdots\text{Cl})/\text{\AA}$	$\theta_1 = \theta_2(^{\circ})$
C(17) - Cl(2)···Cl(2) - C(17)	1-x, 2-y, -z	2.921	4.492	151.43

Table S14 The cyano···Cl interactions of compound **12·CHCl₃**

	Symmetry	$d(\text{N}\cdots\text{Cl})/\text{\AA}$	$d(\text{C}^{\alpha}\cdots\text{Cl})/\text{\AA}$	$d(\text{C}^{\beta}\cdots\text{N})/\text{\AA}$	$\theta_1(^{\circ})$	$\theta_2(^{\circ})$
C ^α - N(1)···Cl(3) - C ^β	x-y, x, 2-z	2.779	3.786	4.285	147.94	149.85

Table S15 The Cl···π interactions of compound **12·CHCl₃**

C - Cl···Cg	Symmetry	$r/\text{\AA}$	$d/\text{\AA}$	$\alpha(^{\circ})$	$\phi(^{\circ})$
C(17) - Cl(1)···Cg(1)	y, 1-x+y, 1-z	3.403(6)	C(1) - C(6)	3.470	104.52(12)
C(17) - Cl(1)···Cg(2)	x, 1+x-y, -1/2+z	3.928(3)	C(7)	3.448	106.87(10)

Table S16 Sums of the van der Waals radii (vdW) of C and phenyl ring with common lp-containing atoms (atom LP)^{S2}

Atom LP	vdW radii/ Å	vdW(LP) + vdW(C ^a)/ Å	vdW(LP) + vdW(Phenyl ring)/ Å
H	1.20	2.90	3.05
O	1.52	3.22	3.37
Cl	1.75	3.40	3.45
S	1.80	3.50	3.65
N	1.55		
C ^a	1.70		
Phenyl ring ^{S3}	1.85		

Reference

S2 Bondi, *J. Phys. Chem.*, 1964, 68, 441–451.

S3 F. Malone, C. M. Murray, M. H. Charlton, R. Docherty and J. Lavery, *J. Chem. Soc., Faraday Trans.*, 1997, 93, 3429–3436.

Table S17 The partial site occupation factors of four compounds

Compound	Split atoms	Occupation factors
7·CHCl₃	C(26)-C(31), S(3), O(4)/C(26')-C(31'), S(3'), O(4')	0.859(16)/0.141(16)
8	C(14)-C(16)/C(14')-C(16')	0.826(8)/0.174(8)
9	C(15)-C(17)/C(15')-C(17')	0.57/0.43
	C(32)-C(34)/C(32')-C(34')	0.66/0.34
	C(14)/C(14B)	0.75/0.25
	O(4), O(4A)/O(4B),	0.375, 0.375/0.25
	O(2)/O(2')	0.69/0.31
	O(6)/O(6')	0.78/0.22
11	C(15)-C(30), S(2), S(3), S(4), O(5), O(6), O(7)/ C(15')-C(30'), S(2'), S(3'), S(4'), O(5'), O(6'), O(7')	0.707(3)/0.293(3)