Supporting Information

Hollow nanocrystals of silicoaluminophosphate molecular sieve synthesized by an aminothermal co-templating strategy

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Table S1. Interaction energies for the SDA in different cavities. (The calculation is based on a unit cell, which includes two different *gme* (*gme*1 and *gme*2) and two different *aft* cages (*aft*1 and *aft*2)

	$E_{\text{interaction energy}}$ (kcal/mol)
One TMA in a <i>gme</i> 1 cavity and One	-30.18
TMA in a <i>gme</i> 2 cavity	
Two TMAs in a <i>gme</i> 1 cavity and Two	247.37
TMAs in a <i>gme</i> 2 cavity	
One TMA in a <i>aft</i> 1 cavity and One TMA	-29.18
in a <i>aft</i> 2 cavity	
One TEA in a <i>aft</i> 1 cavity and One TEA	-53.18
in a <i>aft</i> 2 cavity	
Two TEAs in a <i>aft</i> 1 cavity and Two	-103.2
TEAs in a <i>aft</i> 2 cavity	
Three TEAs in a <i>aft</i> 1 cavity and Three	2424.11
TEAs in a <i>aft</i> 2 cavity	



Fig. S1 SEM images of the calcined samples 1 (0TMA), 2 (0.12TMA), 3 (0.2TMA), 4 (0.3TMA), 5 (0.4TMA) and 6 (0.6TMA).





Fig. S2 XRD patterns of the as-synthesized samples 7 (0.3Si), 4 (0.5Si), 8 (0.75Si), 9 (1.0Si), 10 (0HF), 11 (0.25HF) and 12 (0.5HF) and their corresponding SEM images of the calcined form.



Fig. S3 The XRD pattern of the as-synthesized sample by the route of hydrothermal synthesis (Condition: the gel molar composition of 7TEA/1Al₂O₃/1P₂O₅/0.75SiO₂/0.3TMA/1.0HF/50H₂O at 200 °C for 36 h).



Fig. S4 ²⁷Al and ³¹P MAS NMR of the as-synthesized SAPO-56 (sample 8).



Fig. S5 TG (solid)-DSC (dotted) curves of the sample 8.



Fig. S6 ¹³C MAS NMR spectrum of the SAPO-34 (sample 2.33h). The two peaks are integrated and their ratio is close to 1/1, suggesting that the inclusion in SAPO-34 is only TEA. The peak centered at 48 and 9.5 ppm can be ascribed to the $-CH_2$ - and $-CH_3$ groups of TEA.

The ¹⁹F MAS NMR spectra were recorded on a Bruker AVANCE III 400 WB spectrometer equipped with a 2.5 mm standard bore CP MAS probehead whose X channel was tuned to 376.55 MHz for ¹⁹F, using a magnetic field of 9.39 T at 297 K. The dried and finely powdered samples were packed in the ZrO_2 rotor closed with Kel-F cap which were spun at 20 kHz rate. A total of 2000 scans were recorded with 3 s recycle delay for each sample. All ¹⁹F MAS NMR chemical shifts are referenced to the resonances of trichlorofluoromethane (CFCl₃) standard (d=0.00).



Fig. S7 ¹⁹F MAS NMR spectra of the samples crystallizing at 2.33 h (SAPO-34) and 36 h (SAPO-56).