

## Electronic Supplementary Information

### **Excitation wavelength-induced color tunable and white-light emissions in lanthanide(III) coordination polymers constructed by an environment-dependent luminescent tetrazolate-dicarboxylate ligand**

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**Table S1** Selected bond lengths (Å) and angles (°) for compounds **1-4**

<b>1</b>			
Sm(1)–O(2)#1	2.344(2)	Sm(1)–O(3)	2.406(2)
Sm(1)–O(1)#2	2.353(2)	Sm(1)–O(2W)	2.445(3)
Sm(1)–O(1W)	2.383(3)	Sm(1)–O(4)#3	2.490(2)
Sm(1)–O(3W)	2.389(3)	Sm(1)–O(3)#3	2.616(2)
O(2)#1–Sm(1)–O(1)#2	88.14(9)	O(3)–Sm(1)–O(2W)	88.41(9)
O(2)#1–Sm(1)–O(1W)	97.65(11)	O(2)#1–Sm(1)–O(4)#3	78.34(8)
O(1)#2–Sm(1)–O(1W)	141.75(10)	O(1)#2–Sm(1)–O(4)#3	145.61(10)
O(2)#1–Sm(1)–O(3W)	85.46(11)	O(1W)–Sm(1)–O(4)#3	72.02(10)
O(1)#2–Sm(1)–O(3W)	71.22(10)	O(3W)–Sm(1)–O(4)#3	76.25(9)
O(1W)–Sm(1)–O(3W)	146.70(10)	O(3)–Sm(1)–O(4)#3	117.72(8)

O(2)#1–Sm(1)–O(3)	163.16(9)	O(2W)–Sm(1)–O(4)#3	132.28(8)
O(1)#2–Sm(1)–O(3)	80.77(9)	O(2)#1–Sm(1)–O(3)#3	128.19(8)
O(1W)–Sm(1)–O(3)	83.67(10)	O(1)#2–Sm(1)–O(3)#3	126.77(8)
O(3W)–Sm(1)–O(3)	102.71(10)	O(1W)–Sm(1)–O(3)#3	77.84(10)
O(2)#1–Sm(1)–O(2W)	76.12(9)	O(3W)–Sm(1)–O(3)#3	74.45(9)
O(1)#2–Sm(1)–O(2W)	72.54(9)	O(3)–Sm(1)–O(3)#3	68.57(9)
O(1W)–Sm(1)–O(2W)	72.31(10)	O(2W)–Sm(1)–O(3)#3	144.06(9)
O(3W)–Sm(1)–O(2W)	139.62(10)	O(4)#3–Sm(1)–O(3)#3	50.86(8)

Symmetry transformations used to generate equivalent atoms: for **1**: (#1)  $x, y - 1, z$ ; (#2)  $-x + 1, -y + 2, -z$ ; (#3)  $-x, -y + 1, -z$ .

## 2

Eu(1)–O(3)#1	2.335(2)	Eu(1)–O(1)	2.392(2)
Eu(1)–O(4)#2	2.347(3)	Eu(1)–O(2W)	2.436(2)
Eu(1)–O(3W)	2.368(3)	Eu(1)–O(2)#3	2.480(2)
Eu(1)–O(1W)	2.372(3)	Eu(1)–O(1)#3	2.609(2)
O(3)#1–Eu(1)–O(4)#2	88.27(8)	O(1)–Eu(1)–O(2W)	88.52(9)
O(3)#1–Eu(1)–O(3W)	72.39(9)	O(3)#1–Eu(1)–O(2)#3	146.43(10)
O(4)#2–Eu(1)–O(3W)	85.60(10)	O(4)#2–Eu(1)–O(2)#3	78.46(8)
O(3)#1–Eu(1)–O(1W)	140.88(10)	O(3W)–Eu(1)–O(2)#3	75.92(9)
O(4)#2–Eu(1)–O(1W)	97.59(10)	O(1W)–Eu(1)–O(2)#3	72.06(9)
O(3W)–Eu(1)–O(1W)	146.41(9)	O(1)–Eu(1)–O(2)#3	117.50(8)
O(3)#1–Eu(1)–O(1)	80.59(8)	O(2W)–Eu(1)–O(2)#3	132.16(8)
O(4)#2–Eu(1)–O(1)	163.23(8)	O(3)#1–Eu(1)–O(1)#3	126.82(8)
O(3W)–Eu(1)–O(1)	102.71(10)	O(4)#2–Eu(1)–O(1)#3	128.45(7)
O(1W)–Eu(1)–O(1)	83.61(10)	O(3W)–Eu(1)–O(1)#3	73.90(9)
O(3)#1–Eu(1)–O(2W)	72.00(9)	O(1W)–Eu(1)–O(1)#3	78.23(10)
O(4)#2–Eu(1)–O(2W)	76.07(9)	O(1)–Eu(1)–O(1)#3	68.24(8)

O(3W)–Eu(1)–O(2W)	140.16(9)	O(2W)–Eu(1)–O(1)#3	144.06(8)
O(1W)–Eu(1)–O(2W)	72.04(9)	O(2)#3–Eu(1)–O(1)#3	51.06(8)

Symmetry transformations used to generate equivalent atoms: for **2**: (#1)  $-x + 1, -y + 2, -z$ ; (#2)  $x, y - 1, z$ ; (#3)  $-x, -y + 1, -z$ .

### 3

Gd(1)–O(3)#1	2.318(3)	Gd(1)–O(1)#3	2.365(3)
Gd(1)–O(4)#2	2.322(2)	Gd(1)–O(1W)	2.407(2)
Gd(1)–O(2W)	2.339(3)	Gd(1)–O(2)	2.460(2)
Gd(1)–O(3W)	2.347(3)	Gd(1)–O(1)	2.599(2)
O(3)#1–Gd(1)–O(4)#2	88.04(9)	O(1)#3–Gd(1)–O(1W)	88.30(9)
O(3)#1–Gd(1)–O(2W)	85.29(10)	O(3)#1–Gd(1)–O(2)	78.46(8)
O(4)#2–Gd(1)–O(2W)	71.76(9)	O(4)#2–Gd(1)–O(2)	145.66(9)
O(3)#1–Gd(1)–O(3W)	97.76(11)	O(2W)–Gd(1)–O(2)	75.74(9)
O(4)#2–Gd(1)–O(3W)	141.80(10)	O(3W)–Gd(1)–O(2)	71.93(9)
O(2W)–Gd(1)–O(3W)	146.14(10)	O(1)#3–Gd(1)–O(2)	117.55(8)
O(3)#1–Gd(1)–O(1)#3	163.23(8)	O(1W)–Gd(1)–O(2)	132.24(9)
O(4)#2–Gd(1)–O(1)#3	80.86(9)	O(3)#1–Gd(1)–O(1)	128.67(8)
O(2W)–Gd(1)–O(1)#3	102.90(11)	O(4)#2–Gd(1)–O(1)	126.33(9)
O(3W)–Gd(1)–O(1)#3	83.65(11)	O(2W)–Gd(1)–O(1)	74.02(9)
O(3)#1–Gd(1)–O(1W)	76.36(9)	O(3W)–Gd(1)–O(1)	78.05(10)
O(4)#2–Gd(1)–O(1W)	72.72(9)	O(1)#3–Gd(1)–O(1)	68.04(9)
O(2W)–Gd(1)–O(1W)	140.31(9)	O(1W)–Gd(1)–O(1)	143.73(9)
O(3W)–Gd(1)–O(1W)	72.12(9)	O(2)–Gd(1)–O(1)	51.29(8)

Symmetry transformations used to generate equivalent atoms: for **3**: (#1)  $-x, -y + 2, -z$ ; (#2)  $x + 1, y + 1, z$ ; (#3)  $-x, -y + 3, -z$ .

### 4

Gd(1)–O(3)#1	2.324(2)	Gd(1)–O(1)#3	2.374(2)
Gd(1)–O(4)#2	2.330(2)	Gd(1)–O(1W)	2.414(2)

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Gd(1)–O(2W)	2.350(2)	Gd(1)–O(2)	2.467(2)
Gd(1)–O(3W)	2.347(3)	Gd(1)–O(1)	2.609(2)
Eu(1)–O(3)#1	2.324(2)	Eu(1)–O(1)#3	2.374(2)
Eu(1)–O(4)#2	2.330(2)	Eu(1)–O(1W)	2.414(2)
Eu(1)–O(2W)	2.350(2)	Eu(1)–O(2)	2.467(2)
Eu(1)–O(3W)	2.347(3)	Eu(1)–O(1)	2.609(2)
O(3)#1–Gd(1)–O(4)#2	87.86(8)	O(1)#3–Gd(1)–O(1W)	88.44(8)
O(3)#1–Gd(1)–O(2W)	85.58(9)	O(3)#1–Gd(1)–O(2)	78.46(7)
O(4)#2–Gd(1)–O(2W)	71.65(8)	O(4)#2–Gd(1)–O(2)	145.66(8)
O(3)#1–Gd(1)–O(3W)	97.70(9)	O(2W)–Gd(1)–O(2)	75.97(8)
O(4)#2–Gd(1)–O(3W)	141.71(9)	O(3W)–Gd(1)–O(2)	71.96(9)
O(2W)–Gd(1)–O(3W)	146.32(9)	O(1)#3–Gd(1)–O(2)	117.72(7)
O(3)#1–Gd(1)–O(1)#3	163.08(8)	O(1W)–Gd(1)–O(2)	131.97(8)
O(4)#2–Gd(1)–O(1)#3	80.83(8)	O(3)#1–Gd(1)–O(1)	128.53(7)
O(2W)–Gd(1)–O(1)#3	102.55(9)	O(4)#2–Gd(1)–O(1)	126.57(7)
O(3W)–Gd(1)–O(1)#3	83.82(9)	O(2W)–Gd(1)–O(1)	73.90(8)
O(3)#1–Gd(1)–O(1W)	76.14(8)	O(3W)–Gd(1)–O(1)	78.14(9)
O(4)#2–Gd(1)–O(1W)	72.80(8)	O(1)#3–Gd(1)–O(1)	68.34(8)
O(2W)–Gd(1)–O(1W)	140.37(8)	O(1W)–Gd(1)–O(1)	143.86(8)
O(3W)–Gd(1)–O(1W)	71.92(9)	O(2)–Gd(1)–O(1)	51.16(7)
O(3)#1–Eu(1)–O(4)#2	87.86(8)	O(1)#3–Eu(1)–O(1W)	88.44(8)
O(3)#1–Eu(1)–O(2W)	85.58(9)	O(3)#1–Eu(1)–O(2)	78.46(7)
O(4)#2–Eu(1)–O(2W)	71.65(8)	O(4)#2–Eu(1)–O(2)	145.66(8)
O(3)#1–Eu(1)–O(3W)	97.70(9)	O(2W)–Eu(1)–O(2)	75.97(8)
O(4)#2–Eu(1)–O(3W)	141.71(9)	O(3W)–Eu(1)–O(2)	71.96(9)
O(2W)–Eu(1)–O(3W)	146.32(9)	O(1)#3–Eu(1)–O(2)	117.72(7)
O(3)#1–Eu(1)–O(1)#3	163.08(8)	O(1W)–Eu(1)–O(2)	131.97(8)

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O(4)#2-Eu(1)-O(1)#3	80.83(8)	O(3)#1-Eu(1)-O(1)	128.53(7)
O(2W)-Eu(1)-O(1)#3	102.55(9)	O(4)#2-Eu(1)-O(1)	126.57(7)
O(3W)-Eu(1)-O(1)#3	83.82(9)	O(2W)-Eu(1)-O(1)	73.90(8)
O(3)#1-Eu(1)-O(1W)	76.14(8)	O(3W)-Eu(1)-O(1)	78.14(9)
O(4)#2-Eu(1)-O(1W)	72.80(8)	O(1)#3-Eu(1)-O(1)	68.34(8)
O(2W)-Eu(1)-O(1W)	140.37(8)	O(1W)-Eu(1)-O(1)	143.86(8)
O(3W)-Eu(1)-O(1W)	71.92(9)	O(2)-Eu(1)-O(1)	51.16(7)

Symmetry transformations used to generate equivalent atoms: for 4: (#1)  $-x, -y, -z$ ; (#2)  $x + 1, y + 1, z$ ; (#3)  $-x, -y + 1, -z$ ; (#4)  $x - 1, y - 1, z$ .

**Table S2** CIE coordinates (x, y), CRI and CCT for compound **1**

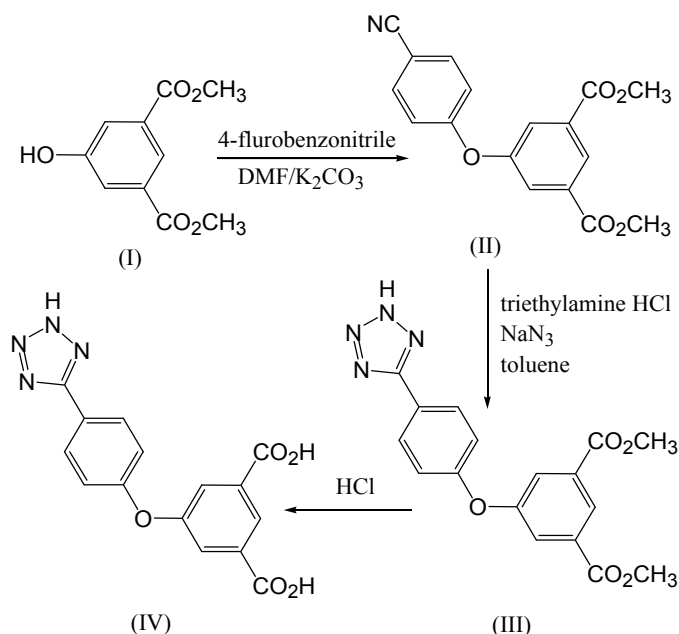
$\lambda_{\text{ex}}/\text{nm}$	(x, y)	CRI	CCT/K
340	(0.53, 0.36)	33	1728
345	(0.49, 0.35)	43	1833
350	(0.44, 0.34)	65	2301
352	(0.41, 0.33)	73	2739
354	(0.38, 0.32)	79	3480
356	(0.35, 0.32)	84	4454
358	(0.33, 0.31)	87	5748
360	(0.31, 0.31)	90	7000
365	(0.29, 0.30)	94	8716
370	(0.26, 0.30)	87	10958
384	(0.25, 0.30)	78	11849

**Table S3** CIE coordinates (x, y), CRI and CCT for compound **4**

$\lambda_{\text{ex}}/\text{nm}$	(x, y)	CRI	CCT/K
340	(0.48, 0.34)	56	1915
345	(0.45, 0.35)	67	2301
350	(0.41, 0.34)	75	3025
355	(0.36, 0.34)	84	4177
360	(0.33, 0.33)	89	5489
365	(0.31, 0.32)	92	6957
370	(0.29, 0.31)	94	8206

**Table S4** CIE coordinates(x, y) for the Eu(III)-doped Gd(III) frameworks

$\lambda_{\text{ex}}/\text{nm}$	<b>Eu<sub>0.009</sub>Gd<sub>0.99</sub></b> 1 (x, y)	<b>Eu<sub>0.02</sub>Gd<sub>0.9</sub></b> 8 (x,y)	<b>Eu<sub>0.04</sub>Gd<sub>0.9</sub></b> 6 (x, y)	<b>Eu<sub>0.06</sub>Gd<sub>0.9</sub></b> 4 (x, y)	<b>Eu<sub>0.07</sub>Gd<sub>0.9</sub></b> 3 (x, y)	<b>Eu<sub>0.11</sub>Gd<sub>0.8</sub></b> 9 (x, y)
340	(0.40, 0.47)	(0.42, 0.45)	(0.43, 0.43)	(0.45, 0.39)	(0.43, 0.41)	(0.48, 0.33)
345	(0.39, 0.47)	(0.40, 0.45)	(0.42, 0.43)	(0.43, 0.40)	(0.41, 0.40)	(0.45, 0.34)
350	(0.38, 0.46)	(0.39, 0.44)	(0.40, 0.43)	(0.40, 0.39)	(0.39, 0.39)	(0.40, 0.33)
355	(0.36, 0.44)	(0.36, 0.42)	(0.37, 0.41)	(0.37, 0.38)	(0.35, 0.37)	(0.36, 0.32)
360	(0.34, 0.41)	(0.34, 0.39)	(0.34, 0.38)	(0.33, 0.36)	(0.32, 0.34)	(0.32, 0.31)
365	(0.31, 0.38)	(0.31, 0.37)	(0.30, 0.35)	(0.31, 0.34)	(0.29, 0.32)	(0.30, 0.30)
370	(0.28, 0.35)	(0.28, 0.34)	(0.28, 0.33)	(0.29, 0.32)	(0.27, 0.30)	(0.28, 0.30)

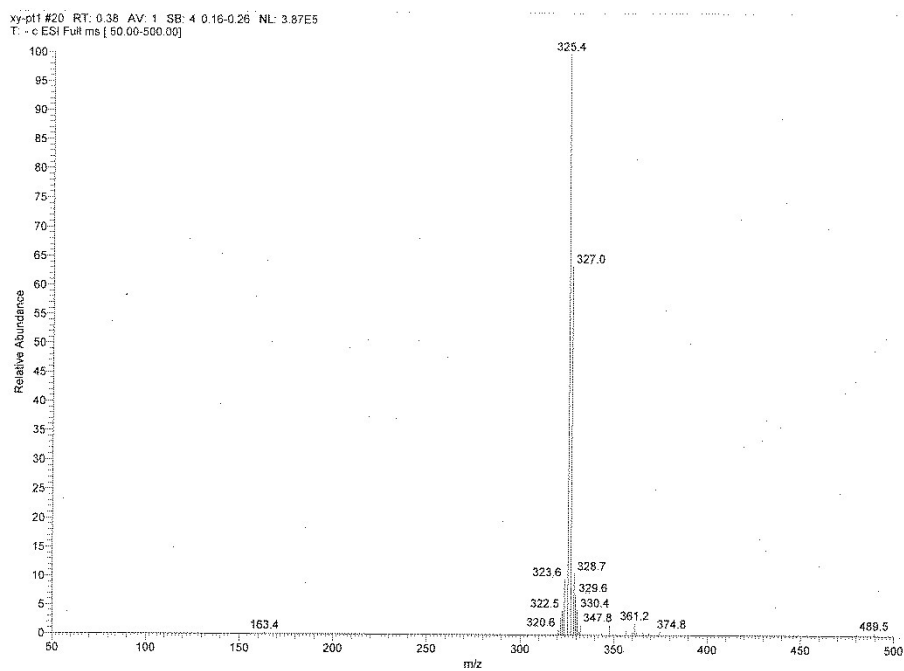


**Scheme S1** The synthetic route of H<sub>3</sub>TPIA

**Synthesis of dimethyl 5-(4-cyanophenoxy)isophthalate (II).** The mixture of 4-fluorobenzonitrile (12.1 g, 0.1 mol), dimethyl 5-hydroxyisophthalate I (10.5 g, 0.05 mol) and K<sub>2</sub>CO<sub>3</sub> (13.8 g, 0.1 mol) in 500 mL DMF was refluxed for overnight under N<sub>2</sub> atmosphere and then poured into cold water. Pure product of II was obtained from recrystallization in EtOH with a yield of 95%. Anal. Calcd for C<sub>17</sub>H<sub>13</sub>NO<sub>5</sub>: C, 65.59; H, 4.21; N, 4.50%. Found: C, 65.42; H, 4.15; N, 4.33%.

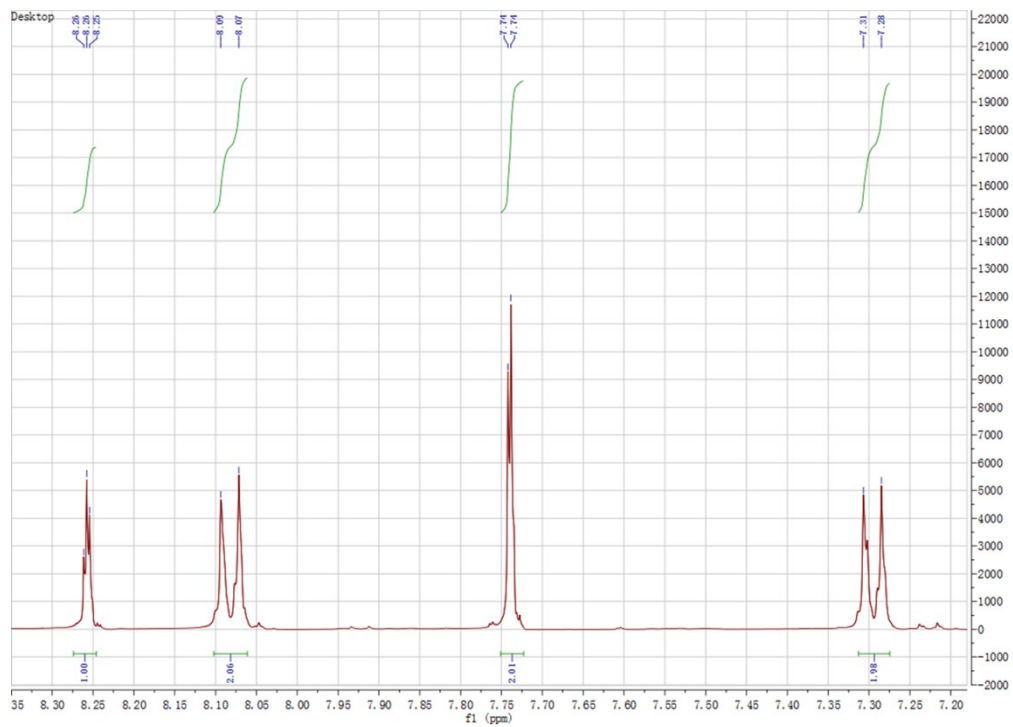
**Synthesis of dimethyl 5-(4-(2H-tetrazol-5-yl)phenoxy)isophthalate (III).** The mixture of II (3.1 g, 0.01 mol), NaN<sub>3</sub> (2.0 g, 0.03 mol), and triethylamine hydrochloride (4.1 g, 0.03 mol) in 100 mL toluene was refluxed for 3 days under N<sub>2</sub> atmosphere and then 100 mL 0.1 M NaOH was dropped. The water solution was separated and acidified with dilute hydrochloric acid. White solid was obtained and recrystallized in EtOH with a yield of 90%. ESI-MS: m/z [M - H]<sup>-</sup>, 353.3 (calcd for C<sub>17</sub>H<sub>14</sub>N<sub>4</sub>O<sub>5</sub>, 354.3). Anal. Calcd for C<sub>17</sub>H<sub>14</sub>N<sub>4</sub>O<sub>5</sub>: C, 57.63; H, 3.98; N, 15.81%. Found: C, 57.52; H, 3.90; N, 15.83%.

**Synthesis of 5-(4-(tetrazol-5-yl)phenoxy) isophthalic acid IV (H<sub>3</sub>TPIA).** The mixture of **III** (1.8 g, 0.005 mol) and 2 mL concentrated hydrochloric acid in 100 mL H<sub>2</sub>O was refluxed for 1 day and then filtrated. The white solid was recrystallized in EtOH with a yield of 92%. ESI-MS: m/z [M - H]<sup>-</sup>, 325.4 (calcd for C<sub>15</sub>H<sub>10</sub>N<sub>4</sub>O<sub>5</sub>, 326.3). Anal. Calcd for C<sub>15</sub>H<sub>10</sub>N<sub>4</sub>O<sub>5</sub>: C, 55.22; H, 3.09; N, 17.17%. Found: C, 55.15; H, 2.96; N, 17.05%.

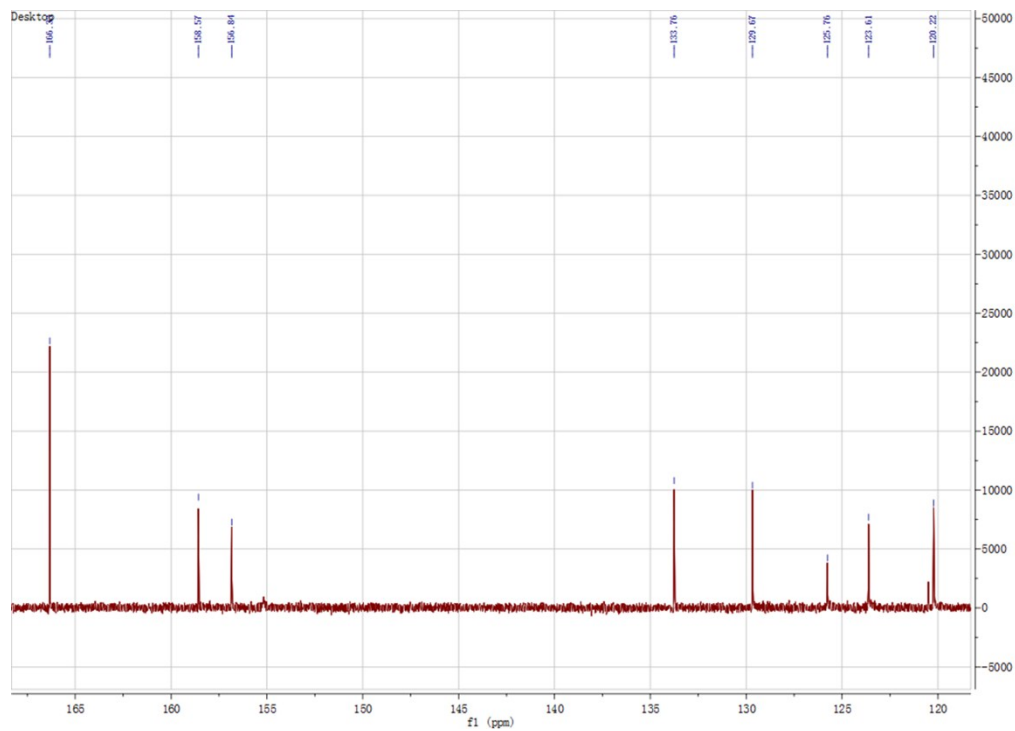


**Fig. S1** ESI-MS of H<sub>3</sub>TPIA

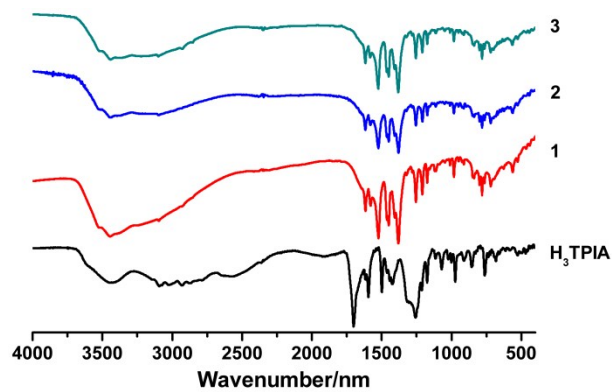




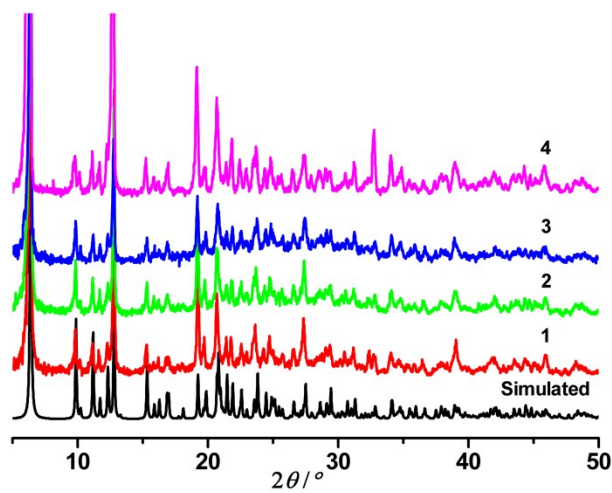
**Fig. S2**  $^1\text{H}$  NMR spectrum of  $\text{H}_3\text{TPIA}$  in  $\text{DMSO-d}_6$



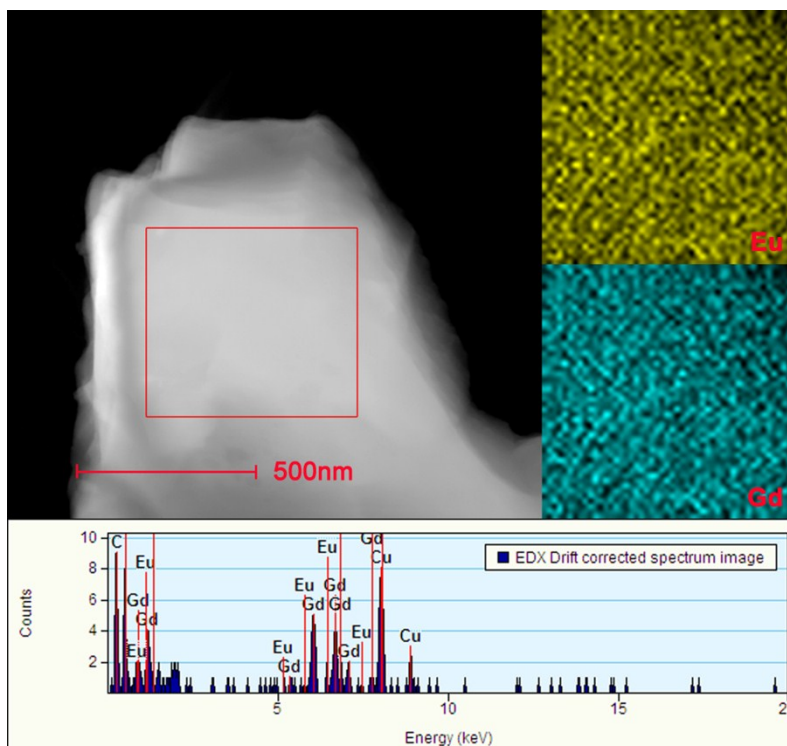
**Fig. S3**  $^{13}\text{C}$  NMR spectrum of  $\text{H}_3\text{TPIA}$  in  $\text{DMSO-d}_6$



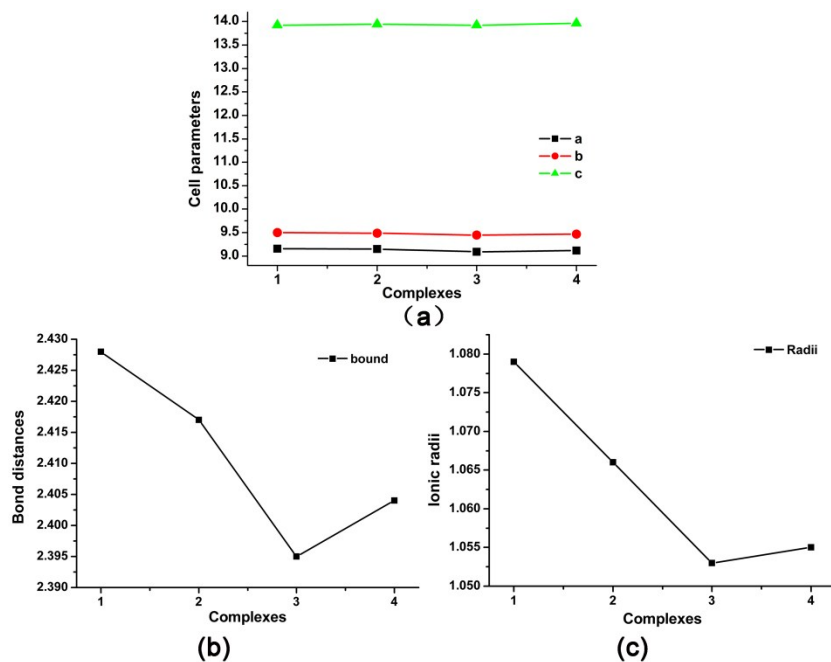
**Fig. S4** FT-IR of H<sub>3</sub>TPIA and 1–3.



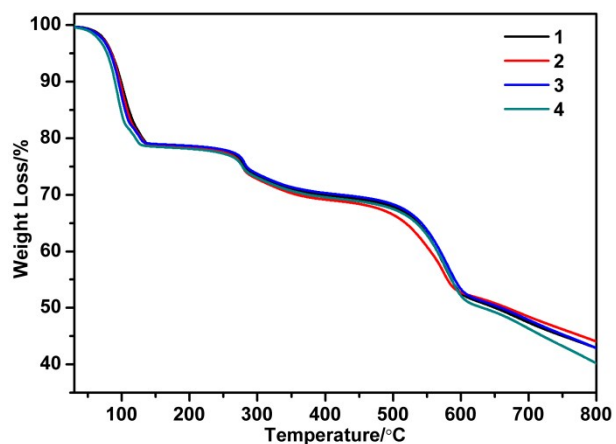
**Fig. S5** PXRD patterns for the as-synthesized compounds 1–4 and the simulated one from single-crystal X-ray diffraction data.



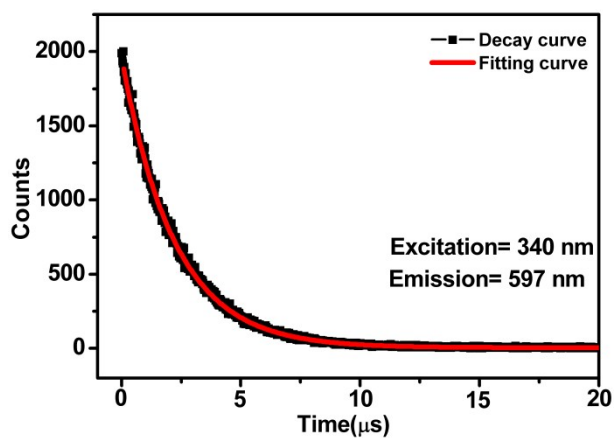
**Fig. S6** Electron microscopy mapping studies of a portion of the  $[(\text{Eu}_{0.12}\text{Gd}_{0.88})(\text{TPIA})(\text{H}_2\text{O})_3]\cdot 5.5\text{H}_2\text{O}$  (**4**).



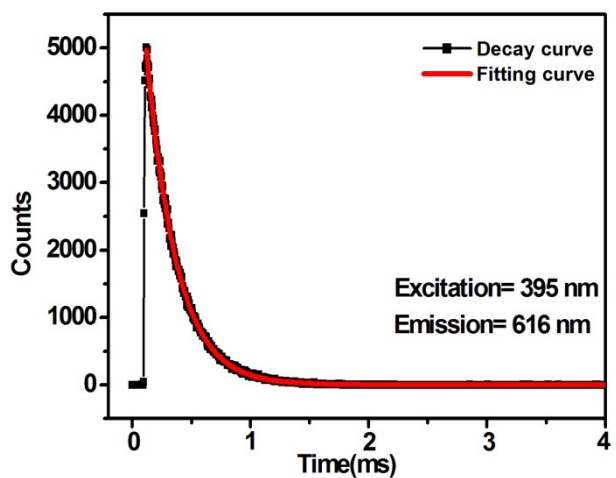
**Fig. S7** (a) Cell parameters. (b) Crystallographically identified Ln–O bond distances. (c) Ideal Ln(III) ionic radii in mono- and bimetallic Ln-MOFs (normalized for mixed Ln(III) ions according to corresponding Ln(III) ion content in mixed-Ln-MOFs **4**).



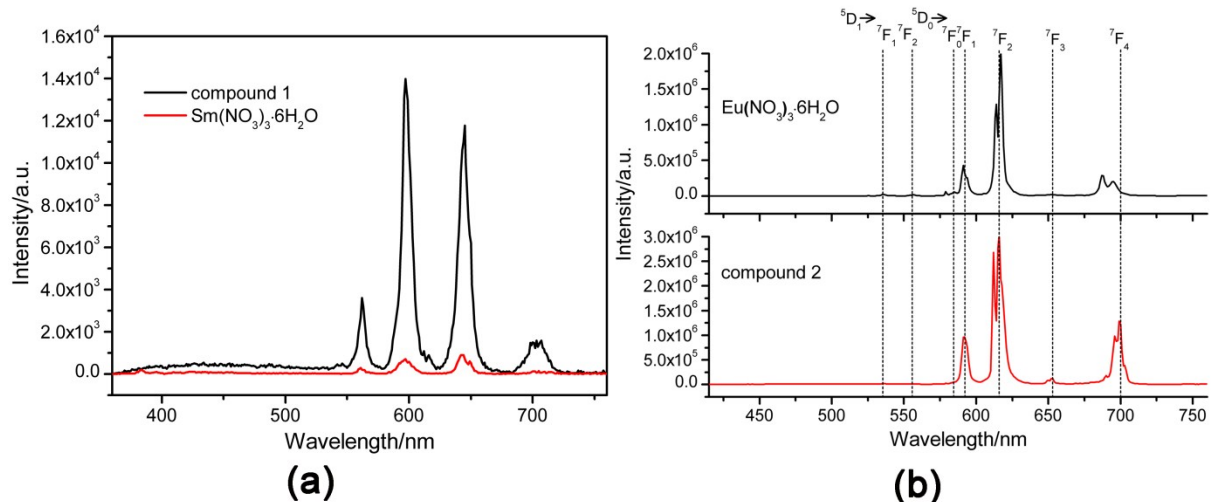
**Fig. S8** The TGA curves of compounds **1–4**.



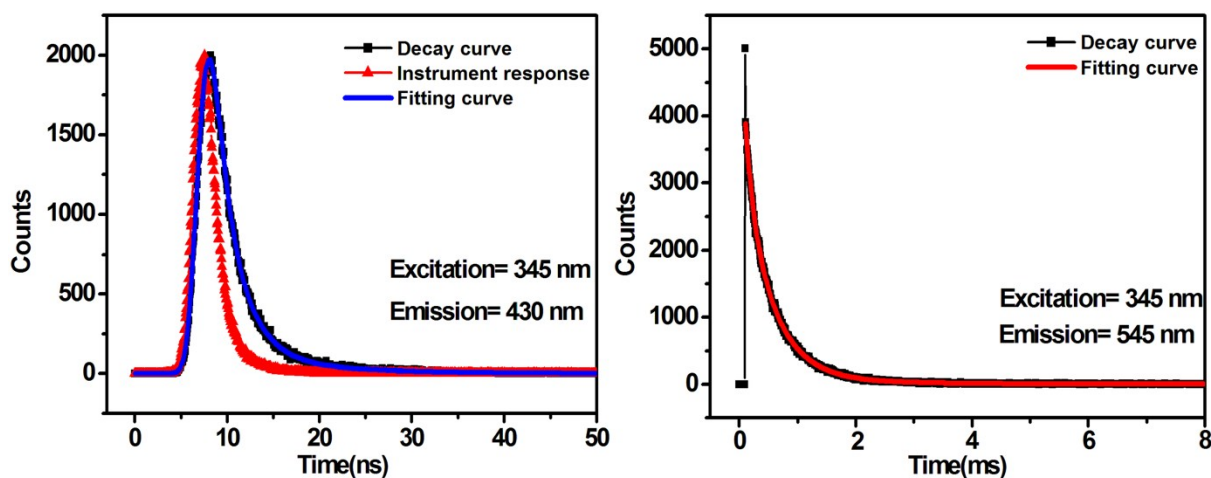
**Fig. S9** The fluorescent decay curves for **1**



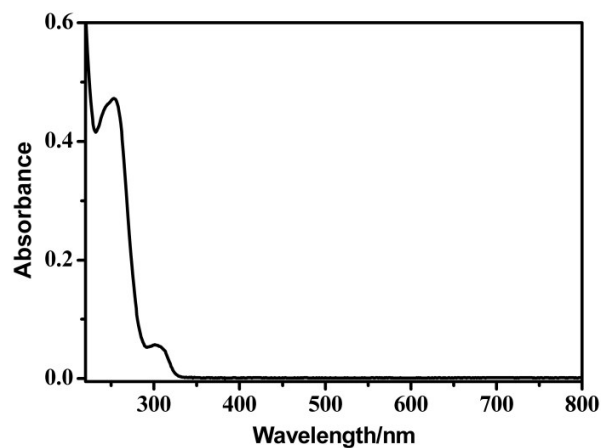
**Fig. S10** The fluorescent decay curves for **2**



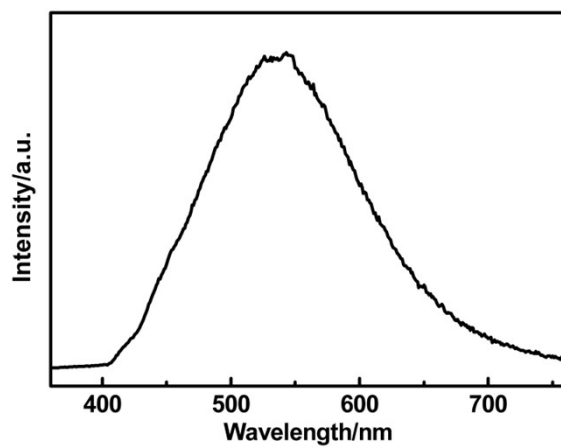
**Fig. S11** (a) Emission spectra of compound **1** and  $\text{Sm}(\text{NO}_3)_3 \cdot 6\text{H}_2\text{O}$  when excited at 340 nm; (b) Emission spectra of compound **2** and  $\text{Eu}(\text{NO}_3)_3 \cdot 6\text{H}_2\text{O}$  when excited at 395 nm.



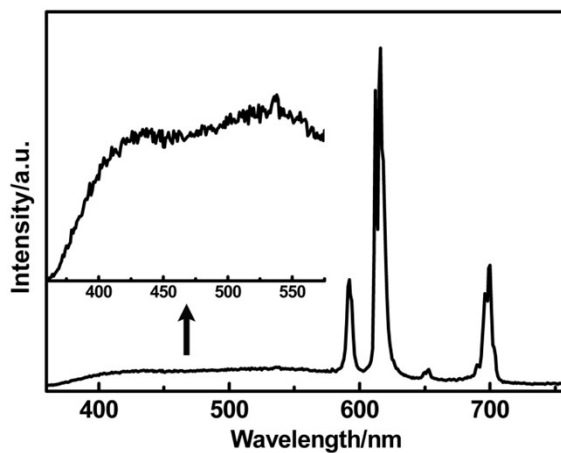
**Fig. S12** The fluorescent decay curves for **3**



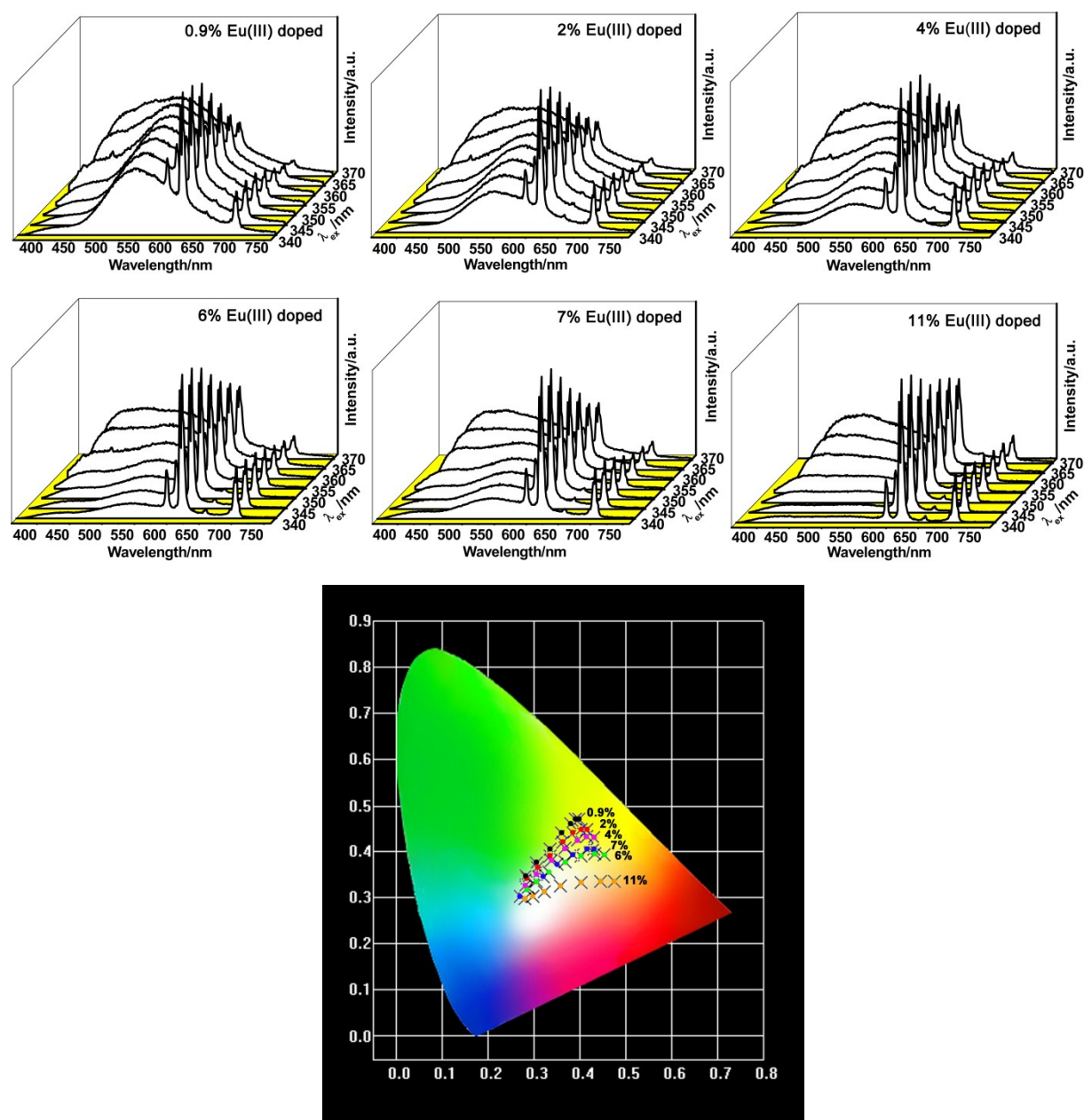
**Fig. S13** UV-Vis absorption spectra of the free ligand H<sub>3</sub>TPIA in methanol solution.



**Fig. S14** The emission spectrum of **3** at 77K



**Fig. S15** The emission spectrum of **4** excited at 340 nm



**Fig. S16** The emission spectra (upper) and 1931 CIE chromaticity diagrams (lower) of different doping concentrations of Eu(III) ion embedded in **3** excited at different wavelengths.