

Supplementary Material (ESI) for CrystEngComm

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**pH, solvent and metal ions induced octamolybdate-based
metal–organic complexes decorated by pyridyl-carboxylate
ligand containing amide group †**

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Table S1 Selected bond distances (Å) and angles (°) for compounds 1–3.

Compound 1			
Cu(1)-O(5)#1	1.942(3)	O(1)-Cu(1)-O(16)#3	86.88(11)
Cu(1)-N(6)#1	1.961(3)	N(1)-Cu(1)-O(16)#3	87.41
Cu(1)-N(1)	1.981(3)	O(5)#1-Cu(1)-O(16)#3	94.49
Cu(1)-O(1)	1.984(2)	O(3)#2-Cu(1)-O(16)#3	167.77
Cu(1)-O(3)#2	2.342(3)	N(6)#1-Cu(1)-O(16)#3	91.72
Cu(1)-O(16) #3	2.655(3)	N(2)#2-Cu(2)-O(1W)#2	89.95
Cu(2)-O(2)#2	2.002(2)	O(1W)-Cu(2)-O(1W)#2	180
Cu(2)-O(2)	2.002(2)	O(2)#2-Cu(2)-O(2)	180
Cu(2)-N(2)#2	2.016(3)	O(2)#2-Cu(2)-N(2)#2	86.64(12)
Cu(2)-N(2)	2.016(3)	N(2)-Cu(2)-O(1W)	90.05
Cu(2)-O(1W)	2.453(3)	O(2)#2-Cu(2)-O(1W)	79.26(12)
Cu(2)-O(1W)#2	2.453(3)	O(2)-Cu(2)-N(2)	86.64(12)
O(5)#1-Cu(1)-N(6)#1	83.93(11)	N(2)#2-Cu(2)-N(2)	180
O(5)#1-Cu(1)-N(1)	97.64(11)	O(2)#2-Cu(2)-O(2)	180
N(6)#1-Cu(1)-N(1)	178.26(13)	O(2)-Cu(2)-N(2)#2	93.36(12)
O(5)#1-Cu(1)-O(1)	178.41(12)	O(2)#2-Cu(2)-N(2)	93.36(12)
N(6)#1-Cu(1)-O(1)	95.25(11)	O(2)-Cu(2)-O(1W)	79.26(11)
N(1)-Cu(1)-O(1)	83.19(11)	N(2)#2-Cu(2)-O(1W)	90.05(11)
O(5)#1-Cu(1)-O(3)#2	97.42(11)	O(1W)#2-Cu(2)-O(1W)	180.00(11)
N(6)#1-Cu(1)-O(3)#2	92.23(11)	N(2)-Cu(2)-O(1W)#2	89.95(11)
N(1)-Cu(1)-O(3)#2	88.31(11)	O(2)#2-Cu(2)-O(1W)#2	79.26(11)
O(1)-Cu(1)-O(3)#2	81.24(10)	O(2)-Cu(2)-O(1W)#2	100.74(12)
Symmetry code for 1: #1 x-1, y+1, z; #2 -x+1, -y-1, -z+1; #3 x, 1 + y, z.			
Compound 2			
Cu(1)-O(1W)#2	1.976(3)	O(21)#4-Cu(1)-O(1)	88.09(12)
Cu(1)-O(1W)	1.976(3)	O(1W)#2-Cu(1)-O(1)#2	89.13(12)
Cu(1)-O(21)#3	1.985(3)	O(1W)-Cu(1)-O(1)#2	90.87(12)
Cu(1)-O(21)#4	1.985(3)	O(21)#3-Cu(1)-O(1)#2	88.09(12)
Cu(1)-O(1)	2.390(3)	O(21)#4-Cu(1)-O(1)#2	91.91(12)
Cu(1)-O(1)#2	2.390(3)	O(1)-Cu(1)-O(1)#2	180
Cu(2)-N(1)	1.961(3)	O(18)-Cu(2)-O(18)#3	180
Cu(2)-O(18)	1.943(3)	O(18)-Cu(2)-N(1)#3	96.66(13)

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Cu(2)-O(18)#3	1.943(3)	O(18)#3-Cu(2)-N(1)#3	83.34(13)
Cu(2)-N(1)#3	1.961(3)	O(18)-Cu(2)-N(1)	83.34(13)
Cu(2)-O2	2.611(3)	O(18)#3-Cu(2)-N(1)	96.66(13)
Cu(2)-O2#3	2.611(3)	N(1)#3-Cu(2)-N(1)	180
O(1W)#2-Cu(1)-O(1W)	179.999(2)	N(1)-Cu(2)-O(2)	93.96(12)
O(1W)#2-Cu(1)-O(21)#3	93.99(13)	N(1)#3-Cu(2)-O(2)	86.04(12)
O(1W)-Cu(1)-O(21)#3	86.01(13)	O(18)-Cu(2)-O(2)	100.76(12)
O(1W)#2-Cu(1)-O(21)#4	86.01(13)	O(18)#3-Cu(2)-O(2)	79.25(12)
O(1W)-Cu(1)-O(21)#4	93.99(13)	N(1)-Cu(2)-O(2)#3	86.04(12)
O(21)#3-Cu(1)-O(21)#4	179.998(1)	N(1)#3-Cu(2)-O(2)#3	93.96(12)
O(1W)#2-Cu(1)-O(1)	90.87(12)	O(18)-Cu(2)-O(2)#3	79.24(12)
O(1W)-Cu(1)-O(1)	89.13(12)	O(18)#3-Cu(2)-O(2)#3	100.76(12)
O(21)#3-Cu(1)-O(1)	91.91(12)	O(2)-Cu(2)-O(2)#3	180

Symmetry code for **2**: #2 $-x-2, -y, -z+2$; #3 $-x-1, -y, -z+2$; #4 $x-1, y, z$

Compound 3

Co(1)-O(1)	2.040(3)	N(2)-Co(1)-N(1)	89.42(12)
Co(1)-O(2)	2.050(3)	O(1)-Co(1)-O(10)#2	90.09(11)
Co(1)-O(1W)	2.120(3)	O(2)-Co(1)-O(10)#2	91.81(11)
Co(1)-N(2)	2.130(3)	O(1W)-Co(1)-O(10)#2	88.96(12)
Co(1)-N(1)	2.133(3)	N(2)-Co(1)-O(10)#2	92.25(11)
Co(1)-O(10)#2	2.160(3)	N(1)-Co(1)-O(10)#2	168.16(12)
Co(2)-O(3W)#3	2.081(4)	O(3W)#3-Co(2)-O(3W)	180
Co(2)-O(3W)	2.081(4)	O(3W)#3-Co(2)-O(4W)#3	87.90(17)
Co(2)-O(4W)#3	2.097(3)	O(3W)-Co(2)-O(4W)#3	92.10(17)
Co(2)-O(4W)	2.097(3)	O(3W)#3-Co(2)-O(4W)	92.10(17)
Co(2)-O(2W)#3	2.110(4)	O(3W)-Co(2)-O(4W)	87.90(17)
Co(2)-O(2W)	2.110(4)	O(4W)#3-Co(2)-O(4W)	180
O(1)-Co(1)-O(2)	173.51(12)	O(3W)#3-Co(2)-O(2W)#3	91.7(2)
O(1)-Co(1)-O(1W)	94.34(13)	O(3W)-Co(2)-O(2W)#3	88.3(2)
O(2)-Co(1)-O(1W)	91.89(12)	O(4W)#3-Co(2)-O(2W)#3	88.64(14)
O(1)-Co(1)-N(2)	95.70(12)	O(4W)-Co(2)-O(2W)#3	91.36(14)
O(2)-Co(1)-N(2)	78.04(11)	O(3W)#3-Co(2)-O(2W)	88.3(2)
O(1W)-Co(1)-N(2)	169.88(13)	O(3W)-Co(2)-O(2W)	91.7(2)
O(1)-Co(1)-N(1)	78.08(13)	O(3W)#3-Co(2)-O(2W)#3	91.7(2)
O(2)-Co(1)-N(1)	100.00(13)	O(3W)-Co(2)-O(2W)#3	88.3(2)
O(1W)-Co(1)-N(1)	91.45(13)	O(4W)#3-Co(2)-O(2W)#3	88.64(14)

Symmetry code for **3** #2 $x+1, y, z$; #3 $-x, -y, -z$.

Table S2. Selected hydrogen-bonding geometry (Å, °) for compounds **1–3**.

	D–H⋯A	D–H	H⋯A	D⋯A	D–H⋯A
Compound 1	C(24)–H(24A)⋯O(6)	0.93	2.38	3.3036	170
Compound 2	C(8)–H(8A)⋯O(5)	0.93	2.47	3.1204	127
Compound 3	N(5)–H(5B)⋯O(5)	0.86	1.95	2.7812	163

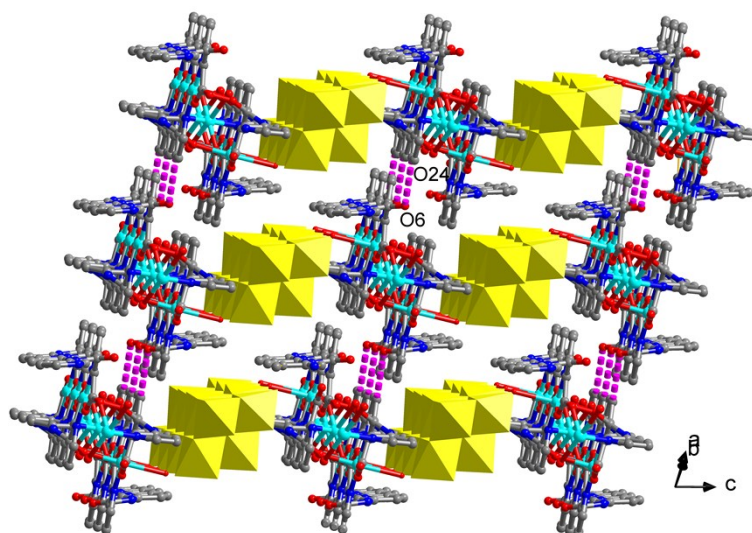


Fig. S1 The 3D supramolecular structure in 1.

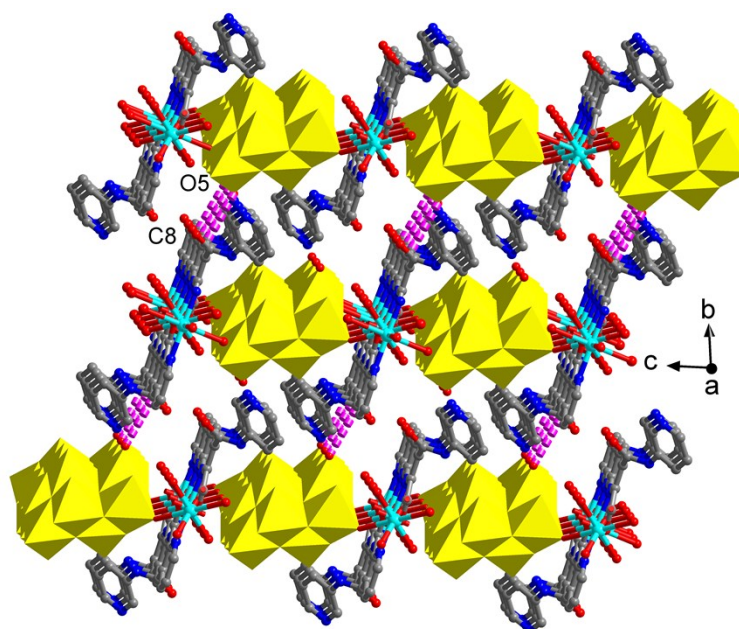


Fig. S2 The 3D supramolecular structure in 2.

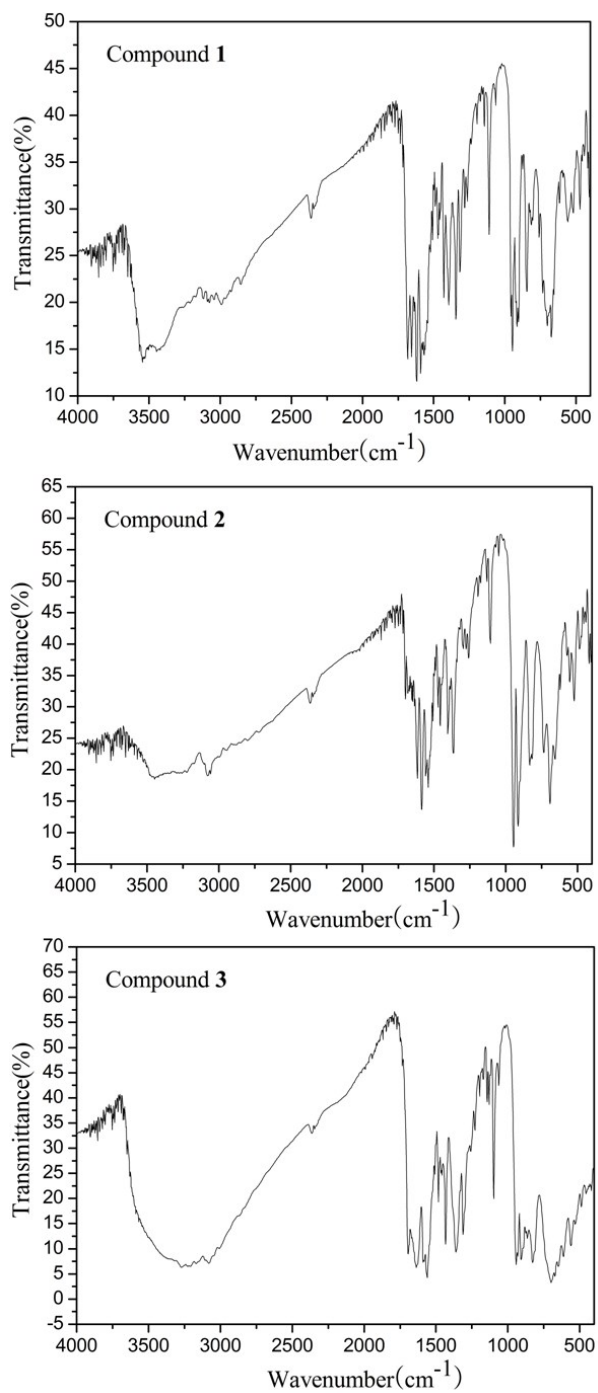


Fig. S3. The IR spectra of compounds 1-3.

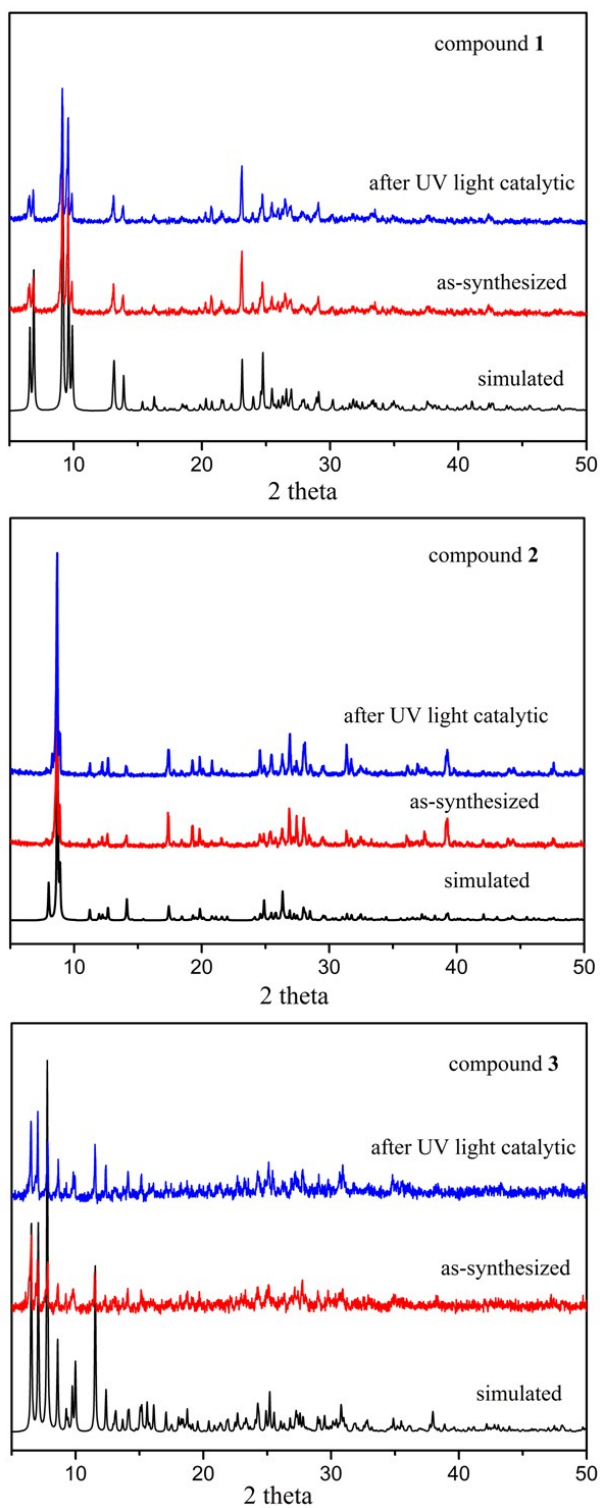


Fig. S4. The PXRD of compounds 1–3.

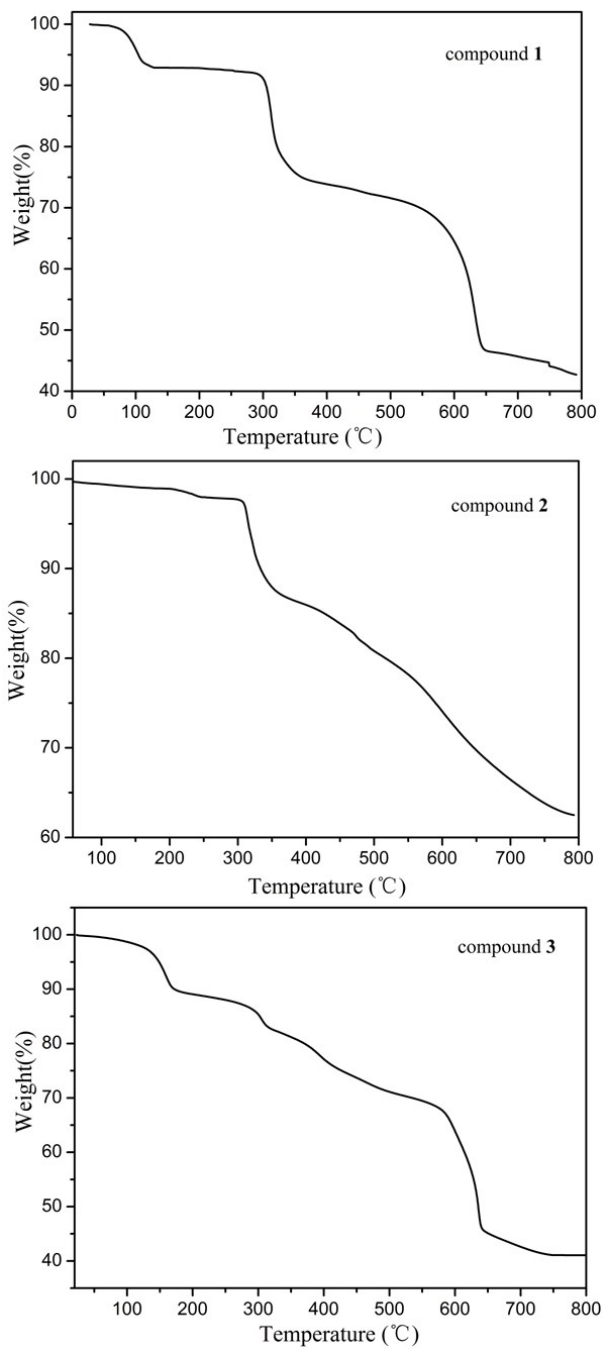


Fig. S5. The TG curves of compounds 1–3.

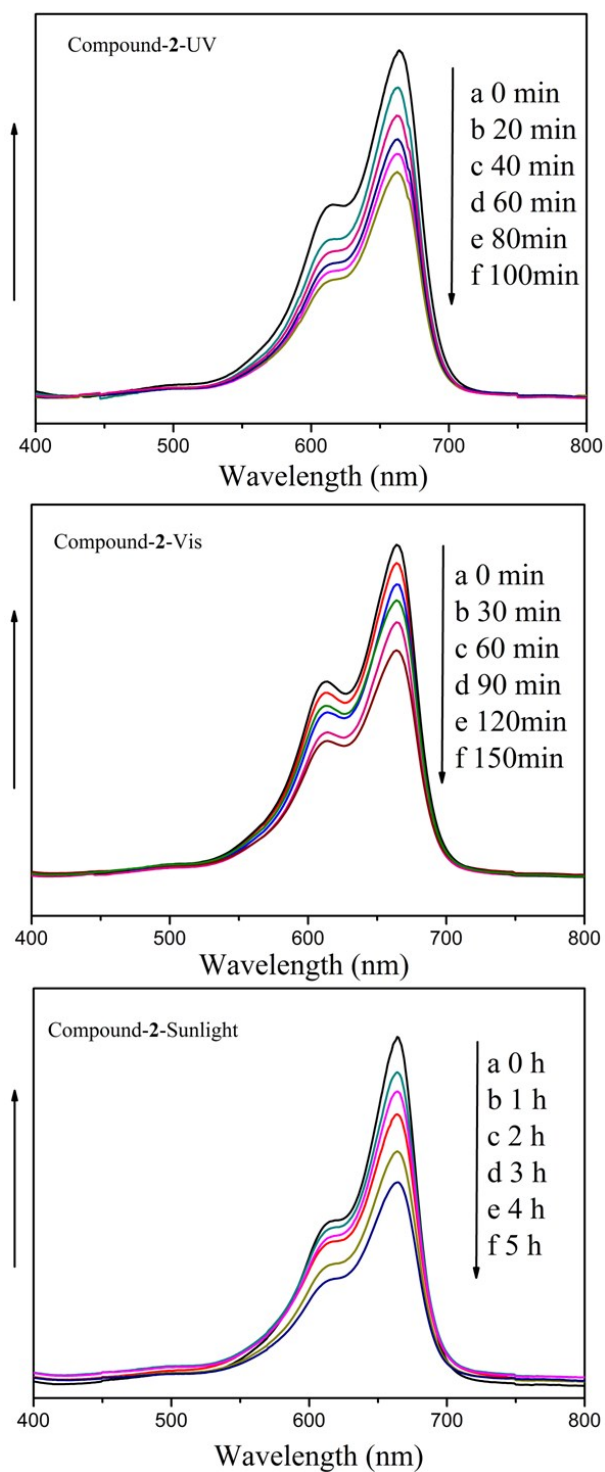


Fig. S6. Absorption spectra of the MB solution during the decomposition reaction under UV (a), visible (b) and sunlight (c) irradiation in the presence of the compound 2.

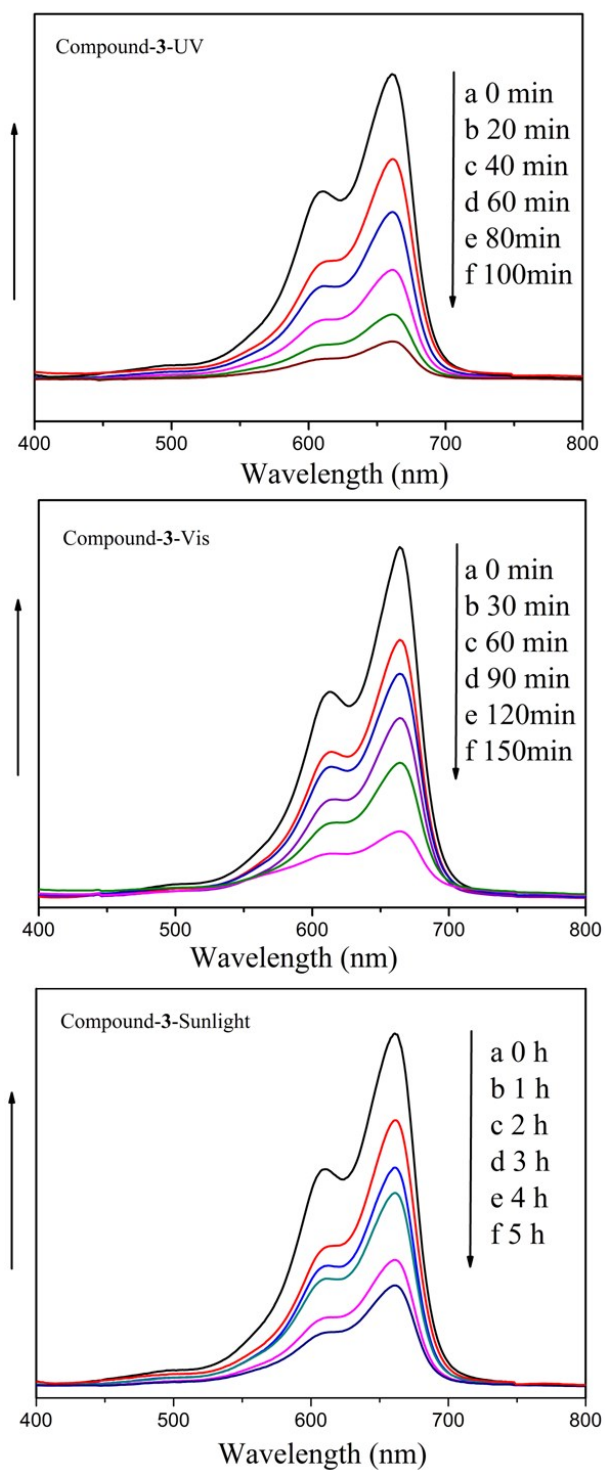


Fig. S7. Absorption spectra of the MB solution during the decomposition reaction under UV (a), visible (b) and sunlight (c) irradiation in the presence of the compound 3.